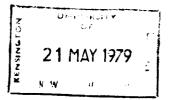
FACULTY OF BIOLOGICAL SCIENCES AND FACULTY OF SCIENCE COMBINED 1974 HANDBOOK

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COMBINED FACULTY OF BIOLOGICAL SCIENCES AND FACULTY OF SCIENCE

1974 HANDBOOK

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FACULTIES OF BIOLOGICAL SCIENCES AND SCIENCE

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INTRODUCTION

The Science Course is offered in a unit system in all three years of the pass degree. In this system, the traditional subjects, such as Mathematics, are broken up into smaller units: first year subjects into two units, second year subjects into three units and third years subjects into four units.

First year subjects, which count as two units, differ from those of later years in that they must be taken as a whole. In other words one unit cannot be taken without the other. All students in the Science Course must enrol in Mathematics I, which is offered in three versions each of which counts as two units: Mathematics I, Higher Mathematics I and Mathematics IT. One only is required, but care must be taken in making the choice. In general, Mathematics IT considerably limits the choice of units in following years.

Students proceeding to a degree in Science will be associated principally with schools within the Faculties of Biological Sciences and Science. However, in accordance with the regulations, students may elect to take subjects from schools in other faculties.

Some schools do not offer a full range of level III units in the evening. Students in the part-time course are urged to take note of these Schools, since a science major involving any of them will only be possible for day-time attendances. The Schools concerned are marked below with \ddagger .

Students seeking advice should contact the representative of the relevant School. A list appears below:

Faculty of Applied Science

‡School of Applied Geology Mr. G. J. Baldwin

Faculty of Arts

School of Geography*	Professor J. A. Mabbutt
School of Philosophy**	Professor C. L. Hamblyn
\$\$ \$\$ \$\$ \$\$ \$\$ \$\$ \$\$ \$\$ \$\$ \$\$ \$\$ \$\$ \$\$	Dr. J. Saunders

Faculty of Biological Sciences

General and Human Biology* (prerequisite for all other un	its
in this Faculty excepting Psychology)	Dr. A. E. Wood
School of Psychology	Mr. P. J. Cleary (Science Course)
	Mrs. N. Binks (Psychology)
\$School of Biochemistry†	Professor E. O. P. Thompson
‡School of Biological Tech-	
nology	Professor B. J. Ralph
\$School of Botany†	Dr. M. M. Hindmarsh
\$\$ School of Microbiology	Professor J. Vincent
\$\$chool of Zoology\$	Mrs. Patricia Dixon

Faculty of Engineering

School of Mechanical and Industrial Engineering	
(Engineering I*)	Associate Professor R. G. Robertson
School of Electrical	

Engineering (Computer	
Science [†])	 Mr. K. A. Robinson

Faculty of Medicine	
\$School of Anatomy†	Associate Professor B. R. A. O'Brien
\$School of Human Genetics	Mr. A. E. Stark
\$\$Chool of Physiology	Dr. D. G. Garlick

Faculty of Science	
School of Applied Physics and Optometry	Professor C. J. Milner (Applied Physics)
	Associate Professor J. Lederer (Optometry)
School of Chemistry	Mr. W. J. Dunstan
‡School of Mathematics	Associate Professor W. E. Smith
School of Physics	Mr. K. Mann

In addition to the Science subjects, all undergraduates in Science are required to pass in three subjects in General Studies. A wide choice is available and students should consult the Department of General Studies handbook which is provided free of charge.

Students who wish to be admitted with advanced standing should obtain the necessary forms from the Admissions Office. Copies of recommended courses may be obtained from the Faculty Office (Room 57, Main Building).

> C.J.Q. N.C.S.

* First year level only

- ** First and Second year levels only
 - † Second and Third year levels only
 - ‡ Schools which do not offer a full range of level III units in the evening

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CALENDAR OF DATES FOR 1974

Session 1:	March 4 to May 19
	May Recess May 20 to May 26
	May 27 to June 16
	Midyear Recess June 17 to July 21
Session 2:	July 22 to August 25
	August Recess August 26 to September 1
	September 2 to November 3
	Study Recess November 4 to November 10

JANUARY

Tuesday 1	New Year's Day—Public Holiday
Friday 11	Last day for application for review of results of annual examinations
	Last day for application for admission under "show cause" rules
Monday 14	Timetable for deferred examinations available
Tuesday 15	Last day for acceptance of applications from students graduating in 1974 for admission to University degrees and diplomas
Friday 18	Last day for application for deferred examinations
	Last day for acceptance of applications to enrol by new students and students repeating first year
Monday 28	Australia Day—Public Holiday
Tuesday 29	Deferred examinations begin
FEBRUARY	
Friday 8	Last day for students to appeal against exclusion under the re-enrolment rules
Saturday 9	Deferred examinations end
Monday 18	Enrolment period begins for new students and students repeating first year
Monday 25	Enrolment period begins for students re-enrolling (second and later years)
	Deferred examination results available
MARCH	
Friday 1	Last day for application for review of deferred examination results
	Last day for students with deferred examinations to appeal against exclusion under the re-enrolment rules
Monday 4	Session 1 begins
	Faculty of Biological Sciences meeting, 2 p.m.

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MARCH

Friday 15	Last day for acceptance of enrolments by new students (late fee payable)
Tuesday 19	Faculty of Science meeting, 2 p.m.
Wednesday 20	Board of Studies in Science meeting, 2 p.m.
Friday 22	Last day for application for review of results of deferred examinations
Friday 29	Last day for changes in course programmes
	Last day for acceptance of enrolments by students re-enrolling (late fee payable)

APRIL

Thursday 4	Last day for discontinuation without failure of subjects which extend over the first session only
Friday 12 to Monday 15 Thursday 25	Easter Anzac Day—Public Holiday
1AY	

Μ

Tuesday 7	Provisional timetable for June/July examinations published
Tuesday 14	Last day for acceptance of corrected enrolment details forms
Monday 20	May Recess begins
	Last day for students to advise of examination time- table clashes
Sunday 26	May Recess ends
	Last date for discontinuation without failure of subjects which extend over the academic year

JUNE

Monday 3	Faculty of Biological Sciences meeting, 2 p.m.
Tuesday 4	Timetable for June/July examinations published
Tuesday 11	Faculty of Science meeting, 10 a.m.
Sunday 16	Session 1 ends
Monday 17	Queen's Birthday—Public Holiday
	Midyear Recess begins
Tuesday 18	Midyear examinations begin
Sunday 30	Last date for acceptance of applications for re-admission after exclusion under rules govern- ing re-enrolment

JULY

Tuesday 2	Midyear examinations end
Wednesday 17	Board of Studies in Science meeting, 2 p.m.
Sunday 21	Midyear Recess ends
Monday 22	Session 2 begins

AUGUST

Thursday 1	Foundation Day
Monday 12	Faculty of Biological Sciences meeting, 2 p.m.

AUGUST

Thursday 22	Last day for discontinuation without failure of subjects which extend over the second session only
Monday 26	August Recess begins
-	Holiday for non-academic staff

SEPTEMBER

Sunday 1	August Recess ends	
-	published	for annual examinations
-	forms	corrected enrolment details
Monday 23	Last day for students timetable clashes	to advise of examination

OCTOBER

	Faculty of Science meeting, 2 p.m.
Monday 7	Eight Hour Day—Public Holiday
Wednesday 23	Board of Studies in Science meeting, 2 p.m.
Monday 28	Faculty of Biological Sciences meeting, 2 p.m.
Tuesday 29	Timetable for annual examinations published

NOVEMBER

Monday 4	Study Recess begins
Sunday 10	Session 2 ends
Monday 11	Annual examinations begin

DECEMBER

Tuesday 3	Annual examinations end
Wednesday 25	Christmas Day—Public Holiday
Thursday 26	Boxing Day—Public Holiday

1975

Session 1:	March 3 to May 11
	May Recess: May 12 to May 18
	May 19 to June 15
	Midyear Recess: June 16 to July 20

Session 2: July 21 to August 24 August Recess: August 25 to August 31 September 1 to November 2 Study Recess: November 3 to November 9

JANUARY

Wednesday 1	New Year's Day—Public Holiday		
Friday 10	Last day for application for review of results of		
annual examinations			

JANUARY		
Monday 13	Timetable for deferred examinations published	
Friday 17	Last day for application for deferred examinations	
	Last day for acceptance of applications to enrol by new students and students repeating first year	
Monday 27	Australia Day—Public Holiday	
Tuesday 28	Deferred examinations begin	
FEBRUARY		
Saturday 8	Deferred examinations end	
Monday 17	Enrolment period begins for new students and students repeating first year	
Friday 21	Results of deferred examinations available	
Monday 24	Enrolment period begins for students re-enrolling (second and later years)	

THE ACADEMIC YEAR

The academic year is divided into two sessions, each containing 14 weeks for teaching. There is a recess of five weeks between the two sessions as well as short recesses of one week within each of the sessions. Session 1 commences on the first Monday of March.

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P. D. Lark, BEc Syd., MSc PhD N.S.W., ASTC, ARACI

LECTURERS

D. S. Alderdice, MSc Syd., PhD Lond. Ruby Foon, MSc PhD Melb. W. D. Johnson, BSc Syd., MSc N.E., PhD N.S.W. B. J. Orr, MSc Syd., PhD Brist., ARACI A. D. Rae, MSc PhD Auck., ANZIC

SENIOR TUTOR

M. P. Boggard, BSc PhD Syd.

TUTOR

R. O. Pascual, BSPharm Philippines, MA PhD Col.

First Year Chemistry

DIRECTOR OF FIRST YEAR CLASSES IN CHEMISTRY June C. Griffith, MSc N.S.W., PhD Syd.

LECTURER

C. R. Taylor, BSc Syd.

SENIOR TUTOR

P. S. K. Chia, MSc PhD N.S.W.

TUTOR DEMONSTRATORS

V. C. Bien, MSc Syd. G. B. Morris, BSc Syd.

CHIEF LECTURE DEMONSTRATOR V. Shuk, DipChemIng Kiev

MICRO-ANALYST (SCHOOL) E. Challen, DrIng Berl., ARACI

PROFESSIONAL OFFICERS (SCHOOL) J. Bell, BE N.S.W., ASTC D. C. Craig, BSc Syd., MSc N.S.W. V. Djohadze, BSc N.S.W R. J. Finlayson, BSc N.S.W., ARACI A. M. Ingster, IngESE Paris, AMIRE(Aust.) N. Inflation, and the second se R. B. Rogers, BSc N.S.W., ARMTC G. T. See, BSc N.S.W., ASTC J. Sussman, BSc Syd. M. K. Withers, MSc N.S.W.

SCHOOL OF MATHEMATICS

- PROFESSOR OF APPLIED MATHEMATICS AND HEAD OF SCHOOL V. T. Buchwald, BSc Manc., MSc PhD Lond.
- PROFESSOR OF APPLIED MATHEMATICS J. M. Blatt, BA Cinc., PhD Corn. and Prin., FAA, FAPS
- PROFESSOR OF PURE MATHEMATICS G. Szekeres, DiplChemEng Bud., FAA
- PROFESSOR OF STATISTICS A. M. Hasofer, BEE Faruk, BEc PhD Tas., MIEAust
- DIRECTOR OF FIRST YEAR STUDIES Associate Professor A. H. Low, MSc DipEd Syd., PhD N.S.W.
- ADMINISTRATIVE ASSISTANT Pamela J. Monk, BSc N.E.
- PROFESSIONAL OFFICER G. J. P. Tomis, AB Carrol, Montana, BA Montana, MA Arizona

Department of Pure Mathematics

Associate Professor J. L. Griffith, BA MSc DipEd Syd.

SENIOR LECTURERS

S. A. Morris, BSc Qld., PhD Flin.

- J. St. A. Sandiford, MSc Syd.
- A. J. Van der Poorten, BA BSc PhD MBA N.S.W.

LECTURERS

C. D. Cox, BSc DipEd Qld. S. A. R. Disney, BA Adel., DPhil Oxon. P. W. Donovan, BA Syd., DPhil Oxon. J. D. Gray, BA Syd., PhD N.S.W. D. C. Hunt, BSc Syd., MSc PhD Warw. R. K. James, BSc PhD Syd. J. H. Loxton, MSc Melb. E. S. Noussair, BA Cairo, PhD Br.Col. J. F. Price, MSc Melb., PhD A.N.U. D. G. Tacon, BSc N'cle.(N.S.W.), PhD A.N.U.

POSTDOCTORAL TEACHING FELLOW

L. J. Dickson, BS Seattle, PhD Prin.

SENIOR TUTORS

Mary R. Freislich, BA Rand., MA N.S.W. Mrs. Agnes V. Nikov, DiplMath DiplEd Bud.

TUTORS

M. C. Cullinan, BSc Adel. Susan J. M. Cullity, BSc W.Aust.

Department of Applied Mathematics

ASSOCIATE PROFESSORS

I. H. Sloan, BA BSc Melb., MSc Adel., PhD Lond. W. E. Smith, MSc Syd., BSc Oxon., PhD N.S.W., MInstP

SENIOR LECTURER

K. Okamoto, BS PhD Tokyo and Louisiana State

LECTURERS

M. L. Banner, BE MEngSc Syd., PhD Johns H.

B. J. Burn, MSc Otago, PhD Camb.

B. S. Goh, BSc PhD Cant. E. J. Moore, MSc W.Aust., PhD Harv.

A. H. Opie, BSc DipEd Melb., PhD Monash

SENIOR TUTORS

M. Brender, MSc McG. Nola G. Cooper, BSc Melb., PhD N.S.W.

TUTORS

Felicity A. Dewar, BSc Qu. C. Kidd, BSc Syd.

HONORARY ASSOCIATE

B. V. Hamon, BSc BE Syd., MAIP

Department of Statistics

ASSOCIATE PROFESSORS OF MATHEMATICAL STATISTICS J. B. Douglas, MA BSc DipEd Melb. C. A. McGilchrist, BSc BEd Qld., MSc PhD N.S.W.

SENIOR LECTURERS

A. G. L. Elliott, BSc W.Aust.

P. J. Staff, BSc DipEd Syd., MSc PhD N.S.W.

M. K. Vagholkar, MSc Bom., PhD Lond., DIC

LECTURERS

P. J. Cooke, MSc N.E., MS PhD Stan.

R. B. Davis, BSc Syd., MSc N.S.W., DipEd N.E.

J. A. Eccleston, BSc Syd., MSc Man., PhD Corn.

TUTORS

V. J. Gebski, BA N.S.W. G. J. Newell, BSc N.S.W.

K. G. Russell, BA Macq.

Department of Engineering Mathematics

ASSOCIATE PROFESSOR

S. J. Prokhovnik, BA MSc Melb.

SENIOR LECTURERS

C. M. Groden, DiplMath Zür.

C. B. Kirkpatrick, MSc Syd., AInstP

LECTURERS

M. G. Greening, MA Lond. D. E. Mackenzie, BSc Tas. D. A. Mustard, BSc Syd., MSc N.S.W. W. J. Pretorius, MSc Rhodes, DIC

TUTORS

D. S. Craig, BSc *Qld*. A. T. Daoud, BSc *R'dg*.

J. T. Goozeff, MSc N.S.W.

B. C. Raphael, DipTechSc N.S.W.I.T.

SCHOOL OF PHYSICS

PROFESSOR OF PHYSICS AND HEAD OF SCHOOL E. P. George, BSc PhD Lond., DSc N.S.W., FInstP, FAIP

PROFESSORS OF EXPERIMENTAL PHYSICS
H. J. Goldsmid, BSc PhD DSc Lond., FInstP, FAIP
K. N. R. Taylor, BSc PhD Birm., FInstP

PROFESSOR OF THEORETICAL PHYSICS Vacant

ASSOCIATE PROFESSORS

D. Haneman, DSc Syd., PhD R'dg., FAIP

J. C. Kelly, BSc Syd., PhD R'dg., FInstP, AAIP

J. F. McConnell, MSc Syd., PhD N.S.W., MInstP, AAIP

L. G. Parry, BSc DipEd Syd., MSc PhD N.S.W., MInstP, AAIP

H. F. Pollard, MSc W.Aust., PhD N.S.W., MInstP, MAAS, MASA, AAIP

EXECUTIVE ASSISTANT TO HEAD OF SCHOOL K. Mann, BSc Qld., MSc N.S.W.

DIRECTOR OF FIRST YEAR STUDIES J. E. Giutronich, BSc Syd., PhD N.S.W., AAIP

ADMINISTRATIVE OFFICER

C. C. Rosario

SENIOR LECTURERS

H. G. L. Coster, MSc PhD Syd., MInstP, AAIP

J. E. Giutronich, BSc Syd., PhD N.S.W., AAIP

J. R. Hanscomb, BSc Qld., MSc PhD N.S.W., AAIP, GradInstP

N. R. Hansen, BSc DipEd Syd., MSc N.S.W., MInstP, AAIP

L. B. Harris, BSc Lond., BA DipEd Durh., PhD N.S.W., AIM, AInstP

B. R. Lawn, BSc PhD W.Aust., GradInstP

R. G. Simons, BSc Syd., MSc Tel Aviv, PhD N.S.W.

LECTURERS

J. I. Dunlop, BSc PhD N.S.W., AAIP, MAAS ATI

- C. T. Grainger, BSc DipEd Syd., MSc N.E., PhD N.S.W., MInstP, AAIP
- E. Harting, BSc PhD N.S.W., ASTC
- Veronica J. James, BA BSc Qld., PhD N.S.W., AAIP
- P. J. Jennings, BSc W.Aust., PhD Harv. K. Mann, BSc Qld., MSc N.S.W.
- K. H. Marsden, BSc Lond., MSc N.S.W., MInstP, ARCS, AAIP
- P. Mitchell, BSc PhD Adel., AAIP
- J. Oitmaa, BSc PhD N.S.W., AAIP
- G. L. Paul, MSc Syd., PhD Edin., AAIP
- G. J. Russell, BSc PhD N.S.W., GradInstP, GradAIP
- J. R. Shepanski, MSc Syd., AAIP
- A. M. Stewart

SENIOR TUTOR

I. R. Dunn, BSc BA Melb., MIEEE

SENIOR TUTOR DEMONSTRATOR

M. D. Knight, BSc N.S.W.

TUTOR DEMONSTRATORS

E. P. Eyland, BSc N.S.W., BD Lond. R. K. Katsch, BSc DipEd Syd. Z. Kerestes, BSc Syd. D. Littler, BSc N.S.W. R. Outhred, MSc Syd., PhD N.S.W. A. Pavey, BSc N.S.W. P. Pick, BSc PhD Syd. M. Varady, BSc Svd. D. J. Wheeler, BSc N.S.W., GradAIP

TEACHING FELLOWS

R. L. Calvert, BSc S'ton., MSc Qu.

- J. T. H. Ho, BSc H.K.
- J. L. Menendez-Cortinas, LicSci(Phys) Barcelona
- L. B. Shaw, BSc N.S.W., GradAIP
- C. Uher, BSc N.S.W., GradAIP

HONORARY ASSOCIATES

J. S. Dryden, MSc Melb., PhD Lond., DIC, FAIP G. H. Godfrey, MA BSc Syd., FInstP, FAIP, HonFIO J. L. Symonds, BSc Adel., PhD Birm., FInstP, FAIP G. K. White, MSc Syd., DPhil Oxon., FAA, FInstP, FAIP

HONORARY VISITING FELLOW

L. Lynch, BSc PhD N.S.W., GradAIP

PROFESSIONAL OFFICERS

H. Hofer, PhD Vienna, AAIP V. Kastalsky, BSc PhD N.S.W., ASTC, MInstP, AAIP B. Pantic, DipEng Belgrade, MIEAust F. G. M. Steenbeeke, DiplMechEng Arnhem T.H.

ADMISSIONS AND ENROLMENT PROCEDURE

ADMISSIONS OFFICE

The Admissions Office which is located in the Chancellery on the upper campus provides intending students (both local and overseas) with information regarding courses, admission requirements, scholarships and enrolment. Office hours are from 9.00 a.m. to 1.00 p.m. and 2.00 p.m. to 5.00 p.m. Monday to Friday and an evening service is provided during the enrolment period.

Applications for special admission, admission with advanced standing and from persons relying for admission on overseas qualifications are processed by the Admissions Office. The Office also receives applications from students who wish to transfer from one course to another, resume their studies after an absence of twelve months or more, or seek any concession in relation to a course in which they are enrolled. It is essential that the closing dates for lodgment of applications are adhered to, and, for further details the sections on "Rules Relating to Students" and "Enrolment Procedure for Undergraduate Courses" should be consulted.

Applications for admission to undergraduate courses from students who do not satisfy the requirements for admission (see section on "Requirements for Admission"), from students seeking admission with advanced standing, and from students who have had a record of failure at another university, are referred by the Admissions Office to the Admissions Committee of the Professorial Board.

Students seeking to register as higher degree candidates should discuss their proposals initially with the Head of the School in which they wish to register. An application is then lodged on a standard form and the Admissions Office, after obtaining a recommendation from the Head of the School, refers the application to the appropriate Faculty or Board of Studies Higher Degree Committee.

ADMISSIONS PROCEDURE

Details of the procedure to be followed by students seeking entry to first year courses at the University may be obtained from the Admissions Office or the Metropolitan Universities Admissions Centre.

Persons seeking entry to first year courses in one or more of the three Universities in the Sydney Metropolitan Area (Macquarie University, The University of New South Wales and the University of Sydney) are required to lodge a single application form with the Metropolitan Universities Admissions Centre, Third Floor, 13-15 Wentworth Avenue (near Liverpool Street), Sydney. Postal address: P.O. Box 7049, G.P.O., Sydney 2001. Telephone: 26 6301. On the application form, provision is made for applicants to indicate preferences for courses available in any of the three Universities. Students are notified individually of the result of their applications and provided with information regarding the procedures to be followed in order to accept the offer of a place at this University and complete their enrolment at the Enrolment Bureau, Unisearch House, 221 Anzac Parade, Kensington.

ENROLMENT PROCEDURE FOR UNDERGRADUATE COURSES

In 1974, it will be necessary for the University to limit entry into each Faculty and Board of Studies.

The enrolment procedure for the different classes of undergraduate students is as follows:

First Enrolments

(a) New South Wales residents already qualified for admission and persons who are applying for enrolment on the basis of qualifications gained or about to be gained outside New South Wales must lodge an application for enrolment with the Metropolitan Universities Admissions Centre, 13-15 Wentworth Avenue, Sydney (P.O. Box 7049, G.P.O., Sydney 2001) by 26th October, 1973.

(b) New South Wales residents qualifying for admission by the 1973 New South Wales Higher School Certificate Examination or the 1974 Sydney University Matriculation Examination and those who have attended a University in New South Wales in 1973 must apply for enrolment to the Metropolitan Universities Admissions Centre, 13-15 Wentworth Avenue, Sydney (P.O. Box 7049, G.P.O., Sydney 2001) by 18th January, 1974.

Students whose applications for enrolment are accepted will be required to complete their enrolment at a specified appointment time before the start of Session 1. Compulsory fees must be paid on the day of the appointment. However, in special circumstances and provided class places are still available, students may be allowed to complete their enrolment after the prescribed week subject to the payment of a late fee.

Application forms for enrolment and details of the application procedures may be obtained on application to the Registrar, P.O. Box 1, Kensington 2033.

Failure in First Year

First year students who failed all subjects at the 1973 Annual Examinations and who were not granted any deferred examinations will NOT follow the above procedure. They are required to "show cause" why they should be allowed to continue in the course, and should await instructions in writing from the Registrar as to the procedure.

Later Year Enrolments

All students enrolling other than for the first time and not included above must attend at the time and place during Enrolment Week as set out in the booklet published each year on enrolment procedures. Enrolment forms for these students will be prepared and available at the enrolment centre.

All full-time and part-time students in the Pure and Applied Chemistry course are required to complete an enrolment form in the last fortnight of Session 2. The forms may be obtained from the office of the School of Chemistry (Mr. A. Funnell). An appointment will then be made to complete enrolment in accordance with the special arrangements made by the school. These arrangements are also published in the booklet on enrolment procedures.

A late fee of \$10 will be incurred by students failing to enrol during Enrolment Week.

Miscellaneous Subjects

(students not proceeding to a degree or diploma)

Students may be accepted for enrolment in miscellaneous

subjects provided the University considers that the subject/s will be of benefit to the student and there is accommodation available. Only in exceptional circumstances will subjects taken in this way count towards a degree or diploma.

Students seeking to enrol in miscellaneous subjects should obtain a letter of approval from the Head of the appropriate School or his representative permitting them to enrol in the subject concerned. The letter should be given to the enrolling officer at the time of enrolment. Where a student is under exclusion he may not be enrolled in any miscellaneous subjects unless given approval by the Professorial Board.

Unless otherwise instructed, students who have obtained written permission to enrol should attend the Unisearch House enrolment centre (see next pages) on Friday 1st March, 2.00 p.m. to 6.00 p.m.

Students unable to enrol at the above time may enrol by attending the Admissions Office, Chancellery, at the times shown below, with a written permission to enrol from the Head of School.

Week commencing 4th March:	Monday to Friday 9.30 a.m. to 1.00 p.m. 2.00 p.m. to 4.30 p.m. 5.30 p.m. to 8.00 p.m.
Week commencing 11th March:	Monday to Friday 9.30 a.m. to 1.00 p.m. 2.00 p.m. to 4.30 p.m. Wednesday and Friday 5.30 p.m. to 8.00 p.m.

Preliminary Enrolment

Board of Studies in Science: Science Course

Before the end of Session 2, each student must obtain his or her personal enrolment form and 1974 programme form (form SC74) plus instruction forms from the Faculty of Science Office, Room 57, Main Building.

After notification of the annual examination results, the student should complete form SC74 and lodge it, together with re-enrolment form filled in as far as possible, at the Science Faculty office not later than 18th January, 1974. Students whose programme forms and re-enrolment forms are not received by 18th January, 1974, must re-enrol at a late re-enrolment session and an additional charge will be made.

Pure and Applied Chemistry Course and Optometry Course

Before proceeding on vacation students are required to attend the Office of the School to complete the necessary preliminary enrolment procedures.

Enrolment forms must be completed as far as possible and left with the School office.

BSc in Psychology Course

Each student must obtain his or her personal enrolment form and Personal Programme FORM AP/RE from the School of Psychology. The forms are available from 22nd October, 1973. After notification of the annual examination results the student should indicate the subjects already completed and the proposed programme for 1974 on FORM AP/RE and forward this, together with his enrolment form (completed except for the entry of subjects) to reach the Enrolment Officer, School of Psychology, not later than Friday, 18th January, 1974.

Enrolment Timetable

Board of Studies in Science: Science Course

After fulfilling preliminary enrolment requirements, students should complete their re-enrolment at Unisearch House in accordance with the following timetable:

Full-time Course

Year 2 & Year 1 Repeats

L · · · · ·	
Surnames A to G Surnames H to M	Wednesday 27th February 9.30 a.m. to 12.30 p.m. 2.00 p.m. to 4.30 p.m.
Surnames N to R Surnames S to Z	Thursday 28th February 9.30 a.m. to 12.30 p.m. 2.00 p.m. to 4.30 p.m.
Year 3	
Surnames A to J	Monday 25th February 2.00 p.m. to 4.30 p.m.
Surnames K to R Surnames S to Z	Tuesday 26th February 9.30 a.m. to 12.30 p.m. 2.00 p.m. to 4.30 p.m.
Year 4	
All students	Friday 1st March 9.30 a.m. to 12.30 p.m. 2.00 p.m. to 4.30 p.m.

New Students with Advanced Standing	Friday 1st March 9.30 a.m. to 12.30 p.m. 2.00 p.m. to 4.30 p.m.
Part-time Course	
Stage 2 & Stage 1 Repeats	Monday 25th February 6.00 p.m. to 8.00 p.m.
Stage 3 and Stage 4 Students	Tuesday 26th February 6.00 p.m. to 8.00 p.m.
Stage 5 & Later Stages	Wednesday 27th February 6.00 p.m. to 8.00 p.m.
New Students with Advanced Standing	Thursday 28th February 6.00 p.m. to 8.00 p.m.

Pure and Applied Chemistry Course

Students in the course are required to attend Unisearch House in accordance with the following timetable:

a.	Full-time Course	
	Year 2 & Year 1 repeats	Monday 25th February 2.00 p.m. to 4.30 p.m.
	Year 3	Wednesday 27th February 9.30 a.m. to 12.30 p.m.
	Year 4	Friday 1st March 9.30 a.m. to 12.30 p.m.
b.	Part-time Course	
	Stage 1 repeats & Stages 2 and 3	Tuesday 26th February 2.00 p.m. to 4.30 p.m. 6.00 p.m. to 8.00 p.m.
	Stage 4	Wednesday 27th February 2.00 p.m. to 4.30 p.m.
	Stages 5, 6 & later	Tuesday 26th February 2.00 p.m. to 4.30 p.m. 6.00 p.m. to 8.00 p.m.
c.	New Students with Advanced Standing	Wednesday 27th February 2.00 p.m. to 4.30 p.m.

Optometry Course

All students are required to attend Unisearch House, 221 Anzac Parade (across from Main Campus).

All students

Monday 25th February 2.00 p.m. to 4.30 p.m.

Enrolment Centre

Science	Unisearch House
Pure and Applied Chemistry	221 Anzac Parade
Optometry	(across from Main Campus)

School of Psychology

BSc in Psychology students must attend for re-enrolment at the School of Psychology, The Sciences Building, as follows:

Full-time Students	
Year 2 and Year 1 repeats	Tuesday 26th February 10.00 a.m. to 12 noon
Years 3 and 4	Tuesday 26th February 2.00 p.m. to 4.00 p.m.
Part-time Students All Stages	Wednesday 27th February 6.00 p.m. to 7.30 p.m.

Students who are unable to attend personally should send a representative at the specified time with a letter of authority to collect their form for them.

Students who fail to do this or fail to attend personally will be required to attend one of the late enrolment periods.

Students who have completed the final examinations but have a thesis still outstanding are required to enrol for the period necessary to complete the thesis and to pay the requisite fees.

Course details must be completed during the prescribed Enrolment Week. For details of fee requirements, including late fee provisions, see under Fees.

Enrolment Centre

School of Psychology

The Sciences Building

Level 10

Final Dates for Completion of Enrolment

No enrolments will be accepted from *new students* after the end of the second week of Session 1 (15th March, 1974) except with the express approval of the Registrar and the Head of the School concerned; no *later year enrolments* will be accepted after 31st March without the express approval of the Registrar which will be given in exceptional circumstances only.

UNIVERSITY UNION CARD

All students other than miscellaneous students are issued with a University Union membership card. This card must be carried during attendance at the University and shown on request.

The number appearing on the front of the card above the student's name is the student registration number used in the University's records. This number should be quoted in all correspondence.

The card must be presented when borrowing from the University libraries, when applying for travel concessions and when notifying a change of address. It must also be presented when paying fees on re-enrolment each year when it will be made valid for the year and returned. Failure to present the card could result in some inconvenience in completing re-enrolment.

A student who loses a Union card must notify the University Union as soon as possible.

New students will be issued with University Union cards at the University Union Enquiry Desk as soon as practicable after payment of fees. In the meantime, fees receipt form should be carried during attendance at the University and shown on request. A period of at least three weeks should be allowed to elapse after payment of fees before making application for the card. Cards will not be posted under any circumstances. A person who seeks to become a candidate for any degree of Bachelor of the University must first have qualified for matriculation and have satisfied the requirements for admission to the particular Faculty, Course or Subject* chosen.

In addition to complying with these conditions, candidates must be selected before being permitted to enrol in a course. In 1974, it will be necessary for the University to limit the number of students enrolling in all undergraduate courses.

Special Assistance for Aboriginal Students

The University may admit suitably qualified persons of Aboriginal descent outside of any quota restrictions.

Upon receipt of an application under this provision, the University will assess the applicant's potential to cope with University studies, and will make Student Counsellors available to discuss the choice of a course and subsequent career opportunities.

All enquiries relating to this scheme should be directed to the Registrar.

Matriculated Student

A candidate who has satisfied the conditions for matriculation and for admission to a course of study shall be classed as a "matriculated student" of the University, after enrolment.

A person who has satisfactorily met the conditions for admission may be provided with a statement to that effect on the payment of the prescribed fee.

SECTION A

General Matriculation and Admission Requirements

1. A candidate may qualify for matriculation by attaining in recognized matriculation subjects at one New South Wales

^{*} Students who have gained a 2S level pass in Mathematics and/or Science, while being admitted, may have difficulties in their chemistry, physics and mathematics studies and are strongly advised to attend the bridging courses in these subjects.

Higher School Certificate Examination or at one University of Sydney Matriculation Examination a level of performance determined by the Professorial Board from time to time.

2. The level of performance required to qualify for matriculation shall be

(a) passes in at least five recognised matriculation subjects, one of which shall be English and three of which shall be at Level 2 or higher;

and

- (b) the attainment of an aggregate of marks, as specified by the Professorial Board, in not more than five recognized matriculation subjects, such marks being coordinated in a manner approved by the Board.
- 3. The following subjects, and such other subjects as may be approved by the Professorial Board from time to time, shall be recognized matriculation subjects:—

English	Greek	Chinese
Mathematics	Latin	Japanese
Science	French	Hebrew
	German	Dutch
Modern History		Art
	Bahasa Indonesia	Music
Geography	Spanish	Industrial Arts
Economics	Russian	

- 4. A candidate who has qualified to matriculate in accordance with the provisions of Clauses 1, 2 and 3 may be admitted to a particular Faculty, course or subject provided that:—
 - (a) his qualification includes a pass at the level indicated in the subject or subjects specified in Schedule A as Faculty, course or subject prerequisites;

or

- (b) the requirements regarding these particular Faculty, course or subject prerequisites, as specified in Schedule A, have been met at a separate Higher School Certificate or University of Sydney Matriculation Examination.
- 5. Notwithstanding any of the provisions of Clauses 1 to 4, the Professorial Board may grant matriculation status to any candidate at the Higher School Certificate or University of Sydney Matriculation Examination who has reached an acceptable standard and may admit him to any Faculty, course or subject.

NOTE

- 1. For the purposes of clause 2 (a), Mathematics and Science BOTH PASSED at first level or second level full course shall together count as three subjects.
- 2. For the purposes of clause 2 (b), Mathematics and Science TAKEN either singly or together at first level or second level full course shall each count as one and one half subjects.

SCHEDULE A

FACULTY OR COURSE	FACULTY OR COURSE PREREQUISITES
Applied Science (excl. Applied Geography and Wool and Pastoral Sciences courses) Biological Sciences Engineering Industrial Arts Course Medicine Military Studies (Engineering course and Applied Science course) Science Bachelor of Science (Education)	 (a) Science at Level 2S or higher AND (b) either Mathematics at Level 2F or higher OR Mathematics at Level 2S, provided that the candidate's performance in this subject and his general level of attainment are at standards acceptable to the Professorial Board.
Architecture Applied Geography (Biogeography and Pedology specializations) Wool and Pastoral Sciences courses	 (a) Science at Level 2S or higher AND (b) Mathematics at Level 2S or higher
Applied Geography (Economic Geography specialization)	Either Mathematics at Level 2F or higher OR Mathematics at Level 2S, provided that the candidate's performance in this subject and his general level of attainment are at standards acceptable to the Professorial Board.

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FACULTY OR COURSE	FACULTY OR COURSE PREREQUISITES
Arts	English at Level 2 or higher
Commerce	 (a) Mathematics at Level 2S or higher AND (b) either English at Level 2 or higher OR English at Level 3, provided that the candidate's performance in this subject and his general level of attainment are at standards acceptable to the Professorial Board.
Law Combined Jurisprudence/Law Combined Arts/Law Combined Commerce/Law	Nil Nil As for Arts As for Commerce
Military Studies (Arts Course)	English at Level 2 or higher; OR English at Level 3, provided that the candidate's performance in this subject and his general level of attainment are at standards acceptable to the Professorial Board, and provided that a candidate so qualified shall not enrol in a course of English literature.
Social Work course	English at Level 2 or higher OR English at Level 3, provided that the candidate's performance in the subject and his general level of attainment are at standards acceptable to the Professorial Board, and provided that a candidate so qualified shall not enrol in English I.

SUBJECT	SUBJECT PREREQUISITES			
1.011—Higher Physics I 1.001—Physics I	As for Faculty of Science			
2.001—Chemistry I 17.011—Human Biology 25.001—Geology I 25.111—Geoscience I	Science at Level 2S or higher			
10.011—Higher Mathematics I	Mathematics at Level 2F or higher			
10.001-Mathematics I	Either Mathematics at Level 2F or higher			
	OR Mathematics at Level 2S, provided that the candidate's performance in the subject and his general level of attainment are at standards acceptable to the Professorial Board.			
10.021—Mathematics IT	Mathematics at Level 2S or higher			
15.102—Economics II	As for Faculty of Commerce			
50.111—English I 51.111—History IA 51.121—History IB	English at Level 2 or higher			
56.111—French I	French at Level 2 or higher			
59.111-Russian I	Russian at Level 2 or higher			
64.111—German I	German at Level 2 or higher			
65.111—Spanish I	Spanish at Level 2 or higher			
59.001—Russian IZ 64.001—German IZ 65.001—Spanish IZ	A foreign language, other than that in which enrolment is sought, at Level 2 or higher			

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SECTION B

Supplementary Provisions for Matriculation

Notwithstanding the provisions of Section A above, candidates may be accepted as "matriculated students" of the University under the conditions which are listed in the University Calendar. FEES

Payment of Fees

As from 1st January, 1974, no fees for tuition will be payable. Other fees and charges will still be payable. These include those charges raised to finance the expenses incurred in operating student activities such as the University Union, the Students' Union, the Sports Association and the Physical Education and Recreation Centre. Late fees are also charged where a student fails to observe required procedures by the appropriate time. Charges may also be payable, sometimes in the form of a deposit, for the hiring of kits of equipment which are lent to students for their personal use during attendance in certain subjects. Accommodation charges and costs of subsistence on excursions, field work etc. and for hospital residence (medical students) are payable in appropriate circumstances. In order to become a student member of the University in any particular course of study it is necessary to meet the entrance requirements for the course and to enrol formally in it. To effect enrolment it is necessary to present a duly completed and authorized enrolment form to the University cashier together with where payable, either the appropriate fees, or an authority authorizing those fees to be charged to some other person or institution.

Completion of Enrolment

All students are required to attend the appropriate enrolment centre during the prescribed enrolment period* for authorization of course programme. Failure to do so will incur a late fee of \$10.

First year students (including students repeating first year) must complete enrolment (including fee payment) before they are issued with class timetables or permitted to attend classes. A first year student who has been offered a place in a course to which entry is restricted and who fails to complete enrolment at the appointed time may lose the place allocated.

Fees should be paid during the prescribed enrolment period but will be accepted during the first two weeks of Session 1. (For late fees see below.) No student is regarded as having completed an enrolment until fees have been paid. Fees will not be accepted

^{*} The enrolment periods for Sydney students are prescribed annually in the leaflets on enrolment procedures.

(i.e. enrolment cannot be completed) from new students in yearlong courses after the end of the second week of Session 1 (i.e. 15th March, 1974), and after 31st March from students who are re-enrolling, except with the express approval of the Registrar, which will be given in exceptional circumstances only.

Students enrolling for the first time in any year at the commencement of Session 2 are required to pay all fees due within the first two weeks of that session. Student Activities fees due will be one half of the annual fees.

These arrangements also apply to medical students and although the structure of the academic year in the later years of the course in Medicine differs from that followed in other courses, medical students are required to observe the same dates for payment as apply to students in other courses.

Assisted Students

Scholarship holders or sponsored students who have not received an enrolment voucher or appropriate letter of authority from their sponsor at the time when they are enrolling should complete their enrolment paying their own fees. A refund of fees will be made when the enrolment voucher or letter of authority is subsequently lodged with the Cashier.

Extension of Time

Any student who is unable to pay fees by the due date may apply in writing to the Deputy Registrar (Student Services) for an extension of time. Such application must state year or stage, whether full-time or part-time, and the course in which enrolment is sought, describe clearly and fully the reasons why payment cannot be made and what extension is required, and must be lodged before the date on which a late fee becomes payable. Normally the maximum extension of time for the payment of fees is one month for fees due in Session 1 and for one month from the date on which a late fee becomes payable in Session 2.

Where an extension of time is granted to a first year student in Session 1, such student may only attend classes on the written authority of the Registrar, but such authority will not normally be given in relation to any course where enrolments are restricted.

Failure to Pay Fees or Other Debts

Any student who fails to pay prescribed fees or charges or is otherwise indebted to the University and who fails to make a satisfactory settlement of his indebtedness upon receipt of due notice ceases to be entitled to the use of University facilities. Such a student is not permitted to register for a further session, to attend classes or examinations, or to be granted any official credentials.

No student is eligible to attend the annual examinations in any subject where any portion of his fees for the year is outstanding after the end of the fourth week of Session 2 (16th August, 1974).

In very special cases the Registrar may grant exemption from the disqualification referred to in the two preceding paragraphs upon receipt of a written statement setting out all relevant circumstances.

STUDENT FEES*

All undergraduate students and students taking miscellaneous subjects (with the exception of External students) will be required to pay ----

University Union**

\$20.00 — entrance fee

Student Activities Fees University Union**

Sports Association**

\$30.00 — annual subscription

\$4.00 — annual subscription

\$7.00 --- annual subscription

Students' Union** Miscellaneous

\$17.00 --- annual fee

(The miscellaneous fee is used to finance expenses generally of a capital nature relating to student activities. Funds are allocated to the various student bodies for projects recommended by the Student Affairs Committee and approved by the University Council.)

Depending on the subject being taken, students may also be required to pay ---

Pathology Instrument Kit — \$10. (Refundable on return in satisfactory condition.)

Special Examination Fees

Examinations conducted under				
special circumstances	 \$11	for	each	subject.
Review of examination results	 \$11	for	each	subject.

^{*} Fees quoted in the schedule are current at the time of publication and may be amended by the Council without notice.
** Life members of these bodies are exempt from the appropriate fee or

fees.

LATE FEES

Session 1 — First Enrolments	
Fees paid in the late enrolment period and before commencement of Session 1	\$10
Fees paid during the 1st and 2nd weeks of Session 1	\$20
Fees paid after the commencement of the 3rd week of Session 1 with the express approval of the Registrar and Head of the School concerned	\$40

Session 1 — Re-enrolments

Failure to attend enrolment centre during enrolment week	\$10
Fees paid after the commencement of the 3rd week of Session 1 to 31st March	\$20
Fees paid after 31st March where accepted with the express approval of the Registrar	\$40

Session 2 — All Enrolme	nts						
Fees paid in 3rd and	l 4th	weeks o	f Session	2	••••	••••	\$20
Fees paid thereafter						••••	\$40

WITHDRAWAL FROM COURSE

- 1. Students withdrawing from a course are required to notify the Registrar in writing.
- 2. Where notice of withdrawal from a course is received by the Registrar before the first day of Session 1 a refund of all fees paid will be made.
- 3. On notice of withdrawal:
 - (a) a partial refund of the University Union Entrance Fee will be made on the following basis: any person who has paid the entrance fee in any year and who withdraws from membership of the University Union after the commencement of Session 1 in the same year, or who does not renew his membership in the immediately succeeding year may on written application to the Warden receive a refund of half the entrance fee paid.

- (b) A partial refund of other Student Activities Fees will be made on the following basis:
 University Union \$7.50 in respect of each half session.
 University of New South Wales Students' Union where notice is given prior to the end of the fifth week of Session 1, \$3.50; thereafter no refund.
 University of New South Wales Sports Association where notice is given prior to the fifth week a full refund is made; thereafter no refund.
 Miscellaneous Student Activities Fee \$4.25 in respect of each half session.
- 4. Where initial enrolment is made at commencement of Session 2 in any year and the student subsequently withdraws, a refund of fees based on the above rules may be made.

Cashier's Hours

The cashier's office is open for the payment of fees from 9.30 a.m. to 1.00 p.m., and from 2.00 p.m. to 4.30 p.m. Monday to Friday. It is open for additional periods during the first four weeks of Session 1. Students are advised to consult noticeboards for details.

GENERAL CONDUCT

Acceptance as a member of the University implies an undertaking on the part of the student to observe the regulations, by-laws and other requirements of the University, in accordance with the declaration signed at the time of the enrolment.

In addition, students are expected to conduct themselves at all times in a seemly fashion. Smoking is not permitted during lectures, in examination rooms or in the University Library. Gambling is also forbidden.

Members of the academic staff of the University, senior administrative officers, and other persons authorised for the purpose, have authority, and it is their duty, to check and report on disorderly or improper conduct or any breach of regulations occurring in the University.

ATTENDANCE AT CLASSES

Students are expected to be regular and punctual in attendance at all classes in the course or subject in which they are enrolled. All applications for exemption from attendance at lectures or practical classes must be made in writing to the Registrar.

In the case of illness or of absence for some other unavoidable cause a student may be excused by the Registrar from nonattendance at classes for a period of not more than one month, or on the recommendation of the Dean of the appropriate Faculty for any longer period.

Applications to the Registrar for exemption from re-attendance at classes, either for lectures or practical work, may only be granted on the recommendation of the Head of the appropriate School. The granting of an exemption from attendance does not carry with it exemption from payment of fees.

Application forms for exemption from lectures are available at the Admissions Office and should be lodged there (with a medical certificate where applicable). If session examinations have been missed this fact should be noted in the application. Where a student has failed a subject at the annual examinations in any year and re-enrols in the same course in the following year, he must include in his programme of studies for that year the subject in which he has failed. This requirement will not be applicable if the subject is not offered the following year; is not a compulsory component of a particular course; or if there is some other cause, which is acceptable to the Professorial Board, for not immediately repeating the failed subject.

Where a student has attended less than eighty per cent of the possible classes, he may be refused permission to sit for the examination in that subject.

INDEBTEDNESS TO THE UNIVERSITY

Any student who is indebted to the University and who fails to make a satisfactory settlement of his indebtedness upon receipt of due notice ceases to be entitled to membership and privileges of the University. Such a student is not permitted to register for a further session, to attend classes or examinations, or to be granted any official credentials.

In very special cases the Registrar may grant exemption from the disqualification referred to in the preceding paragraph upon receipt of a written statement setting out all relevant circumstances.

COURSE TRANSFERS

Students wishing to transfer from one course to another must apply on an application form obtainable from the Admissions Office, Chancellery, by Friday, 18th January. As the number of places in each course is limited, failure to apply by 18th January, 1974, will probably result in the application for transfer being unsuccessful.

Students whose applications to transfer are successful are required to comply with the enrolment procedures for the year/ stage of the new course in which they expect to enrol. Unless otherwise instructed they must present the letter granting approval of the transfer to the enrolling officer.

Students who have not received advice regarding their application to transfer before the date on which they are required to enrol should check with the Admissions Office.

Students should also advise the Enrolling Officer of the School in which they are enrolled of their intention to transfer.

ADMISSION WITH ADVANCED STANDING

Any person who makes application to register as a candidate for any degree or other award granted by the University may be admitted to the course of study leading to such degree or award with such standing on the basis of previous attainments as may be determined by the Professorial Board.

Students should consult the University Calendar for complete details regarding "Admission with Advanced Standing".

CHANGES IN COURSE PROGRAMMES AND WITHDRAWAL FROM SUBJECTS

Students seeking approval to substitute one subject for another (including change of session), add one or more subjects to their programme or discontinue part or all of their programme must make application to the Registrar through the Head of the School responsible for the course on forms available from School offices. The Registrar will inform students of the decision. Application to enrol in additional subjects must be submitted by 31st March.

Approval of withdrawal from subjects is not automatic, each application being determined after considering the circumstances advanced as justifying withdrawal.

It is emphasized that:

- 1. Withdrawal from a subject, tuition in which extends over the academic year, at any time after the May recess;
- 2. withdrawal from a subject, tuition in which extends over only one session, at any time after one month from the commencement of the subject; or
- 3. failure to sit for the examinations in any subject in which the student has enrolled,

shall be regarded as failure to satisfy the examiners in the subject, unless written approval to withdraw without failure has been obtained from the Registrar.

If a student applies after the following dates, to withdraw from a subject he will most likely be awarded a failure in the subject: Subject taken over Session 1 only 4th April 1974 Subject taken over Session 2 only 22nd August 1974 Subject taken over both sessions 26th May 1974

STUDENT RECORDS

All students will receive enrolment details forms by 30th April and 2nd September. It is not necessary to return the forms unless any information recorded thereon is incorrect. Amended forms must be returned to the Examinations and Student Records Section by 14th May and 16th September respectively. Amendments notified after the closing date will not be accepted unless exceptional circumstances exist and approval is obtained from the Registrar. Amended forms returned to the Registrar will be acknowledged in writing within fourteen days.

RESUMPTION OF COURSES

Students wishing to resume their studies after an absence of twelve months or more are required to apply to the Admissions Office for permission to re-enrol by 18th January, 1974. Students re-enrolling in this way will normally be required to satisfy conditions pertaining to the course at the time of re-enrolment. This condition applies also to students who have been re-admitted to a course after exclusion under the rules restricting students re-enrolling.

ANNUAL EXAMINATIONS

Formal examinations in most faculties are held in June-July and November-December. Provisional timetables including the dates and times of examinations are posted on the central notice boards in the Wallace Wurth Medical School, Biological Sciences Building, the Chancellery, Central Lecture Theatre Block, Dalton (Chemistry) Building, Main Building (Mining and Physics), outside the Science Theatre and in the Western Grounds Area on 7th May and 10th September. Students must advise the Examinations Unit (Chancellery) of clash of examinations by 20th May and 23rd September. Final timetables will be displayed, and individual copies available for students, on 4th June and 29th October.

Misreading of the timetable is not an acceptable excuse for failure to attend an examination.

A student suffering from a physical disability which puts him at a disadvantage in written examinations should apply to the Registrar in writing, as early as possible, for special provisions to be made for him to take examinations. The request should be supported by medical or other evidence.

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Examinations are conducted in accordance with the following rules and procedure:-----

- (a) Candidates are required to obey any instruction given by an examination supervisor for the proper conduct of the examination.
- (b) Candidates are required to be in their places in the examination room not less than ten minutes before the time for commencement.
- (c) No bag, writing paper, blotting paper, manuscript or book, other than a specified aid, is to be brought into the examination room.
- (d) No candidate shall be admitted to an examination after thirty minutes from the time of commencement of the examination.
- (e) No candidate shall be permitted to leave the examination room before the expiry of thirty minutes from the time the examination commences.
- (f) No candidate shall be re-admitted to the examination room after he has left it unless during the full period of his absence he has been under approved supervision.
- (g) A candidate shall not by any improper means obtain, or endeavour to obtain, assistance in his work, give, or endeavour to give, assistance to any other candidate, or commit any breach of good order.
- (h) Smoking is not permitted during the course of examinations.
- (i) All answers must be in English unless otherwise directed. Foreign students who have the written approval of the Officer-in-Charge of Examinations may use standard translation dictionaries.
- (j) A candidate who commits any infringement of the rules governing examinations is liable to disqualification at the particular examination, to immediate expulsion from the examination room and to such further penalty as may be determined in accordance with the By-laws.

A student who through serious illness or other cause outside his control is unable to attend an examination is required to bring the circumstances (supported by a medical certificate or other evidence) to the notice of the Registrar not later than seven days after the date of the examination, and may be required to submit to medical examination. A student who attempts an examination yet claims that his performance is prejudiced by sickness on the day of the examination, must notify the Registrar or Examination Supervisor, before, during or immediately after the examination and may be required to submit to medical examination.

A student who believes that his performance at an examination has been affected by serious illness during the year or by other cause outside his control, and who desires these circumstances to be taken into consideration in determining his standing is required to bring the evidence (supported by medical certificates or other evidence) to the notice of the Registrar not later than seven days after the date of the examination.

In the assessment of a student's progress, consideration is given to work in laboratory and class exercises and to any term or other tests given throughout the year, as well as to the results of written examinations.

Final examination results are posted to the term addresses of students and it is therefore essential that any change of address be advised to the Examination and Student Records Section. Results are also posted on School notice boards and in the foyer of the Library. No examination results will be given by telephone.

Examination results may be reviewed for a fee of \$11 a subject, which is refundable in the event of an error being discovered. Such a review will consist primarily in ensuring that all questions attempted by candidates have been marked and that the total of all marks awarded are correct. Applications for review must be submitted on the appropriate form to the Examinations and Student Records Section, together with the necessary fee by the date indicated on the notification of results.

Examination Results

Graded Passes

Passes will be graded as follows:

High Distinction (indicates a quite superior performance).

Distinction (indicates a superior performance).

Credit (indicates a good, but not superior performance).

Pass (indicates the achievement of an acceptable minimum level of competence in relation to the course objectives).

Pass Conceded

A pass conceded may be granted to students where the mark in the subject is slightly below the required standard and whose overall performance warrants it.

Terminating Pass

A terminating pass may be granted where the mark for the subject is below the required standard. A terminating pass will not permit a student to progress further in the subject or to enrol in any other subject for which a pass in the subject is a co-requisite or prerequisite. A student granted a terminating pass may attempt a deferred examination, if available, to improve his performance, but if the student fails the deferred examination, the terminating pass will stand.

DEFERRED EXAMINATIONS

Deferred examinations may be granted in the following cases:---

- (i) When a student through illness or some other acceptable circumstance has been prevented from taking the annual examination or has been placed at a serious disadvantage during the annual examinations. Applications for deferred examination in this category must be lodged with the Registrar with appropriate evidence of the circumstances (e.g., medical certificate) not later than seven days after the examination concerned. All such applications shall be reported to the Head of the School responsible for the subject. Before a deferred examination is granted on medical grounds, regard shall be paid to the student's class and assignment work in the subject, to his general performance in the year, and to the significance of the annual examination in compiling the composite mark.
- (ii) To help resolve a doubt as to whether a student has reached the required standard in a subject.
- (iii) To allow a student by further study to reach the required standard in a subject. The granting of a deferred examination in such cases will be based on the general quality of the student's performance.
- (iv) Where a student's standing at the annual examinations is such that his progression or graduation could depend on his failure in one subject only, then his position in that subject shall be again reviewed with a view to determining whether a deferred examination may be granted notwithstanding his failure otherwise to qualify for such concession.

Deferred examinations must be taken at the centre in which the student is enrolled, unless he has been sent on compulsory industrial training to remote country centres or interstate. In this case the student must advise the Registrar, on a form available from his school or the Enquiry Desk, Chancellery, of relevant particulars, before leaving for his destination, in anticipation that deferred examination papers may have to be forwarded to him. Normally, the student will be directed to the nearest University for the conduct of the deferred examination.

A student eligible to sit for a deferred examination must lodge with the Accountant an application accompanied by the fee of \$8 per subject, by the date indicated on the notification of results.

Conceded Deferred Examination

A conceded deferred examination, may be granted to a student where the mark in the subject is below the standard at which deferred examinations have been granted in the subject but whose overall performance warrants such a concession.

APPLICATION FOR ADMISSION TO DEGREE OR DIPLOMA

Applications for admission to a degree or diploma of the University must be made on the appropriate form in a student's final year. Applicants should ensure that they have completed all requirements for the degree or diploma, including industrial training where necessary. Any variation such as cancelling of application in order to proceed to an honours degree or submission of an application following discontinuation of honours programme, must be submitted in writing to the Registrar no later than 30th January.

RESTRICTION UPON STUDENTS RE-ENROLLING

The University Council has adopted the following rules governing re-enrolment with the object of requiring students with a record of failure to show cause why they should be allowed to re-enrol and retain valuable class places. These rules apply retrospectively from 1st January, 1971.

1. (i) A student shall show cause why he should be allowed to repeat a subject in which he has failed more than once. (Failure in a deferred examination as well as in the initial examination counts, for the purpose of this regulation, as one failure). Where such subject is prescribed as a part of the student's course he shall be required to show cause why he should be allowed to continue the course.

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Notwithstanding the provisions of Clause 1(i)

- (ii) A student enrolled in the first year or first stage of any course who has failed in more than half the programme in which he is enrolled for that year or stage shall be required to show cause why he should be allowed to continue in the course.
- (iii) A student enrolled in the first year of the Medical course who has failed in more than one subject of that year shall be required to show cause why he should be allowed to continue in the Medical course.
- (iv) The provisions of sections (ii) and (iii) of this rule shall be deemed to apply to any student on transfer from another course or institution whose programme of studies in the first year of enrolment immediately following transfer is comprised of subjects so chosen that half or more of such subjects are listed in the University Calendar as first year subjects.
- 2. Notwithstanding the provisions of Clause 1, a student shall be required to show cause why he should be allowed to continue a course which he will not be able to complete in the time set down in the following schedule:—

Number of	Total time allowed from
years in	first enrolment to
course	completion (years)
3	5
4	6
5	8
6	9
7	11
8	12

3.

No full-time student shall, without showing cause, be permitted to continue a course unless all subjects of the first year of his course are completed by the end of his second year of attendance. No student in the Faculty of Arts shall, without showing cause, be permitted to continue a course unless he completes four subjects by the end of his second year of attendance. No full-time student in the Bachelor of Social Work course shall without showing cause be permitted to continue with the course unless he completes the equivalent of four full subjects by the end of his second year of attendance. No part-time student in a course in which progression is by stage shall without showing cause be permitted to continue a course in which he will not be able to complete all subjects of the first two stages by the end of his fourth year of attendance and all subjects of the third year and fourth stages of his course by the end of his seventh year of attendance.

No part-time student in the Science course shall without showing cause be permitted to continue a course in which he will not be able to complete level one Mathematics and six other level one units by the end of his fourth year of attendance and fourteen units inclusive of at least three at level two of his course by the end of his seventh year of attendance.

No student in the Faculty of Medicine shall, without showing cause, be permitted to continue with the medical course unless he completes the second year of the course by the end of his third year of attendance, and the third year of the course by the end of his fourth year of attendance.

- 4. A student who has a record of failure in a course at another University shall be required to show cause why he should be admitted to this University. A student admitted to a course at this University following a record of failure at another University shall be required to show cause, notwithstanding any other provisions in these rules, why he should be permitted to continue in that course if he is unsuccessful in the annual examinations in his first year of attendance at this University.
- 5. Any student excluded under any of the Clauses 1-3 may apply for re-admission after two academic years and such application shall be considered in the light of any evidence submitted by him.
- 6. A student wishing "to show cause" under these provisions shall do so in writing to the Registrar. Any such application shall be considered by a committee, hereinafter referred to as the Re-enrolment Committee appointed by the Professorial Board, which shall determine whether the cause shown is adequate to justify his being permitted to continue his course or re-enrol as the case may be.

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- 7. The Vice-Chancellor may on the recommendation of the Re-enrolment Committee exclude from attendance in a course or courses any student who has been excluded from attendance in any other course under the rules governing re-enrolment and whose record at the University demonstrates, in the opinion of the Re-enrolment Committee and the Vice-Chancellor, the the student's lack of fitness to pursue the course nominated.
- 8. A student who has failed, under the provisions of Clause 6 of these rules, to show cause acceptable to the Re-enrolment Committee why he should be permitted to continue in his course, and who has subsequently been permitted to re-enrol in that course or to transfer to another course, shall also be required to show cause, notwithstanding any other provisions in these rules, why he should be permitted to continue in that course if he is unsuccessful in the annual examinations immediately following the first year of resumption or transfer of enrolment as the case may be.
- 9. Any student who is excluded from attendance in any course or subject under the provisions of these rules may appeal to an Appeal Committee constituted by Council for this purpose. The decision of the Appeal Committee shall be final.
- 10. The notification to any student of a decision by the Re-enrolment Committee to exclude the student from attendance in any course or subject shall indicate that the student may appeal against the decision to an Appeal Committee. In lodging such appeal the student shall ensure that a complete statement is furnished of all grounds on which the appeal is based and shall indicate whether or not the student wishes to appear in person before the Appeal Committee.

In considering an appeal the Appeal Committee, on the basis of the student's academic record and the stated grounds of appeal, shall decide:

- (i) whether there are grounds which justify the Committee seeing the student in person, or
- (ii) whether there is sufficient information available to the Committee to allow decision without seeing the student in person

and so proceed to determine the application accordingly.

RE-ADMISSION AFTER EXCLUSION

Applications for re-admission must be made on the standard form and lodged with the Registrar not later than 30th June of the year prior to that for which re-admission is sought. An application should include evidence of appropriate study in the subjects (or equivalents) on account of which the applicant was excluded. In addition, evidence that the circumstances which were deemed to operate against satisfactory performance at the time of exclusion are no longer operative or are reduced in intensity, should be furnished. An applicant may be required to take the annual examinations in the relevant subjects as qualifying examinations in which case re-admission does not imply exemption from the subject. Late applications cannot be considered where, in the opinion of the University, insufficient time will be available for the student to prepare himself for any qualifying examinations which may be required.

It should be noted that a person under exclusion may not be enrolled in miscellaneous subjects unless he has received the approval of the Professorial Board on the recommendation of the Admissions Committee.

Persons who intend applying for re-admission to the University at a future date may seek advice as to ways in which they may enhance their prospects of qualifying for re-admission. Enquiries should be made on a form obtainable from the Examinations and Student Records Section, and lodged with the Registrar.

OWNERSHIP OF STUDENTS' WORK

The University reserves the right to retain at its own discretion the original or one copy of any drawings, models, designs, plans and specifications, essays, theses or other work executed by students as part of their courses, or submitted for any award or competition conducted by the University.

CHANGE OF ADDRESS

Students are requested to notify the Student Records Section of the Registrar's Division of any change in their address, as soon as possible. Failure to do this could lead to important correspondence not reaching students. The University cannot accept responsibility if official communications fail to reach students who have not notified their change of address. A Change of Address Advice form is available at Faculty and School offices and at the Enquiry Counters on the Ground Floor of the Chancellery Building.

NOTICES

Official University notices are displayed on the notice boards and students are expected to be acquainted with the contents of those announcements which concern them.

LOST PROPERTY

All enquiries concerning lost property should be made to the Superintendent on Extension 2503 or to the Lost Property Office at the Union.

PARKING WITHIN THE UNIVERSITY GROUNDS

Because of the limited amount of parking space available, only the following categories of students may apply for a permit; motor cycle owners (annual fee \$3.90); higher degree students (limited issue, annual fee \$7.80); postgraduate, and senior undergraduate students who have completed three years of a full-time or parttime course (annual fee \$3.90). A permit will allow access to the campus between 5 p.m. and 11 p.m. on weekdays and during library hours on Saturdays, Sundays and public holidays. Enquiries should be made to the Property Section, Room 240, The Chancellery Building, or phone 663 0351, extension 2920. It should be noted that increasing demand for parking space may require the imposition of further restrictions.

APPLICATION OF RULES

General

Any student who requires information on the application of these rules or any service which the University offers may make inquiries from the Admissions Office, the Student Counselling Centre or the Registrar.

Appeals

Section 5(c) of Chapter III of the By-laws provides that "Any person affected by a decision of any member of the Professorial Board (other than the Vice-Chancellor) in respect of breach of discipline or misconduct may appeal to the Vice-Chancellor, and in the case of disciplinary action by the Vice-Chancellor, whether on appeal or otherwise, to the Council".

THE LIBRARY

The University library is on the upper campus and adjacent to the Chancellery, and the Arts and Commerce Buildings. The Bio-Medical Library is in the Biological Sciences Building with a branch at Prince Henry Hospital ('Phone: 661 0111). The Law Library is temporarily housed on the 4th Floor of the Science Building on the upper campus.

The Library's Undergraduate Collection covers the teaching and research interests of the Faculty, and students are expected to read widely and critically from it.

It is recommended that students attend the "Introduction to the Library" which is held at advertised times during Orientation Week and the first week of Session 1. The "Introduction" uses audio-visual aids to describe the physical layout of the undergraduate library and the services available to readers.

Copies of the booklet, Guide to the Library, are available on request.

Students who are interested in a subject approach to information may attend a course which outlines methods of searching for information in libraries. This course runs for eight hours over a period of one week.

Individual assistance for readers with specific library problems is provided by the *Reader Assistance Unit* which is located in the foyer.

Staff and students must use a machine readable identification card to borrow from the main University Library. Personal identification is required in the other libraries listed. For students a current union card is acceptable. Staff must apply to the Library for a library card.

THE UNIVERSITY UNION

The University Union, housed in the circular building and joined by a courtyard to an adjacent rectangular building, is located near the entrance to the Kensington campus from Anzac Parade. The third building in the Union complex was completed in 1971. Membership of the Union is compulsory for all registered students of the University and is also open to all members of staff and graduates of the University.

On the lower campus the range of facilities provided by the Union includes a cafeteria service and other dining facilities, a large shopping centre, cloak room, banking and hairdressing facilities, showers, a women's lounge, common, games, reading, meeting, music, practice, craft and dark rooms. Photocopying, sign printing and stencil cutting services are also available. On the upper campus there is a cafeteria and coffee bar on the ground floor of the Sciences Building, a vending area and lounge off the Science Plaza, a similar facility off the Commerce Courtyard and a Snack Bar at the Golf House on the corner of High and Botany Streets.

The Union also sponsors and conducts courses in many facets of the arts including weaving, photography, creative dance and yoga.

STUDENT ACCOMMODATION

The Kensington Colleges

Accommodation for students is provided within the group of The Kensington Colleges which comprise Basser College, Goldstein College and Philip Baxter College. The group houses 450 men and women students, as well as staff members. Tutors in residence provide tutorial assistance in a wide range of subjects.

Board and residence fees, which are payable on a session basis, amount to \$336 per session. Intending students should apply in writing to the Master, Box 24, Post Office, Kensington, N.S.W. 2033, from whom further information is available.

International House

International House accommodates over 180 students of whom half are Australian; the remaining half is made up of students from some 20 different countries. First-year students who have come to the University straight from school are not eligible for residence because preference is given to mature undergraduates and postgraduate students. Fees are \$24 per week.

Students should apply as soon as possible if they wish to reside at International House at a later date. They should write to the Warden, International House, P.O. Box 88, Kensington, N.S.W. 2033 for information.

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New College

This Church of England College is the first of the independent Colleges on the Campus of the University. There are no religious tests, and accommodation is available for 220 men and women in single study-bedrooms. Fees are \$25 per week and may change in 1974.

Enquiries should be addressed to The Master, New College, Anzac Parade, Kensington, N.S.W. 2033.

Warrane College

This College, an affiliated Roman Catholic residential college, was completed in 1970, and provides accommodation for 200 students and fourteen resident tutors.

Basic fees are \$28 per week for board and residence, payable on a session basis, and a registration fee of \$20. Fees may change in 1974. Intending students should write to The Master, Warrane College, Box 123, P.O. Kensington, N.S.W. 2033.

Shalom College

Shalom College, which opened in 1973, provides accommodation for 86 men and women students. The basic fee for residents is \$28 a week although this may change in 1974. Non-resident membership is available to students who wish to avail themselves of the Kosher dining room and tutorial facilities.

Applications for residence and further information should be addressed to The Master, Shalom College, The University of New South Wales, Box 1, P.O. Kensington, N.S.W. 2033.

Other Accommodation

Students requiring other than Residential College accommodation may make personal application to the Housing Officer (Ext. 3260) at the Student Amenities Unit. Current lists are kept of accommodation available at recognized boarding houses, private homes, and in serviced and unserviced apartments.

STUDENT AMENITIES UNIT

The Amenities Unit is concerned with student welfare and its activities are associated with sport and recreation, travel and student accommodation. It works in close liaison with the Sports Association, assisting the various clubs, and administers sporting facilities for both grade and social competitions. The Unit also has the added responsibility of the Physical Education and Research Centre where attractive recreational programmes for students and staff are provided. Concessional application forms for all types of travel may also be obtained at the Enquiry Desk in the Chancellery or at the Student Amenities Unit. A Housing Officer is also available to assist students with any off-campus accommodation problems.

Location: The Student Amenities Unit is located in Hut B at the foot of Basser Steps.

Phone: 663 0351, Extension 2235 Sports Association; 3271 Physical Education and Recreation Centre; 3261 Travel; and 3260 Accommodation.

STUDENT EMPLOYMENT UNIT

The Student Employment Unit offers assistance with career employment for final year students and graduates of the University. This service includes the mailing of regular job vacancy notices to registered students and a campus interview programme for final year students. Careers advice and assistance is also available to undergraduates. Assistance is offered in finding vacation employment which gives either course related experience or industrial training experience where this is a course requirement. Information and advice regarding cadetships, undergraduate and postgraduate scholarships is also available.

The Service is located in the Chancellery on the ground floor. Telephone: 663 0351 ext. 3259 for employment and careers advice, or 663 0351 ext. 2086 for cadetships and industrial training information.

CHAPLAINCY SERVICE

This service is provided for the benefit of students and staff by five Christian Churches and by the Jewish congregation. Chaplains are in attendance at the University at regular times. A Chapel is also available for use by all denominations.

The University Chapel and full-time chaplains are located in Hut F near the Chemistry Building. They may be contacted by phone at the following extensions: Anglican, 2684; Jewish, 3273; Roman Catholic, 2379; Churches of Christ, Methodist and Seventh Day Adventist, 2683.

STUDENT HEALTH UNIT

A student health and first aid centre is situated within the University. It is staffed by two qualified medical practitioners, assisted by a nursing sister and two secretaries. This medical service, although therapeutic, is not intended to replace private or community health services. Thus, where chronic or continuing conditions are revealed or suspected, the student is referred to a private practitioner or to an appropriate hospital for specialist opinion and/or treatment. The health service is not responsible for fees incurred in these instances. The service is confidential and students are encouraged to attend for advice on matters pertaining to health.

The service is available to all enrolled students by appointment, free of charge between 9 a.m. and 5 p.m. Mondays to Fridays, and additionally to part-time students from 6 p.m. to 8 p.m. on Tuesdays and Thursdays during session. For staff members, immunizations are available, and first-aid service in the case of injury or illness on the campus.

The centre is located in Hut E on the northern side of the campus in College Road.

Appointments may be made by calling at the centre or by telephoning extension 2679 or 3275 during the above hours.

STUDENT COUNSELLING AND RESEARCH UNIT

The Student Counselling and Research Unit offers a free, confidential counselling service to help students, individually or in groups, to deal with problems, and to make plans and decisions associated with their personal, academic, and vocational progress.

Interviews, and group programmes, are available between 9 a.m. and 8 p.m. each week-day. Appointments may be made at the Unit, which is located at the foot of Basser Steps, or by ringing 663 0351, extensions 2600-2605 between 9 a.m. and 5 p.m.

FINANCIAL ASSISTANCE TO STUDENTS

In addition to the Tertiary Allowances Scheme financed by the Australian Government (see Scholarships for details), the following forms of assistance are available.

(a) The Students' Union and the University have co-operated to provide assistance to students who are in financial difficulties which are considered likely to prejudice their progress with their studies.

Three main forms of assistance are available:

1. Deferment of Payment of Fees

Deferments may be granted for a short period, usually one month, without the imposition of a late fee penalty, provided the deferment is requested prior to the due date for fee payments.

In exceptional circumstances the University may consider granting deferments for up to twelve months or even longer. In cases where payment is deferred to 31st December, examination results will not be published or made available until such time as the outstanding fees are paid. Where deferments are granted to a date beyond 31st December, the University may require the student to enter into a formal agreement to repay the fees.

2. Short Term Cash Loans

Donations from the Students' Union, the University Union and other sources have made funds available for urgent cash loans not exceeding \$100.00. These loans are normally repayable within one month.

3. Long Term Cash Loans

An amount of up to \$300.00 is available from this fund, repayable usually after twelve months or within twelve months of graduation or upon withdrawal from the course. This scheme is funded jointly by the University and the Students' Union. Students are required to enter into a formal agreement with the University to repay such a loan.

(b) Early in 1973 the Australian Government made funds available to the University to provide loans to students in financial difficulty. The loans are to provide for living allowances and other approved expenses associated with attendance at University. Under this scheme allowances are paid approximately monthly during the academic year. Repayment usually commences after twelve months of graduation or upon withdrawal from the course. Students are required to enter into a formal agreement with the University to repay the loan.

From the same source of funds as mentioned in the preceding paragraph students who are in extremely difficult financial circumstances may apply for assistance by way of non-repayable grant. In order to qualify for a grant a student must generally show that the financial difficulty has arisen from misfortune beyond his control.

Applications for all forms of assistance may be made personally to the Deputy Registrar (Student Services).

FINANCIAL ASSISTANCE TO ABORIGINAL STUDENTS

Financial assistance is available from a number of sources to help Aboriginal students. Apart from the Australian Government's Tertiary Allowances Scheme, to be introduced in 1974, there is a Commonwealth Aboriginal Study Grant Scheme. Furthermore, the University may assist Aboriginal students with some essential living expenses in exceptional circumstances.

All enquiries relating to this scheme should be directed to the Deputy Registrar (Student Services).

UNIVERSITY CO-OPERATIVE BOOKSHOP LTD.

Membership is open to all students, on payment of a fee of \$5, refundable when membership is terminated. Members receive an annual rebate on purchases of books.

THE STUDENTS' UNION

The Students' Union was formed in 1952 as an organization, duly recognised by the University Council, to represent the student body and to provide a central organization for the administration of student activities. In the words of its constitution, "The Students' Union is formed for the purpose of advancing the interests of University men and women, facilitating their general scientific and technical education, and fostering a University spirit among them."

The Union affords a recognised means of communication between the student body and the University administration, and represents its members in all matters affecting their interests. It aims to promote the cultural, educational and recreational life of the University and to encourage a permanent interest among graduates in the life and progress of the University.

Membership of the Union is compulsory for all registered students of the University and the annual subscription is \$7.

The Students' Union is governed by a Council consisting of student representatives from the various faculties of the University, representatives of Life Members, overseas students, and of the University and the Sports Association. The Council is elected annually.

THE SPORTS ASSOCIATION

The Sports Association is a student organization within the University, and it caters for a variety of competitive sports for both men and women.

In December 1952 the University Council approved the establishment of the Sports Association which consisted of five clubs. As the University has grown, the Association has expanded, and today includes over thirty clubs.

The controlling body of the Association is the General Committee which consists of a President, Secretary, Treasurer, eight Vice-Presidents and two delegates from each of the affiliated clubs.

Membership of the Association is compulsory for all registered students, and the annual subscription is \$4.

PHYSICAL EDUCATION AND RECREATION CENTRE

The Physical Education and Recreation Centre consists of eight squash courts and a main building. The latter has a large gymnasium and ancillary practice rooms for fencing, table tennis, judo and weightlifting. The Supervisor of Physical Recreation is responsible for this Centre and provides a recreational programme for both students and staff. Those who desire to participate in the recreational programmes should contact the Supervisor on Extension 3271.

STUDENT CLUBS AND SOCIETIES

Students have the opportunity of joining a wide range of clubs and societies. Affiliated with the Students' Union are the School and Faculty associations, and the numerous religious, social and cultural clubs. There are also many sporting clubs (33) affiliated with the Sports Association.

Clubs and societies seeking to use the name of the University in their title, or seeking University recognition, must submit their constitutions either to the Students' Union or the Sports Association if they wish to be affiliated with either of these bodies, or to the Registrar for approval by the University Council.

THE UNIVERSITY REGIMENT

Enquiries should be made to the Adjutant at the Regimental Depot in Day Avenue just west of Anzac Parade.

THE N.S.W. UNIVERSITY SQUADRON

Enquiries should be made to the Commanding Officer at Squadron Headquaters at the corner of City and Darlington Streets, Darlington 2008.

ROYAL AUSTRALIAN NAVY

Enquiries should be made to the Royal Australian Naval Liaison Officer, Professor J. S. Ratcliffe, Commander, R.A.N.V.R., at the School of Chemical Engineering. Phone 663 0351, ext. 2406.

SCHOLARSHIPS, BURSARIES, CADETSHIPS AND PRIZES

SCHOLARSHIPS

Students undertaking courses in the Faculty of Science are eligible to apply for the following scholarships. Not all scholarships are offered each year. Towards the end of December prospective applicants should enquire from the Student Employment and Scholarships Unit which scholarships are available.

Except where otherwise specified, applications on the forms obtainable from the Admissions Office ('phone: 663 0351, ext. 2485) must be lodged with the Registrar, the University of New South Wales, P.O. Box 1, Kensington 2033. A separate application must be lodged for each category of scholarship.

In addition to these scholarships, a number of industrial organizations and Government Departments sponsor students at the University. In the past, such students generally had their University fees paid by the employer and were employed at cadet rates of pay during the course. However, these schemes, details of which are set out in Section B of the University Calendar, are currently under review and interested persons should contact the relevant organization or department for up-to-date information.

TERTIARY ALLOWANCES SCHEME

In 1974, no new awards will be offered under the Commonwealth University Scholarship Scheme. Instead a new system of Australian Government Assistance for tertiary students, called the Tertiary Allowances Scheme, will operate. This scheme will apply to students who commence approved courses—in 1974 as well as those who commenced their courses earlier.

Means-tested living and other allowances will be available to full-time students enrolled in an approved course who satisfy certain academic and residence requirements, are unbonded and who do not receive assistance in excess of \$350 from other scholarships. No age limit will apply.

Students enrolled in the following types of university courses will be eligible for assistance:

- Undergraduate and postgraduate bachelor degree courses
- Postgraduate diplomas
- Combined bachelor degree courses offered by institutions
- Master's qualifying courses

Benefits

Means-tested Living Allowance: The maximum rates of living allowance are \$850 per annum for students living at home and \$1,400 per annum for students living away from home.

The maximum rates of living allowance will be paid where the adjusted family income is equal to or less than \$5,300 per annum. The adjusted family income is assessed by subtracting from the gross income of both parents business expenses and an amount of \$450 for each dependent child other than the student.

When the adjusted family income exceeds \$5,300 p.a. the amount of living allowance will be reduced by \$2 for every \$10of income until the family income exceeds \$10,600 per annum. After this level, the living allowance will be reduced by \$3 for every \$10 of income.

A concession may be made where there are other children in the family undertaking tertiary education with scholarship assistance from schemes other than the Tertiary Allowances Scheme of less than \$350 p.a.

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Students qualifying for living allowance will also receive the following allowances where appropriate:

Incidentals Allowance: The Incidentals Allowance of \$100 is designed to help the students meet the cost of those fees which have not been abolished — the Students' Union, University Union and sports fees — and other expenses associated with their studies.

Travel Allowance: Students whose home is in the country may be reimbursed the cost of three return trips per year, during vacation time.

Dependent's Allowance: This is made up of allowances of \$8 per week for a dependent spouse and \$4.50 per week for each child.

How to apply: Higher School Certificate students will be able to obtain application forms from their school. Students who do not already hold a scholarship may obtain forms from the

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Admissions Office, or from The Regional Director, New South Wales State Office, Department of Education, Sydney Plaza Building, 59 Goulburn Street, Sydney, N.S.W. 2000. (Telephone 2 0929).

N.B.: Current Commonwealth Scholarship holders will have application forms posted to them. All forms should be returned to the above address as soon as possible.

AUSTRALIAN POSTGRADUATE AWARDS

Australian Postgraduate Course Awards

The Australian Government provides a number of awards for full-time postgraduate study in courses leading to the degree of Master by formal course work. Persons permanently domiciled in Australia who are under 45 years of age on 1 January of the year in which the award is to be taken up and who are University graduates or will graduate in the current academic year, are eligible for the awards. Award holders receive a living allowance of \$2,900 paid over the academic year. Other allowances may also be paid in certain cases.

Application for awards tenable at the University must be lodged with the Registrar by 30 September each year.

Australian Postgraduate Research Awards

The Australian Government also provides each year a number of awards for full-time postgraduate study and research. The awards are renewable annually up to a maximum duration of two years in the case of a candidate for a Masters degree or three years in the case of a PhD candidate. In special circumstances, a PhD candidate may be granted an extension of tenure into a fourth year. Persons permanently domiciled in Australia who are under 35 years of age on 1 January of the year in which the award is to be taken up and who are University graduates or will graduate in the current academic year, are eligible for the awards. Award holders receive a living allowance of \$3,050 per annum. Other allowances may also be paid in certain cases. The closing date for applications is 31 October each year.

OTHER AWARDS

Scholarships in Optometry

The Australian Optometrical Association (New South Wales Division) and Gibb & Beeman (Spectacle Makers) Pty. Ltd., offer annually one scholarship each to the value of \$500 per annum. These scholarships are available to students who desire to enrol in the full-time degree course in Optometry leading to the degree of Bachelor of Optometry at the University of New South Wales. Applicants must be residents of New South Wales.

The Australian Optometrical Association offers annually a scholarship open to any student enrolling in the first or second year of the full-time degree course in Optometry, and provides him benefits of \$250 p.a. if enrolled in Year 1 and \$500 p.a. if enrolled in subsequent years. If his home address is more than 50 miles away from the University he receives an additional allowance of up to \$150 p.a. and his first class return surface travel ticket to his home each year. The scholarship is open to students whose parents are ordinarily permanent residents of Australia or who are themselves permanent residents of Australia.

Further details are available in the University Calendar.

Applications on the prescribed form available from the Registrar should be lodged with the Registrar by the 14th January each year.

The Fell Scholarship (University Residential Colleges)

The Fell Scholarship is available to any undergraduate who is or will be in residence at one of the Colleges under the administration of Kensington Colleges Ltd. during the year of the award. The annual value of the Scholarship is \$100. It may be held concurrently with other scholarships.

In awarding the scholarship the academic merit and financial need of the applicant will be taken into consideration.

Applications must be made on the appropriate form and lodged with the Master, Kensington Colleges Ltd., Box 24, P.O., Kensington 2033 (telephone 663 0651).

BURSARIES

Bursaries Awarded by the Bursary Endowment Board

A number of Bursaries tenable at the University are awarded to candidates of merit at the Higher School Certificate Examination whose family income falls within certain limits prescribed by the Bursary Endowment Board.

Applications should be made to the Secretary, Bursary Endowment Board, Box 7077, G.P.O., Sydney 2001.

CADETSHIPS

Sponsored Students

Many private industrial and commercial organisations sponsor students in Science courses. The conditions under which students are sponsored vary from company to company, but in general the company meets all compulsory fees. Industrial training is generally undertaken with the sponsoring company.

Students are advised to consult the Admissions Office or the Student Counselling Unit at Kensington for further details concerning scholarships and cadetships and for information concerning companies sponsoring students.

School/Department	Donor/Name of Prize	Value \$	Awarded for
General	Sydney Technical College Union Award	50.00	Leadership in the development of student affairs and academic proficiency through- out the course.
	University of New South Wales Alumni Association	Statuette	Achievement for community benefit— students in their final or graduating year.
School of Botany	E. O. Tout Memorial	40.00	Best aggregate any five units offered by School of Botany.
School of Chemistry	Abbott Laboratories Pty. Ltd.	50.00	2.622 Organic Chemistry II.
	Australian Chemical Holdings Ltd.	21.00	2.001 Chemistry I.
	Australian Consolidated Industries Ltd.	30.00	Subject selected by Heau of School.
	Borden Chemical Co. (Aust.) Pty. Ltd.	50.00	Subject selected by Head of School.
	Chamber of Manufactures of New South Wales	10.00	Subject selected by Head of School.
	C.S.R. Chemicals Ltd.	100.00	Chemistry Honours.
	Inglis Hudson Bequest	6.00	2.611 Organic Chemistry I.
	Merck, Sharp & Dohme (Aust.) Pty. Ltd.	52.50 52.50	Chemistry—Level 2 Units Science Course. Chemistry—Level 3 Units Science Course.
	The Nestlé Co. (Aust.) Ltd.	20.00	Subject selected by Head of School.

PRIZES (continued)

School/Department	Donor/Name of Prize	Value \$	Awarded for	
School of Chemistry (cont.)	Parke Pope	10.50	Subject selected by Head of School.	THE
	V. S. Rawson	10.50	Subject selected by Head of School.	C
	Science Association	(i) 10.50 (ii) 10.50	Subject selected by Head of School. Subject selected by Head of School.	UNIVERSITY
	Tooheys Ltd.	10.00	Subject selected by Head of School.	R
	Tooth & Co. Ltd.	10.00	Subject selected by Head of School.	Ĩ
	Univeler Aust. Pty. Ltd.	21.00	2.322 Physical Chemistry II.	~
	George Wright	10.50	2.001 Chemistry I—Full-time students only.	OF
School of Mathematics	School of Mathematics	25.00	Higher Mathematics I.	NEW
		25.00	Higher Pure Mathematics II.	S
		25.00	Higher Applied Mathematics II.	SOUTH
		25.00	Higher Pure Mathematics III.	TΗ
		25.00	Higher Applied Mathematics III.	\$
	The Broken Hill Proprietary Co. Ltd.	50.00	Higher Theory of Statistics II.	A
	The W.D. & H.O. Wills (Aust.) Ltd. Prize	50.00	Higher Theory of Statistics III.	WALES
	I.C.I. Australia Ltd.	50.00	Theory of Statistics IV.	
	Statistical Society of Australia (New South Wales Branch)	20.00	General proficiency—Theory of Statistics subjects.	

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Department of Optometry	Australian Optometrical Association	30.00	Subject selected by Head of School.			
	Chamber of Manufactures of New South Wales	10.00	Subject selected by Head of School.			
	Arthur Cocks & Co. Ltd.	10.50	Subject selected by Head of School.			
	Contavue Laboratories Pty. Ltd.	Trial fitting set of contact lenses	Subject selected by Head of School.			
	Filmer Sceats	25.00	31.812 Optometry II.			
	International Optical Corporation Ltd.	21.00 21.00	Subject selected by Head of School. Subject selected by Head of School.			
	L. G. Darcey Memorial	25.00	31.811 Optometry I.			
	Martin Wells Pty. Ltd.	30.00 50.00 30.00	31.821 Special Anatomy and Physiology. 31.831 Diseases of the Eye. Final Year Essay.			
	G. Nissel & Co. Aust. Pty. Ltd.	150.00	31.813 Optometry III and 31.841 Clinical Optometry. Contact Lenses sections.			
	Optical Products Pty. Ltd.	21.00	Subject selected by Head of School.			
	Opticians and Optometrists' Association of N.S.W.	25.00	Subject selected by Head of School.			
School of Physics	Head of School's Prize in Physics	20.00	Area selected by Head of School.			
	Physics Staff	60.00	Physics III.			
	Physics IV Prize	40.00	Physics IV.			
	School Prize for Physics II	40.00	Physics II.			
School of Psychology	Australian Psychological Society	50.00	A Year IV Psychology subject selected by Head of School.			

There are two types of courses available within the Faculties of Science and Biological Sciences. The first is the Science Course, which allows a student to select sequences from a variety of the sciences. The second type of course is of a more specialized nature. Such courses are offered in Pure and Applied Chemistry, Optometry and Psychology. Details of each of these courses are given below.

SCIENCE COURSE

The Science Course is administered by the Dean of the Faculty of Science through his nominated representative on behalf of the Schools within the Faculty of Biological Sciences and the Faculty of Science, as well as the Schools of Applied Geology, Anatomy, Physiology and Pharmacology, History and Philosophy of Science, and the Department of Electronic Computation (School of Electrical Engineering).

The pass degree (Bachelor of Science) is based on a unit structure. A unit in experimental subjects comprises 90 hours of lectures, tutorials and laboratory work, and in theoretical subjects comprises an equivalent loading of lectures and tutorials. A unit may be of 14 or 28 weeks' duration.

The unit structure has been chosen to allow flexibility in the choice of a course of study and the regulations have been framed so that a student may choose a pattern suitable for:—

- (1) a general scientific education.
- (2) the training of science teachers.
- (3) professional training in a specific discipline.
- (4) professional training in interdisciplinary areas.

Units are grouped according to levels. Level I subjects are all double units. Level II units normally follow after level I prerequisites. Level III units in most cases follow after level II prerequisites.

The requirements of a pass degree may be met by completing units in accordance with the regulations set out below and which constitute a major in one of the disciplines of the Schools of the Faculties of Science and Biological Sciences, or the Schools of Applied Geology (Faculty of Applied Science), Physiology or Anatomy (Faculty of Medicine). Some units may also be included from Schools in the Faculties of Arts and Engineering. A major normally includes four level III units chosen from those offered by a particular School.

All students are required to complete three General Studies subjects. Patterns and outlines of these subjects are listed in the Department of General Studies Handbook, which is available free of cost.

The minimum time required to complete a pass degree is three years' full-time study or an equivalent period part-time. Some subject groupings cannot, however, be completed in the minimum time due to timetable difficulties.

A student may be admitted, subject to meeting conditions defined in the regulations, to an honours course which involves an extra year of full-time study or two years of part-time study. Those intending to seek admission to an honours year should consult the Head of the appropriate school on completion of the first year subjects.

Any arrangement of units to be studied must be approved by the Dean of the Faculty of Science. Advice on recommended course patterns may be obtained from the Education Officer of the School in which a student intends to major.

REGULATIONS GOVERNING THE SCIENCE COURSE

1. Definitions

The Science course is administered by the Dean of the Faculty of Science through his nominated representative.

The pass degree is based on a unit structure. A unit may be of 14 or 28 weeks' duration, and units are grouped according to levels. Level I subjects are all double units, level II units normally follow after level I prerequisites and level III units, in most cases, follow after level II prerequisites. A major sequence normally includes four level III units chosen from those offered by a particular school, although a number of schools offer more than four such units.

A prerequisite unit is one which must be completed prior to enrolment in the unit for which it is prescribed. A co-requisite unit is one which must either be completed successfully before or be studied concurrently with the unit for which it is prescribed. An excluded unit is one which cannot be counted together with the unit which excludes it towards the degree qualification. In exceptional circumstances, on the recommendation of the head of the appropriate school, the Dean of the Faculty of Science may waive or vary a particular prerequisite or co-requisite.

CARE SHOULD BE TAKEN IN THE CHOICE OF UNITS TO ENSURE THAT THE PATTERN COMPLIES WITH THE REGULATIONS SET OUT IN SECTION 3(a). CERTAIN COMBINATIONS OF UNITS CANNOT BE COMPLETED IN THE MINIMUM TIME DUE TO THE RESTRICTIONS OF TIMETABLES. COPIES OF TYPICAL COURSE PATTERNS ARE AVAILABLE FROM THE FACULTY OFFICE.

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2. Regulations governing the Science course

(a) Requirements for a pass degree

In order to qualify for admission to the degree of Bachelor of Science under these regulations a candidate shall attend classes and satisfy the examiners in Science units and General Studies subjects chosen as follows-

- (i) At least 23 Science units shall be included from the list set out in section 3(a) and three General Studies subjects from the list in section 3(b).
- (ii) The 23 Science units shall comply with the prerequisites, co-requisites and exclusion conditions set out in section 3(a) and also shall conform to the following restrictions:

not less than 8 units, nor more than 10 units may be from level I:

not less than 4 units may be from level III, and these four shall be chosen from related disciplines.

- (iii) One of 10.001 Mathematics I, OR 10.011 Higher Mathematics I, OR 10.021 Mathematics IT shall be included.
- (iv) In addition to the specific prerequisites listed in Clause 3(a), additional general prerequisites are required by some schools as a preliminary to certain advanced level units. These units, which are scheduled below, should be taken in the first year of enrolment together with compulsory mathematics. Eight units are normally taken in first year.

School of Chemistry 1.001 or 1.011

School of Applied Geology 2.001 Chemistry.

1.001 or 1.011 and

School of	2.001 Chemistry.
Biochemistry	17.011 Human Biology and 17.021 Com-
School of Botany	parative Functional Biology; plus one other subject.
School of Microbiology	<i>Note:</i> In making their choice students should consider carefully, in their first year,
School of Zoology	the requirements of level II and level III units.
School of Anatomy	17.011 Human Biology and 17.021 Comparative Functional Biology.
School of Physiology	2.001 Chemistry and 17.011 Human Biology and 17.021 Com- parative Functional Biology.
(v)	Only one from each of the following subjects/ units may be included:
	(a) 12.001 Psychology or 26.121 Psychology.
	(b) 52.111 Philosophy or 26.521 Philosophy.

- (c) Any unit listed in Section 3(a) or the equivalent unit offered at Wollongong University College which contains similar syllabus material.
- (vi) A full-time student is required to complete the appropriate level I Mathematics and six other approved level I units in the first two years of attendance or else show cause to the satisfaction of the Professorial Board why he should be allowed to re-enrol. The remaining units of the course may be completed in any order consistent with the requirements concerning prerequisite and co-requisite units as set out in Clause 3(a).
- (vii) The proposed course must be approved by the Dean of the Faculty of Science or his representative at enrolment. In special circumstances, the Dean may grant a student permission to defer enrolment in certain level I units until the second year of the course. Where any alteration in the course approved at enrolment is desired, the student must obtain the approval of the Dean or his representative for the new course.

(b) Requirements for an honours degree

- (i) In order to qualify for admission to the honours degree of Bachelor of Science a candidate shall:
 - 1. Satisfy the requirements for a pass degree but without proceeding to graduation;
 - 2. Undertake an extra year of full-time or two extra years of part-time study.
- (ii) Admission to an honours course is granted by the Head of School. Students wishing to proceed to an honours degree must apply to the Head of the appropriate school on completion of pass degree requirements.
- (iii) A suitably qualified candidate may be admitted to an honours course in one of the following:

Anatomy Applied Mathematics Applied Physics Biochemistry Biological Technology Botany Chemistry Computer Science Entomology Geoscience History and Philosophy of Science** Microbiology Physics Physiology Psychology† Pure Mathematics Theory of Statistics Zoology

- (iv) To qualify for admission to an honours course, a student must have completed successfully 8 level III units in the pass degree course* except that in special cases the Head of the appropriate school may approve entry without such a qualification.
- (v) Further to requirements listed in paragraph 2(b) (iv), to qualify for entry into an honours year a student must have completed any special units at required grades as determined by the Head of the School, prior to admission to the Honours year.

^{*} For the honours course in Applied Physics the corresponding normal requirement is both (a) at least six level III units to be completed and (b) at least eight units at levels II and III to be completed at Credit grade or better or in the respective Higher version.

[†] The Honours subject is 12.014 Psychology IV.

^{**} The honours year will be offered in 1976 and subsequent years.

In order to ascertain any such special conditions, a student contemplating honours is advised to consult the Head of School at the end of the first vear of study.

(vi) Upon admission to the honours course a student must attend lectures, read and engage in laboratory work as required by the Head of School.

3. Schedule of Units

(a) Science units

These are listed under the Schools which provide the instruction and are divided into levels. Students must observe the prerequisites and co-requisites. Some Schools offer higher units to which special prerequisites apply and which are designed to lead to honours. Students contemplating honours studies must ensure that they have selected appropriate units. Some units are terminating so that students taking these may not qualify to continue studies in that School. When selecting terminating units students must ensure that a choice of a major sequence is still available. Note that many units are of half year duration so that it is necessary to choose units which give a balanced programme of study over the year.

The Dean of the Faculty has the power to vary in exceptional cases the prerequisites and/or co-requisites set down below on the recommendation of the Head of the appropriate school.

See following pages B8-B27.

(b) General Studies

Turn to page B28.

FACULTY OF SCIENCE

SCHOOL OF PHYSICS

No.	Name	Level	Unit Value	When Offered	Hours p.w.	Prerequisites‡	Co-requisites‡	Excluded
1.001	Physics I	I	2	Full yr.	6	Sc. Faculty Ent.		
1.011	Higher Physics I	IH	2	Full yr.	6	Sc. Faculty Ent.		
	PHYSICS LEVEL II							
1.112A	Electromagnetism	II	1	Session 2	6	1.001, 10.001 1.001, 10.001	10.211A	1.122A
1.112B	Modern Physics	п	1	Session 1	6		10.211A	1.122B, 1.212C
1.112C	Waves in Continuous Media and Thermodynamics	II	1	Full yr.	2	1.001, 10.001	10.211A	1.122C
1.212T	Physics IIT (any two of 1.212A, 1.212B, 1.212C, 1.212D)	п	1	Full yr.	3	1.001 or 1.011; 10.001 or 10.011 or 10.021		1.112B (excluded by 1.212C only)
	HIGHER PHYSICS LEVEL	п						
1.122A	Electromagnetism	IIH	1	Session 2	6	1.011, 10.001	10.211A	1.112A
1.122B	Quantum Physics	IIH	1	Session 1	6	1.011, 10.001	10.211A	1.112B
1.122C	Thermodynamics and Mechanics	ΠН	1	Full yr.	2	1.011, 10.001	10.211A	1.112C

No.	Name	Level	Unit Value	When Offered	Hours p.w.	Prerequisites [‡]	Co-requisites:	Excluded	· T
1.113A	PHYSICS LEVEL III Wave Mechanics and Spectroscopy	III	1	Session 1	6	1.112B, 1.112C 10.211A		1.123A and 1.123D 2.023A and 10.222F	FACULTIES OF
1.113B	Electromagnetic Fields and Physical Optics	ш	1	Session 2	6	1.112A, 10.211A			
1.113C	Statistical Mechanics and Solid State	III	1	Session 1	6	1.112B and 1.112C	1.113A	1.123B and 1.123C	DOGIC
1.113D	Astrophysics and Nuclear Physics	III	1	Session 2	6	1.112B	1.113A§ or 10.222F	1.123C	AL SC
	HIGHER PHYSICS LEVEL	ш							IEI
1.123A	Quantum Mechanics	ШН	1	Session 1	6	1.122B, 1.122C, 1.122A, 10.211A, 10.111A, 10.111B		1.113A, 2.023A 10.222F	ACES A
1.123B	Electromagnetic Theory and Statistical Mechanics	ШН	1	Session 1	6	1.122C, 1.122A 10.211A		1.113C, 10.212C, 10.222C	BIOLOGICAL SCIENCES AND SCIENCE
1.123C	Solid State and Nuclear Physics	IIIH	1	Session 2	6	1.122B, 10.211A	1.113A or 1.123A or 10.222F	1.113C and 1.113D	ENCE
1.123D	Atomic Physics and Spectroscopy	IIIH	1	Session 2	6	1.122B, 1.122A 10.211A	1.123A or 10.222F	1.11 3A	ВÀ

SCHOOL OF PHYSICS (Continued)

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No.	Name	Level	Unit Value	When Offered	Hours p w	Prerequisites‡	Co-requisites‡	Excluded
	PHYSICS LEVEL III SUPPI	EME	TAR	Y UNITS				
.133A	Electronics	III**	1	Session 1	6	1.001 or 1.011		
.143A	Biophysics	III	1	Session 1	5	1.112C		
.143B	Solid State Devices and Electronics	ш	1	Session 2	6	1.133A		
143C	Magnetism	ш	1	Session 2	5	1.112A, 1.112B 10.211A		
143D	Conceptual Framework of Physics	III	1	Session 2	5	1.112C† 1.112A, 1.112B		
143E	Electrical and Optical Properties of Solids	III	1	Session 2	5		1.113C	1.123D
143F	Marine Acoustic and Seismic Methods (Oceanography Unit)	III	1	Full yr.	3	10.211A or 10.221A or 10.031		
153A	Hydrodynamics and Magnetohydrodynamics	IIIH	1	Full yr.	4	1.122A, 1.122C 10.211A, 10.111A 10.111B		
153B	Relativity and Electro- magnetism	IIIH	1	Full yr.	4	1.122A and 1.122C 10.211A, 10.111A 10.111B		10.212C a 10.222C

SCHOOL OF PHYSICS (Continued)

This unit may be a co-requisite in special cases.

[‡] Students should note the additional mathematics prerequisite to units of Higher Physics III. Where a unit is specified at level II as a prerequisite or co-requisite the level IIH unit may be substituted. Students must apply to the Head of School for admission to Physics honours and they must have completed at least Physics units 1.123A, 1.123B, 1.123C and 1.123D.

** This unit may be taken in second year of the course provided prerequisites have been completed.

§ This co-requisite may be waived under certain circumstances subject to the approval of the School of Physics.

No.	Name	Level	Unit Value	When Offered	Hours p.w	Prerequisites	Co-requisites	Excluded
2.001	Chemistry I	I	2	Full yr.	6	H.S.C. Science 2S		AC
2.002A‡	Physical Chemistry	II	1	Half yr.	6	2.001 and 10.001 or 10.011 or 10.021 and 1.001 or 1.011 or 1.041 or 1.031 or 1.061		
2.002B‡	Organic Chemistry	п	1	Half yr.	6	2.001 and 10.001 or 10.011 or 10.021		
2.002C‡	Inorganic Chemistry	Π	1	Half yr.	6	2.001 and 10.001 or 10.011 or 10.021		
2.003A	Physical Chemistry	ш	1	Half yr.	6	2.002A		2.013A, 2.023A a
2.003B	Organic Chemistry	ш	1	Half yr.	6	2.002B		
2.003C	Inorganic Chemistry	III	1	Half yr.	6	2.002C		
2.003D	Analytical Chemistry	III	1	Half yr.	6	2.002A, 2.002C		
2.003E	Nuclear and Radiation Chemistry	ш	1	Half yr.	6	2.002A*, 2.002C*		
2.013A	Theoretical Chemistry	III	1	Half yr.	6	2.002A, and 10.031 or 10.211A		2.003 A, 2 .023A
2.023A§	Chemical Physics	ш	1	Full yr.	3	10.211A (or equiv.) and 2.002A or 1.112B		1.113A, 1.123A 2.013A, 2.003A
2.033A**	Macromolecules	ш	1	Half yr.	6	2.002A and 2.002B		2.013A, 2.003A

SCHOOL OF CHEMISTRY

* If taken as one unit independently, prerequisites may be waived subject to the approval of Head of School.

‡ All three level II units must be taken by students majoring in Chemistry.

§ This is a unit which may be taken in conjunction with units of Applied Mathematics or Physics. It cannot be included as a Chemistry level III unit.

** Entry to this course requires permission from the Head of School.

B11

SCHOOL OF MATHEMATICS

No.	Name	Level	Unit Value	When Offered	Hours p.w.	Prerequisites	Co-requisites	Excluded*
			MAT	HEMATICS				
10.001	Mathematics I	Ι	2	Full yr.	6			
10.011	Higher Mathematics I	ĪH	$\overline{2}$	Full yr.	Ğ			
0.021	Mathematics IT	IT	22	Full yr.	Ğ			
0.031‡	Mathematics	ĪĪ	ī	Full yr.	ž	10.001 or 10.011		
			-	J	-	or 10.021 Credit		
0.032§	Mathematics	III	1	Full yr.	2	10.031		
			-	1 un j.,	~	10.051		
			PURE N	IATHEMAT	201			
	Pure Mathematics Level II		I ORD II		105			
0.111A	Linear Algebra	п	1	Full yr.	2	10.001 or 10.011		10. 121A
0.111B	Analysis	ÎÎ	i	Full vr.	2	10.001 or 10.011		10.121A
0.111C	Algebra and Geometry	ÎÎ	î	Full yr.	2	10.001 or 10.011	10.111A.	10.1216
			-	I uli yl.	2	10.001 0/ 10.011	10.111A, 10.111B.	
							10.211A	10.121A
	Higher Pure Mathematics Leve	а П†					10.211A	10.121A
0.121A	Algebra	IIH	1	Full yr.	2 1	10.011		10.111A
0.121B	Real and Complex Analysis	ÎÎĤ	î	Full yr.	$\frac{23}{2\frac{1}{2}}$	10.011		10.111A
0.121C¶	Number Theory and Geometry	ÎĤ	i	Full yr.	$\frac{21}{2\frac{1}{2}}$	10.011	10.121A,	
			-	I uli yl.	22	10.011	10.121A, 10.121B.	10.111C,
							10.121B, 10.221A or	10.112A
							10.221A or 10.211A	
	Pure Mathematics Level III***						10.211A	
0.112A	Number Theory and Algebra	III	1	Full yr.	2	10.111A	10.111C	10.121A,
	·····, ·····		-	I ull yl.	4	10.1117	10.1110	10.121A, 10.121C
0.112B	Real Analysis	III	1	Full vr.	2	10.111B		10.121C 10.121B
0.112C	Differential Geometry	ĪĪĪ	ĩ	Full yr.	ĩ	10.111A, 10.211A,		10.1210
			-	- un yr.	~	10.111B		10.122C
0.112D	Topology and Set Theory	ш	1	Full yr.	2	10.001 or 10.011	10.111A,	10.1220
			-	,	-	10.001 0/ 10.011	10.111B,	
							10.211A	
							10.21IA	

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B1

SCHOOL OF MATHEMATICS (Continued)

No.	Name	Level	Unit Value	When Offered	Hours p.w.	Frerequisi	tes Co-req	uisites	Excluded*
0.112E	Complex Analysis and Differential Equations	III	1	Full yr.	2	10.111B, 10	.211A		10.122E
	Higher Pure Mathematics	Level III**						•	
0.122A	Algebra	IIIH	1	Full yr.	2 1	10.121A			10.112A
0.122B	Integration and Functional								
	Analysis	IIIH	1	Full yr.	2 1	10.121B			10.112E
).122C	Topology and Differential		_		• •				40.4400
1007	Geometry	IIIH	1	Full yr.	2 1	10.121A, 10	.1218	10.1128	, 10.11 2 C
).122E	Complex Analysis and	ШН	1	Full yr.	2 1	10.121B			10 1125
	Differential Equations n this column is counted the corre			•	-				10.112E
Mathema Students Students Mathematic	on to Higher Pure Mathematics II atics I may, subject to the approvi- majoring in Physics who wish to aiming at Honours in Pure Math- is 10.031 is included for students of the study of the subject of the students of the subject of the subject of the students of the subject of the subject of the students of the subject of the subject of the students of the subject of the subject of the students of the subject	al of the Head take Higher Pu ematics must ta testring to attem	of the Sch re Mathem ake 10.121A mpt only of	atics II should , B and C and ne level II Ma	natics, be pe l attempt 10 1 either 10.2 thematics un	0.121A. 10.121B a 221A or 10.211A. ht If other level	nd either 10.221A II units in Pure	Mathemat or 10.211 Mathemati	A. cs, Applie
Mathema 2. Students 3. Students 3. Students Mathematic Mathematic Mathematic In special of Students in special * Students	atics I may, subject to the approv- majoring in Physics who wish to aiming at Honours in Pure Mathe is 10.031 is included for students of	al of the Head take Higher Pr ematics must ta desiring to atten vill not be cour esiring to atten vill not be cour pleted as a leve ts should consu- n of the Head o attempt a le	of the Sch rre Mathem ake 10.121A mpt only on ited. apt only on ited el III unit i ult with the of the Sch vel III Pur	atics II should , B and C and ne level II Mai e level III Mat for students pr School of Ma nool. e Mathematics	natics, be pe l attempt 10 d either 10.2 thematics un chematics un coceeding to athematics p s unit unless	121A. 10 121B a 121A or 10.211A. hit If other level hit. If other level honours in math rior to enrolment s they have comp	nd either 10.221A II units in Pure III units in Pure matics. Pre- and co-rec	Mathemat or 10.211 Mathemati Mathemati quisites ma	A. cs, Applied cs, Applied y be varied
Mathema 2. Students 3. Students Mathematic Mathematic Mathematic Mathematic * Students * Students 10.111A,	atics I may, subject to the approv- majoring in Physics who wish to aiming at Honours in Pure Mathe ss 10.031 is included for students of sare taken, 10.031 Mathematics w is 10.032 is included for students d ss are taken, 10.032 Mathematics w ircumstances 10.121C may be com wishing to attempt Level IIIH unit circumstances with the permission will not normally be permitted t 10.111B and 10.211A and are com	al of the Head take Higher Pr ematics must ta desiring to atten vill not be cour esiring to atten vill not be cour pleted as a lev- ts should consu- n of the Head o attempt a le- neurrently attem	of the Sci re Mathem ke 10.121A mpt only on ted pt only on ted el III unit i lit with the of the Sci wel III Pun mpting the r	atics II should , B and C and ne level II Mat e level III Mat for students pr School of Ma nool. e Mathematics emaining units	natics, be pe 1 attempt 10 1 either 10.2 thematics un cocceding to athematics p 2 unit unless 3 of these th	121A. 10 121B a 221A or 10.211A. ait If other level hit. If other level honours in math rior to enrolment s they have comp aree units.	nd either 10.221A II units in Pure III units in Pure lematics. Pre- and co-rec pleted at least or	Mathemata or 10.211 Mathemati Mathemati quisites ma ne level II	A. ics, Applied ics, Applied y be varied unit from
Mathema 2. Students 3. Students 3. Athematic Mathematic Mathematic Mathematic In special of in special * Students * Students	atics I may, subject to the approv- majoring in Physics who wish to aiming at Honours in Pure Mathe s: 10.031 is included for students of s are taken, 10.031 Mathematics w s: 10.032 is included for students d s are taken, 10.032 Mathematics w ircumstances 10.121C may be com wishing to attempt Level IIIH unit circumstances with the permission will not normally be permitted t	al of the Head take Higher Pre- ematics must ta desiring to attein vill not be cour pleted as a lev ts should consu- n of the Head o attempt a le currently attem	of the Sci re Mathem ke 10.121A mpt only on ted pt only on ted el III unit i lit with the of the Sci wel III Pun mpting the r	atics II should , B and C and ne level II Mat e level III Mat for students pr School of Ma nool. e Mathematics emaining units	natics, be pe 1 attempt 10 1 either 10.2 thematics un cocceding to athematics p 2 unit unless 3 of these th	121A. 10 121B a 121A or 10.211A. hit If other level hit. If other level honours in math rior to enrolment s they have comp	nd either 10.221A II units in Pure III units in Pure matics. Pre- and co-rec	Mathemata or 10.211 Mathemati Mathemati quisites ma ne level II	A. ccs, Applied ccs, Applied y be varied
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		SCHO	OL OF	MATHEN	MATICS	(Continued)			_ B1
No.	Name	Level	Unit Value	When Offered	Hours p.w.	Prerequisites	Co-requisites	Excluded**	4
	Higher Applied Mathemat	ics Level	Ш	<u> </u>					
10.221A	Mathematical Methods	IIH	1	Full yr.	2 1	10.011†		10.211A	
10.221B	Analytical Dynamics	IIH	1	Session 1	4	10.011†, 1.011†	10.221A or	10.211B	<u>ت</u>
10.221C	Hydrodynamics	ШΗ	1	Session 2	4	10.011†, 1.011†	10.211A 10.221A or	10.211C	THE
10.2210	Hydrodynamics	1111	1	50331011 2	-	10.0111, 1.0111	10.211A		
	Applied Mathematics Leve	el III							UNIVERSITY
10.212A	Numerical Analysis	III	1	Full yr.	1 1	10.111A, 10.211A		10.222A	7
10.212B	Continuum Mechanics	III	1	Full yr.	1 1	10.111A, 10.111B		10.222B	Ē
						& 10.211A, B, C			, RS
10.21 2D	Mathematical Methods	III	1	Full yr.	1 1	10.211A, 10.111A,		10.032, 10.222E	ち ロ
		***		F 11	4.1	10.111B	0.0114	10.222E	
10.212L	Optimization Techniques	III	1	Full yr.	1 1	10.111A, 10.111B, 1	0.211A		<u>Q</u>
10.412	Dynamical and	III	1	Full yr.	2	1.001 or 1.011 &		ş	
	Physical Oceanography	111	1	Fun yr.	2	10.211A or 10.221A			Ξ
						or 10.031			Ň
	Higher Applied Mathemat	ics Level	Ш						NEW SOUTH WALES
10.222A	Numerical Analysis	IIIH	1	Full yr.	11	10.111A (or better)		10.212A	2
10.222B	Continuum Mechanics	IIIH	1	Full yr.	2	10.111A, B,		10.212B	Ē
10.2220	000000000000000000000000000000000000000		-			10.221A, B, C			Ξ
10.222C	Maxwell's Equations and	IIIH	1	Full yr.	2	10.221A, 10.121B‡,		1.113B	×
	Special Relativity					1.001		1.123B, 1.153B	2
10.222D	Mathematical Methods	IIIH	1	Full yr.	2	10.221A, 10.121A		10.212D	ų,
				•		10.121B‡			•
10.222F	Quantum Mechanics	IIIH	1	Full yr.	2	10.221A, 10.121A,	10.222D	1.113A, 1.123A	
	-					10.121B‡			

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* The evening course for 10.211B runs at 2 hours per week throughout the year.
* A student who gains a superior pass in 10.001 Mathematics I and/or 1.001 Physics I may apply to proceed to Higher Applied Mathematics units.
* 10.111A, B and 10.211A with a sufficiently good pass may be substituted as a prerequisite in place of 10.121A, B and 10.221A.
§ It is recommended that one of the following be taken concurrently: 10.211C or 10.21C or 1.143F.
** If a unit in this column is counted the corresponding unit in the first column may not be counted.

SCHOOL OF MATHEMATICS (Continued)

No.	Name	Level	Unit Value	When Offered	Hours p.w.	Prerequisites (all units named except as governed by or)	Co-requisites	Excluded*
			STA	TISTICS				
10.311	Theory of Statistics Level II Probability and Random Variables Sampling Distributions and Estimation Tests of Hypotheses and Regression	II	3	Full yr.	7	10.001 or 10.011 or 10.021 Cr		10.321, 10.331
10.321	Higher Theory of Statistics Level II Probability and Random Variables Sampling Distributions and Estimation Tests of Hypotheses and Regression	IIH	3	Full yr.	8	10.001 or 10.011		10.311, 10.331
10.331	Statistics SS	II	1	Full yr.	2	10.001 or 10.011 or 10.021 Cr		10.311, 10.321
10.312A	Theory of Statistics Level III Stochastic Processes and Applications	111	1	Session 2	4	10.311 or 10.321 or 10.331; 10.111A or 10.121A; 10.111B or 10.121B; 10.211A or 10.221A		10.322A
10.312B	Experimental Design (Applications) and Sampling	III	1	Session 1	4	10.311 or 10.321 or 10.331 (normally Cr.)	10.211A or 10.221A	10.322B
10.312C	Experimental Design (Theory) and Project	III	1	Session 1	4	10.311 or 10.321; 10.111A or 10.121A; 10.111B or 10.121B; 10.211A or 10.221A	10.312B or 10.322B*	10.322C
10.312D	Probability Theory and Contingency Tables	III	1	Session 2	4	10.311 or 10.321; 10.111A or 10.121A; 10.111B or 10.121B; 10.211A or 10.221A		10.322D

No.	Name	Level	Unit Value	When Offered	Hours p.w.	Prerequisites (all units named except as governed by or)	Co-requisites	Excluded*	0
	Histor Theory of Statistics I and III	STAT	FISTI	CS (Continu	ued)				Ļ
10.322A	Higher Theory of Statistics Level III Stochastic Processes and Applications	IIIH	1	Session 2	4 1	J		10.312A	тне с
10.322B	Experimental Design (Applications) and Sampling	IIIH	1	Session 1	4 1	10.321; 10.111A or 10.121A;		10.312B	UNIVE
10.322C	Experimental Design (Theory) and Project	IIIH	1	Session 1	4 1	10.111B or 10.121B; 10.211A or 10.221A	10.322B†	10.312C	K STI Y
10.322D	Probability Theory and Contingency Tables	IIIH	1	Session 2	4 1			10.312D	OF NEW
* If a unit	in this column is counted the corresponding un	it in th	e first	column may	not be co	ounted.			×

SCHOOL OF MATHEMATICS (Continued)

SCHOOL OF APPLIED PHYSICS AND OPTOMETRY

† Plus any	two level III Pure Mathematics of			ics units.	ND OPT	OMETRY		
No.	Name	Level	Unit Value	When Offered	Hours p.w.	Prerequisites	Co-requisites	Excluded Excluded
31.113A	Physics of Materials	III	1	Session 1 and Full yr.	6 3	1.112B or 1.122B and 2.001 or 2.011		
31.113B	Physics of Measurement	III	1	Full yr.	3	1.112B or 1.122B		
31.113C	Applications of Radiation	III	1	Session 2	6	1.112B or 1.122B		

FACULTY OF BIOLOGICAL SCIENCES SCHOOL OF PSYCHOLOGY*

No.	Name	Level	Unit Value	When Offered	Hours p.w.	Prerequisites	Excluded
12.001	Psychology I	I	2	Full yr.	5	Sc. Faculty Entrance	
12.152	Research Methods II	II	1	Full yr.	3	12.001	
12.252	Learning II	п	ł	Not offered in 1974	3	12.001	
12.302	Personality II	II	ł	Half yr.	3	12.001	
12.322	Motivation II	II	ł	Half yr.	3	12.001	
12.372	Psychological Assessment II	II	1 /2	Half yr.	3	12.001	
12.412	Physiological Psychology II	II	$\frac{1}{2}$	Half yr.	3	12.001	12.402
12.452	Human Information Processing II	II	ł	Half yr.	3	12.001	
12.472	Perception II	II	ł	Half yr.	3	12.001	
12.502	Social Psychology II	II	1	Half yr.	3	12.001	
12.552	Developmental Psychology II	II	$\frac{1}{2}$	Half yr.	3	12.001	
12.602	Abnormal Psychology II	II	1	Half yr.	3	12.001	
12.153	Research Methods IIIA	III	1	Half yr.	6	12.152	
12.163	Research Methods IIIB	III	1	Half yr.	6	12.153	

SCHOOL OF PSYCHOLOGY* (Continued)

No.	Name	Level	Unit Value	When Offered	Hours p.w.	Prerequisites	Excluded
12.173	Psychological Issues	ш	1	Half yr.	6	12.001	
12.253	Learning IIIA	ш	1	Half yr.	6	12.001	
12.263	Learning IIIB	ш	1	Half yr.	6	12.253	
12.303	Personality IIIA	III	1	Half yr.	· 6	12.001	
12.313	Personality IIIB	ш	1	Half yr.	6	12.001	
12.323	Motivation IIIA	III	1	Half yr.	6	12.001	
12.373	Psychological Assessment (Testing) IIIA	ш	1	Half yr.	6	12.372	12.042
12.383	Psychological Assessment (Psychometric Theory) IIIB	ш	1	Not offered in 1974	6	12.372	
12.413	Physiological Psychology IIIA	ш	1	Hal f yr.	6	12.001	12.402
12.423	Physiological Psychology IIIB	ш	1	Hal f yr.	6	12.412	
12.453	Human Information Processing IIIA	ш	1	Half yr.	6	12.001	
12.463	Human Information Processing IIIB	ш	1	Not offered in 1974	6	12.452	
12.473	Perception IIIA	ш	1	Half yr	6	12.001	

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No.	Name	Level	Unit Value	When Offered	Hours p.w.	Prerequisites	Excluded
12.483	Perception IIIB	ш	1	Half yr.	6	12.472 or 12.473	
12.503	Social Psychology IIIA	III	1	Half yr.	6	12.001	
12.513	Social Psychology IIIB	III	1	Half yr.	6	12.502	
12.553	Developmental Psychology IIIA	ш	1	Half yr.	6	12.001	
12.563	Developmental Psychology IIIB	III	1	Not offered in 1974	6	12.552	
12.603	Abnormal Psychology IIIA	III	1	Half yr.	6	12.001	
2.613	Abnormal Psychology IIIB	III	1	Not offered in 1974	6	12.602	
12.623	Guidance & Counselling III	III	1	Half yr.	6	12.372 or 12.373	
12.653	Industrial Psychology III	III	1	Half yr.	6	12.001	
12.703	Psychological Techniques III	III	1	Half yr.	6	12.001	12.042
12.713	Behavioural Control and Modification	III	1	Not offered in 1974	6	12.001	

SCHOOL OF PSYCHOLOGY* (Continued)

* (a) In any year some level II and level III units will be mutually exclusive. Details of these exclusions and of advisable sequences will be announced II before the end of the preceding academic year so that students are fully informed prior to their lodging of pre-enrolment forms and their finalising of enrolment. All students proposing to undertake level II and level III units should consult with the School's Science Course advisers before completing enrolment.

(b) Not all courses will be offered each year.

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No.	Name L		nit alue	When Offered	Hours p.w.	Prerequ	isites	Co-requisites	Exclude
17.011	Human Biology	1 1	1	Session 1	6	Sc. Faculty	Entrance	7.021, 2.001, 10.001 or 10.011 or if level II or level II Biology Units in the of Biological Science be taken subsequent	I Faculty es are to
17.021	Comparative Functional Biology	II	L	Session 2	6	Sc. Faculty	Entrance	17.011, 2.001, 10.001, or 10.011 or if level II or level II Biology Units in the Biological Sciences a taken subsequently.	I Faculty of
17.012	General Ecology		1	Session 1	6	17.001 or 1 and 17.021			
This unit n	nay be taken in either second or thi	rd year o	of the	Science Cou	rse provided	that prerequisi	tes have bee	n completed.	
		6	СПС	OT OF B	IOCHEM	ICTDV+			
No.	Name		CHC Un 1 Val		IOCHEM When Offered	ISTRY‡ Hours p.w.		uisites**	Co-requisit
	Name Chemistry of Biologically Important Molecules		Un 1 Val	ue	When	Hours	Prerect For any 1 17.001†	evel II unit:	
41.101A	Chemistry of Biologically	Leve	Un 1 Val	1 S	When Offered	Hours p.w.	Prerect For any 1 17.001†	evel II unit: and 17.021	Co-requisit 41.101B 41.101A
41.101A 41.101B	Chemistry of Biologically Important Molecules	Leve II	Un 1 Val	ut ue 1 \$ 1 \$	When Offered Session 1	Hours p.w. 6	$\begin{cases} For any 1 \\ 17.001^{\dagger} \\ or 17.011 \\ and 2.001 \end{cases}$	evel II unit: and 17.021	41.101E
No. 41.101A 41.101B 41.101C 41.102A	Chemistry of Biologically Important Molecules Metabolism	Leve II II II	Un 1 Val	ue1 \$	When Offered Session 1 Session 1	Hours p.w. 6 6	$\begin{cases} For any 1 \\ 17.001^{\dagger} \\ or 17.011 \\ and 2.001 \\ both 41.1 \\ \\ For any 1 \\ 41.101A, \\ 41.101C \\ \end{cases}$	evel II unit: and 17.021 †	41.101B

[‡] Third level units available only during the daytime.
 [•] 41.101A may be taken as a single unit under special circumstances and at the discretion of the Head of School.
 [•] In exceptional circumstances a student may apply to the Head of School for variation of the prerequisite.
 [†] Terminating pass not acceptable.

SCHOOL OF BIOLOGICAL TECHNOLOGY

No.	Name L	evel	Unit Value	When Offered	Hours p.w.	Prerequisites*
2.102	Fermentation Technology * In exceptional circumstances	III astu	1 dent may a	Session 2 pply to the Head of Schoo	6 I for variation of t	44.102 he prerequisite.
			SCHO	OL OF BOTANY†		
No.	Name	Lev	Unit rel Value		Hours p.w.	Prerequisites
3.101A	Genetics and Biometry	I	1	Session 1	6	j 17.001
3.101B	Plant Evolution and Ecology	I	1	Session 1	6	{ <i>or</i> 17.011 and 17.021
3.101C	Plant Physiology	11	[1	Session 2	6	17.001 or 17.011 and 17.02 2.001 or 1.001** or 1.031** or 1.041**
3.102A	Advanced Genetics	Ľ	(I * 1	Session 2	6	43.101A
3.102B	Plant Taxonomy	IJ	I* 1	Session 2	6	43.101B; 43.101A pre- or co-requisite
3.102C	Plant Physiology & Biochemistry	I	II* 1	Session 1	6	41.101A; 41.101B; 43.101C
3.102D	Mycology	IJ	I* 1	Session 2	6	17.001 or 17.011 & 17.021
3.102E	Environmental Botany Plant Pathology lents taking four or more units in the		1* 1	Session 2	6	17.001 or 17.011 & 17.021 1.001** or 1.031** or 1.041**
3.102F	Plant Pathology	IJ	I* 1	Session 1	6	17.001 or 17.011 & 17.021
BIOC	ents taking four or more units in the themistry, or Chemistry, or Physics, or M el courses conducted by the School of Bo	lainen	natics.			

* These units may be taken in either second or third year of the Science course provided that prerequisites have been completed.

** This unit may be taken as a co-requisite in some circumstances.

SCHOOL OF MICROBIOLOGY⁺

No.	Name	Level	Unit Value	When Offered	Hours p.w.	Prerequisites*	
44.101	Introductory Microbiology	Π	1	Session 2	6	17.00 or 17.011 & 17.021	
44.111	Microbiology**	II	1	Full yr.	3		
44.102	General Microbiology	III	2	Session 1	12	44.101, 43.101A, 41.101A and 41.101B	
44.112	Applied Microbiology	ш	2	Session 2	12	44.102	
44.122	Immunology	III	1	Session 2	6	17.001 <i>or</i> 17.011 and 17.021; 41.101A, 41.101B	
44.132	Virology	ш	1	Session 2	6	44.102	5
† All level	units available only during the daytin	ne.					11

* In exceptional circumstances a student may apply to the Head of School for variation of the prerequisite.

0 ** For students not intending to major in Microbiology and not taking level II Biochemistry. This unit is not acceptable as a prerequisite for level III T Microbiology, except on the recommendation of the Head of School.

SCHOOL OF ZOOLOGY[†]

Microbio	ogy, except on the recommendation	n of the Head	of School			-	
			SCHOOI	L OF ZOOLOGY†			
No.	Name	Level	Unit Value	When Offered	Hours p.w.	Prerequisites	Co-requisites
45.101 A	Genetics and Biometry (see Botany)	II	1	Session 1	6]	17.001 or 17.01 2.001 or 2.011	11 & 17.021;
45.101B	Invertebrate Zoology	п	1	Session 2	6	10.001 <i>or</i> 10.011 10.021	or
45.101C	Vertebrate Zoology	II	1	Session 2	6)	10.021	
45.101D*'	Field Ecology	п	1	Session 2	6*	43.101A	45.101B or 45.101C

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SCHOOL OF ZOOLOGY[†] (Continued)

No.	Name	Level	Unit Value	When Offered	Hours p.w.		Prerequisites Co	o-requisites	Excluded
45.102A	Marine Ecology	III	1	Session 1	6		17.001 or 17.011 and 17.021 plus three other first year science subjects.		
45.102B	Animal Behaviour	III	1	Session 2	6		43.101A/45.101A		
45.102C	Comparative and Environmental Physiology	III	1	Session 2	6		41.101A & B; 1.001 or 1.011 or 1.041	45.1010	2
45.102D‡	Comparative Reproductive Physiology	III	1	Session 2	6		41.101A & B and 45.101C		70.012E
45.102E 45.102F**	Invertebrate Physiology Invertebrate Behaviour	III III	1 1	Session 1 Session 2	6 6	}	As for 45.101B		
45.201A	Insect Structure and Classification	111	1	Session 1	6		45.101A & 45.101B		
45.201B	Insect Physiology	III	1	Session 1	6		45.201A		
45.201C	Applied Entomology	III	1	Session 2	6		45.201B		
45.201D	Project	ш	1	Session 2	6		45.201B		

NOTE: Students taking four or more units in the School of Zoology must take Genetics and Biometry 43/45.101A and at least two level II units of Biochemistry, or Chemistry, or Physics or Mathematics, or Geology. This unit includes a two-week camp in November/December.

† Third level courses conducted by the School of Zoology are available only during the daytime to part-time students enrolling for the first time in 1973 or later.

\$ May not be counted towards a degree which includes 70.012E Comparative Embryology.

** Not available in 1974.

FACULTY OF APPLIED SCIENCE SCHOOL OF APPLIED GEOLOGY

No.	Name	Level	Unit Value	When Offered	Hours p.w.	Prerequisites Co-requisites
25.111*	Geoscience I	Ι	2	Full yr.	6	Sc. Faculty Ent. 2.001
25.112A**	Geoscience IIA	II	2	Full yr.	6	25.111 2.001
25.112B**	Geoscience IIB	II	1	Full yr.	3	25.111 2.001
25.113A***	Geoscience IIIA	II	[2	Full yr.	6	25.112A and 25.112B
25.113B***	Geoscience IIIB	II	12	Full yr.	6	25.112A and 25.112B
25.603A†	Geological Oceanography	III	[1	Full yr.	3	25.111 and 25.112B

* Three field tutorials, up to five days in all, are an essential part of the course. Attendance is compulsory.

** Field work: approximately seven days will be spent on field tutorials throughout the year. Attendance is compulsory,

*** Field work is an essential part of the course and consists of approximately ten days of field tutorials. Attendance is compulsory.

† Compulsory field work to be arranged.

SCHOOL OF GEOGRAPHY

No.	Name	Level V	Unit Value	When Offered	Hours p.w.	P rerequisites	Co-requisites
27.031	Geography IS	I	2	Full yr.	6	Sc. Faculty Ent.	
27.103*	Climatology	II	1	Session 2	5 1	1.001 and 27.031	
27.203*	Biogeography	п	1	Session 1	5 1	27.031	
27.413*	Geomorphology	п	1	Session 1	5 1	27.031 or 25.111 or 25.001	

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SCHOOL OF GEOGRAPHY (Continued)

No.	Name		Level '	Unit Value	When Offered	Hours p.w.	Prerequisites	Co-requisites
27.423*	Pedology		Ш	1	Session 2	5 1	2.001 and 27.031 or 25.111 or 25.001	
* Field wor	k (to be arranged by the School of	f Geograp	hy) is a co	mpulso	ry component of e	each unit.	<u></u>	
			FA	CUL	FY OF ART	ſS		
			SCHO	OL C	OF PHILOSOP	HY		
No.	Name	Level	Unit Value	w	hen Offered	Hours p.w.	Prerequisites	Co-requisites
52.111	Philosophy I	I	2		Full yr.	4	Sc. Faculty Ent.	
52.112	Philosophy II	II	3		Fuli yr.	4	52.111	
52.122	Philosophy II (Honours)	IIH	3		Full yr.	5	52.111	
	SCHOOL		ISTOPV		PHILOSOPH	V OF SC	TENCE	
			Unit			Hours		
No.	Name	Level	Value	w	hen Offered	p.w.	Prerequisites	Co-requisites
62.012	The Origins of Modern Science	п	1		Session 1	6	A pass in <i>two</i> of 1.001, 17.011 and 10.001, 25.111, 1 10.011, 10.021, 27.031, 12.001	d 17.021, 2,001,

SCHOOL OF HISTORY AND PHILOSOPHY	OF SCIENCE	(Continued)
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No.	Name	Level	Unit Value	When Offered	Hours p.w.	Prerequisites	Co-requisites	
62.022	The Social History of the Scientific Movement	II	1	Session 1		<i>10.001</i> , 25.111,	nd 17.021, 2,001,	
62.032	Philosophy of Science	II	1	Session 2	6	10.011, 10.021, 27.031, 12.001		1

FACULTY OF ENGINEERING

SCHOOL OF MECHANICAL AND INDUSTRIAL ENGINEERING

No	Name	Level	Unit Value	When Offered	Hours p.w.	Prerequisites	Co-requisites
5.010	Engineering A	I	1	Session 1	6	Sc. Faculty Entrance	5.020 or 5.030
5.020	Engineering B	I	1	Session 2	6	•	5.010
5.030	Engineering C	I	1	Session 2	6		5.010
		SCHO	DOL OF	ELECTRICAL EN	GINEERII	NG	
5.601A	Introduction to Computing	II	1	Session 1	5	10.001	
5.601 A *	Introduction to Computing	II	1	Full yr.	2 1	10.001	
5.601 B	Assembler Programming & Non-numeric Computing	II	1	Session 2	5	10.001	6.601A
5.602A	Computer Systems I	III	1	Session 1	5	6.601B	
602B	Computer Systems II	ш	1	Session 2	5		6.602A
5.602C	Computer Applications	ш	1	Session 1	5	6.601A	
5.602D Offered of	Programming Languages nly in the evening.	III	1 .	Session 2	5	6.601A	

FACULTY OF MEDICINE

SCHOOL OF ANATOMY

No.	Name	Level	Unit Vaiue	When Offered	Hours p.w.	Prerequisites	Co-requisites
70.011A	Mammalian Histology	II	1	Session 1	6	17.011 and 17.021	
70.011B	Mammalian Embryology	II	1	Session 2	6	17.011 and 17.021	
70.011C	Systematic Anatomy I	II	1	Full yr.	3	17.001, 70.011A*	
70.012A	Systematic Anatomy II	III	1	Full yr.	3	70.011A*, 70.011C*	
70.012B	Systematic Anatomy III	III	1	Full yr.	3	70.011A*, 70.011C*	
70.012C	Systematic Anatomy IV	III	1	Full yr.	3	70.011A*, 70.011C*	
70.012D	Comparative Histology	III	1	Session 1	6	70.011A	
* In some o	circumstances this subject may be	taken as	s a co-requ	isite rather than a prere	quisite.		
			SCHOO	L OF HUMAN GE			(2.102.4
78.201	Population Genetics Theory	III	SCHOO 1	L OF HUMAN GE Session 2	NETICS 5	43.101A/45.101A, 10.001 or 10.011	43.102A
78.201 73.011A	Theory		1		5	10.001 or 10.011	43.102A

NOTE: The above represent the normal prerequisites for the courses in Physiology, but the Head of School may recommend that students with a good N academic record be granted exemption from them.

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(b) General Studies

Almost all undergraduates in Faculties other than Arts and Law are required to complete a General Studies programme. Courses (in addition to the Faculties of Arts and Law) which do not have this requirement are Bachelor of Science in Psychology, Bachelor of Science in Economic Geography, Bachelor of Science (Education) and Bachelor of Health Administration. The Department of General Studies publishes its own Handbook which is available free of charge. All details regarding General Studies courses and requirements are contained in it, and students are advised to obtain a copy. All enquiries about General Studies should be made to the General Studies Office, Room G15, Morven Brown Building (663-0351 Extn. 2091).

Students shall select *three* general studies subjects (see General Studies handbook *and* School of History and Philosophy of Science in Subject Information and Textbook Lists section in this handbook); in addition, honours students shall be required to complete an Advanced General Studies Elective.

4. Pattern of Studies

In general, a student should select a course which is adequately distributed over the six half years of study. Typical course patterns are available from the Faculty Office.

A suggested pattern of study is:---

First year: The appropriate two units of level I Mathematics and six other level I units including those essential to the intended major sequence of units.

Second year: One general studies elective and eight units from level II or six units from level II and two from level I.

Third year: Two general studies electives and at least four level III units. The other units could be level II or III.

Fourth year: For an honours degree, an advanced general studies elective and such requirements as specified by the Head of the appropriate School.

5. Part-time Study

A student must select the units and general studies electives in accordance with these regulations save that Clause 2a(vi) is modified so that he must complete level I Mathematics and

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6 other level I units in the first four years of enrolment or else show cause to the satisfaction of the Professorial Board why he should be allowed to re-enrol.

RULES GOVERNING ADMISSION TO THE SCIENCE DEGREE COURSE WITH ADVANCED STANDING FOR THE PURPOSE OF OBTAINING A DOUBLE DEGREE

1. Undergraduates* of the University of New South Wales who have satisfied the examiners in at least the first two years of a degree course extending over four or more years and approved by the Faculty of Science for the purpose of double degrees, may be admitted to the Science degree course with advanced standing. Such undergraduates' performance shall have been of a high standard and their admission shall be subject to the approval of the Dean of the Faculty of Science.

2. Students so admitted who have satisfied the examiners in General Studies subjects and/or Science course units shall be given advanced standing in such General Studies subjects and no more than 14 such Science course Units.

3. Students so admitted may be granted exemption from two other level II Science Units on the basis of other subjects completed by them.

4. In order to qualify for the award of the degree of B.Sc., students so admitted with advanced standing shall be required to complete the appropriate General Studies subjects and no less than four units of either level II or level III and four other level III units in accordance with the Science course regulations.

The units submitted for the Bachelor's degree under these regulations must include at least four level III units chosen from related disciplines in accordance with the Science course regulations. One of Mathematics 10.021 or 10.001 or 10.011 must be included in the course.

^{*}In Rule 1, the word "undergraduates" includes graduands, i.e., a person may be admitted under these rules if he has met all requirements for a first degree which has not yet been conferred on him, and his admission under these rules shall be no bar to the subsequent award for the first degree.

RULES GOVERNING ADMISSION TO THE SCIENCE DEGREE COURSE WITH ADVANCED STANDING

1. Graduates of the University of New South Wales may be admitted to the Science degree course with exemption in all General Studies subjects completed by them and in no more than twelve Science course units completed by them.

2. Undergraduates of the University of New South Wales who transfer from another course to the Science degree course, may be admitted to the Science degree course with exemption in all General Studies subjects completed by them and in all Science course units completed by them. Further, where an undergraduate has completed a subject which contains the syllabus material of a Science course unit (or units) the Dean, with the agreement of the Head of the School offering the Science course unit (or units) may allow the unit (or units) so covered to be counted to a Bachelor of Science degree.

An undergraduate transferring to the Science course must take Mathematics 10.021 or 10.001 or 10.011 during his first year of enrolment in the course unless one of them has previously been completed.

3. Graduates or undergraduates of other universities or of other approved tertiary institutions may be admitted to the Science degree course with advanced standing.

4. Students admitted under Rule 3 who have satisfied the examiners in units of the same title or subject matter as Science course subjects in this University may, subject to the approval of the appropriate Heads of School, be granted exemption in no more than eleven Science course units but not including level III Science course units.

5. Notwithstanding the provisions of Rules 1, 2, 3 and 4, Faculty may determine a special programme to be completed by a student who wishes to be granted advanced standing for an honours degree of Bachelor of Science in this University.

PURE AND APPLIED CHEMISTRY COURSE

This course also leads to the Bachelor of Science degree, but provides a study in depth of one field only. It may be taken either as a full-time or part-time course.

Full-time Course

This course may be taken at pass or honours standard. The pass course requires full-time attendance at the University for three years. An additional year is required for the honours course.

First year is similar to Year I of Chemistry in the Science course. In the second year a core of chemistry subjects, similar in content to chemistry units in the science course, but treated in greater depth and with extended practical work, is supplemented by science units offered by the Faculties of Science, Applied Science and Biological Sciences. It will be possible to choose between a wide range of such units, which may include a further first year subject if desired. The most widely chosen electives are some combination of mathematics units or a group of biological science units, such as the level II Biochemistry units in the Science course. It is possible that some elective units at this level may be offered by the School of Chemistry in later years.

The first half of third year will consist of a further development of the core course in four chemistry subjects. At this stage the student will have studied chemistry to an extent comparable to the student who graduates from the Science course with a major in chemistry, but the Pure and Applied Chemistry student will have studied at rather greater depth. In the second half of the third year, students in the Pure and Applied Chemistry course will select three advanced elective subjects. While most of the electives available are provided by the School of Chemistry, the needs of students who see their future in the less technical areas of industry will be met by the provision of electives with an applied bias.

Third year electives are normally of 112 hours, equivalent to eight hours per week for the half year. They are arranged in four groups, corresponding to areas of scientific interest. Not more than two electives may be chosen from one group; at least one must be chosen from electives offered by the School of Chemistry and any pre- or co-requisites must be observed.

1.	Physical and theoretical chemistry chemical physics, mathematics an statistics.	
2.	Organic chemistry, biochemistry	2.633
3.	Inorganic, analytical, nuclear ar radiation chemistry	nd 2.433, 2.533 2.811
4.	Applied chemistry, interdisciplinar	ry 2.513, 2.711 2.911
es off	ered by School of Chemistry	Prerequisites
Phy	ysical Chemistry	2.322* or 2.303*
Th	eoretical Chemistry	2.302* or 2.322*
Inc	organic Chemistry	2.422*
An	alytical Chemistry	2.522*
An	alytical Biochemistry	2.522*
Or	ganic Chemistry	2.622*
	2. 3. 4. Phy The Inc An An	 statistics. Organic chemistry, biochemistry Inorganic, analytical, nuclear an radiation chemistry Applied chemistry, interdisciplination es offered by School of Chemistry Physical Chemistry Theoretical Chemistry Inorganic Chemistry Analytical Chemistry

2.311, 2.411

2.311, 2.411, 2.511 and 2.611

2.411

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2.711 Solid State Chemistry

2.811 Nuclear and Radiation Chemistry

2.911 Applied Chemistry

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* May be taken as co-requisites if necessary.

	TOL			for 1	occione
		Hours	per weel	Lab.	005510115
YEAR	1 .		Lec.	Tut.	Total
1.011	Higher Physics I or Physics I	}	3	3	
2.001	Chemistry I	•••••	2	4	
10.011 10.001 10.021	Higher Mathematics Mathematics I Mathematics IT	I or or }	4	2	
Plus on	ne of				
5,010 5,020 5,030	Engineering A and Engineering B or Engineering C	}	3	3	
17 011	or Human Biology and	1			
17.021	Comparative Functional Biology	}	2	4	
	or		•		
25.111	or		2	4	
27.031	Geography IS		2	4	
YEAR	2				
2.311	Physical Chemistry		$1\frac{1}{2}$	3	4 1
2.411	Inorganic Chemistry		1	3 2 3 3	3
2.511	Analytical Chemistry		1	3	4
2.611	Organic Chemistry		1 1	3	4 1 9
	Science Electives* General Studies Elect	ive			11
	General Studies Dieet				
					26 1

391. PURE AND APPLIED CHEMISTRY FULL-TIME COURSE

* Recommended elective subjects in Second Year. Pre- and co-requisites for these subjects are shown under the Science course and must be observed.

MATHEMATICS 10.031 Mathematics 10.031 Statistics SS 10.111A 10.111B Mathematics II 10.211A 10.211A<	2 2 6
PHYSICS 1.212 Physics IIT	3
BIOLOGICAL SCIENCES 17.011 Human Biology 17.021 Comparative Functional Biology 41.101A Chemistry of Biologically Important Molecules 41.101B Metabolism 41.101B Biochemical Control 41.101A Introductory Microbiology 73.011A Principles of Physiology * One session only.	6 6* 6* 6 3 6

GEOI 25.111 25.112 25.112	Geoscience I A Geoscience IIA			
	Hours	per we	ek for 2 Lab.	sessions
YEAR	3	Lec.	Tut.	Total
2.322	Physical Chemistry*	1	2	3
2.422	Inorganic Chemistry	1	2	3
2.522	Analytical Chemistry	1	2	3 3
2.622	Organic Chemistry	1	2	
	Advanced Elective Subjects†			12
	Two General Studies Electives			3
				27
* 4100	matively 2.013A Theoretical Chemistry (1, $\frac{1}{2}$,	11)		21
			danaa	with the
	e to be selected from the following list in pings and other requirements detailed earlier:	i accoi	uance	with the
grou	pings and other requirements defance earner.	Le	c./Tut.	Lab.
2.333	Physical Chemistry		2	2
2.303	Theoretical Chemistry		2	2
2.433	Inorganic Chemistry		1	3
2.533	Analytical Chemistry		1	3
2.513	Analytical Biochemistry		1	3
2.633	Organic Chemistry		11	2 1
2.711	Solid State Chemistry		1	3
2.811	Nuclear and Radiation Chemistry		2	3 2
2.911	Applied Chemistry	•••••	4	2

YEAR 4 — HONOURS

Consult School for details.

Part-time Course

The part-time course in Pure and Applied Chemistry is equivalent to the full-time course and extends over six part-time years, leading to the degree of Bachelor of Science. Honours may be awarded on the completion of an additional year of full-time study or, in special circumstances, an additional two years of part-time study.

The part-time course has been designed for students employed in the chemical industry but employment in this industry is not obligatory for entrance to the course.

This course allows a student to choose electives from other faculties such as Commerce or Applied Science. Areas such as industrial chemistry, management and technical services can thus be covered by those students who feel that their vocational interests lie in one particular region.

CENTOCY

391. PURE AND APPLIED CHEMISTRY PART-TIME COURSE

STAGES 1 AND 2

Two of the following subjects will be taken in the first year and the other two in the second year (as directed).

		Hours per		for 2 sessions Lab.
1 011	Higher Physics I or	1	Lec.	
1.001	Physics I	}	3	-
			2	4
10.001 10.021	Mathematics I or Mathematics IT	·	4	2
Plus one	e of			
5.010 5.020 5.030		}	3	3
17 011	or Human Biology and	2		
17.021	Comparative Functional Biology	}	2	4
05 111	or Geoscience I		•	
25.111	or		2	4
27.031 * Three		o five days in all, are an e	2 ssenti	4 al part of the

course.

	Hours	per we	ek for 2	sessions
			Lab.	
STAG	E 3	Lec.	Tut.	Total
2.311	Physical Chemistry	11	3	41
2.411	Inorganic Chemistry		2	3
	Science Electives*			6
* See :	footnote under Second Year Full-time Course.			13 1
STAG	E 4			
2.511	Analytical Chemistry	1	3	4
2.611	Organic Chemistry		3	41
	Science Elective*			3
	General Studies Elective			1 1

* See footnote under Second Year Full-time Course. 13

,

Hours per week for 2 sessions

			Lab.	
		Lec.	Tut.	Total
STAG	Е 5			
	Physical Chemistry* Inorganic Chemistry	1 1	2 2	3 3
2.522	Analytical Chemistry Organic Chemistry	1 1	2 2	3 3 3
2.022	General Studies Elective			11
				13 1
* Alter	matively 2.013A Theoretical Chemistry (1, 1,	1 1).		

STAGE 6

Advanced Elective Subjects** General Studies Elective	12 1 1
	13 1
** Three to be selected. See list and regulations under Third Yes	ar Full-

** Three to be selected. See list and regulations under Third Year Fulltime course.

Honours

The requirements for admission to the honours course are the same as for the full-time honours course. A student wishing to do honours on a part-time basis may complete the honours year over two part-time years. Students are, however, advised to make every effort to do the honours year full time.

OPTOMETRY COURSE

The Department of Optometry provides a four year full-time course in Optometry leading to the degree of Bachelor of Optometry, which may be awarded at the pass or honours level. The first year of the course involves a study in the fundamental sciences of physics, chemistry, mathematics and general and human biology. Students who have completed the first year of a science course including physics, chemistry, mathematics and general and human biology or zoology at any Australian university are qualified for admission to the second year of the course. Second, third and fourth years are devoted to professional training in optometry including clinical optometry in the final year.

395. OPTOMETRY—FULL-TIME COURSE

Bachelor of Optometry

	Hours pe	r week	for 2 sessions Lab.
YEAR	1	Lec.	Tut.
1.001	Physics I	3	3
2.001	Chemistry I	2	4
10.001 10.011 10.021	Mathematics I or	4	2
17.011 17.021	Human Biology General Functional Biology	2	4
		11	13
YEAR	2		
31.811	Optometry I	4	4
31.821	Special Anatomy and Physiology	3	3 3
73.011A	General Studies Elective	5 1	3
	General Studies Elective		·
		11	10 1
YEAR	-		
12.001	Psychology I	3	2
	Optometry II	8	7
31.831	Diseases of the Eye Two General Studies Electives	2 2	1 1
		15	11

	Hours p	er week	for 2 sessions
YEAR	4	Lec.	Lab. Tut.
12.741	Psychology	2	0
31.813	Optometry III	6	0
31.841	Clinical Optometry*	1	14
74.001	Indication for Medical Referral†	1	0
	Advanced General Studies Elective	1	$\frac{1}{2}$
		11	14 1
* Lootu	res cappo often first 0 moster		,

* Lectures cease after first 9 weeks. † Lectures commence after first 9 weeks.

CONDITIONS FOR THE AWARD OF THE DOUBLE DEGREE OF BSc BOptom IN THE FACULTY OF SCIENCE

- 1. Undergraduates* of the University of New South Wales who have satisfied the examiners in at least the first two years of the Optometry degree course may be admitted to the Science degree course with advanced standing for the purpose of qualifying for the double degree of BSc BOptom. Such undergraduates' performance shall have been of a high standard and their admission shall be subject to the approval of the Dean of the Faculty of Science.
- 2. In order to qualify for the award of the degree of BSc, students so admitted shall be required to complete the appropriate general studies subjects and no less than four units of either level II or level III and four other level III units, in accordance with the Science Course regulations.

The units submitted for the Bachelor's degree under these regulations must include at least four level III units chosen from related disciplines in accordance with the Science course regulations.

3. In order to qualify for the award of the degree of BOptom, students so admitted shall complete the requirements of the Optometry degree course.

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^{*}In Rule 1, the word "undergraduates" includes graduands, i.e., a person may be admitted under these rules if he has met all requirements for a first degree which has not yet been conferred on him, and his admission under these rules shall be no bar to the subsequent award of the first degree.

BACHELOR OF SCIENCE IN PSYCHOLOGY COURSE

The four year course in Psychology, which leads to the degree of Bachelor of Science, is designed to meet the requirements of students who intend to become professional psychologists, as either practitioners or research workers. It provides extensive study of psychological theory and practice, supported by an appropriate selection of other subjects.

The course is available on a full-time basis only.* Entry into the course is subject to a quota which is determined from time to time.

In the fourth year, students undertake a programme of study which includes courses selected from the School's advanced electives. Electives are offered in the major areas of general psychology and in a number of applied fields, e.g. clinical, industrial, human factors, and educational. In addition, the student must complete a research thesis or project.

Details of qualifications for admission to the Psychology BSc Course, the course requirements for Pass and Honours at graduation and rules governing admission with Advanced Standing are given below. Hours of attendance for the main subjects available in the course are shown in the Schedule of Course Subjects, followed by some recommended course patterns.

RULES GOVERNING THE PSYCHOLOGY BSc COURSE

- I. Applicants for admission to the Course must be matriculated to this University; and also have either satisfied the entrance requirements for the Faculty of Science or, alternatively, have passed Mathematics I or General and Human Biology.
- II. (A) In order to qualify for admission to the degree of BSc in Psychology under these regulations a candidate must attend classes and satisfy the examiners in the following subjects:—

^{*} Any student who enrolled in the former BSc in Applied Psychology Course prior to 1973 on a part-time basis may continue enrolling on that basis, provided that the course of study is completed within the minimum time plus two years.

1. Each of:----

12.001 Psychology I 12.042 Psychology IIA

and

A total value of 11 units of Psychology (Level II and III)

(In special cases, the Head of the School of Psychology or his representative may approve of the substitution of any other appropriate course or equivalent units).

and

12.004 Psychology IV.

2. Five other subjects (or their equivalent in units) selected to meet the following requirements:

(a) that they shall include at least one of:

- (i) 10.011 Higher Mathematics I, or
 - 10.001 Mathematics I, or
 - 10.021 Mathematics IT

or

(ii) 17.011 Human Biology, and

17.021 Comparative Functional Biology. [They may include both (i) and (ii).]

- (b) that they shall include at least one of:
 - 53.111 Sociology I
 - 15.101 Economics I
 - 54.111 Political Science I

52.111 Philosophy I

or with the approval of the Head of the School of Psychology, one other Arts I subject or two General Studies electives.

(c) that they shall include at least one subject which together with the subject meeting the requirements of (a) or (b) immediately above constitutes a recognized sequence of two courses.

Recognized sequences are:

(i) 10.001 Mathematics I, followed by three Mathematics level II units (10.111A, 10.111B, 10.211A) or by 10.311 Theory of Statistics II.

- (ii) 17.001 General and Human Biology (or both of 17.011 Human Biology and 17.021 Comparative Functional Biology) followed by 12.402 Physiological Psychology, or by the equivalent of one subject chosen from the following units according to the regulations of the Faculty of Biological Sciences:
 - 41.101A Chemistry of Biologically Important Molecules
 - 41.101B Metabolism
 - (41.101A and 41.101B must be taken together, and count as two units)
 - 41.101C Biochemical Control
 - 45.101A Genetics and Biometry
 - 45.101C Vertebrate Zoology
 - 73.011A Principles of Physiology I (equivalent to 2 units)
- (iii) 53.111 Sociology I, followed by 53.112 Sociology II
 - 15.101 Economics I, followed by 15.102 Economics II
 - 54.111 Political Science I, followed by 54.112 Political Science II
 - 52.111 Philosophy I, followed by 52.112 Philosophy II.
- (B) The proposed course must be approved by the Head of the School of Psychology or his representative prior to or during enrolment. The courses must be chosen in such a way as will fit in with the timetable.
- (C) Progression in the Course shall be by subjects, and the subjects in the Course may be completed in any order consistent with the requirements concerning prerequisites and co-requisites for the subjects chosen.

III. Prerequisites and Co-requisites

Before enrolling in any course (or equivalent units of a subject) the student shall have attended the classes and shall have satisfied the examiners in all relevant prerequisite subjects.

B42 THE UNIVERSITY OF NEW SOUTH WALES

The student should refer to the appropriate Faculty Handbook or to the Calendar for a statement of subject prerequisites and/or co-requisites.

IV. The award of BSc in Psychology at graduation shall be at either Pass level or with Honours after a minimum of four years of full-time study.

RULES GOVERNING ADMISSION TO THE PSYCHOLOGY BSc COURSE WITH ADVANCED STANDING

- 1. Graduates of the University of New South Wales may be admitted to the Psychology BSc degree course with exemptions from no more than five subjects or their unit equivalents completed by them. No more than two Psychology subjects may be included in the subjects exempted.
- 2. Undergraduates of the University of New South Wales who transfer from another course to the Psychology BSc course may be admitted to the Psychology BSc course with exemption in no more than seven Psychology BSc course subjects or their unit equivalents.
- 3. Graduates or undergraduates of other universities may be admitted to the Psychology BSc course with advanced standing.
- 4. Students admitted under Rule 3 who have satisfied the examiners in subjects of the same title or subject matter as those permissible in the Psychology BSc course may, subject to the approval of the appropriate Heads of School, be granted exemption in no more than five subjects, of which no more than two may be Psychology subjects.

RECOMMENDED PSYCHOLOGY BSc COURSE PATTERNS

The course requirements have been so designed that they allow for:

(a) a solid core of psychology to equip the psychologist-intraining with psychological theory, skill in experimentation and psychological techniques by way of the equivalent of 6 compulsory psychology subjects [although the student may choose from a number of level II, III and IV units];

- (b) some supporting studies in mathematics and/or biology, of which a minimum of one course is compulsory;
- (c) some supporting studies in the social sciences, of which a minimum of one course is compulsory; and
- (d) the special needs, interests and academic or vocational background of the individual student to be considered when the balance of the five supporting subjects (or their equivalents in units) is selected in consultation with the Head of School or his representative.

For this reason, no course patterns are prescribed. The patterns to be completed by students who are admitted with advanced standing will take into account the subjects credited.

Students commencing university studies for the first time will arrange their patterns of supporting subjects in consultation with the Head of the School or his representative before completing enrolment. For such full-time students, some examples of patterns, based on supporting subject variants, are suggested below:

COMPULSOR	v	Year I	Year II	Year III	Year IV	B44
SUBJECTS FO COURSES	Ŷ	12.001	3 Psychology units value 12.042	8 Psychology units value	12.004	4
MAIN SUPPO SUBJECT: Pure Mathematics	ORTING (2 Yrs.)		{10.111A, 10.111B and {10.211A			THE UNIVERSITY
		Social Science Subject I Any approved level I Subject		An approved level I or II Subject (or equiv. units)		OF
	(3 Yrs.)	10.001	10.111A unit 10.111B unit 10.211A unit	10.112A unit 10.112B unit 10.112E unit 10.111C unit		NEW SOUTH WALES
		A Social Science Subject I Any approved level I Subject				H WALES
Statistics	(2 Yrs.)	10.001 A Social Science Subject I Any approved level I Subject	10.311	An approved level I or II Subject (or equiv. units)		

		Year I	Year U	Year III	Year IV
Biochemistry		17.011 and 17.021 2.001 10.001 or 10.021	41.101A 41.101B 41.101C	A Social Science Subject I	
Zoology	(2 Yrs.)	17.011 and 17.021 2.001 10.001 or 10.021	45.101A unit 45.101C unit		
			A Social Science Subject I	·····	
Physiology (2 Yrs.	(2 Yrs.)	17.011 and 17.021	73.011A		
	A Social Science Subject I Any approved level I Subject	Any approved level I or II Subject (or equiv. units)			
	(2 Yrs.)	17.011 and 17.021 2.001	73.011A		
		10.001 or 10.011	10.331		
		or 10.021	or A Pure Maths II Unit	A Social Science Subject I	
Social Science Subject	(2 Yrs.)	A Social Science Subject (A) I 10.011 or 10.001 or 10.021 or 17.011 and 17.021 Any approved level I Subject	Social Science Subject (A) II An approved level I or II Subject (or equivalent units)		

	Year I	Year II	Year III	Year IV
(3 Yrs.	A Social Science Subject (A) I 10.011 or 10.001 or 10.021 or 17.011 and 17.021 Any approved level I Subject	Social Science Subject (A) II	Social Science Subject (A) III	
General	Social Science Subject (A) I		Social Science Subject (A) II	
	17.011 and 17.021	12.402	or	
	10.001 or 10.021		Social Science Subject B (I)	
			or	
			Any approved level I or II Subject	

	No.	Subject or Unit	Level	Hours p.w.	When Offered	Prerequisites	Co-requisites
	12.001	Psychology I	I	5	Full yr.		
	12.042	Psychology IIA*	II	6	Full yr.	12.001	12.012
PSYCHOLOGY	12.004	Psychology IV	IV	15	Full yr.	All other Course requirements	
	12.402	Physiological Psychology	II	4	Full yr.	12.001, 17.001	
	12.042**	11 Psychology units	II & III				
	10.001	Mathematics I	I	6	Full yr.		
	10.011	Higher Mathematics I	I	6	Full yr.		
MATHEMATICS §	10.021	Mathematics IT	Ι	6	Full yr.		
	10.111A 10.111B 10.211A	Mathematics II	II	6	Full yr.	10.001 or 10.011	
	10.311	Theory of Statistics II	п	7	Full yr.	10.001 or 10.011 or 10.021 C	r.
	10.321	Higher Theory of Statistics II	II	8	Full yr.	10.001 or 10.011	
HUMAN BIOLOGY	17.011	Human Biology	I	6	Session 1		
HOMMIN BIOLOGI	17.021	Comparative Functional Biology	I	6	Session 2		
BIOCHEMISTRY UNITS §	41.101A	Chemistry of Biologically Important Molecules	II	6	Session 1	17.001	41.101B
	41.101B	Metabolism	II	6	Session 1	2.001	41.101A
	41.101D	Biochemical Control	II	6	Session 2	10.001 or 10.011 or 10.021	{41.101A {41.101B

SCHEDULE: MAIN PSYCHOLOGY BSc COURSE SUBJECTS

	No.	Subject or Unit	Level	Hours p.w.	When Offered	Prerequisites	Co-requisites	B48
ZOOLOGY UNITS §	45.101A	Genetics and Biometry	II	6	Session 1	17.001		
	45.101C	Vertebrates	II	6	Session 2	2.001 10.001 or 10.011 or 10.021		THE
PHYSIOLOGY UNITS §	73.011A	Principles of Physiology (Equiv. Unit Value = 2)	II	6	Full yr.	17.001 2.001 10.001 or 10.011 or 10.021		E UNIVERSITY
ECONOMICS	15.101 15.102	Economics I Economics II	I II	3 4	Full yr. Full yr.	15.101		SITY OF
PHILOSOPHY	52.111 52.112	Philosophy I Philosophy II	I II	4 5	Full yr. Full yr.	52.111		NEW
SOCIOLOGY	53.111 53.112	Sociology I Sociology II	I II	4 4 1	Full yr. Full yr.	53.111		SOUTH WALES
POLITICAL SCIENCE	54.111 54.112	Political Science I Political Science II	I II	3 1 31	Full yr. Full yr.	54.111		'ALES

* Day-time attendance for tutorials and practical work, including visits to institutions, etc., is required.

** For details of level II and level III units, including pre- and co-requisites, refer to Science Course details under Faculty of Science.

§ For details of level II and level III units, including pre- and co-requisites, refer to Science course details under Faculty of Science. If units are taken, three level II units are equivalent to one level II subjects; four level III units are equivalent to one level III subject.

POSTGRADUATE COURSES

On completion of a first degree course (BSc) the student may wish to proceed to a higher degree. This usually entails two or three years' research under direction. A limited number of Scholarships are available at this and other Universities, and these are competitive.

The regulations governing higher degrees are to be found in the University Calendar. A candidate thinking of undertaking such a course should first discuss the matter with the Head of the School in which he wishes to study.

A course in Food and Drug Analysis is offered by the School of Chemistry on a part-time basis over two years and leads to a diploma (DipFDA). The course is designed to provide systematic training at an advanced level for chemists who wish to extend their acquaintance with analytical techniques, and is thus suitable for those who wish to practise as public analysts. The School also offers a formal graduate course for the degree of Master of Chemistry (MChem) in Analytical Chemistry, on a full-time basis for one year. It will not operate in 1974, but will run full-time in 1975. The programme may also be extended in the future to part-time students. For full details see Calendar.

The School of Psychology offers a postgraduate formal course leading to the award of Master of Psychology (MPsychol). It is available to selected graduates with Honours in Psychology and provides professional training in either Experimental Clinical Psychology or Psychodynamic Clinical Psychology.

The School of Biological Technology, conjointly with the School of Chemical Engineering, offers a course in biochemical engineering which leads to the award of a postgraduate diploma (DipBiochemEng). The course may be completed in one year of full-time study or part-time over two years and is intended for graduates in chemical engineering, chemistry, biological sciences and agriculture.

The School also offers advanced treatments of important areas of Biotechnology in a postgraduate formal course leading to the

award of Master of Science (Biotechnology) (MSc(Biotech)). It may be undertaken by graduates with honours in Biotechnology or those who have completed the preliminary or qualifying programmes available in the School. The course is of one year's duration full time but may be completed over a longer period by part-time study.

The School of Mathematics offers a postgraduate course which covers a wide range of statistical theory and practice. It leads to the award of the degree of Master of Statistics (MStats), and is available on a two-year full-time basis or on a four-year part-time basis.

The course provides advanced training for practising statisticians, and is available to graduates with a pass degree in statistics or an honours degree in a related field (commonly mathematics) with supporting study in statistics. Honours graduates in statistics may be exempted from a maximum of half the course.

The School of Physics offers a postgraduate course, with an emphasis on Solid State Physics, which leads to the award of MPhysics. The course may be completed in one year of full-time study or two years of part-time study.

A formal graduate course for the degree of Master of Optometry (MOptom) is given by the School of Applied Physics and Optometry. For details see page B54.

SUBJECT INFORMATION AND TEXTBOOK LISTS

The following pages list details of textbooks, subject descriptions, etc. Reference books are not included here but the reference lists will be made available by the various schools. Information concerning general studies subjects is contained in the Handbook of the Board of General Studies which is available free of charge.

SCHOOL OF ANATOMY

The School of Anatomy offers three Level II units and four Level III units for Science students. Level II units comprise Mammalian Histology, Mammalian Embryology and Systematic Anatomy I (cardio-pulmonary). The Level III units are Systematic Anatomy II (locomotion), Systematic Anatomy III (alimentary and urogenital), Systematic Anatomy IV (neuroendocrine) and Comparative Histology. Students who major in Anatomy and who attain an adequate standard may proceed to a BSc degree with honours. Each Anatomy unit is offered once during the year as a day course only.

For details of level, unit value, when offered, hours per week, prerequisites and co-requisites, see page B27.

70.011A Mammalian Histology

Cell form and tissue structure. Cell structure and function. Cell function and evolution. Epithelial cells and tissue. Connective tissues and connective tissue cells. Muscle cell and muscle tissue. Nerve cell and nervous tissue. Cellular interrelations. Structure of organs and organ systems. Skin and derivatives. Development and structure of teeth. Circulatory System. Oral cavity. Alimentary canal and associated glands. Respiratory system. Urinary system. Eye, ear. Reproductive system.

TEXTBOOK

Bloom, W. & Fawcett, D. W. A Textbook of Histology. 9th ed. Saunders, 1968.

70.011B Mammalian Embryology

History of embryology and its development as a science. The mammalian reproductive system. Gametogenesis. Fertilisation and cleavage. Development and implantation of blastocyst. Development of embryonic disc, embryonic membranes, placenta. Comparative mammalian placentation. Human embryogenesis. Development of human foetus. Characteristics of external form. Teratology. Human organology. Comparative mammalian development. Biochemistry and embryogenesis.

TEXTBOOK

Arey, L. B. Developmental Anatomy. 7th ed. Saunders, 1965.

70.011C Systematic Anatomy I

Introduction to terms and concepts in systematic anatomy. Introduction to nervous system. Nerves of the thorax. Walls of the thorax. Thoracic movement. Circulation. Pericardium and heart. Blood vessels. Vascular distribution. Vascular anastomoses. The lymphatic system. Spleen and thymus. Skull, nasal cavity and sinuses. Larynx. Speech. Trachea and bronchi. Pleura and lungs. Living and radiological anatomy of air passages and thorax. Biomechanics of respiration.

TEXTBOOK

Basmajian, J. V. Primary Anatomy. 6th ed. Williams & Wilkins Co., Baltimore, 1970.

70.012A Systematic Anatomy II

Bone, cartilage and joints. Fascia and muscle. Bones, muscles and joints of limbs, trunk, head and neck. Course and distribution of peripheral nerve trunks. Vascular arrangements of the limbs. Biomechanics of movement, posture and locomotion. The hand as an organ. Living and radiological anatomy of trunk and limbs.

TEXTBOOK

Basmajian, J. V. Primary Anatomy. 6th ed. Williams & Wilkins Co., Baltimore, 1970.

70.012B Systematic Anatomy III

Mouth and salivary glands. Pharynx. Oesophagus. Abdomino-pelvic cavity. Peritoneum. Stomach, intestine, liver, pancreas. Kidney, ureter, bladder, urethra. Male and female reproductive organs. Innervation, blood supply and lymph drainage of gastro-intestinal and urogenital systems. Living and radiological anatomy of alimentary and urogenital organs.

TEXTBOOK

Basmajian, J. V. Primary Anatomy. 6th ed. Williams & Wilkins Co., Baltimore, 1970.

70.012C Systematic Anatomy IV

The neurons, neuronal satellite cells. Functional anatomy of the central nervous system. Blood supply of central nervous system. Organs of special sense. Endocrine glands. Principles of peripheral nerve distribution.

TEXTBOOK

Barr, M. L. The Human Nervous System: An Anatomical Approach. Harper & Row. Paperback.

70.012D Comparative Histology

Comparative cellular and intracellular structure and function. Comparative tissue structure and function. Vertebrate and invertebrate cells and tissues. Detailed comparative study of skin and derivatives. Bone and skeletal structure. Haemopoietic tissues. Cells of circulating blood and tissue fluids. Blood vascular system. Muscle. Nervous tissue and sense organs. Alimentary system and associated glands. Excretory system. Genital and reproductive tissue.

TEXTBOOK

Bloom, W. & Fawcett, D. W. A Textbook of Histology. 9th ed. Saunders, 1968.

SCHOOL OF APPLIED PHYSICS AND OPTOMETRY

DEPARTMENT OF APPLIED PHYSICS

There are significant and increasing numbers of opportunities for employment of physicists along with other technologists in the research-anddevelopment laboratories and other departments of Australian industrial firms. The kind of work done by industrial physicists is described as applied physics; and the Department of Applied Physics in this University has been set up to bring together industrial scientists and students and staff of the University.

The Department currently offers three level III units in the Science Course (31.113A, B and C). While these are intended as part of preparation for applied physics honours study, they are equally suitable for science students, whether majoring in physics or not, who have an interest in the application of physics in technology and have completed the pre-requisite units (1.112B or 1.122B; and 2.001 for 31.113A).

Undergraduates who are majoring in Physics in the Science Course and whose interest is in applying their subject are offered the opportunity to achieve a BSc with Honours in Applied Physics on the basis of the fourthyear course which the Department conducts.

In accordance with Science Course regulations, suitably-qualified students may apply to the Head of the School for admission to the Honours year on completing pass degree requirements. Suitable qualifications include, besides the major in Physics, completion of the majority of a range of Science Course subjects and units which give appropriate support to applied physics study. A recommended pass degree programme is:

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Year 1	1.001 (or 1.011) 2.001	Physics I Chemistry I
	5.001	Engineering I (2 units each)
	10.001 (or 10.011)	Mathematics I
Year 2	1.112 (or 1.122)	Physics II (units A, B & C)
	10.111 (or 10.121)	Pure Mathematics II (units A & B)
	10.211 (or 10.221)	Applied Mathematics II (unit A)
	2 further units from t	he "Preferred List" below
Year 3	1.113 (or 1.123 Higher Physics III (units A, B, C and D))	Physics III (units A, B and C)
	31.113	Applied Physics III (units A, B & C)
	Two (or one) further page to comprise a	r units from the "Preferred List" on next total of 8 units for the year.

"Preferred	List" of Science Course units
Level I*	17.011 General Biology 17.021 Comparative Functional Biology } { (1 unit each)
	12.001 Psychology I 25.001 Geology I (2 units each)
Level II	1.212T Physics (including "A" option)2.002 Chemistry II (units A, B & C)6.601A Introduction to Computing10.311T Statistics
Level III	1.133A Electronics 1.143 Physics (units A, B, C, E and F)

One only of 6.601B, 6.602C and 6.602D

The honours course comprises lectures, laboratory studies, and project work in areas of the application of physics to practical objectives, some of which areas may be chosen by the student from a number of electives. Also included is "Introduction to Industrial Practice" in which aspects of the work of scientists in industry will be critically studied in depth. In this course, and in other parts of the honours year work, the Department has the support of a staff of visiting lecturers from industry.

Graduates with honours in applied physics, or in physics, may register as research students in the Department working for the MSc or the PhD degree. Research work in the Department is directed towards practical objectives. Students working part-time or externally in appropriate fields for the MSc are welcome and given full encouragement.

Graduates not holding an appropriate honours degree either must present evidence of research ability, or must complete a qualifying course prescribed by the Department, before being accepted as higher-degree research students.

Students coming from outside the Science Course should note the "Rules governing Admission to the Science Degree Course with Advanced Standing" and in particular Rule 5, which relates to admission for the purpose of obtaining an honours degree. The "special programme" which Faculty would be recommended to prescribe, in the case of a pass graduate or graduand with a major in physics, would normally comprise one year of preparatory studies followed by the normal applied physics honours year. Depending on circumstances, the preparatory work might be accomplished by one year's full-time study, or might involve more than one year if part-time.

31.113A Physics of Materials I

A study of the physical properties of all types of materials in relation to structure.

TEXTBOOK

Van Vlack, L. H. Materials Science for Engineers. Addison-Wesley, 1970.

^{*} The Science Course Regulations (see 2(a)(ii)) require that not less than 8 nor more than 10 units be from level I. Completion of a 24-unit pass degree programme (as indicated), and a majority of graded

Completion of a 24-unit pass degree programme (as indicated), and a majority of graded passes in the level II and level III units, is normally required for admission to the Honours year.

[†] Prior to 1974, 17.001 General and Human Biology (2 units).

REFERENCE BOOKS

- Andrews, E. H. Fracture in Polymers. Oliver & Boyd.
- Bowden, F. P. & Tabor, D. Friction and Lubrication of Solids. O.U.P., 1964.
- Cottrell, A. H. The Mechanical Properties of Matter. Wiley.
- Dekker, A. J. Solid State Physics. Macmillan.
- Di Benedetto, A. T. The Structure and Properties of Materials. McGraw-Hill.
- Ferry, J. D. Viscoelastic Properties of Polymers. Wiley.
- Hutchinson, T. S. & Baird, D. C. The Physics of Engineering Solids. Wiley. Meares, P. Polymers: Structure and Bulk Properties. Van Nostrand.
- Ritchie, P. D., ed. The Physics of Plastics. Iliffe.
- Tabor, D. Gases, Liquids & Solids. Penguin.
- Treloar, L. R. G. The Physics of Rubber Elasticity. O.U.P.
- Wulff, J. et al. The Structure and Properties of Materials, 4 vols., Wiley, 1965.

31.113B Physics of Measurement

The general principles of measuring physical quantities and analysing measurements. Techniques of measurement, their scope and limitations.

TEXTBOOK

Cook, N. H., & Rabinowicz, E. Physical Measurement and Analysis. Addison-Wesley, 1963.

REFERENCE BOOKS

Blatt, J. M. Introduction to FORTRAN IV Programming, Goodyear, 1971.

- Brookes, A. M. P. Basic Instrumentation for Engineers and Physicists. Pergamon, 1968.
- Braddick, H. J. J. The Physics of Experimental Method, Chapman & Hall, 1956.
- Brophy, J. J. Basic Electronics for Scientists, McGraw-Hill, 1966. Brownlee, K. A. Industrial Experimentation, Chemical Publishing Co., 1953.
- Cooper, B. E. Statistics for Experimentalists, Pergamon, 1969.
- Cross, A. W. Experimental Microwaves. Sanders Electronics, Stevenage, 1970.
- Norton, H. N. Handbook of Transducers for Electronic Measuring Systems. Prentice-Hall, 1969.
- Rowe, J. An Introduction to Digital Electronics. Electronics Australia, 1970. Sears, F. W. Optics. Addison-Wesley, 1956.
- Standards Association of Australia. Australian Standard 1000-1970. The International System (SI) Units and their Application. S.A.A., Sydney, 1970.
- Whittaker, E. & Robinson, G. The Calculus of Observations. Blackie, 1952.

31.113C Applications of Radiation

Long-wave to short-wave electromagnetic radiation; some uses of electron beams and other radiations.

TEXTBOOK

No set text.

REFERENCE BOOKS Birchon, D. Optical Microscope Techniques. Newnes, 1961. Brown, J. G. X-rays and their Applications. Iliffe, 1966.

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Hackforth, H. L. Infra-Red Radiation. McGraw-Hill, 1960. Harrison, G. R. et al, Practical Spectroscopy. Blackie, 1949. Lipson, S. G. & Lipson, H. Optical Physics. C.U.P., 1969. McDowell, C. A. Mass Spectrometry. McGraw-Hill. Sears, F. W. Optics. Addison-Wesley, 1956. Stephens, R. W. B. & Bate, A. E. Acoustics and Vibrational Physics.

Arnold, 1966.

Part of each unit will comprise relevant laboratory work and other exercises. In these, students will be to a significant extent associated with the current programme of research work of the Department.

DEPARTMENT OF OPTOMETRY

The following courses are offered by the Department.

(a) A four-year full-time course leading to the degree of Bachelor of Optometry at either pass or honours level. This degree (BOptom) fulfils the requirements defined in the N.S.W. Optometrists (Amendment Act, 1963), and is the only course of professional training for Optometrists given in this State. Full details of the course appear earlier in this handĥook

(b) An extended undergraduate course leading to the double degree BSc/BOptom.

(c) A formal graduate course for the degree of Master of Optometry (MOptom). This course involves the study of three elective postgraduate subjects and advanced clinical optometry, together with the preparation of a thesis on an assigned project. It may be completed in one year of full-time study, or in the case of practising optometrists, in two or three vears of part-time study.

(d) Facilities for individual research are available and students who are considered as eligible may enrol with the university as candidates for the degrees of Master of Science or Doctor of Philosophy.

Further information on the foregoing may be obtained from the brochures issued by the Department of Optometry.

31.811 Optometry I

Geometrical and Physical Optics-Extension of Physics I content on the nature of light, reflection, refraction, thin lenses, optical instrument, dispersion and colour.

Lens systems and thick lenses, Interference, Diffraction, Polarisation, Photometry.

Mechanical Optics and Optical Dispensing-The manufacture and properties of spectacle lens materials. The optical properties of spherical, cylindrical, sphero-cylindrical, and prismatic spectacle lenses. Bifocal and multifocal lenses. Protective lenses. Frame measurements. Optical dispensing. Magnifying spectacles, and magnifying glasses. Lens aberrations and spectacle lens design. Lens measuring and lens testing instruments.

Physiological Optics-Optical system of the eye; the retinal image, visual acuity. Refraction of the eye; hyperopia, myopia, astigmatism, aphakia. Presbyopia. Anisometropia. The schematic eye. Theory of subjective refraction. Aberrations of the eye. Entopic phenomena. Accommodation and convergence. Binocular vision, stereoscopy.

TEXTBOOKS Emsley, H. H. Visual Optics. Vols. I & II. Butterworths. Fincham, W. H. A. Optics. Butterworths. Jalie, M. The Principles of Ophthalmic Lenses. Assoc. of Dispensing Opticians, London. REFERENCE BOOKS Andrews, C. H. Optics of the Electromagnetic Spectrum. Prentice-Hall. Bennett, A. G. Ophthalmic Lenses. Hatton. Curry, C. Wave Optics. Arnold. Emsley, H. H. Aberrations of Thin Lenses. Hatton. Helmholtz, H. Physiological Optics. Dover. Jenkins, F. A. & White, H. E. Fundamentals of Optics. McGraw-Hill. Morgan, J. Introduction to Geometrical and Physical Optics. McGraw-Hill. Rossi, B. Optics. Addison-Wesley. Sears, F. W. Optics. Addison-Wesley.

Smith, F. G. & Thomson, J. H. Optics. Wiley.

31.812 Optometry II

External and Internal Examination of the Eye-Case history and symptoms. Signs of local and/or general disease. Examination methods and instruments. Optometrical photography. Facial measurements and frame fitting.

Examination of Visual Functions—Theory and practice of perimetry. Criteria of norms. Interpretation of field defects. Evaluation of light and colour sense.

Refraction—Theory and practice of keratometry, objective and subjective refraction, prescribing special visual aids. Theory of design and construction of apparatus.

Orthoptics and Pleoptics-Assessment of binocular sensory and motor functions. Diagnosis and treatment of anomalies. Instrumentation,

Reading Deficiency-The reading process and its anomalies. Remedial training. Instrumentation.

Lighting-Elements of illumination engineering. Assessment of visibility. Sight conservation.

TEXTBOOKS

Aust, W. The Conservative Management of Squint. Karger.

Bier, N. Correction of Sub-Normal Vision. 2nd ed. Butterworths.

Clayton, G. H. Spectacle Frame Dispensing. Assoc. of Dispensing Opticians, London.

Harrington, D. O. The Visual Fields. Mosby.

REFERENCE BOOKS

Berens, C. & Zuckerman, J. Diagnostic Examination of the Eye. Lippincott. Berliner, S. Biomicroscopy of the Eye. Hoeber. Davson, H. The Eye. Vols. 3 & 4. Academic.

Doggart, J. H. Ocular Signs in Slit Lamp Microscopy. Kimpton,

Dowaliby, M. Modern Eyewear-Fashion and Cosmetic Dispensing. Professional Press.

Duke-Elder, S. The Practice of Refraction. Churchill.

Duke-Elder, S. Textbook of Ophthalmology. Kimpton. Giles, G. H. The Practice of Orthoptics. Hammond. Giles, G. H. The Principles and Practice of Refraction. Hammond.

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Hartridge, H. Recent Advances in the Physiology of Vision. Churchill. Lyle, K. & Jackson, S. Orthoptics. Lewis.

Ogle, K. N. Researches in Binocular Vision. Saunders.

Sasieni, L. S. The Principles and Practice of Optical Dispensing and Fitting. Hammond.

Shapero, M., Cline, D. & Hofstetter, H. W. Dictionary of Visual Science. Chilton.

Schonell, F. J. Backwardness in the Basic Subjects. Oliver & Boyd. Smith, W. Clinical Orthoptic Procedure. Mosby. Spache, G. D. Towards Better Reading. Garrard.

Tait, E. F. Textbook of Refraction. Saunders.

Traquair, H. M. An Introduction to Clinical Perimetry. Kimpton.

Wolff, M. D. & Wolff, J. A. Remedial Reading. McGraw-Hill.

31.813 Optometry III

Industrial Optometry-Job analysis and standardization of visual requirements. Occupational visual aids. Vision screening. Industrial hazards and industrial eye protection.

Contact Lenses-Theory and practice of prescribing haptic and corneal lenses. Instruments.

Theory of Spectacle Lenses and Optical Instruments-Advanced geometrical optics and spectacle lens design. Aberrations and their control. The elements of macroscopic and microscopic systems.

Advanced Visual Physiology and Physiological Optics-Recent advances in anatomy and physiology. An introduction to electro-physiology. Aetiology of refractive errors. Theories of colour perception and its anomalies. Evaluation of diagnostic tests. Theories of space perception. Distortion of stereoscopic space. Stereoptics.

Comparative Ophthalmology and Ocular Evolution—The anatomy and physiology of invertebrate and vertebrate visual organs. Evolution of binocular vision.

History of Optics-Discussion of the development of optics, ophthalmology and optometry against the background of a short history of science. Optometrical and interprofessional ethics.

TEXTBOOKS

Bennett, A. G. Optics of Contact Lenses. Association of Dispensing Opticians, London.

Burnham, R. W., Hanes, R. M. & Bartleson, C. J. Color: A Guide to Basic Facts and Concepts. Wiley.

Mandel, R. B. Contact Lens Practice: Basic and Advanced. Thomas.

REFERENCE BOOKS

Brandt, F. H. The Psychology of Seeing. Philosophical Library. Brindley, G. S. Physiology of the Retina and the Visual Pathway. Arnold. Conrady, A. E. Applied Optics and Optical Design. Dover.

Corson, R. Fashion in Eyeglasses. Owen.

Davson, H. The Eye. Vols. 1 & 4. Academic.

Duke-Elder, S. System of Ophthalmology. Vol. 1. Kimpton. Duke-Elder, S. & Perkins, E. S. Transparency of the Cornea. Blackwell.

Emsley, H. H. Aberrations of Thin Lenses. Hatton. Gibson, J. J. Perception of the Visual World. Houghton-Mifflin. Graham, C. H. Vision and Visual Perception. Wiley. Grosvenor, T. P. Contact Lens Theory and Practice. Professional Press. Hartridge, H. Recent Advances in the Physiology of Vision. Churchill.

Haynes, P. L. Encyclopaedia of Contact Lens Practice. International Optics.

Hirsch, M. J. & Wick, R. W. Vision of the Ageing Patient. Hammond. Hofstetter, H. W. Industrial Vision. Chilton.

Holmes, C. Guide to Occupational and other Visual Needs. Silverlake.

Ittelson, W. H. The Ames Demonstrations in Perception. Princeton. Johnson, B. K. Optics and Optical Instruments. Hatton. Julesz, B. Foundations of Cyclopean Perception. Chicago U.P., 1971.

Le Grand, Y. Light, Colour and Vision. Chapman & Hall. Martin, L. C. Technical Optics. Pitman.

Mazow, B. Synopsis of Corneal Contact Lens Fitting for Optometrists. Burgess.

Murray, H. D. Colour in Theory and Practice. Chapman & Hall.

Ogle, K. N. Researches in Binocular Vision. Saunders.

Optical Society of America. The Science of Color. Crowell. Polyak, S. The Vertebrate Visual System. Chicago U.P. Prince, J. H. Comparative Anatomy of the Eye. Thomas.

Prince, J. H. Ocular Prosthesis. Livingstone.

Sorsby, A. A Short History of Ophthalmology. Staples. Thomas, C. I. The Cornea. Thomas.

Thomas, P. F. Conoid Contact Lenses. Corneal Lens Corporation, Sydney.

Wright, W. D. Researches in Normal and Defective Colour Vision. Kimpton.

Wright, W. D. The Measurement of Colour. Hilger.

Wyszecki, G. & Stiles, W. S. Color Science. Wiley.

31.821 Special Anatomy and Physiology

Histology, Anatomy, and Embryology of the Eye and Associated Strucand nerves of the orbit and associated structures. The blood vessels, muscles, and nerves of the orbit and associated structures. The motor and sensory pathways associated with the visual apparatus. Elementary embryology and the detailed development of the eye and adnexae. Developmental defects of the eye and adnexae.

Physiology of the Eye and Vision-Physiology of the eyelids and lacrimal apparatus, cornea, aqueous humour and intra-ocular pressure, iris and pupil, lens and accommodation, retina and photo-chemistry of vision. Sensory responses to occular stimulation, luminosity curve, flicker, afterimages, and contrast phenomena. Visual acuity. Light- and dark-adaptation, photopic and scotopic vision. Colour vision and colour blindness. Eye movements, binocular vision, and stereopsis. Theories of vision, visual perception.

TEXTBOOKS

Moses, R. A. Adler's Physiology of the Eye. Mosby. Wolff, E. The Anatomy of the Eye and Orbit. Lewis.

REFERENCE BOOKS

Davson, H. The Eye. Vols. 1, 2 & 3. Academic. Duke-Elder, S. System of Ophthalmology. Vols. 2, 3 & 4. Kimpton. Keeney, A. H. Chronology of Ophthalmic Development. Thomas. Mann, I. The Development of the Human Eye. C.U.P. Mann, I. Developmental Abnormalities of the Eye. C.U.P. Polyak, S. L. The Retina. Chicago U.P. Rushton, W. A. H. Visual Pigments in Man. Liverpool U.P. Spooner, J. D. Ocular Anatomy. Hatton.

31.831 Diseases of the Eye

Introductory Bacteriology and Pathology-Pathogenic organisms, infection, immunity, allergic manifestations. Antiseptics and germicides, anti-biotics. Pathological tissue changes; cysts, neoplasms. Diseases of the blood, arteries, veins, heart, lungs, and kidneys. Venereal diseases. Diseases of the nervous system.

The Aetiology, Pathology, Diagnosis and Prognosis of Diseases of the Eye and Adnexae—Diseases of the eyelids, lacrimal apparatus, orbit, conjunctiva, cornea, sclera, uveal tract, lens, vitreous, retina, and optic nerve, Glaucoma, Ocular injuries, Sympathetic ophthalmia, Disease resulting from blood-borne infection. Disturbances of vision of central origin. Disturbances of ocular motility. Developmental abnormalities.

The Ocular Manifestation of Systemic Diseases-Ocular manifestations of: tuberculosis, syphilis, disorders of metabolism, dental sepsis, diseases of the kidneys, cardiovascular system, blood, endocrine system, central nervous system, phakomatoses and hereditary syndromes.

TEXTBOOKS

Lyle, T. K. & Cross, A. G. May & Worth's Manual of Diseases of the Eye. Bailliere.

Passmore, R. & Robson, J. S. eds. A Companion to Medical Studies. Vol. 2. Blackwell.

Perkins, E. S. & Hansell, P. An Atlas of Diseases of the Eye. Churchill.

REFERENCE BOOKS

Ballantyne, A. J. & Michaelson, I. C. Textbook of the Fundus of the Eye. Livingstone.

Doggart, J. H. Ocular Signs in Slit-Lamp Microscopy. Kimpton.

Doggart, J. H. Ophthalmic Medicine. Kimpton.

Duke-Elder, S. Textbook of Ophthalmology. Kimpton. Larsen, H. W. Atlas of the Fundus of the Eye. Munsgaard.

Nover, A. The Ocular Fundus. Lea & Febiger, Philadelphia.

SCHOOL OF BIOCHEMISTRY

Biochemistry involves a study of the chemistry of living organisms, and it is a subject where those interested in biology and those interested in chemistry work together to increase our understanding of life.

Some of the most spectacular achievements of recent times have been in the unravelling of the chemistry and function of very large molecules, macromolecules, of proteins, nucleic acids and polysaccharides which occur in living organisms. This has resulted in a better understanding of the transmission of hereditary information and the adaptation of organisms to the environment by mutations and natural selection.

Major areas of interest in Biochemistry at the present time involve a study of the chemistry of these large molecules involved in body processes such as growth, movement and reproduction. The formation and breakdown of these large molecules is known as metabolism and necessarily includes the chemical processes, involving both large and small molecules present in foodstuffs, which provide the necessary energy and the simple molecules or monomers that are then used in biosynthesis of these larger molecules which constitute the organism.

Basic to any understanding of the reactions of living organisms is the process of promoting chemical reactions under physiological conditions, that is, at low temperatures. A considerable emphasis is placed on understanding the hundreds of different types of enzymes, large protein molecules, each specifically designed for promoting one particular chemical reaction, the speed with which they function and the factors which control their operation.

The integration and control of biochemical reactions involves the study of hormones, or chemical messengers, that are synthesised in certain glands and exert their effects on cells, often situated in distant parts of the body, after being transported in the circulating blood or other vascular fluid.

A knowledge of Biochemistry is essential in maintaining the health of living organisms and is vital to the study of Medicine. There is an overlap with other biological sciences so that Biochemistry is a co-requisite or prerequisite for study in the disciplines of Biological Technology, Botany, Microbiology, Physiology and Zoology. Biochemistry on the other hand draws particularly on a background of Biology and Chemistry and some knowledge of Mathematics and Physics.

There are excellent prospects for advanced training at honours level, involving an additional year's training, mainly in research, and for graduate research work for the Master of Science degree or for the Doctorate in Philosophy.

Biochemistry plays a role in so many facets of human activity that for students majoring in Biochemistry there are good employment opportunities in educational institutions, in many research areas, e.g., in Universities, C.S.I.R.O., Public Health and hospital organizations, and in industries concerned with food, pharmaceuticals and agriculture.

For details of level, unit value, when offered, hours per week, prerequisites and co-requisites, see page B20.

41.101A Chemistry of Biologically Important Molecules

The chemical properties of amino acids, peptides and proteins, carbohydrates, nucleic acids and lipids and the biological roles of these compounds. The nature and function of enzymes. Practical work to illustrate the lecture course.

TEXTBOOKS

Barker, R. Organic Chemistry of Biological Compounds. Prentice-Hall, 1971.

Bohinski, R. C. Modern Concepts in Biochemistry. Allyn & Bacon, 1973.

Montgomery, R. & Swenson, C. A. Quantitative Problems in the Biological Sciences. Freeman, 1969.

REFERENCE BOOK

White, A., Handler, R. & Smith, E. L. Principles of Biochemistry. 5th ed. McGraw-Hill, 1973.

41.101B Metabolism

The intermediary metabolism of carbohydrates, lipids and nitrogenous compounds. The molecular mechanism of gene expression and protein synthesis. Practical work to illustrate the lecture course.

TEXTBOOKS

As for 41.101A plus McGilvery, R. W. Biochemistry: A Functional Approach. Saunders, 1970.

REFERENCE BOOKS As for 41.101A.

41.101C Control Mechanisms

The relation between structure and function of enzymes, hormones, vitamins and membranes. Photosynthesis. Metabolic networks and control mechanisms. Practical work to illustrate the lecture course.

TEXTBOOKS

As for 41.101B plus Frieden, E. & Lipner, H. Biochemical Endocrinology of the Vertebrates. Prentice-Hall, 1971.

REFERENCE BOOKS As for 41.101B.

41.102A Biochemistry of Macromolecules and Cell Biochemistry

Polysaccharides and glycoproteins including bacterial cell walls. Chemistry and biology of polynucleotides. Methods of amino acid and nucleic acid sequence analysis. Protein structure and synthesis. Active centres of some proteins. Sub-unit organization of proteins. Cellular degradation. Practical work to illustrate the lecture course and to provide experience in modern biochemical techniques.

TEXTBOOKS

- Barker, R. Organic Chemistry of Biological Compounds. Prentice-Hall, 1971.
- Frieden, E. & Lipner, H. Biochemical Endocrinology of the Vertebrates. Prentice-Hall, 1971.
- The Molecular Basis of Life. An Introduction to Molecular Biology. Readings from Scientific American. Freeman, 1968.
- White, A., Handler, R. & Smith, E. L. Principles of Biochemistry. 5th ed. McGraw Hill, 1973.

or

Lehninger, A. L. Biochemistry: The Molecular Basis of Cell Structure and Function. Worth Publishers Inc., 1970.

Wold, F. Macromolecules: Structure and Function. Prentice-Hall, 1971.

REFERENCE BOOKS

Bernhard, S. The Structure and Function of Enzymes. Benjamin, 1968.

Davidson, J. N. The Biochemistry of the Nucleic Acids. 7th ed. Methuen, 1972.

Watson, J. D. The Molecular Biology of the Gene. 2nd ed. Benjamin, 1970.

41.102B Metabolic Pathways and Control Mechanisms

Haemoproteins, and electron transport, photosynthesis, photophosphorylation and oxidative phosphorylation. The nature and function of coenzymes. Interrelationships in mammalian intermediary metabolism. Biochemical control mechanisms including hormones and allosteric interactions. Enzyme kinetics. Practical work to illustrate the lecture course and to provide experience in modern biochemical techniques.

TEXTBOOKS

As for 41.102A above.

REFERENCE BOOKS

McGilvery, R. W. Biochemistry: A Functional Approach. Saunders, 1970.

41.121 Biochemistry

Physical and chemical properties and functional roles of the principal biological constituents of man. Enzymology, energetics, metabolism of principal cell constituents in the organs and tissues of man, multicellular organisation, metabolic and hormonal regulation and whole body metabolism. The biochemistry of body fluids and specialized tissues. Energy storage in man, whole body metabolic economy and nutrition. Practical work to illustrate the lecture course.

TEXTBOOKS

Bohinski, R. C. Modern Concepts in Biochemistry. Allyn & Bacon, 1973. Frieden, E. & Lipner, H. Biochemical Endocrinology of the Vertebrates. Prentice-Hall, 1971.

McGilvery, R. W. Biochemistry: A Functional Approach. Saunders, 1970. Montgomery, R. & Swenson, C. A. Quantitative Problems in the Biochemical Sciences, Freeman, 1969.

REFERENCE BOOKS

Bondy, P. K. ed. Duncan's Diseases of Metabolism. 6th ed. Saunders, 1969. Christensen, H. N. pH and Dissociation: A programmed text. 2nd ed. Saunders, 1964.

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SCHOOL OF BIOLOGICAL TECHNOLOGY

Biotechnology is the interface between the biological sciences and their applications in industry, medicine and agriculture. It is an inter-disciplinary area in which considerable use is made of the techniques and methodologies of mathematics, the physical sciences and engineering.

Biotechnology is a firmly established area of study in relation to fermentation technology, enzyme technology and engineering, and the biodeterioration and biodegradation of materials, both organic and inorganic. In these areas, the principal basic biological sciences involved are biochemistry, microbiology and genetics, and considerable use is made of mathematical modelling and computer techniques, and of chemical engineering concepts. Other areas of biotechnology overlap with biomedical engineering in which various of the engineering disciplines intermesh with the medical sciences. Most of the activities of the School are collaborative with other schools and departments of the University, and in some areas with relevant industry.

The School offers one level III subject, Fermentation Technology, as an option to students taking major sequences in Microbiology and Biochemistry in the Science course (the subject is also available to students in the Food Technology course). The School also contributes to fourth year subjects in the Industrial Chemistry course (Processes) and in Electrical Engineering (Biomedical Engineering).

An honours programme in the fourth year of the Science course can be undertaken in the School by students who have reached a satisfactory standard in biochemical or microbiological subjects in the third year of the course. A graduate diploma course in Biochemical Engineering is offered in collaboration with the School of Chemical Engineering and is open to students in relevant disciplines. A Master's course (MSc(Biotech)) by formal study is offered for honours graduates in Biotechnology or other graduates who have reached the required entrance standards by appropriate routes. The course is of one year's duration full-time, but may be completed over a longer period on a part-time basis.

Registration for the research degrees of Master of Science or Doctor of Philosophy is offered to honours graduates in relevant disciplines or to those graduates who have completed the preliminary or qualifying programmes available in the School. *Financial support for higher degree studies is available.

The School is not interested in producing narrow specialists, but in the training of graduates who, by participation in formal courses and research programmes of a collaborative kind, are equipped to identify and solve a wide range of problems, and who are experienced in the multi-disciplinary approach and are appreciative of its potentialities.

42.102 Fermentation Technology

An introduction to the basic factors involved in the operation of microbial processes on an industrial scale, including: The selection, main-

^{*} See Postgraduate Scholarships in Section C of the Calendar.

tenance and improvement of micro-organisms; the influence of physical and chemical factors on the microbial environment; the control of environmental factors; the effects of operational patterns in batch and continuous flow cultivation; the harvesting, purification and standardisation of products; process optimisation; disposal of waste materials; an examination of selected microbial processes for chemical, pharmaceutical and food production, against the basic characteristics of large-scale fermentation processes practical exercises, including the operation of various types of fermenters, to illustrate the principal aspects of the lecture course.

TEXTBOOKS

- Aiba, S., Humphrey, A. E. & Millis, N. Biochemical Engineering. Academic, 1965.
- Casida, L. E. Jr. Industrial Microbiology. Wiley, 1968.
- Kubitschek, H. E. Introduction to Research with Continuous Cultures. Prentice-Hall, 1970.

REFERENCE BOOKS

- Blakebrough, N. ed. Biochemical and Biological Engineering Science. Vols. 1 & 2. Academic, 1968.
- Ghosh, T. & Fiechter, A. Advances in Biochemical Engineering. Springer-Verlag, 1971.
- Rhodes, A. & Fletcher, D. Principles of Industrial Microbiology. Pergamon, 1966.

Solomons, G. L. Materials and Methods in Fermentation. Academic, 1969. Webb, F. C. Biochemical Engineering. Van Nostrand, 1964.

For details of level, unit value, when offered, hours per week, prerequisites and co-requisites, see page B21.

42.103 Biological Technology (Honours)

Advanced formal training in selected areas of biochemistry and/or microbiology and participation in one of the school's research projects.

Botany is concerned with all aspects of the structure and function of plants and the relation of plants to the environment. Knowledge gained by investigations in these fields is important in agriculture, forestry and conservation, as well as in understanding the fundamental properties of biological material.

The major aspects of the subject which are taught in undergraduate courses in the School are Plant Genetics, Plant Physiology and Biochemistry, Plant Morphology, Ecology, Environmental Botany, Mycology and Plant Pathology. Any of these courses are usually combined with appropriate subjects in Biochemistry, Microbiology and Zoology. By this means, students may complete their studies with a broad spread over a number of biological disciplines, or may concentrate more in botanical aspects, with other subsidiary supporting subjects.

Fourth year Honours courses are provided for students wishing to specialise in a particular branch of Botany.

Research facilities are available within the School of postgraduate study leading to a degree of Master of Science or Doctor of Philosophy.

Careers for graduates in Botany include teaching, at secondary or tertiary level, scientific and technological work in food and drug industries, and investigational, research or extension work in the science and agriculture laboratories of State or Commonwealth organizations.

For details of level, unit value, when offered, hours per week, prerequisites and co-requisites, see page B21.

43.101A/45.101A Genetics and Biometry

Analysis of the mitotic cycle; replication of DNA and its organization in the chromosomes; linkage, non-meiotic recombination; mutation, structural changes, polyploidy, aneuploidy; population genetics; cytoplasmic inheritance; episomes; gene structure and function. An introduction to statistical methods and their application to biological data, including an introduction to analysis of variance and experimental design.

TEXTBOOKS

Clarke, M. C. Statistics and Experimental Design. Arnold, 1969.

Rohlf, F. T. & Sokal, R. Statistical Tables. Freeman, 1969.

Srb, A. M., Owen, R. D. & Edgar, R. S. General Genetics. 2nd ed. Freeman, 1965.

This unit is offered jointly by the Schools of Botany and Zoology.

43.101B Plant Evolution and Ecology

A study of the evolution of vegetative form and structure of vascular plants; an examination of their organization into terrestrial communities; identification, evolution and distribution of elements of the Australian flora. Students are required to attend field excursions, all of which form an integral part of the course.

Beadle, N. C. W., Evans, O. E. & Carolin, R. C. Flora of the Sydney Region. Reed, 1972.

Fahn, A. Plant Anatomy. Pergamon, 1967.

Whittaker, R. H. Communities and Ecosystems. Collier MacMillan, N.Y., 1970.

43.101C Plant Physiology

A general introduction to the physiology of the whole plant including a consideration of photosynthesis, inorganic nutrition, transport, translocation, physiology of growth and development, and plant growth substances and their application in agriculture.

TEXTBOOKS

Galston, A. W. & Davies, P. J. Control Mechanisms in Plant Development. Prentice-Hall, 1970.
Richardson, M. Translocation in Plants. Arnold, 1968.
Sutcliffe, J. Plants and Water. Arnold, 1968.
Whittingham, C. P. Photosynthesis. O.U.P., 1971.

43.102A Advanced Genetics

Human genetics including chromosome analysis, genetics of haemoglobin variation and drug response. Twin studies. Serum and enzyme polymorphisms. DNA studies including polarity and transcription concepts. Hypothesis of genetic recombination. Evolutionary genetics. Allelic complementation and fine structure. Polyploid cytogenetics, particularly wheat cytogenetics. Genetics of pathogenicity. Quantitative genetics. Heritability estimates and selection. Students are required to attend to long-term experiments outside formal class time.

43.102B Plant Taxonomy

Considers the assessment analysis and presentation of data for classifying plants both at the specific and supra-specific level. Students are required to attend field excursions all of which form an integral part of the course.

TEXTBOOKS*

- Beadle, N. C. W., Evans, O. D. & Carolin, R. C. Handbook of the Vascular Plants of the Sydney District and Blue Mountains. Authors, Armidale, 1962.
- Cronquist, A. The Evolution and Classification of Flowering Plants. Nelson, 1968.
- Heywood, V. H. Plant Taxonomy. The Institute of Biology's Studies in Botany No. 5. Arnold, 1970.

43.102C Plant Physiology and Biochemistry

Includes the pathway of carbon in leaves, lipid metabolism, hormone physiology and the cell wall. Modifications may include items of current interest in plant physiology. Projects are related to the above topics or to research in progress within the School, and may require attendance outside the hours set down in the timetable.

^{*} Students should consult lecturers in the course before purchasing textbooks.

43.102D Mycology

General structure and ultrastructure of the fungal cell. Morphology of members of the major taxonomic groups. Spore liberation, dispersal, deposition and germination. Cytology, genetics and patterns of life cycles. Growth and differentiation of hyphae and fruit bodies. Response to nutritional and environmental conditions for growth and reproduction. Transport processes, metabolism and metabolic products. Ecological considerations of fungi in specialised habitats.

TEXTBOOKS

Alexopoulos, C. J. Introductory Mycology. Wiley, 1962. Burnett, J. H. Fundementals of Mycology. Arnold, 1968.

43.102E Environmental Botany

An introduction to the soil and atmospheric environment in which terrestrial plants exist and a study of the behaviour and response of the flowering plant to its environment, both in nature and agriculture. Students are required to attend field excursions all of which form an integral part of the course.

43.102F Plant Pathology

History of plant pathology; pathogenic organisms; symptoms of disease. Specific diseases caused by fungi, nematodes, bacteria and viruses. Hostpathogen relationships including stages of infection, evolution of hostpathogen relationships, adaptation for successful parasitism, resistance mechanisms and genetics of resistance. Control of diseases by the use of fungicides, nematicides, crop rotation and breeding for resistance.

TEXTBOOKS

Alexopoulos, C. J. Introductory Mycology. Wiley, 1962. Walker, J. C. Plant Pathology. 3rd ed. McGraw-Hill, 1968.

SCHOOL OF CHEMISTRY

Chemistry is the science of materials, their properties and their transformations. As such, it is both an experimental and a theoretical science. Chemistry provides a common language for the experimental sciences, comparable with the language of quantitative scientific thought provided by mathematics, and is central among them, lying between physics on the one hand, and biology on the other. The interdependence of chemistry and other sciences is exemplified in the fields of biochemistry, chemical physics, geochemistry and chemical engineering. Additional to its intrinsic value, chemistry provides the basis of modern technology, through its contributions to medicine, industry and agriculture.

Career opportunities in chemistry are available for graduates in chemical industry, particularly in the research and development, control and management sections. Opportunities are also available in the universities and tertiary institutes, and in secondary teaching. Further opportunities are provided within Commonwealth and State departments, and within research organizations including the CSIRO and the AAEC.

Chemistry forms a part of many undergraduate courses offered, for example, Chemistry in the Science course, and Pure and Applied Chemistry. Additionally, there are courses within the Faculty of Applied Science, such as Industrial Chemistry, Ceramic Engineering, Food Technology, Chemical Engineering, Textile Technology and Metallurgy, which are predominantly concerned with technological aspects of chemistry.

The School of Chemistry provides two main undergraduate courses, namely (1) Pure and Applied Chemistry, and (2) Chemistry (as a co-major) in the Science Course. Both courses lead to the BSc degree.

A study of **Chemistry (as a co-major) in the Science course** involves a study of two branches of science to an advanced level. For example, a combination of level III Chemistry with level III Mathematics will provide a useful basis for later specialization in X-ray crystallography or theoretical chemistry; a combination of level III Chemistry with level III Geology will be of assistance to those who later wish to specialize in geochemistry. Another possibility is to combine level III Chemistry with level III Biochemistry units. These courses are suitable for those who wish to acquire advanced knowledge of two fields of study, or of interdisciplinary subjects. The Science course, as an alternative to the BSc(Ed) course, is also suitable for those planning to teach Chemistry at the secondary level. On a full-time basis, the Science course may be taken in three years (pass) or four years (with honours). On a part-time basis, however, the Science course may, according to the choice of subjects, require seven years (pass).

The aim of the **Pure and Applied Chemistry** course is to provide both depth and choice of subject matter at pass and honours level, to meet the needs of students who will become professional chemists. The course consists of a study of the fundamental principles of chemistry and of electives which deal with topics in contemporary fields of chemistry. It may be taken either full-time (three years for pass, four years for honours) or part-time (six years for pass, eight years for students taking the parttime course to find employment in some branch of the chemical industry. The role of basic scientific research in the creation of modern industrial society is widely accepted. The usual introduction to research in chemistry is provided by the honours degree (in either the Science course, or the Pure and Applied Chemistry course), which may be followed by a higher research degree in Chemistry (e.g., MSc, PhD). These degrees are aimed at those whose interests are in research and/or teaching. Alternatively, postgraduate training in chemistry is provided through formal Diploma or Master's courses (e.g. the Diploma in Food and Drug Analysis, and the MChem in Analytical Chemistry).

REQUIREMENTS FOR HONOURS IN CHEMISTRY

Students desiring admission to the honours course must apply in writing to the Head of the School not later than 30th November of the year in which the third year of the full-time (or equivalent stage of the part-time) course is completed.

The requirement for admission to the honours course is a sufficiently meritorious record in the work of the pass degree.

The major part of the work for honours will consist of a research project on which a written thesis is submitted. There is also some formal course work. Attendance will be required at such lectures and seminars as the Head of the School directs. Honours will not be awarded in any particular branch of the subject, but in chemistry as a whole.

For admission to the honours chemistry course in Science, the applicant must complete at least eight level III units, of which at least four must be in Chemistry. Students who, at the beginning of their third year are already interested in taking honours in chemistry, are advised to seek guidance from the School about the most appropriate subject to accompany the level III Chemistry units.

Prospective Honours students in the Pure and Applied Chemistry course should seek guidance before choosing their final year elective subjects.

2.001 Chemistry I

Classification of matter and theories of the structure of matter. Atomic structure, the periodic table and chemical behaviour. Chemical bonds and molecular structure. Equilibrium and change in chemical systems. The structure, nomenclature and properties of organic compounds. Reactions of organic compounds.

TEXTBOOKS

Aylward, G. H. & Findlay, T. J. V. SI Chemical Data. Wiley, 1971

Chemistry I Laboratory Manual. Univ. of N.S.W., 1974.

Kneen, W. R., Rogers, M. J. W. & Simpson, P. Chemistry: Facts, Patterns and Principles. Addison-Wesley, 1972.

Schaum Outline Series. Theory and Problems of College Chemistry, SI (Metric) ed. McGraw-Hill.

REFERENCE BOOKS

Barrow, G. M., Kenney, M. E., Lassila, J. D., Litle, R. L. & Thompson, W. E. Understanding Chemistry. Benjamin, 1969.

Brown, G. I. A New Guide to Modern Valency Theory. Longmans, 1967. Eastwood, F. W., Swan, J. M. & Yonatt, J. B. Organic Chemistry. A First University Course in Twelve Programs. Science Press, 1967. Gray, H. B. & Haight, G. P. Basic Principles of Chemistry. Benjamin, 1967.

Pauling, L. College Chemistry. 3rd ed. Freeman, 1964.

Runquist, O., Cresswell, C. J. & Head, J. T. Chemical Principles. A Programmed Text. Burgess Pub. Co., 1968.

Sisler, H. H., Van der Werf, C. A. & Davidson, A. W. College Chemistry. 3rd ed. Collier-Macmillan, 1967. Vogel, A. I. Macro and Semimicro Qualitative Analysis. 4th ed. Longmans,

1954.

2.002A Chemistry II (Physical Chemistry)

Quantum mechanics; molecular energy and thermodynamics; chemical application of thermodynamics; surface and colloid chemistry.

TEXTBOOKS

Aylward, G. H. & Findlay, T. J. V. SI Chemical Data. Wiley, Sydney, 1971.

Barrow, G. M. Physical Chemistry. 3rd ed. McGraw-Hill, 1973.

Shaw, D. J. Introduction to Colloid and Surface Chemistry. 2nd ed. Butterworth, 1970.

REFERENCE BOOKS

Adamson, A. W. Textbook of Physical Chemistry. Academic, 1973.

Alexander, A. E. & Johnson, P. Colloid Science. O.U.P., 1950. Barrow, G. M. Structure of Molecules. Benjamin, 1963.

Daniels, F. et al. Experimental Physical Chemistry. 7th ed. McGraw-Hill, 1970.

Daniels, F. & Alberty, R. A. Physical Chemistry. 3rd ed. Wiley, 1966. Glasstone, S. Textbook of Physical Chemistry. 2nd ed. Van Nostrand, 1948. Moore, W. J. Physical Chemistry. 4th or 5th ed. Longman, 1963 or 1972. Shoemaker, D. P. & Garland, C. W. Experiments in Physical Chemistry. 2nd ed. McGraw-Hill, 1967.

2.002B Chemistry II (Organic Chemistry)

Chemistry of the more important functional groups: aliphatic hydrocarbons, monocyclic aromatic hydrocarbons, halides, alcohols, phenols, aldehydes, ketones, ethers, carboxylic acids and their derivatives, nitro compounds, amines and sulphonic acids.

TEXTBOOKS

Morrison, R. T. & Boyd, R. N. Organic Chemistry. 3rd ed. Int. Stud. Ed. Allyn & Bacon, 1973.

Only if proceeding to 2.003B Chemistry III (Organic Chemistry):

Vogel, A. I. Elementary Practical Organic Chemistry, Pt. II. Qualitative Organic Analysis. Longman, 1957.

2.002C Chemistry II (Inorganic/Analytical Chemistry)

Chemistry of non-metals; chemistry of typical metals; transition metals, lanthanides and actinides; introduction to nuclear chemistry. Quantitative inorganic analysis.

B72

- Fischer, R. B. & Peters, D. G. Quantitative Chemical Analysis. Saunders, 1968.
- Jolly, W. L. The Chemistry of the Non-Metals. Prentice-Hall, 1966.
- Quagliano, J. V. & Vallarino, L. M. Coordination Chemistry. Heath & Co., Lexington, 1969.

REFERENCE BOOKS

Basolo, F. & Johnson, R. Coordination Chemistry. Benjamin, 1964.

Carswell, D. J. Introduction to Nuclear Chemistry. Elsevier, 1967.

Cotton, F. A. & Wilkinson, G. Advanced Inorganic Chemistry. 2nd ed. Wiley, 1966.

2.003A Chemistry III (Physical Chemistry)

Physico-chemical aspects of spectroscopy—quantum mechanical approach; electronic and vibrational spectra; nuclear magnetic resonance and electron spin resonance spectroscopy. Chemical kinetics—transition state theory; theories of uni-molecular reactions; chemistry of excited species.

TEXTBOOKS

Banwell, C. N. Fundamentals of Molecular Spectroscopy. 2nd ed. McGraw-Hill, 1972.

Barrow, G. M. Physical Chemistry. 2nd ed. McGraw-Hill, 1966.

Dixon, R. N. Spectroscopy and Structure. Methuen, 1965.

Laidler, K. J. Chemical Kinetics. 2nd ed. McGraw-Hill, 1965.

REFERENCE BOOKS

Benson, S. W. Thermochemical Kinetics. Wiley, 1968.

Calvert, J. G. & Pitts, J. N. Photochemistry. Wiley, 1966.

- Carrington, A. & McLachlan, A. D. Introduction to Magnetic Resonance. Harper & Row, 1967.
- Daniels, F. et al. Experimental Physical Chemistry. 6th or 7th ed. McGraw-Hill, 1962 or 1970.
- Gardiner, W. C. Rates and Mechanisms of Chemical Reactions. Benjamin, 1969.

Glasstone, S., Laidler, K. J. & Eyring, H. Theory of Rate Processes. McGraw-Hill, 1941.

King, G. W. Spectroscopy and Molecular Structure. Holt, Rinehart & Winston, 1964.

Shoemaker, D. P. & Garland, C. W. Experiments in Physical Chemistry. 2nd ed. McGraw-Hill, 1967.

Weston, R. E. & Schwarz, H. A. Chemical Kinetics. Prentice-Hall, 1972.

2.003B Chemistry III (Organic Chemistry)

Stereochemistry of acyclic systems. Alicyclic Chemistry: the synthesis and properties of monocyclic systems, conformational aspects of cyclohexane and related systems, rearrangement reactions, and the chemistry of fused and bridged polycyclic compounds. *Heterocyclic Chemistry*: the chemistry of pyridine, quinoline, isoquinoline, and benzopyran and its derivatives; the chemistry of pyrrole, furan, and thiophene and their benzo derivatives; the chemistry of pyrimidine, imidazole and pyrazole,

- 1. Morrison, R. T. & Boyd, R. N. Organic Chemistry. 3rd ed. Int. Stud. Ed. Allvn & Bacon, 1973. or
- Roberts, J. D. & Caserio, M. C. Basic Principles of Organic Chemistry. Benjamin, 1964.
- 2. Tedder, J. M., Nechvatal, A., Murray, A. W. & Carnduff, J. Basic Organic Chemistry. Pt. 3. Wiley, 1970.
- 3. Vogel, A. I. Elementary Practical Organic Chemistry. Pt. II. Qualitative Organic Analysis. Longmans, 1957.

REFERENCE BOOKS

Acheson, R. M. An Introduction to the Chemistry of Heterocyclic Com-Pounds. 2nd ed. Wiley Int. Ed., 1967.
Eliel, E. L. Stereochemistry of Carbon Compounds. McGraw-Hill, 1962.
Eliel, E. L., Allinger, N. L., Angyal, S. J. & Morrison, G. A. Conforma-

- tional Analysis. Interscience, 1965.
- Gould, E. S. Mechanism and Structure in Organic Chemistry, Holt, Rinehart & Winston, 1959.

Hallas, G. Organic Stereochemistry. McGraw-Hill, 1965.

- March, J. Advanced Organic Chemistry: Reactions, Mechanisms and Structure. McGraw-Hill, 1968. Sykes, P. A Guidebook to Mechanism in Organic Chemistry. 3rd. ed.
- Longman, 1971.

Whitham, G. H. Alicyclic Chemistry. Oldbourne.

2.003C Chemistry III (Inorganic Chemistry)

Molecular structure determination, with particular reference to complex salts, optical activity, crystal structure, systematic chemistry of the lanthanides and transition elements, further chemistry of nitrogen, sulphur and the halogens.

TEXTBOOKS

Cotton, F. A. & Wilkinson, G. Advanced Inorganic Chemistry. 2nd ed. Wiley, 1966.

Vogel, A. I. A Textbook of Macro & Semi-micro Qualitative Inorganic Analysis. Longmans.

REFERENCE BOOKS

Bailar, J. C. Chemistry of the Coordination Compounds. Reinhold, 1960. Barnard, A. K. Theoretical Basis of Inorganic Chemistry. McGraw-Hill, 1965.

Dwyer, F. P. & Mellor, D. P. Chelating Agents and Metal Chelates. Academic, 1964.

Lewis, J. & Wilkins, R. Modern Coordination Chemistry. Interscience, 1959. Pauling, L. Nature of the Chemical Bond. 3rd. ed. Cornell, 1960.

Sienko, M. J. & Plane, R. A. Physical Inorganic Chemistry. Benjamin, 1965. Wells, A. F. Structural Inorganic Chemistry. 3rd ed. O.U.P., 1962.

2.003D Chemistry III (Analytical Chemistry)

Ionic equilibria in solution; advanced qualitative analysis; advanced electrochemical analysis; advanced spectrophotometry; separations and preconcentrations.

- Eckschlager, K., Chalmers, R. A. trans. ed. Errors and Measurement in Chemical Analysis. Van Nostrand, 1969.
- Ewing, G. W. Instrumental Methods of Chemical Analysis. McGraw-Hill, 1969.
- Fischer, R. B. & Peters, D. G. Quantitative Chemical Analysis. Saunders, 1968

2.003E Chemistry III (Nuclear and Radiation Chemistry)

Nuclear structure, reactions, transformations; radioactive properties and measurements, radiations, isotopes, radio-chemical techniques.

TEXTBOOKS

Carswell, D. J. Introduction to Nuclear Chemistry. Elsevier, 1967. or

Friedlander, G., Kennedy, J. & Miller, J. M. Nuclear and Radiochemistry. 2nd ed. Wiley, 1964.

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Harvey, B. Introduction to Nuclear Physics and Chemistry. Prentice Hall, 1962.

REFERENCE BOOKS

Ausloos, P. Fundamental Processes in Radiation Chemistry. Interscience, 1968.

Denaro, A. R. & Jayson, G. G. Fundamentals of Radiation Chemistry. Butterworths, 1972.

Farley, S. Elements of Pulse Circuits. Methuen, 1955.

Haissinsky, M. Nuclear Chemistry and its Applications. Tuck, D. C., trans. Masson, 1957.

Sharpe, J. Nuclear Radiation Detectors. Methuen, 1964. Spinks, J. W. T. & Woods, R. J. An Introduction to Radiation Chemistry. Wiley, 1964. Taylor, D. The Measurement of Radioisotopes. Methuen, 1959.

2.013A Theoretical Chemistry

A fundamental approach to wave mechanics—operators; solving the Schrödinger wave equation; variation and perturbation methods; many-electron problem; vector coupling; allowed transitions. Chemical kinetics transition state theory; theories of unimolecular reactions; chemistry of excited species.

TEXTBOOKS

Golding, R. M. Applied Wave Mechanics. Van Nostrand, 1969. Laidler, K. J. Chemical Kinetics, 2nd ed. McGraw-Hill, 1965.

REFERENCE BOOKS

Amdur, I. & Hammes, G. G. Chemical Kinetics. McGraw-Hill, 1966.

- Calvert, J. G. & Pitts, J. N. Photochemistry. Wiley, 1966. Daniels, F. et al. Experimental Physical Chemistry. 7th ed. McGraw-Hill, 1970.
- Gardiner, W. C. Rates and mechanisms of Chemical Reactions. Benjamin, 1969.

Glasstone, S., Laidler, K. J. & Eyring, H. Theory of Rate Processes. McGraw-Hill, 1941.

Shoemaker, D. P. & Garland, C. W. Experiments in Physical Chemistry. 2nd ed. McGraw-Hill, 1967.

2.023A Chemical Physics

Wave mechanics-linear operators: Schrödinger wave equation, applications, methods of solution; variation principle; linear combinations; perturbation theory. The many-electron problem-central field method; electron spin; Fermi-Dirac statistics; angular momentum operators; Coulomb repulsion two-electron operator; spin-orbit coupling; Russell-Saunders and jj coupling; Zeeman effect; vector coupling and Wigner coefficients; allowed transitions. Group theory—symmetry operations; matrix representation; irreducible representation; characters of a group; non-rigid molecules; antisymmetry operations.

TEXTBOOK

B76

Golding, R. M. Applied Wave Mechanics. Van Nostrand, 1969.

2.004 Chemistry IV (Science Honours)

Advanced lectures and research project.

2.022 Chemistry II(M)

Units 2.002A (Physical Chemistry) and 2.002C (Inorganic Chemistry) of 2.002 Chemistry II (Science).

2.033A Macromolecules

Structural classes. Macromolecules in solution: determination of molecular size, conformation. Macromolecules in the solid state; methods of investigation.

TEXTBOOK

Van Holde, K. E. Physical Biochemistry. Prentice-Hall, 1971.

REFERENCE BOOKS

Amdur, I. & Hammes, G. C. Chemical Kinetics. McGraw-Hill, 1966.

Caldin, E. F. Fast Reactions In Solution. Blackwell, 1964.

Mahler, H. R. & Cordes, E. H. Biological Chemistry. Harper & Row, 1971. Poland, D. & Sheraga, H. A. Theory of Helix-Coil Transitions In Biopoly-mers. Academic, 1970.

Tanford, C. Physical Chemistry of Macromolecules. Wiley, 1961.

2.091 Project

For Honours students in Pure and Applied Chemistry.

2.211 Applied Organic Chemistry

A discussion of selected topics at advanced level of commercially important groups of organic materials.

Theoretical chemistry, physical properties, thermal and photo-initiated processes are treated together with methods of examination in an overall unit approach correlating structure with behaviour. Emphasis is placed on breakdown to model systems.

Subject-matter covers: Theory of physical techniques, refractometry, polarimetry, etc., from basis of additivity, spectroscopy, visible and ultraviolet definitions and chromophores. Fatty acids with emphasis on unsaturation, thermal and oxidative polymerizations and alkyd resins, analysis of mixtures, isomerizations, clath-rates, etc. Waxes, sterols. Essential oils, review of terpenoid materials, mono-, sesqui-, and diterpenoids, analytical chemistry, correlation of physical constants with composition, methods for individual classes of constituents. Alkaloids and fine chemicals, assays and basis of group separation procedures. Vitamins, synthesis, assays and analysis, stability and determination. Antibiotics, treatment of commonly used materials, penicillin, etc. Chromatography of organic materials, theory of processes. Solvent-solute relationships. Synthetic and vegetable insecticides, e.g., pyrethrum and organophosphorus group. Synthetic resins and high polymers, polymerization processes-initiation and inhibitor efficiencies and measurements, thermal and photo degradation, identification and analysis problems. Sulphur in the vulcanization process. Mechanisms in sulphur chemistry. Ozone-processes. Anti-ozonants, mode of action,

TEXTBOOK

No set text.

REFERENCE BOOKS

Carney, T. P. Laboratory Fractional Distillation. McMillan, 1949.

Flory, P. J. Principles of Polymer Chemistry. Cornell, 1953.

Heftmann, E. Chromatography. 2nd ed. Reinhold, 1967. Kan, R. O. Organic Photochemistry. McGraw-Hill, 1966.

Kharasch, N. ed. The Chemistry of Organic Sulphur Compounds. Vol 1, 1961.

Lenz, R. W. Organic Chemistry of Synthetic Polymers. Interscience, 1967. Markley, K. L. The Fatty Acids. 2nd ed. Interscience, 1960-67.

Pinder, A. R. Chemistry of the Terpenes. Chapman & Hall, 1960. Pryor, W. A. Mechanisms of Sulphur Reactions. McGraw-Hill, 1962. Schwarz, J. C. P. Physical Methods in Organic Chemistry. Oliver & Boyd, 1964.

Scott, G. Atmospheric Oxidation and Antioxidants. Elsevier, 1965.

Solomon, D. H. Organic Film Formers. Wiley, 1967.

2.221 Chemistry and Enzymology of Foods

Subject matter covers areas similar to 2.261 Chemistry and Enzymology of Foods with reduction in scope and depth. Emphasis is continued on the integration of different areas of chemistry.

TEXTBOOK

No set text.

REFERENCE BOOKS

Heftmann, E. Chromatography. 2nd ed. Reinhold, 1967.

Joslyn, M. A. Methods in Food Analysis. Academic, 1950.

Karrer, P. & Jucker, E. Carotenoids. Elsevier, 1950.

Markley, K. L. The Fatty Acids. 2nd ed. Interscience, 1960-67. Neurath, H. The Proteins. Vols. I-IV. 2nd ed. Academic, 1963-68. Pigman, W. The Carbohydrates. Academic, 1957. Winton, A. L. & Winton, K. B. Structure and Composition of Foods. Wiley, 1932.

Subsidiary lists are supplied from the Department.

2.261 Chemistry and Enzymology of Foods

Covers the chemistry of food constituents at an advanced level and provides a correct appreciation of the relationship between the chemistry and enzymology associated with the origin and post-harvest handling of the foodstuff. Treatment is given of deteriorative changes in colour and texture occurring during processing and storage. Analytical procedures, chemical and physical are discussed where necessary, integrated with the remainder of the subject matter.

General classification of constituents, role of moisture. Fixed oils and fats, rancidity of enzymic and autoxidative origin, anti-oxidants-natural and synthetic-theories on mechanisms of action, carbohydrates reactivity, role in browning processes, carbohydrate polymers, starch structure, enzymic susceptibility and mode of action, estimation, pectic substances and other gelling agents, gel structure. Proteins, sulphur chemistry of proteins, position in cereal chemistry, bleachers and improvers, theories on mode of action, redox and displacement reactions. Colour systems, origin, development and chemistry of natural food pigments, carotenoids, chlorophyll, etc. Stability and estimations, enzymic degradation and enzymic browning, vitamins, preservatives.

TEXTBOOK

No set text.

REFERENCE BOOKS

Gunstone, F. D. An Introduction to the Chemistry and Biochemistry of the Fatty Acids and their Glycerides. Chapman & Hall, 1968.

Heftmann, E. Chromatography. 2nd ed. Reinhold, 1967.

Reinnann, E. Chromatography. 2nd ed. Reinnold, 1907.
Karrer, P. & Jucker, E. Carotenoids. Elsevier, 1950.
Markley, K. L. The Fatty Acids. 2nd ed. Interscience, 1960-67.
Neurath, H. The Proteins. Vols. I-IV. 2nd ed. Academic, 1963-68.
Pigman, W. The Carbohydrates. Academic, 1957.
Reed, G. Enzymes in Food Processing. Academic, 1966.
Schultz, H. W. ed. Carbohydrates and Their Roles. Avi. Pub. Co., 1969.
Schultz, I. C. P. Physical Matheds in Organic Chamistry, Oliver & Boyd

Schwarz, J. C. P. Physical Methods in Organic Chemistry. Oliver & Boyd, 1964.

Scott, G. Atmospheric Oxidation and Antioxidants. Elsevier, 1965.

Walton, H. F. Principles and Methods of Chemical Analysis. 2nd ed. Prentice-Hall, 1964.

Willard, H. H., Merritt, L. L. & Dean, J. A. Instrumental Methods of Analysis, 4th ed. Van Nostrand, 1965.

Subsidiary lists are supplied from the Department.

2.303 Theoretical Chemistry

Advanced physico-chemical topics of a theoretical nature; in two equal strands: (a) Wave mechanics, development and applications of group theory. (b) Any strand from 2.333 Physical Chemistry.

TEXTBOOK

(a)

Golding, R. M. Applied Wave Mechanics. Van Nostrand, 1969.

2.311 Physical Chemistry I

Quantum mechanics; molecular energy and thermodynamics; chemical statistics, chemical application of thermodynamics; surface and colloid chemistry.

B78

Adamson, A. W. Textbook of Physical Chemistry, Academic, 1973. Aylward, G. H. & Findlay, T. J. V. SI Chemical Data. Wiley, 1971.

REFERENCE BOOKS

Alexander, A. E. & Johnson, P. Colloid Science. O.U.P., 1950.

Barrow, G. M. Structure of Molecules. Benjamin. 1963.

Daniels, F. et al. Experimental Physical Chemistry. 7th ed. McGraw-Hill, 1970.

Daniels, F. & Alberty, R. A. Physical Chemistry. 3rd ed. Wiley, 1966.

Glasstone, S. Textbook of Physical Chemistry. 2nd ed. Van Nostrand, 1948.

Moore, W. J. Physical Chemistry. 4th or 5th ed. Longman, 1963 or 1972. Shaw, D. J. Introduction to Colloid and Surface Chemistry. 2nd ed. Butter-

- worths, 1970. Shoemaker, D. P. & Garland, C. W. Experiments in Physical Chemistry.
- 2nd ed. McGraw-Hill, 1967.

2.322 Physical Chemistry II

Subject description, text and reference book lists as for 2.003A Chemistry III (Physical Chemistry).

2.333 Physical Chemistry

Advanced physico-chemical topics, to be chosen from two of the following strands: (a) Statistical thermodynamics; its application to gases, liquids and chemical equilibria; states of matter. (b) Infrared, Raman, microwave and electronic spectroscopy; lasers; optical properties of molecules. (c) Non-ideal thermodynamics, electrode processes and electrolyte solution equilibria. (This series is intended to cover topics of interest in inorganic, organic and analytical chemistry.) (d) Physico-chemical proper-ties of macromolecular systems; colligative and electrokinetic properties and conformation in solution; solid state structure and properties.

(A strand chosen as part of 2.303 Theoretical Chemistry cannot be chosen as part of this subject.)

TEXTBOOKS

(b)

Banwell, C. N. Fundamentals of Molecular Spectroscopy. 2nd ed. McGraw-Hill, 1972.

Dixon, R. N. Spectroscopy and Structure. Methuen, 1965. (d)

Van Holde, K. E. Physical Biochemistry. Prentice-Hall, 1971.

REFERENCE BOOKS

(a)

Andrews, F. C. Equilibrium Statistical Mechanics. Wiley, 1963.

Hill, T. L. Introduction to Statistical Thermodynamics. Addison-Wesley. 1960.

Knox, J. H. Molecular Thermodynamics. Wiley, 1971. Moelwyn-Hughes, E. A. Physical Chemistry. 2nd ed. Pergamon, 1961.

Tabor, D. Gases, Liquids and Solids. Penguin, 1969.

(b)

Chang, R. Basic Principles of Spectroscopy. McGraw-Hill, 1971.

Eyring, H., Walter, J. & Kimball, G. E. Ouantum Chemistry, Wiley, 1944.

Heavens, O. S. Optical Masers. Methuen, 1964.

- Herzberg, G. Molecular Spectra and Molecular Structure. II. Infrared and Raman Spectra of Polyatomic Molecules. Van Nostrand, 1945.
- Herzberg, G. Molecular Spectra and Molecular Structure. III. Electronic Spectra and Electronic Structure of Polyatomic Molecules. Van Nostrand, 1966.

Herzberg, G. Spectra and Structures of Simple Free Radicals. Cornell, 1971.

- King, G. W. Spectroscopy and Molecular Structure. Holt, Rinehart & Winston, 1964.
- Wilson, E. B., Decius, J. C. & Cross, P. C. Molecular Vibrations. McGraw-Hill, 1955.

(d)

Amdur, I. & Hammes, G. C. Chemical Kinetics. McGraw-Hill, 1966. Caldin, E. F. Fast Reactions In Solution. Blackwell, 1964. Mahler, H. R. & Cordes, E. H. Biological Chemistry. Harper & Row, 1971.

Poland, D. & Scheraga, H. A. Theory of Helix-Coil Transitions in Biopolymers. Academic, 1970.

Tanford, C. Physical Chemistry of Macromolecules. Wiley, 1961.

2.391 Basic Diffraction Theory

Symmetry, point and space groups, crystal lattices. Scattering of X-rays by electrons, atoms and ideal lattices. Bragg's law. Effect of temperature and finite atom size, mosaicity, extinction, intensity formula. Structure factors, reciprocal lattice-geometrical interpretaion of Bragg's law.

TEXTBOOK

Klug, H. P. & Alexander, L. E. X-Ray Diffraction Procedure. Wiley, 1954.

2.392 Structure Determination Methods

Phase problem. Patterson function, heavy atom method, superposition methods and isomorphous replacement. Transform approach. Direct methods, inequalities, equalities and statistical approach. Refinement, difference Fourier syntheses, differential syntheses, least squares. Accuracy, standard deviations of bond lengths and angles, best fit planes. Significance levels. "R" factor.

TEXTBOOK

Buerger, M. J. Crystal Structure Analyses. Wiley, 1960.

2.393 Recording Methods

Photographic; powder, focussing methods, Laue, oscillation, Weissenberg procession. Counter methods; powder, parafocussing, three and four circle goniostats, linear diffractometers.

TEXTBOOK

Henry, N. F. M., Lipson, H & Wooster, W. A. The Interpretation of X-Ray Diffraction Photographs. 2nd ed. Macmillan, 1960.

2.394 Crystal Optics

Use of microscopics, polarising. Optical goniometer, birefringence, optical diffraction.

TEXTBOOK

Phillips, F. C. Introduction to Crystallography. 3rd ed. Longmans, 1963.

B80

2.411 Inorganic Chemistry I

Chemistry of the non-metals including B, C, Si, N, P, S, Se, Te, halogens, and noble gases. Chemistry of the metals of groups IA, IIA, and A1. Typical ionic, giant-molecule and close-packed structures. Transition metal chemistry, including variable oxidation states, paramagnetism, Werner's theory, isomerism of six- and four-coordinate complexes, chelation, stabilization of valency states. Physical methods of molecular structure determination. Chemistry of Fe, Co, Ni, Cu, Ag, Au.

TEXTBOOKS

1.

Jolly, W. L. The Chemistry of the Non-Metals. Prentice Hall, 1966. Larsen, E. M. Transitional Elements. Benjamin, 1965.

Quagliano, J. V. & Vallarino, L. M. Coordination Chemistry. Heath, Lexington, 1969.

or 2.

Cotton, F. A. & Wilkinson, G. Advanced Inorganic Chemistry. 2nd ed. Wiley, 1966.

REFERENCE BOOKS

Bailar, J. C. Chemistry of Coordination Compounds. Reinhold, 1960.

- Barnard, A. K. Theoretical Basis of Inorganic Chemistry. McGraw-Hill, 1965.
- Basolo, F. & Johnson, R. Introduction to Coordination Chemistry. Benjamin, 1964.
- Graddon, D. P. An Introduction to Coordination Chemistry. 2nd ed. Pergamon, 1968.
- Jones, M. M. Elementary Coordination Chemistry. Prentice-Hall, 1964.
- Vogel, A. A Textbook of Macro & Semi-micro Qualitative Inorganic Analysis. Longmans.
- Wells, A. F. Structural Inorganic Chemistry. 3rd ed. O.U.P., 1962.

2.422 Inorganic Chemistry II

Chemistry of groups VA, VIA, VIIA, the lanthanides and actinides. More advanced chemistry of groups IIIB, IVB, VB, VIB, VIB and inert gases. Crystal field theory, formation constants of complex ions, unusual oxidation states of transition metals.

TEXTBOOK

Cotton, F. A. & Wilkinson, G. Advanced Inorganic Chemistry. 2nd ed. Wiley, 1966.

REFERENCE BOOKS

Bailar, J. C. Chemistry of Coordination Compounds. Reinhold, 1960.

- Barnard, A. K. Theoretical Basis of Inorganic Chemistry, McGraw-Hill, 1965.
- Dwyer, F. & Mellor, D. P. Chelating Agents & Metal Chelates. Academic 1964.
- Lewis, J. & Wilkins, R. G. Modern Coordination Chemistry. Interscience, 1960.

Sienko, M. J. & Plane, R. A. Physical Inorganic Chemistry. Benjamin, 1965.

Wells, A. F. Structural Inorganic Chemistry. 3rd ed. O.U.P., 1962.

2.433 Inorganic Chemistry

(i) Reaction mechanisms involving metal complexes. (ii) Thermodynamics of complex formation. (iii) Spectroscopic methods for investigating metal complexes, including infrared, electronic, NMR and Mossbauer spectroscopy. (iv) π -complexes.

TEXTBOOK

Cotton, F. A. & Wilkinson, G. Advanced Inorganic Chemistry. 2nd ed. Wiley, 1966.

REFERENCE BOOKS

- Adams, D. M. & Raynor, J. B. Advanced Practical Inorganic Chemistry. Wiley, 1965. Basolo, F. & Pearson, R. Mechanism of Inorganic Reactions. 2nd ed. Wiley,
- 1965.
- Chemical Society of London. Special Publication No. 17, 1964. Stability Constants of Metal Ion Complexes.
- Cotton, F. A. Chemical Applications of Group Theory. Wiley, 1963.
- Dwyer, F. P. & Mellor, D. P. Chelating Agents & Metal Chelates. Academic, 1964.

- Edwards, J. O. Inorganic Reaction Mechanisms. Benjamin, 1964. Hannay, N. B. Solid State Chemistry. Prentice-Hall, 1967. Lewis, J. & Wilkins, R. Modern Coordination Chemistry. Interscience, 1959. Nakamoto, K. Infrared Spectra of Inorganic and Coordination Compounds.
- 2nd ed. Wiley/Interscience, 1970. Nakamoto, K. & McCarthy, P. J. Spectroscopy & Structure of Metal Chelate
- Compounds. Wiley, 1968. Wells, A. F. Structural Inorganic Chemistry. 3rd ed. O.U.P., 1962.

2.511 Analytical Chemistry I

Sampling; data evaluation; ionic equilibria in solution; electrochemical analysis; volumetric analysis; spectroscopy in analytical chemistry.

TEXTBOOKS

- Ewing, G. W. Instrumental Methods of Chemical Analysis. McGraw-Hill, 1969.
- Fischer, R. B. & Peters, D. G. Quantitative Chemical Analysis. Saunders, 1968.

2.513 Analytical Biochemistry

(i) Electrochemistry of biological systems. (ii) Spectroscopy and fluorimetry. (iii) Chromatography of biological systems. (iv) Microscopy.

TEXTBOOK

Bender, G. T. Chemical Instrumentation: A Laboratory Manual Based on Clinical Chemistry, Saunders, 1972.

2.522 Analytical Chemistry II

Solution chromatography; gas chromatography; advanced electrochemical analysis; emission spectroscopy; instruments in analytical chemistry; precision absorption spectrophotometry in solution; evaluation and development of a spectrophotometric method; literature of analytical chemistry.

- Chalmers, R. A. Aspects of Analytical Chemistry. Contemporary Science. Oliver & Boyd, 1968. Paperback.
- Eckschlager, K. Errors and Measurements in Chemical Analysis. Chalmers, R. A. trans. ed. Van Nostrand, 1969.
- Ewing, G. W. Instrumental Methods of Chemical Analysis. McGraw-Hill, 1969.

Hamilton, L. F., Simpson, S. & Ellis, D. W. Calculations of Analytical Chemistry. 7th ed. McGraw-Hill, 1969.

2.533 Analytical Chemistry III

Kinetics in analytical chemistry; emission and absorption spectroscopy in flames; spectrometric methods (IR, Mass, XRF, electron probe and NMR); chemical analysis of organic and biological materials; differential thermal analysis; complexes in analytical chemistry; automation and data processing in analytical chemistry.

TEXTBOOKS

- Chalmers, R. A. Aspects of Analytical Chemistry. Contemporary Science. Oliver & Boyd, 1968. Paperback.
- Eckschlager, K. Errors and Measurements in Chemical Analysis. Chalmers, R. A. trans. ed. Van Nostrand, 1969.
- Ewing, G. W. Instrumental Methods of Chemical Analysis. McGraw-Hill, 1969.
- Hamilton, L. F., Simpson, S. & Ellis, D. W. Calculations of Analytical Chemistry. 7th ed. McGraw-Hill, 1969.
- Kolthoff, I. M., Sandell, E. B., Meehan, E. J. & Bruckenstein, S. Quantitative Chemical Analysis. Macmillan, 1969.
- Schwarzenbach, G. & Flaschka, H. Complexometric Titrations. Irving, H. M. trans. 2nd ed. Methuen, 1969.

2.611 Organic Chemistry I

As for 2.002B Chemistry II (Organic Chemistry).

TEXTBOOKS

Morrison, R. T. & Boyd, R. N. Organic Chemistry. 3rd ed. Int. Stud. Ed. Allyn & Bacon, 1973.

Only if proceeding to 2.622 Organic Chemistry II:

Vogel, A. I. Elementary Practical Organic Chemistry. Pt. II. Qualitative Organic Analysis. Longman, 1957.

2.622 Organic Chemistry II

Subject description, text and reference book lists as for 2.003B Chemistry III (Organic Chemistry).

2.633 Organic Chemistry

(i) Spectroscopic methods in organic chemistry: emphasis on the correlations of spectra with structure. Ultraviolet spectroscopy: absorption laws and techniques. Infrared spectroscopy: experimental techniques. Absorption by common structural groupings. Nuclear magnetic resonance

spectroscopy: the NMR phenomenon, the chemical shift, shielding mechanisms, spin-spin interactions. (ii) Synthetic methods in organic chemistry: reduction by hydrogenation, diimide, metal hydrides and dissolving metal systems. Enamines. Olefin formation. (iii) High energy chemistry: mass spectrometry in the elucidation of the structures of organic compounds. Photochemistry: radical processes, oxidation coupling of phenols, reactions of carbenes.

TEXTBOOK

Dyke, S. F., Floyd, A. J., Sainsbury, M. & Theobald, R. S. Organic Spectroscopy. Penguin Books, 1971.

REFERENCE BOOKS

Budzikiewicz, H., Djerassi, C. & Williams, D. H. Mass Spectrometry of Organic Compounds. Holden-Day, 1967.

Carruthers, W. Some Modern Methods of Organic Synthesis. C.U.P., 1971. Colthup, N. B., Daley, L. H. & Wiberley, S. E. Introduction to Infrared and Raman Spectroscopy. Academic, 1964.

Gilchrist, T. L. & Rees, C. W. Carbenes, Nitrenes and Arynes. Nelson, 1969.

Jackman, L. M. & Sternhell, S. Applications of Nuclear Magnetic Resonance Spectroscopy in Organic Chemistry. Pergamon, 1969.

Kan, R. O. Organic Photochemistry. McGraw-Hill, 1966.

Mathieson, D. W. ed. Nuclear Magnetic Resonance for Organic Chemists. Academic, 1967.

Scott, A. I. Interpretation of the Ultraviolet Spectra of Natural Products. Pergamon, 1964.

Williams, D. H. & Fleming, I. Spectroscopic Methods in Organic Chemistry. McGraw-Hill, 1966.

2.711 Solid State Chemistry

(i) Symmetry, diffraction and determination of crystal structures. (ii) Typical structures, lattice defects, deviations from stoichiometry, semi-conduction. (iii) Electronic structure and physical properties of solids, solid state reactions, surface properties and catalysis. Applications of EPR, NMR and mass spectrometry.

TEXTBOOKS

Bond, G. C. Catalysis by Metals. Academic, 1965.

Greenwood, N. N. Ionic Crystals, Lattice Defects and Non-stoichiometry. Butterworths, 1969.

Moore, W. J. Seven Solid States. Benjamin, 1967.

2.811 Nuclear and Radiation Chemistry

For the student who requires a general foundation in the subject, which he can later apply to other fields. Topics are: Fundamental particles; structure and properties of the nucleus; nuclear reactions and radioactive decay; origin, properties and measurement of nuclear radiations; nuclear instrumentation; preparation and applications of separated stable isotopes and of radioisotopes; radiation chemistry; radiochemical techniques; carbon dating and geochronology; the transuranium elements.

TEXT AND REFERENCE BOOKS

As for 2.003E Chemistry III (Nuclear and Radiation Chemistry).

2.911 Applied Chemistry

(a) Information utilization: introductory instrumentation and analogue computation (see also 22.143); an advanced treatment of Fortran programming, data reduction and analysis, regression analysis; information retrieval. (b) Chemical resources and environment: ecological relationships between man and the physical and biological world, resources of matter and energy, current and predicted states of human environment, pollution, corrosion; sociological implications of technological advances.

TEXTBOOKS

Hamilton, C. H. Chemistry in the Environment. Freeman, 1973. Schaum Outline Series. Numerical Analysis. McGraw-Hill, 1968.

COMPUTER SCIENCE

FOR STUDENTS IN THE SCIENCE COURSE

Students in the Science course may major in Computer Science. This course is provided by the Department of Computer Science within the School of Electrical Engineering; the course is available on a full-time basis only and leads to the degree of BSc (pass or honours).

Students of sufficient merit who have completed the undergraduate units in Computer Science may be admitted to the honours course in fourth year. Permission to enter the course is granted by the Head of the Department of Computer Science. The honours course consists of prescribed lectures, seminars and reading in the areas of mathematical theory of computation, computer applications, computer logic and organization.

6.601A Introduction to Computer Science

Problem solving: problem formation, construction of mathematical model, algorithm formulation, computer solution. Computer solution using ALGOLW, PL/I or FORTRAN. Classification of algorithms. Simple machine language. Development of symbolic assemblers, high level languages, operating systems. Data representation. Errors. Computability.

TEXTBOOKS

- Ralston, A. Introduction to Programming and Computer Science. McGraw-Hill.
- Sites, R. L. Algol W Reference Manual. U.N.S.W. Students Union (by permission Stanford University).

REFERENCE BOOKS

Bates, F. & Douglas, M. L. Programming Language/One. Prentice-Hall.

Knuth, D. Fundamental Algorithms. The Art of Computer Programming. Vol I. Addison-Wesley.

Ralston, A. Fortran IV Programming: A Concise Exposition. McGraw-Hill. Rice, J. K. & Rice, J. R. Introduction to Computer Science. Holt, Rinehart &

Rice, J. K. & Rice, J. R. Introduction to Computer Science. Holt, Rinehart & Winston.

6.601B Assembler Programming and Non-numeric Processing

Computer structure, machine language, instruction execution, addressing techniques and digital representation of data. Symbolic coding. Manipulation of strings, lists and other data structures.

REFERENCE BOOKS

Griswold, R. E., Poage, J. F. & Polansky, I. P. The SNOBOL 4 Programming Language. Prentice-Hall.

IBM System/360: Principles of Operation. Form A22-6821. IBM.

Stone, H. S. Introduction to Computer Organisation and Data Structures. McGraw-Hill.

6.602A Computer Systems

Switching algebra, simplification of switching functions, synchronous sequential networks, digital systems. Flow tables, cycles, races, hazards. Number systems, codes, computer arithmetic. Memory techniques and organization, microprogramming.

TEXTBOOK

Booth, T. L. Digital Networks and Computer Systems. Wiley.

6.602B Computer Systems II

Organization and components of digital computing systems. Operating system components: processes, their implementation, control and interaction. Handling of interrupts, use of queues. Storage management and addressing techniques. Resource allocation and protection. Input/output control, file systems, data communications. System control job scheduling, error recovery, system interfaces. Software engineering.

REFERENCE BOOKS

Colin, A. J. T. Introduction to Operating Systems. Macdonald/Elsevier.
Genuys, F. ed. Programming Languages. Academic Press.
Knuth, D. E. Fundamental Algorithms, The Art of Computer Programming.
Vol. I. Addison-Wesley.

Rosen, S. Programming Systems & Languages. McGraw-Hill. Yourdon, E. Design of On-Line Computer Systems. McGraw-Hill.

6.602C Computer Applications

A selection of topics from Computer simulation. Modelling of discrete event systems, with applications to queueing; Pseudo random member generation and testing; simulation languages, especially GPSS. Optimization techniques: "hill climbing", critical path method, dynamic programming, linear programming. The simplex and revised simplex methods. Job shop scheduling. Data processing; file and data management systems; use of COBOL; searching and sorting of files. Information retrieval: search on secondary keys, inverted files. Artificial intelligence. Social consequences of computer technology.

REFERENCE BOOKS

Gass, S. I. Linear Programming. McGraw-Hill.

Gordon, G. System Simulation. Prentice-Hill.

Barrodale, I., Roberts, F. D. K. & Ehle, B. L. Elementary Computer Applications. Wiley.

6.602D Programming Languages and Compiling Techniques

Compiling Techniques: data structures; table look-up; language description; lexical analysis; syntax analysis; semantic analysis/code generation; interpretation/program execution.

Programming Languages: a comparative study.

REFERENCE BOOKS

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Aho, A. V. & Ullman, J. D. Theory of Parsing, Translation and Compiling. Vols. I & II. Prentice-Hall.

Foster, J. M. Automatic Syntactic Analysis. Macdonald/Elsevier.

Gries, D. Compiler Construction for Digital Computers. Wiley.

Hopgood, F. R. A. Compiling Techniques. Macdonald/Elsevier.

- Higman, B. A Comparative Study of Programming Languages. Macdonald/ Elsevier.
- Knuth, D. Fundamental Algorithms. The Art of Computer Programming Vol I. McGraw-Hill.
- McKeeman, W. M., Horning, J. J. & Wortman, D. B. A Compiler Generator. Prentice-Hall.

Rosen, S. Programming Systems & Languages. McGraw-Hill.

- Sammet, J. Programming Languages: History and Fundamentals. Prentice-Hall.
- Wegner, P. Programming Languages, Information Structures & Machine Organization. McGraw-Hill.

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5.010 Engineering A

Prerequisite: None.

Engineering Mechanics I: Two and three dimensional force systems, composition and resolution of forces, laws of equilibrium. Statics of rigid bars, pin-jointed frames. Shear force, axial force, bending moment. Simple states of stress. Kinematics of the plane motion of a particle. Kinetics of the plane motion of a particle; equations of motion, dynamic equilibrium, work and energy.

Introduction to Engineering Design: Engineering method, problem identification, creative thinking, mathematical modelling, computer aided design, materials and processes, communication of ideas, the place of engineering in society.

Materials: An introductory course on the production, structure and properties of the main types of engineering materials, with a brief introduction to the process used in shaping and fabricating them.

TEXTBOOKS

Svensson, N. L. Introduction to Engineering Design. N.S.W. U.P. Waldron, K. J. Engineering Mechanics. Wiley. Walshaw, A C. SI Units in Worked Examples. Longman.

REFERENCE BOOKS

Beakley, G. C. & Chilton, E. G. Introduction to Engineering Design and Graphics. Macmillan.

Beer, F. P. & Johnston, E. R. Statics and Dynamics. Vector, ed. McGraw-Hill.

Harrisberger, L. Engineersmanship. Wadsworth.

Higdon, A. & Stiles, W. B. Engineering Mechanics. Vector, ed. Prentice-Hall. Krick, E. V. Introduction to Engineering and Engineering Design. Wiley. McCormick, E. J. Human Engineering. McGraw-Hill.

Meriam, J. L. Statics and Dynamics. Wiley.

For Materials:

Aitchison, L. A History of Metals. Vols. I & II. MacDonald & Evans.

Dennis, W. H. Extractive Metallurgy. McGraw-Hill.

Gilchrist, J. D. Extraction Metallurgy. Pergamon.

Guy, A. C. Physical Metallurgy for Engineers. Addison-Wesley.

Newton, J. Extractive Metallurgy. Wiley.

Street, A. Metals in the Service of Man. Penguin.

Timoshenko, S. History of the Strength of Materials. Van Nostrand.

5.030 Engineering C

Engineering Drawing: Fundamental concepts of descriptive geometry, including reference systems, representation of point, line and plane; fundamental problems of position and measurement. Application of descriptive geometry to certain problems arising in engineering practice. Special emphasis on ability to visualize problems and processes involved in their solution. Instruction in the correct use of drawing instruments and the application of drawing standards. Measurements and dimensioning. Orthographic and isometric projections.

Introduction to Systems and Computers: Introduction to computers to follow the computer work in Mathematics I. To develop: (a) familiarity with algorithms; (b) the use of procedure oriented languages; and (c) an introduction to computing equipment. Systems, Introduction and concepts: concepts and introduction to systems. To give students an appreciation of some of the concepts used in engineering, to relate the concepts to phenomena within their experience, and to illustrate them by case histories and engineering examples. Quantities. Concepts. Components. Systems.

TEXTBOOKS

For Engineering Drawing: Robertson, R. G. Descriptive Geometry. Pitman. Thomson, R. Exercises in Graphic Communication. Nelson.

For Introduction to Systems and Computers:

Karbowiak, A. E. & Huey, R. M. eds. Information, Computers, Machines and Man. Wiley.

B90

17.001A Human Biology

Mankind evolving: primate evolution; background of early man. Evolution of technological man: biological problems associated with communication and tool-making; development of man as a hunting predator. Development of utilization of natural resources: development of man as a pastoralist and farmer; animal and plant domestication. Evolution of urban man, culture, society: reproductive biology and genetics of man; population growth, fluctuation, control; natural history of disease, background of medical and industrial microbiology. Effects of modern society: biology of social stress; effect of society in contemporary environments, planning and control.

TEXTBOOKS

Abercrombie, M. et al. A Dictionary of Biology. Penguin, 1967. Boughey, A. S. Man and the Environment. Macmillan, 1971. Clarke, C. A. Human Genetics and Medicine. Arnold, 1970.

REFERENCE BOOKS

Baker, H. B. Plants and Civilization. Wadsworth, 1970.

Bates, M. Man in Nature. Prentice-Hall, 1964.

Carter, C. O. Human Heredity. Penguin, 1962.

Ehrlich, P. A. & Ehrlich, A. H. Population, Resources, Environment. Freeman, 1972.

Jorgensen, J. G. ed. Biology and Culture in Modern Perspective. Freeman, 1972.

Mulvaney, D. J. & Golson, G. eds. Aboriginal Man and Environment in Australia. A.N.U., 1971.

Nix, H. A. ed. The City as a Life System. Southwood Press, 1973.

Thomas, W. L. ed. Man's Role in Changing the Face of the Earth. Chicago U.P., 1956.

Watt, K. F. Principles of Environmental Science. McGraw-Hill, 1973.

Weiner, J. S. Man's Natural History. Weidenfeld & Nicolson, 1971.

Young, J. A. An Introduction to the Study of Man. Clarendon, 1971.

17.001B Comparative Functional Biology

Maintenance of the organism: gas exchange systems in plants and animals; transport inside organisms; uptake, digestions, absorption; enzymes structure and function. Photosynthesis: process and structural relationships; metabolic systems, energy yields and pathways.

Developing organisms: sexual reproduction in plants and animals, general life cycle patterns; cell development and differentiation in flowering plants and mammals.

Control and co-ordination in organisms: organisms and water, uptake and effects; control mechanisms, urinary systems and kidney structure and function. Stimuli and responses: plant hormones, hormones in vertebrate animals, muscle activity and muscle structure, eye structure and vision mechanism; ear structure and hearing mechanism; nerves, central nervous system, nerve action, brain structure and functioning.

Abercrombie, M. et al. A Dictionary of Biology. Penguin, 1967. Roberts, M. B. V. Biology: A functional approach. Nelson, 1971.

REFERENCE BOOKS

Grenville, H. W. Biology of the Individual. Longman, 1971.

Griffin, D. R. & Novick, A. 2nd ed. Animal Structure and Function. Holt. Rinehart & Winston, 1970.

Kramer, A. ed. Topics in the Study of Life. Harper & Row, 1971.

Ray, P. M. The Living Plant. 2nd ed. Holt, Rinehart & Winston, 1972.

17.012 General Ecology

Evolution and environmental selection in the Australian continent: geological, paleoclimatological, biogeographical and historical background. Organizational structure of biological populations, with special reference to plants, animals and microorganisms. Functional organization of eco-systems: energy budgets, hydrological and biogeochemical cycles. Integrated structure and function of ecosystems: case studies of soil, terrestrial, aquatic and urban ecosystems. Cropping and management of natural resources. Natural history of disease and pest invasion; integrated pest control. Systems analysis and dynamic programming in resource managements and ecological problem-solving.

TEXTBOOKS

Alexander, M. Microbial Ecology. Wiley, 1971. Odum, E. P. Fundamentals of Ecology. Saunders, 1953. Watt, K. E. F. Principles of Environmental Science. McGraw-Hill, 1973.

REFERENCE BOOKS

Ehrlich, P. R. & Ehrlich, A. H. Population, Resources, Environment. Freeman, 1972. Also in paperback.

Kershaw, K. A. Quantitative and Dynamic Ecology. Arnold, 1964. Phillipson, J. Ecological Energetics. Arnold, 1966.

Solomon, M. E. Population Dynamics. Arnold, 1969.

Wagner, R. H. Environment and Man. Norton, 1971.

REQUIREMENTS FOR PRACTICAL WORK

A list of equipment required for practical work is posted on the notice board in the ground floor of the Biological Sciences Building. Students must purchase this material before the first practical class.

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GEOGRAPHY

FOR STUDENTS IN THE SCIENCE COURSE

The geographer studies variations from place to place on the earth arising from the spatial relationships of the phenomena making up man's physical and social environment. Apart from its cultural value, an understanding of these relationships is necessary for the conservation and planned development of physical and economic resources. Geography IS should be of particular interest to those studying concurrently in the physical and biological sciences.

27.031 Geography IS

Part I

Economic Geography: The geographic problems of scale and distance. The relevance of theory and quantitative methods. Economic landscape systems. Patterns and structures of systems of agriculture, manufacturing and tertiary production. Geographic significance of population growth components in modernizing and advanced countries; natural increase, fertility and mortality patterns and internal and international migration. Includes an urban field tutorial of one day.

Laboratory classes consist of the application of statistical methods to areal and point data.

TEXTBOOKS

Toyne, P. & Newby, P. T. Techniques in Human Geography. Macmillan. Wilson, M. G. A. Population Geography. Nelson.

REFERENCE BOOKS

Abler, R., Adams, J. S. & Gould, P. Spatial Organisation. Prentice-Hall.

Chisholm, M. Rural Settlement and Land Use. Hutchinson.

Clarke, J. I. Population Geography. Pergamon.

Cooke, R. U. & Johnson, J. H. Trends in Geography. Pergamon. Paperback.

- Cox, K. R. Man, Location and Behavior: An Introduction to Human Geography. Wiley.
- Davies, W. K. D. ed. The Conceptual Revolution in Geography. Univ. of London Press.
- Demko, G. J., Schnell, G. A. & Rose, H. M. Population Geography: A Reader. McGraw-Hill.
- Dohrs, F. E. & Sommers, L. M. eds. Introduction to Geography: Selected Readings. Crowell. Paperback.
- Ehrlich, P. R. & A. N. Population, Resources, Environment. Freeman, Paperback.
- Haggett, P. Geography: A Modern Synthesis. Harper International Paperback.
- Logan, M. I. & Missen, G. J. New Viewpoints in Urban and Industrial Geography. Reed Education.
- McCarty, H. H. & Lindberg J. B. A Preface to Economic Geography. Prentice-Hall.
- Minshull, R. The Changing Nature of Geography. Hutchinson.

Morgan, W. B. & Munton, R. J. C. Agricultural Geography. Methuen. Paperback. Morrill, R. L. The Spatial Organisation of Society. Wadsworth. Pollard, A. H. Demography: An Introduction. Pergamon. Rose, A. J. Patterns of Cities. Nelson. Taaffe, E. J. Geography. Prentice-Hall, Paperback.

Part II

An introduction to Physical Geography: Controls of landform development, cyclic and equilibrium approaches to landform study; processes and factors of soil formation; the mature soil profile; vegetation structure; factors affecting vegetation distribution; plant and soil succession and the ecosystem; particular reference to the Sydney area. The radiation budget and atmospheric circulation; climatic distribution. Laboratory classes include: weather recording and analysis of climatic data; use of maps and air photos; soil profile description. Two field tutorials.

TEXTBOOKS

Strahler, A. N. An Introduction to Physical Geography. Wiley International. Twidale, C. R. Geomorphology. Nelson. Paperback.

REFERENCE BOOKS

Bird, E. Coasts. A.N.U.P.

Bloom, A. L. The Surface of the Earth. Prentice-Hall.

Branagan, D. & Packham, G. Field Geology of New South Wales. Science. Corbett, J. R. The Living Soil. Martindale.

CSIRO. The Australian Environment. M.U.P.

Dury, G. H. & Logan, M. I. Studies in Australian Geography. Heinemann. Gentilli, J. Sun Climate and Life. Jacaranda.

Hare, F. K. The Restless Atmosphere, Hutchinson, Paperback.

Morisawa, M. Streams, Their Dynamics and Morphology. McGraw-Hill.

Odum, E. P. Ecology. Modern Biology Series. Riley, D. & Young, A. World Vegetation. C.U.P.

Taylor, G. Sydneyside Scenery. A. & R.

Trewartha, G. T. An Introduction to Climate. McGraw-Hill. Tweedie, A. D. Water and the World. Nelson. Paperback.

Twidale, C. R. & Foale, M. R. Landforms Illustrated. Nelson.

Practical classes throughout the year introduce the use of maps and diagrams, air photographs and geographical statistics. The approximate cost to students is about \$5 for field tutorials and about \$8 for the required drawing equipment and a topographic map.

Lecture, laboratory and tutorial arrangements for Geography IA are as follows:

> Hours per week for two sessions 2

> > 11

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Lectures Laboratory Tutorials

27.103 Climatology

Components of the radiation and heat balance of the earth surface as affected by differing atmospheric, soil and surface cover conditions. Factors controlling evaporation and transpiration under freely-available and restricted water supply conditions, and methods for the measurement and estimation of evapotranspiration. Characteristic patterns of energy and water exchange

for differing types of natural or man-modified land surface. Man's modification of factors affecting the local climate in rural and urban settings.

Laboratory work is directed toward developing an appreciation of the operational principles and limitations of instruments commonly used in radiation and water balance studies. An introduction is given to the practical application of energy and water balance models for evaluation of the climatic environment as related to catchment hydrology, agricultural productivity and land resource management problems.

TEXTBOOKS

Miller, D. H. A Survey Course. The Energy and Mass Budget at the Surface of the Earth. Assoc. Amer. Geog.

Sellers, W. D. Physical Climatology. Chicago U.P.

REFERENCE BOOKS

Bradley, E. F. & Denmead, O. T. eds. The Collection and Processing of Field Data. Wiley.

Chang, Jen-Hu. Climate and Agriculture. Aldine.

Chorley, R. J. ed. Water, Earth and Man. Methuen.

Chow, Ven Te. ed. Handbook of Applied Hydrology. McGraw-Hill. Gates, D. H. Energy Exchange in the Biosphere. Harper & Row.

Landsberg, H. E. ed. World Survey of Climatology. Vol. 2 General Climatology. Elsevier.

Munn, R. E. Biometeorological Methods. Academic, 1970.

27.203 Biogeography

Ecosystems, their structure and dynamics. Energy flow and biogeo-chemical cycles. Comparative photosynthetic capacity of plants. Productivity, exploitation, pollution, management and conservation of ecosystems. Man as an ecological agent. Quantitative sampling, measurement and description as an ecological agent. Quantitative sampling, measurement and description of vegetation. Spatial distribution (pattern) of individual species. Associ-ation between species. Ecology of tropical and sub-tropical regions with special reference to Australia. Floristic composition, structure and physi-ognomy of the principal vegetation formations of Australia (rain forest, woodland, shrubland, heath and grassland). Geographical affinities of component species. Environmental and biotic controls. Adaptations of plants to humid and arid conditions. Vegetation management under humid and arid conditions. Fieldwork forms an integral part of the course.

TEXTBOOKS

Kershaw, K. A. Quantitative and Dynamic Ecology. Arnold. Odum, E. P. Fundamentals of Ecology. 3rd ed. Saunders, 1971.

REFERENCE BOOKS

Anderson, R. H. The Trees of New South Wales. N.S.W. Govt. Printer.

Barnard, C. Grasses and Grasslands. Macmillan, 1964.

- Beadle, N. C. W., Evans, O. D. & Carolin, R. L. Flora of the Sydney District. Reed.
- Curtis, J. T. The Vegetation of Wisconsin: an Ordination of Plant Communities. Madison.
- Darlington, P. J. Biogeography of the Southern End of the World. Harvard U.P

Elton, C. S. The Ecology of Invasions by Animals and Plants. Methuen. Eyre, S. R. Vegetation and Soils. A World Picture. 2nd ed. Arnold, 1968.

Eyre, S. R. ed. World Vegetation Types. Macmillan, 1971. Hutchinson, J. The Families of Flowering Plants. Vols. I and II. O.U.P.

Keast, A., Crocker, R. L. & Christian, C. S. eds. Biogeography and Ecology in Australia. Monographiae Biologicae. Vol. 8. W. Junk.

- Lazarides, M. The Grasses of Central Australia. A.N.U. Press.
- Leeper, G. W. The Australian Environment. M.U.P.
- Lemee, G. Precis de Biogeographie. Masson, Paris.
- Moore, R. M. ed. Australian Grasslands. A.N.U. Press.
- Odum, E. P. Ecology. Holt, Rinehart & Winston.
- Quezel, P. La Vegetation du Sahara. Fischer-Verlag. Stuttgart.
- Richards, P. W. The Tropical Rain Forest. C.U.P.
- Schnell, R. Introduction a la Phytogeographie des Pays Tropicaux. Vols I and II. Gautheir-Villars. Paris.
- Slatyer, R. O. & Perry, R. A. eds. Arid Lands of Australia. A.N.U.P.
- Udvardy, M. D. F. Dynamic Zoogeography. Van Nostrand Reinhold.
- Walter, H. Ecology of Tropical and Subtropical Vegetation. Oliver & Boyd.
- Walter, H. Die Vegetation der Erde. In Oko-Physiologisher Betrachtung. Vols I and II. Jena.

Watts, D. Principles of Biogeography. McGraw-Hill.

27.413 Geomorphology

Fluvial processes and valley features. Hillslopes and slope mantles. Coastal, volcanic, structural and neotectonic landforms. Case studies illustrating approaches to geomorphic investigations. Classification and mapping of landforms, including airphoto interpretation. Morphometry. Laboratory study of aeolian, fluvial, beach and colluvial materials.

TEXTBOOKS

Bird, E. F. C. Coasts. A.N.U.P.

Morisawa, M. Streams: their Dynamics and Morphology. McGraw-Hill.

REFERENCE BOOKS

Allen, J. R. L. Physical Processes of Sedimentation. Unwin.

- Brunsden, D. ed. Slopes Form and Process. Inst. Brit. Geographers.
- Chorley, R. G. ed. Spatial Analysis in Geomorphology. Methuen.
- Chorley, R. J. ed. Water Earth and Man. Methuen.
- Doornkamp, J. C. & King, C. A. M. Numerical Analysis in Geomorphology. Arnold.
- Dury, G. H. ed. Rivers and River Terraces. Macmillan.
- Dury, G. H. ed. Essays in Geomorphology. Heinemann.
- Jennings, J. N. Karst. A.N.U.P.
- Jennings, J. N. & Mabbutt, J. A. ed. Landform Studies from Australia and New Guinea. A.N.U.P.
- King, C. A. M. Beaches and Coasts. Arnold.
- King, C. A. M. Techniques in Geomorphology. Arnold.
- Leopold, L. B., Wolman, M. G. & Miller, J. P. Fluvial Processes in Geomorphology. Freeman.
- Miller, V. C. Photogeology. McGraw-Hill.
- Ollier, C. Volcanoes. A.N.U.P.
- Selby, M. J. Slopes and Slope Processes. N.Z. Geog. Soc., Waikato Branch.
- Steers, A. J. ed. Introduction to Coastline Development. Macmillan,
- Thornbury, W. D. Principles of Geomorphology. Wiley.
- Twidale, C. R. Structural Landforms. A.N.U.P.
- Young, A. Slopes. Oliver & Boyd.

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27.423 Pedology

History of Pedology. Morphological, physical and chemical properties of soil. Soil forming processes; rock weathering, silicate formation. Great Soil Groups; soil classification; soil-landscape relations and periodicity. Physical and chemical aspects of soil fertility; nutrient cycles; soil microbiology. Laboratory classes upon the measurement of soil properties; soil profile description; soil survey and mapping; analysis of soil maps. Up to four days field tutorials are an essential part of the course.

TEXTBOOK

Bridges, E. M. World Soils, C.U.P. Corbett, J. R. The Living Soil. Martindale.

REFERENCE BOOKS

Alexander, M. Introduction to Soil Microbiology. Wiley.

Baver, L. D. Soil Physics. Wiley. Bear, F. E. ed. The Chemistry of the Soil. Arnold. Bear, F. E. Soil in Relation to Crop Growth. Reinhold.

Black, C. A. Soil-Plant Relationships. Wiley.

Buckman, H. O. & Brady, N. C. The Nature and Properties of Soils. Macmillan.

Clarke, G. C. & Beckitt, P. H. T. The Study of the Soil in the Field. O.U.P. Northcote, K. H. A Factual Key for the Recognition of Australian Soils. Rellim.

Piper, C. S. Soil and Plant Analysis. Adelaide U.P.

Rose, C. W. Agricultural Physics. Pergamon. Russell, E. W. Soil Conditions and Plant Growth. Longman.

Stace, G. T. et al. A Handbook of Australian Soils. Rellim.

GEOLOGY

FOR STUDENTS IN THE SCIENCE COURSE

Students may major in Geoscience in the Science course (see the regulations governing this course). This course is available on both a full-time and a part-time basis and leads to the degree of Bachelor of Science, Pass or Honours. Students majoring in Geoscience will complete the following subjects:—

First year—25.111 Geoscience I (as for 25.001 Geology I in the Applied Geology degree course).

Second year—25.112A and Geoscience IIA and 25.112B Geoscience B. Third year—25.113A Geoscience IIIA and 25.113B Geoscience IIIB.

HONOURS IN GEOLOGY

Full-time students in the Faculty of Science who have completed the two third year Geology subjects and part-time students who have completed course requirements up to the end of the sixth year and whose programme includes the two third year Geology subjects may apply to the Head of the School of Applied Geology to read for an Honours degree in Geology. Students who have majored in either Physics and Geology or Chemistry and Geology, may also be admitted to the course which would include some studies in Geophysics or Geochemistry respectively.

The Honours course consists of:-

A field assignment with appropriate work in the laboratory on material collected, the results of both the field and laboratory investigations to be presented in a graduation thesis. Advanced lectures, practical work and seminars. Short laboratory assignments on specific problems may be given.

Further details of the Honours course may be had from the Head of School.

Full-time students cover the Honours work in the fourth year of the course. Part-time students are required to commence their field thesis work at the end of the sixth year of their course and advanced laboratory assignments are done in the eighth year along with the further work necessary to complete the field thesis work.

Students seeking to do Honours in Geology must satisfy the Head of the School that they have attained a sufficient standard in their pass course work to indicate their ability to undertake geological studies at a more advanced level.

25.111 Geoscience I

Physical Geology: The structure and main surface features of the earth; geological cycle—processes of erosion, transportation, sedimentation and lithification. Surface and sub-surface water. Weathering, lakes, rivers, glacial phenomena. Vulcanism, earthquakes, orogenesis and epeirogenesis. Introductory physiography. Crystallography and Mineralogy: Introduction to crystal symmetry, systems, forms, habit, twinning. Occurrence, form and physical properties of minerals. Basic structures of silicate minerals. Mineral classification. Descriptive mineralogy. Principal rock forming minerals.

Petrology: Field occurrence, lithological characteristics and structural relationships of igneous, sedimentary and metamorphic rocks. Introduction to coal, oil and ore deposits.

Stratigraphy and Palaeontology: Basic principles of stratigraphy; introductory palaeontology. The geological time scale. The geological history of the Australian continent and more specifically that of New South Wales in introductory outline.

Practical Work: Preparation and interpretation of geological maps and sections. Map reading and use of simple geological instruments. Study of simple crystal forms and symmetry. Applied stereoscopic projection. Identification and description of common minerals and rocks in hand specimen. Recognition and description of examples of important fossil groups. Supplemented by three field tutorials, attendance at which is compulsory.

TEXTBOOKS

Black, R. M. Elements of Palaeontology. C.U.P., 1970. Bickford, M. E. et al. Geology Today. CRM Books. California, 1973.

- Rutley, F. Rutley's Elements of Mineralogy. Rev. Read, H. H. Murby, London.
- Tyrrell, G. W. The Principles of Petrology. Methuen.

25.112A Geoscience IIA

Mineralogy: Principles of optical crystallography; the construction and use of a polarizing microscope. Polymorphism; the crystal chemistry, crystallography and geological occurrence of the main groups of rock forming minerals. Description and recognition of common ore and rock forming minerals in both hand specimen and thin section.

Igneous Petrology: Occurrence, genesis and classification of the commoner igneous rocks. Crystallization of magma. Binary systems. The reaction series. Introduction to micropetrography.

Metamorphic Petrology: Principles, concepts and theories relating to the occurrence, origin and classification of metamorphic rocks. ACF and AKF diagrams. Metamorphic facies. Practical: megascopic and microscopic examination of selected metamorphic rocks. Field Work: at least one field trip to illustrate the above course.

Sedimentary Petrology: The influence of transportation, deposition and diagenesis on the composition, texture and structure of the sedimentary rocks. The classification of detrital sediments. The non-elastic sediments.

Structural Geology: Description of structures, mesoscopic-macroscopic, fractures, joints, faults, folds and their structural elements; foliation, lineation. Introduction to tectonics and plate tectonics. Practical: stereographic projection; analysis of fractures, faults folds and their structural elements; foliation, lineation, strain analysis and rotation problems. Field Work: at least one compulsory field trip to illustrate the above course.

Mineralogy

- Bloss, F. D. An Introduction to the Methods of Optical Crystallography. Holt, Rinehart & Winston, 1967.
- Heinrich, E. W. Microscopic Identification of Minerals. McGraw-Hill, 1965.

Petrology (Igneous, Metamorphic and Sedimentary)

Williams, H., Turner, F. J., & Gilbert, C. M. Petrography. Freeman, 1954. Winkler, H. G. F. Petrogenesis of Metamorphic Rocks. 2nd ed. Springer, 1967.

Structural Geology

- Spencer, E. W. Introduction to the Structure of the Earth. McGraw-Hill, 1969.
- Ragan, D. M. Structural Geology: An Introduction to Geometrical Techniques. Wiley, 1968.

25.112B Geoscience IIB

Palaeontology: Morphology and systematics of major fossil invertebrate phyla (Part 1) and their stratigraphic distribution. Practical: examination of representative fossils from each phylum.

Sedimentary Environments: Environments of deposition and sedimen-tary processes. Classification of sedimentary rocks.

Stratigraphy: Stratigraphic principles. Geosynclines and their evolution. Stratigraphy of selected provinces of Eastern Australia.

TEXTBOOKS

Palaeontology

Moore, R. C., Lalicker, C. G. & Fischer, A. G. Invertebrate Fossils. McGraw-Hill, 1952.

Sedimentary Environments and Stratigraphy

Blatt, H., Middleton, G. & Murray, R. Origin of Sedimentary Rocks. Prentice-Hall.

Dunbar, C. O. & Rodgers, J. Principles of Stratigraphy. Wiley, 1957.
 Brown, D. A., Campbell, K. S. & Crook, K. A. W. Geological Evolution of Australia and New Zealand. Pergamon, 1968.

25.113A Geoscience IIIA

Economic Geology I: Principles and theories of ore formation. Magmatic, hydrothermal, submarine exhalative deposits. Sedimentary deposits including biogenetic alluvial and residual deposits. Metallic and non-metallic economic minerals. Hand specimen and elementary mineragraphic practical work.

X-ray Crystallography: Principles of X-ray diffraction; simple application of X-ray powder cameras and diffractometers. Elementary single crystal camera theory.

Mineralogical Techniques: Optical instrumentation; Berek Compensator, Refractometers, Universal Stage. Volume Distribution analysis. Laboratory methods of mineral separation.

Igneous Petrology: Magma types and differentiation trends. Ternary systems. Effects of load pressure and water vapour pressure on phase equilibria. Micro-petrography of a wide range of igneous rocks.

Tectonics: The geophysical, sedimentological, petrological and structural geological aspects of global geotectonics.

TEXTBOOKS

Economic Geology

Park, C. F., & MacDiarmid, R. A. Ore Deposits. Freeman, 1964. Stanton, R. Ore Petrology. McGraw-Hill, 1972.

X-ray Crystallography

Azaroff, L. V. & Donahue, R. J. Laboratory Experiments in X-ray Crystallography, McGraw-Hill, 1969.

Mineralogical Techniques

As for 25.112B Mineralogy, plus: Zussman, J. ed. Physical Methods in Determinative Mineralogy. Academic, 1967.

Igneous Petrology

Deer, W. A., Howie, R. A. & Zussman, J. Rock-forming Minerals. Long-

mans, 1966. Turner, F. J., & Verhoogen, J. Igneous and Metamorphic Petrology. McGraw-Hill, 1960.

Tectonics

Gaskell, T. F. Physics of the Earth. Thames & Hudson, 1970.

25.113B Geoscience IIIB

Stratigraphy: Advanced stratigraphic principles and techniques. Geochronology. Geosynclines and plate tectonics. Sedimentational and tectonic history of selected provinces in Australia. The theory of continental drift and its stratigraphic implications.

Biostratigraphy: Biostratigraphy and the use of selected fossil groups in stratigraphic correlation.

Geophysics: An introduction to the physics, shape, structure, constitution and dynamics of the earth: seismology, gravity, geodesy, geothermy, geomagnetism, palaeomagnetism. An introduction to main methods of geophysical exploration.

Palaeontology: Morphology and systematics of major fossil inverte-brate phyla and their stratigraphic distribution. Palaeobotany.

Palaeoecology: Elements of palaeoecology and practical applications.

Vertebrate Palaeontology: An introduction to evolution of vertebrates.

Oceanography: Dynamic properties of the oceanic water-masses. Physics and chemistry of sea water. Submarine geology and cartography. Recent sedimentation and its correlation with terrestrial stratigraphy. Sediments of organic origin. Oceanic materials of economic importance.

Field Work: To be held during the year. Includes a geological survey camp held in session 1 and ten days of field instruction. Attendance is compulsory.

TEXTBOOKS

Stratigraphy and Biostratigraphy

As for 25.112B Geoscience IIB, plus

Berry, W. B. N. Growth of a Prehistoric Time Scale Based on Organic Evolution. Freeman, 1968.

Krumbein, W. C. & Sloss, L. L. Stratigraphy and Sedimentation. 2nd ed. Freeman, 1963.
Lahee, F. H. Field Geology. McGraw-Hill, 1952.

Geophysics

Bott, M. H. P. The Interior of the Earth. Arnold, 1971. Garland, G. D. The Earth's Shape and Gravity. Pergamon, 1964. Howell, B. Introduction to Geophysics. McGraw-Hill, 1959. Parasnis, D. S. Principles of Applied Geophysics. Chapman & Hall, 1972. Stacey, F. P. Physics of the Earth. Wiley, 1969.

Palaeontology As for 25.112A Palaeontology.

Palaeoecology Ager, D. V. Principles of Palaeoecology. McGraw-Hill, 1963.

Vertebrate Palaeontology Colbert, E. H. Evolution of the Vertebrates. Wiley. Von Koenigswald, G. H. R. The Evolution of Man. Michigan U.P., 1962.

Oceanography Pickard, G. L. Descriptive Physical Oceanography. Pergamon, 1964.

25.014 Geology IV (Honours)

TEXTBOOKS

Mining and Petroleum Geology

Lawrence, L. J. ed. Exploration and Mining Geology. Aust. Inst. Min. Met., Melbourne, 1965.

Geophysics

Dobrin, N. B. Introduction to Geophysical Prospecting. McGraw-Hill, 1960.

Grant, F. S. & West, G. F. Interpretation Theory in Applied Geophysics. McGraw-Hill, 1964.

Parasnis, D. S. Principles of Applied Geophysics. Methuen, 1962.

25.603A Geological Oceanography

The form and nature of ocean basins; the origin, transport and distribution of suspended matter, igneous and sedimentary rocks of the ocean floor and their distribution; the significance of oceanic igneous rocks, palaeontology, stratigraphical history and correlation of marine sedimentary rocks; magnetism and palaeomagnetism, tectonics of ocean basins.

TEXTBOOK

Gross, M. G. Oceanography. Prentice-Hall, 1972.

REFERENCE BOOKS

Emery, K. O. The Sea off Southern California. Wiley, 1960.

Ericson, D. B. & Wollin, G. The Deep and the Past. Knopf, 1964.

Gass, R. Understanding the Earth. Open University, 1971.

Loeblich, A R. & Tappan, H. Foraminiferida. Vol. C of the Treatise on Invertebrate Palaeontology. Geological Society of America, 1964.

Menard, H. Marine Geology of the Pacific. McGraw-Hill, 1964.

Reports of the Deep Sea Drilling Projects. Vols 1-30. National Science Foundation, 1970-.

SCHOOL OF HISTORY AND PHILOSOPHY OF SCIENCE

The division in educational curricula between science and the humanities obscures the fact that throughout history the natural sciences have been an integral component of general intellectual and cultural development. Until the nineteenth century, for example, the term "philosophy" included science. The principal aims of the History and Philosophy of Science courses in the Faculty of Science are to introduce students to the study of the history of science, to relate the sciences to the cultural environment which have given rise to them, to consider the social history of science, to look at analyses, changing through time, of its conceptual foundations, and to study the impact that the success of science has had on man's understanding of his own nature, and of the nature of knowledge.

In 1974, three level II units (62.012 The Origins of Modern Science, 62.022 The Social History of the Scientific Movement, 62.032 Philosophy of Science) are offered, two in Session 1, one in Session 2: the prerequisites for all three units are a pass in at least *two* of the folowing: 1.001, 17.011 and 17.021, 2.001, 10.001, 25.111, 1.011, 10.011, 10.021, 27.031, 12.001. Each unit has unit value 1, and requires six hours a week for one session, made up of three lectures, one tutorial, and two hours a week of library work.

Four units of History and Philosophy of Science II will be introduced at third-year level in 1975, and an Honours year in 1976. Students, intending to major in History and Philosophy of Science should take all three level II units. There is a small, but growing and distinctive range of career opportunities for students with a good science background combined with History and Philosophy of Science, especially in the field of science writing and science reporting, in the ABC Science Unit, the CSIRO Publications Branch, in industry and the newspapers. The programme is well suited for intending science teachers. A major in History and Philosophy of Science in the Faculty of Science administration arise in industry and the Public Service; this field is underdeveloped in Australia, but is unlikely to remain so. A major in History and Philosophy of Science is also a good foundation for a DipLib (for intending science librarians).

62.012 The Origins of Modern Science

An introductory course dealing with the Scientific Revolution of the seventeenth century, the philosophical issues beeng discussed in their historical context. The course will survey the major achievements of science during the period, the consequences of the Copernican Revolution, the construction of dynamics from Galileo to Newton, Harvey's physiology, the development of theories of light and of concepts such as atmospheric pressure. The cultural and intellectual background of these achievements and their effects on European thought will be discussed. PRELIMINARY READING

Butterfield, H. The Origins of Modern Science. Bell.

TEXTBOOKS

Hall, A. R. From Galileo to Newton 1603-1720. Collins.

Hall, M. B. ed. Nature and Nature's Laws-Documents of the Scientific Revolution. Harper.

Westfall, R. S. The Construction of Modern Science, Wiley.

REFERENCE BOOKS

- Bacon, F. Anderson, F. H. ed. The New Organon and Related Writings. Bobbs-Merrill.
- Bacon, F. Jones, R. F. ed. Essays, Advancement of Learning, New Atlantis and Other Pieces. Odyssey.
- Barfield, O. Saving the Appearances. Harbinger.

Bell, A. E. Newtonian Science. Arnold.

Blake, R. M., Ducasse, C. J. & Madden, E. H. eds. Theories of Scientific Method—The Renaissance through the Nineteenth Century. Washington U.P.

Boas, M. The Scientific Renaissance, 1450-1630. Collins.

Buchdahl, G. The Image of Locke and Newton in the Age of Reason. Sheed & Ward.

Burtt, E. A. The Metaphysical Foundations of Modern Science. Routledge. Clark, G. N. The Seventeenth Century. O.U.P. Collingwood, R. G. The Idea of Nature. O.U.P.

Copleston, F. A History of Philosophy. Vols IV & V. Image, Doubleday.

Descartes, R. Discourse on Method and Other Writings. Penguin.

Descartes, R. Philosophical Writings. Nelson.

Dijksterhuis, E. J. The Mechanisation of the World Picture. O.U.P.

Dugas, R. Mechanics in the Seventeenth Century. Griffon.

Euclid. The Elements. Everyman.

Galilei, G. Dialogue Concerning the Two Chief World Systems. U. of Calif.

Galilei, G. Dialogues Concerning Two New Sciences. Dover.

Galilei, G. Drake, S. ed. Discoveries and Opinions of Galileo. Doubleday. Gillispie, C. C. The Edge of Objectivity. Princeton U.P. Hall, A. R. The Scientific Revolution 1500-1800. Collins.

- Hall, M. B. Robert Boyle on Natural Philosophy. An Essay with Selections from his Writings. Indiana U.P.
- Harvey, W. On the Motion of the Heart and Blood. Everyman.
- Hazard, P. The European Mind 1680-1715. Pelican.

Hesse, M. B. Forces and Fields. Nelson.

Hobbes, T. Leviathan. Pelican. Jones, R. F. The Seventeeth Century. Stanford U.P. Kargon, R. H. Atomism in England from Hariot to Newton. O.U.P.

Kearney, H. Science and Change 1500-1700. World Univ. Lib. Keeling, S. V. Descartes. O.U.P.

Koyré, A. From the Closed World to the Infinite Universe. Johns Hopkins U.P.

Koyré, A. Newtonian Studies. Chicago U.P.

Kuhn, T. S. The Copernican Revolution. Random House. Leibniz, G. W. F. Wiener, P. P. ed. Leibniz Selections. Scribner. Mason, S. F. A History of the Sciences. Collier.

- Ornstein, M. The Role of Scientific Societies in the Seventeenth Century. Chicago U.P.
- Purver, M. The Royal Society: Concept and Realization. Routledge.

Randall, U. H. The Making of the Modern Mind. Houghton Mifflin. Sabra, I. Theories of Light from Descartes to Newton. Oldbourne.

- Taylor, H. O. Philosophy and Science in the Sixteenth Century. Collier.

Thayer, H. Newton's Philosophy of Nature. Hafner.

West, U. F. The Great Intellectual Revolution. Citadel.

Westfall, R. S. Science and Religion in Seventeenth Century England. Newhaven.

Willey, B. The Seventeenth Century Background. Pelican.

Wolf, A. A History of Science, Technology and Philosophy in the Sixteenth and Seventeenth Centuries. Harper.

62.022 The Social History of the Scientific Movement

An account of the growth of the scientific movement, from the early seventeenth to the twentieth century, in relation to: (a) its social and cultural environment and the effects of social structures and social changes upon it; (b) its internal organization; (c) its effects upon society, intellectually and otherwise.

The course deals with such topics as: the different national contexts of the scientific movement; its social composition at various times; its relations with the state in different countries at different times, with the universities and other teaching institutions, and with the professions of medicine and engineering; the communications system in science; the nature and function of scientific societies and academies; the institutionalization and professionalization of science.

TEXTBOOKS

No suitable textbooks are available. Reading lists, selections from primary sources, and other material will be issued during the course.

REFERENCE BOOKS

- Argles, M. South Kensington to Robbins: An Account of English Technical and Scientific Education since 1851. Longman.
- Artz, F. B. The Development of Technical Education in France. Society for the History of Technology, Cleveland.
- Ashby, E. Technology and the Academics. Macmillan.
- Barber, B. Science and the Social Order. Collier.

- Ben-David, J. The Scientist's Role in Society. Prentice-Hall. Buchdahl, G. The Image of Newton and Locke in the Age of Reason. Sheed & Ward.
- Clapp, M. The Modern University. Cornell U.P.
- Farrington, B. Francis Bacon, Philosopher of Industrial Science. Collier.

Haberer, J. Politics and the Community of Science. Van Nostrand.

- Hahn, R. The Anatomy of a Scientific Institution: The Paris Academy of Sciences, 1666-1803. California U.P.
- Haines, R. F. German Influence upon English Education and Science, 1800-1866. Connecticut College.
- Hazard, P. The European Mind, 1680-1715. Penguin.

Hazard, P. European Thought in the Eighteenth Century. Penguin.

Jones, R. F. Ancients and Moderns: A Study of the Rise of the Scientific Movement in Seventeenth-Century England. California U.P.

- Lilley, S. Essays on the Social History of Science. Centaurus. Marsak, L. M. The Rise of Science in Relation to Society. Macmillan. Merz, J. T. A History of European Thought in the Nineteenth Century. Vol 1. Dover.
- Ornstein, M. The Role of Scientific Societies in the Seventeenth Century. Chicago U.P.
- Paulsen, F. The German Universities: Their Character and Historical Development. Macmillan.

Price, D. J. Little Science, Big Science. Columbia U.P.

Roderick, G. W. The Emergence of a Scientific Society in England, 1800-1965. Macmillan.

Rose, H. & Rose, S. Science and Society, Penguin.

Stimson, D. Scientists and Amateurs; A History of the Royal Society, Schuman.

Van Tassel, D. D. & Hall, M. G. Science and Society in the United States. Dorsey.

Vucinich, A. Science in Russian Culture, 2 vols, Owen,

62.032 Philosophy of Science

A general introduction to some of the more fundamental problems of the philosophy of science. Special emphasis is placed on the nature, composition, and structure of scientific theories, and of the relations between theoretical statements and observational data.

A selection of such topics as the following is discussed: the logic of theory construction; the logical structure of theories; the status of scientific laws; the roles of regulative principles, correspondence rules, and methodo-logical directives; the function of models and analogies; the nature of scientific explanation; the status of theoretical entities; the principles of theory establishment and rejection; the axiology and apologetics of science; the dynamics of scientific change, including the structure of scientific revolutions.

Examples selected from the history of the sciences are employed in order to illustrate the philosophical issues examined.

TEXTBOOKS

Basson, A. H. & O'Connor, D. J. Introduction to Symbolic Logic. University Tutorial Press.

Braithwaite, R. B. Scientific Explanation. Harper Torchbook. Kuhn, T. S. The Structure of Scientific Revolutions. 2nd ed. Chicago U.P. Popper, K. R. The Logic of Scientific Discovery. Harper Torchbook.

REFERENCE BOOKS

Burtt, E. A. Metaphysical Foundations of Modern Physical Science, Anchor. Campbell, N. R. Foundations of Science. Dover.

Colodny, R. G. ed. Beyond the Edge of Certainty. Prentice-Hall.

Duhem, P. The Aim and Structure of Physical Theory. Prince-rhan. Hansen, N. R. Patterns of Discovery. C.U.P. Hempel, C. G. Philosophy of Natural Science. Prentice-Hall.

Hesse, M. B. Forces and Fields. Nelson.

Hesse, M. B. Models and Analogies in Science. Sheed & Ward.

Kant, I. Prolegomena to Any Future Metaphysics. Manchester U.P. or Library of Liberal Arts.

Körner, S. The Philosophy of Mathematics. Hutchinson. Medawar, P. B. The Art of the Soluble. Penguin.

Medawar, P. B. Induction and Intuition in Scientific Thought. Methuen. Nagel, E. The Structure of Science. Routledge.

Nash, L. K. The Nature of the Natural Sciences. Little, Brown, & Co.

Popper, K. R. Conjectures and Refutations. Routledge.

Reichenbach, H. The Rise of Scientific Philosophy. California U.P.

Theobold, D.W. An Introduction to the Philosophy of Science. Methuen.

Toulmin, S. Foresight and Understanding. Harper Torchbook. Toulmin, S. The Philosophy of Science. Harper Torchbook.

Whitehead, A. N. Science and the Modern World. Mentor.

62.144 History and Philosophy of Biology

Available as an advanced elective in General Studies to the Fourth Year Honours students of the Faculty of Biological Sciences or with the permission of the Head of the School of History and Philosophy of Science.

Surveys of the history of biological ideas and conceptual developments in later nineteenth and in twentieth century biology, leading to a consideration of some specific issues relevant to the practice of contemporary biology, for example: explanation in the biological sciences; models, metaphors and paradigms; parts and wholes, levels of description, and reductionism; the nature of taxonomy, and the species problem; conceptual relations within the behavioural sciences; the attempt to found an ethic on evolutionary theory and primate ethology.

TEXTBOOKS

Coleman, W. Biology in the Nineteenth Century. Wiley. Kuhn, T. S. The Structure of Scientific Revolutions. Chicago U.P. Monod, J. Chance and Necessity. Collins.

REFERENCE BOOKS

- Agar, W. E. A Contribution to the Theory of the Living Organism. Melbourne U.P.
- Beckner, M. The Biological Way of Thought. California U.P.
- Bunge, M. Metascientific Oueries. Thomas.
- Cohen, R. S. & Wartofsky, M. W. eds. Boston Studies in the Philosophy of Science. Vol II. Humanities Press.
- Cole, F. J. Early Theories of Sexual Generation. Clarendon.

Crowson, R. A. Classification and Biology. Heinemann.

Dunn, L. C. A Short History of Genetics. McGraw-Hill.

Elsasser, W. M. Atom and Organism. Princeton U.P.

Gasking, E. B. Investigations into Generation 1651-1828. Hutchinson.

- Gasking, E. B. The Rise of Experimental Biology. Random House.
- Grene, M. Approaches to a Philosophical Biology. Basic Books.
- Grene, M. ed. Interpretations of Life and Mind. Routledge.

Hughes, A. A History of Cytology. Abelard-Schuman.

- Koestler, A. & Smythies, J. R. eds. Beyond Reductionism. Hutchinson.
- Lakatos, I. & Musgrave, A. eds. Criticism and the Growth of Knowledge. C.U.P.

Minnesota Studies in the Philosophy of Science. Vols I, II & III.

Feigl, H. & Scriven, M. eds.

Vol I Feigl, H. & Scriven, M. eas. Vol II Feigl, H., Scriven, M. & Maxwell G. eds.

Vol III Feigl, H. & Maxwell, G. eds.

- Minnesota U.P.
- Nagel, E. The Structure of Science. Routledge.
- Needham, J. A History of Embryology. C.U.P.
- Olby, R. C. Origins of Mendelism. Constable.
- Pantin, C. F. A. The Relations between the Sciences. C.U.P.
- Rashevsky, N. Mathematical Biophysics. Vol II. Dover.
- Sinnott, E. W. The Problem of Organic Form. Yale U.P.
- Whyte, L. L., Wilson, A. G. & Wilson, D. eds. Hierarchical Structures. Elsevier.
- Woodger, J. H. Biological Principles. Routledge.

FOR STUDENTS IN THE SCIENCE COURSE

The course in population genetics theory is relatively new, having been offered for the first time in 1972. This course is a Level III unit and may be taken by students in their third year. It is designed for students who intend to specialize in population genetics or in a field in which population genetics is applied. It is available as a day course only.

Approximately one-third of the lecture time (2 hours) is reserved for mathematics and statistics. The tutorial time (1 hour) is used to relate the models covered in the main part of the course with descriptive treatments of population processes covered in other courses. Students are expected to prepare material for and take an active part in tutorials. Laboratory time is 2 hours.

78.201 Population Genetics Theory

Models of genetic systems and growth of populations, with essential mathematical and statistical theory; illustrated by examples from human genetics. Limitations of models.

Models of population growth in discrete and continuous time with nonoverlapping and overlapping generations. An extension of the Hardy-Weinberg principle to finite populations and several loci. The concept of inbreeding, calculation of coefficients of consanguinity, effects of inbreeding, effective population number. Fisher's Fundamental Theorem of Natural Selection. Advanced treatment of factors maintaining gene frequency equilibria in populations, including balance between mutation and selection, heterozygotic advantage, and genetic loads. Effects of finite population number, including random gene frequency drift.

TEXTBOOK

Crow, J. F. & Kimura, M. An Introduction to Population Genetics Theory. Harper & Row, 1970.

REFERENCE BOOKS

- Cavalli-Sforza, L. L. & Bodmer, W. F. The Genetics of Human Populations. Freeman, 1971.
- Fraser, A. & Burnell, D. Computer Models in Genetics. McGraw-Hill, 1970.
- Wallace, B. Genetic Load—Its Biological and Conceptual Aspects. Prentice-Hall, 1970.

Traditionally, mathematics is classified into Pure Mathematics, Applied Mathematics and Statistics. The classification is not a very sharp one and there is considerable overlap and interaction between the three branches.

The Pure Mathematician is concerned with the study of mathematics in itself, striving to solve new problems, to attain ever greater insight into the relations between different parts of mathematics, and thus to render the whole structure of mathematics more complete, more transparent, and more unified. Possible applications of his subject to problems in science or industry are not his primary concern, but they interest him in so far as they provide stimuli for the growth of new mathematical theories. The main avenues of employment for a Pure Mathematician are the universities, the teaching services and some research establishments such as the CSIRO.

Applied Mathematics is concerned with the understanding of scientific phenomena by the construction, analysis, and interpretation of mathematical models. Problems may originate not only in the physical and engineering sciences, but also in the social, computing, biological, economic and managament sciences. In the Department of Applied Mathematics at this University there are strong interests in modern theoretical physics, with an emphasis on quantum mechanics, nuclear theory and statistical mechanics; and in environmental mechanics, including theoretical oceanography and related subjects. However, the courses offered include all branches of Applied Mathematics, and specialization need not occur until the honours year.

Statistics is concerned with the evaluation of factual material as a basis for inference and decision making. Its mathematical foundation in the theory of probability, and statistical theory is applicable in various fields where probabilistic models are used to describe the observational results. The Department has strong interests in the areas of inference, sequential analysis, design of experiments, multivariate analysis, stochastic processes and geometric probability.

In the past the employment of mathematicians in Australian industry and commerce was rather uncommon. Over the last decade there has been a change, corresponding to the general recognition of the desirability of making quantitative what was previously merely qualitative. One important factor has been the introduction of high-speed computers, making possible the detailed mathematical analysis of complex practical situations in a way which would not have been possible without them. Courses in mathematics include training in programming for digital computers and in numerical analysis.

THE COURSES AND SUBJECTS PROVIDED BY THE SCHOOL

The School of Mathematics provides courses at the Pass and Honours levels in Pure Mathematics, Applied Mathematics and Theory of Statistics. Full details of the subjects and their relations with other subjects in the Science Course appear in the University Calendar. Any student who feels that he does not understand the situation should consult one of the enrolment officers of the School.

FIRST YEAR MATHEMATICS

10.001 Mathematics I

This is the standard course and is generally selected by the majority of students in the Faculties of Science, Biological Sciences, Engineering and Applied Science who intend to pursue further studies in mathematics, physics or chemistry.

For entry into 10.001 Mathematics I, students are required to have passed H.S.C. Mathematics at Level 2F or higher; or mathematics at Level 2S provided that the student's performance in mathematics and his general level of attainment are at standards acceptable to the Professorial Board. Students at the latter level are advised to undertake a bridging course before the beginning of lectures.

10.011 Higher Mathematics I (Day course only)

Covers all the material in 10.001 Mathematics I, plus other topics, at greater depth and sophistication. Though this course starts where Level 1 of the Higher School Certificate ends, some Level 2F students with ability might find it within their capabilities.

While it is expected that students aiming at the honours level in mathematics will take this course, it would be equally valuable for any mathematically able student whose course requires a considerable amount of mathematics.

10.021 Mathematics IT

This course provides for students who do not intend studying mathematics beyond first year but whose other studies require some knowledge of basic mathematical ideas and techniques. It is particularly designed to meet the needs of such students in Biological Sciences, Commerce, Optometry, Applied Psychology and Wool and Pastoral Sciences.

The course assumes that the student has a mathematical background up to H.S.C. Level 2S Mathematics only; entry is open to all with a pass at this level or better. However, students who select this course should weigh seriously the implications of their choice because *no further mathematical units are normally available*.

HIGHER LEVEL MATHEMATICS

Many subjects in the School are offered at two levels. The higher level caters for students with superior mathematical ability. Where both levels are offered, the highest grade awarded in the ordinary level is Credit, except in exceptional cases.

MATHEMATICS MAJORS IN THE FACULTY OF SCIENCE

Any student who completes at least four level III units in the School of Mathematics is regarded as having majored in Mathematics as part of his BSc degree. Students should consider the merits of combining courses in Pure Mathematics, Applied Mathematics, Statistics and Computer Science in accordance with their future interests. Senior members of staff in the School of Mathematics are available for consultation by students who wish to discuss their courses. If students wish to specialize and major in Pure Mathematics, Applied Mathematics or Theory of Statistics, the following minimum courses are suggested.

(i) Pure Mathematics Majors

In order to major in Pure Mathematics at the ordinary level, a student should pass in seven at least of the following units:

10.211A Applied Mathematics II;

10.111A, 10.111B, 10.111C Pure Mathematics II;

10.112A, 10.112B, 10.112C, 10.112D, 10.112E Pure Mathematics III. In all cases the student must pass complementary units or subjects in accordance with Faculty rules.

(ii) Applied Mathematics Majors

In second year the student should take the level II units 10.211A, 10.211B and 10.211C Applied Mathematics, together with the units 10.111A and 10.111B Pure Mathematics. In third year the student should take the level III units 10.212A, 10.212B, 10.212D and 10.212L Applied Mathematics. Complementary units should be chosen in accordance with Faculty rules.

(iii) Theory of Statistics Majors

In second year the student should take 10.311 Theory of Statistics II; in addition, since he will be expected to take some level III Mathematics units in third year, he should take the level II units 10.111A and 10.111B Pure Mathematics and 10.211A Applied Mathematics.

In third year he should take the level III units 10.312A, 10.312B, 10.312C and 10.312D Theory of Statistics, together with at least two level III Mathematics units (Pure or Applied).

In each year he should also take complementary units in accordance with Faculty rules; the recommended complementary units are 6.601A Introduction to Computing in second year, and one or other of 6.602C Computer Applications or 10.212L Optimization Techniques together with further level III Mathematics units in third year.

HONOURS COURSES IN MATHEMATICS

There are three separate fourth year honours courses: Pure Mathematics, Applied Mathematics and the Theory of Statistics. The four-year course for an honours degree is intended primarily for professional pure mathematicians, statisticians and mathematical physicists or applied mathematicians, but will prove of interest also to intending specialists in fields such as theoretical physics, engineering and all other theoretical sciences. The minimum requirements for each honours course are given below but students seeking an honours degree in mathematics are advised to choose units or courses in mathematics according to their individual interests in consultation with senior members of staff of the School.

(i) Honours Course in Pure Mathematics

In the Faculty of Science in second year the student should attempt 10.121A, 10.121B, 10.121C Higher Pure Mathematics II, and 10.221A Higher Applied Mathematics II. In third year the student should attempt 10.122A, 10.122B, 10.122C and 10.122E Higher Pure Mathematics III.

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In all cases complementary units or subjects must be chosen in accordance with Faculty rules.

Permission to enter fourth year in Pure Mathematics is granted only on the recommendation of one of the Professors of Pure Mathematics. Such recommendation is not usually granted unless the student's record contains a satisfactory number of graded passes.

(ii) Honours Course in Applied Mathematics

In the Faculty of Science in second year the student should attempt 10.221A, 10.221B and 10.221C Higher Applied Mathematics II, 10.121A and 10.121B Higher Pure Mathematics II. The student should also regard the inclusion of 1.122C Thermodynamics and Mechanics as desirable. Students intending to take mathematical physics options in later years of Applied Mathematics are advised to take at least one further level II Higher Physics unit.

In third year, students should attempt 10.222B, 10.222C, 10.222F and 10.222D Higher Applied Mathematics III; and at least one level III higher Pure Mathematics unit.

Students wishing to enter the third year of the honours course are advised to consult members of staff of the Department of Applied Mathematics before enrolment. Permission to enter fourth year is granted on the permission of a Professor of Applied Mathematics.

(iii) Honours Course in Statistics

In the Faculty of Science, in second year the student should take 10.321 Higher Theory of Statistics II, 10.111A and 10.111B Pure Mathematics II, or 10.121A and 10.121B Higher Pure Mathematics II, and 10.211A Applied Mathematics II or 10.221A Higher Applied Mathematics II; the student is strongly recommended to take also 6.601A Introduction to Computing. In third year he should take the level III Units 10.322A, 10.322B, 10.322C and 10.322D Higher Theory of Statistics, together with at least three level III Mathematics units (Pure or Applied).

In all cases complementary units or subjects must be chosen in accordance with Faculty rules.

Students wishing to attempt Third Year honours courses are advised to discuss their courses with a Professor of the Department of Statistics. Permission to enter the Fourth Year course in the Theory of Statistics is granted on the recommendation of a Professor of Statistics. Such permission will not usually be granted unless the applicant has obtained a graded pass in 10.322 Higher Theory of Statistics III and the student's record contains a satisfactory number of graded passes.

MATHEMATICS AS A SUBSIDIARY SUBJECT

The School also provides the sequence of two Units 10.031 and 10.032, at the second and third levels respectively, for students in the Faculty of Science who are mainly interested in the chemical and biological sciences. These courses offer an introduction to mathematical techniques for scientists and engineers. It should be noted, however, that these two units cannot be counted together with any second level or third level units in Pure and Applied Mathematics, except when 10.412 is taken as part of a marine science major.

There is also a single unit in Statistics, 10.331, which is recommended for those scientists who wish to have some knowledge of Statistics but who would not normally wish to proceed to further courses in this subject.

For both the above courses the entry qualification is a pass in 10.001 Mathematics I, but in appropriate cases students who have passed in 10.021 Terminating Mathematics I at a satisfactory level may be given permission to enrol.

SCHOOLTEACHERS

There is no doubt that in order to be well qualified as a high school teacher of mathematics it is desirable that the student should have completed a sequence of Mathematics units similar to that outlined under Pure Mathematics majors at Pass level and that other units should be selected from the Theory of Statistics or Applied Mathematics sequences. This extra work will broaden the prospective teacher's outlook and will certainly improve his teaching.

Those who feel that they may be interested in proceeding to a higher degree after graduation are advised to attempt some of the courses at the higher level.

Pure Mathematics level II, Unit C and Pure Mathematics level III, Unit D, should be of interest to schoolteachers.

STUDENTS WITH LOW MATHEMATICAL QUALIFICATIONS

The School of Mathematics arranges a Bridging Course in Mathematics for those students intending to enrol in Mathematics I and who have inadequate mathematical background. The Bridging Course covers the gap between 2S and 2F Mathematics and is a very useful refresher course generally. The course will be held at the University during the period January to February 1974.

Attention is also directed to the Calculus Bridging Course given over the University of N.S.W. Radio Station VL2UV. The radio course explains the ideas of Calculus and assumes no previous knowledge of the subject.

STUDENTS TRANSFERRING FROM OTHER COURSES

In some cases the mathematical subjects of the Science Course differ quite considerably from the mathematics taught to students following other courses (e.g., Engineering). Students transferring to the Science Course and wishing to obtain credit for work done in previous courses should make application through the Admissions Office as early as possible. The staff of the School will advise students in such cases but this does not relieve the student of the responsibility of making an early application through the correct channels.

SUBJECTS SUBSIDIARY TO MATHEMATICS

As mentioned above, a student wishing to major in Mathematics must pass other Science subjects in accordance with Science Course regulations. In this connection it is worth noting that the Applied Mathematics Course has a considerable content of mathematical physics and there is no doubt that Physics I and/or Physics II would assist the student.

Mathematics Prizes

There are prizes available for certain courses in the School of Mathematics. They are open to all Kensington students proceeding to an undergraduate degree or diploma but will not be awarded if there is no candidate of sufficient merit. An award of \$25 and a suitably inscribed certificate are available in the following subjects: Higher Mathematics I, Higher Pure Mathematics II, Higher Applied Mathematics II, Higher Pure Mathematics III, Higher Applied Mathematics III.

Similarly, there are prizes of up to \$40 available in Theory of Statistics subjects.

MATHEMATICS

10.001 Mathematics I

Calculus, analysis, analytic geometry, linear algebra, an introduction to abstract algebra, elementary computing.

PRELIMINARY READING LIST

Allendoerfer, C. B. & Oakley, C. O. Principles of Mathematics. McGraw-Hill.

Bell, E. T. Men of Mathematics. 2 vols. Pelican.

Courant, R. & Robbins, H. What is Mathematics. O.U.P.

Polya, G. How to Solve It. Doubleday Anchor.

Sawyer, W. W. A Concrete Approach to Abstract Algebra. Freeman.

Sawyer, W. W. Prelude to Mathematics. Pelican.

TEXTBOOKS

Blatt, J. M. Introduction to Fortran IV Programming. Prentice-Hall.

Shields, P. C. Elementary Linear Algebra. Worth.

Thomas, G. B. Calculus and Analytic Geometry. 4th ed. Addison-Wesley.

REFERENCE BOOKS

Blatt, J. M. Basic Fortran IV Programming (IBM/360 Version). Computer Systems (Aust.).

Campbell, H. F. Matrices with Applications. Appleton-Century-Crofts.

Kaplan, W. & Lewis, D. J. Calculus and Linear Algebra. Vols. 1 & 2. Wiley.

Kelly, G. M. Introduction to Linear Algebra and Vector Geometry. Reed Education, Sydney, 1971.

Lange, I. H. Elementary Linear Algebra. Wiley.

Pedoe, D. A Geometric Introduction to Linear Algebra. Wiley.

Purcell, E. J. Calculus with Analytic Geometry. Appleton-Century-Crofts.

Smith, W. K. Limits and Continuity. Collier-Macmillan.

Tetra, B. C. Basic Linear Algebra. Harper & Row.

Zelinsky, D. A First Course in Linear Algebra. Academic.

10.011 Higher Mathematics I

Calculus, analytic geometry, linear algebra, an introduction to abstract algebra, elementary computing.

PRELIMINARY READING LIST

As for 10.001 plus: Arnold, B. H. Intuitive Concepts in Elementary Topology. Prentice-Hall. David, F. N. Games, Gods and Gambling. Griffin. Felix, L. The Modern Aspect of Mathematics. Science. Huff, D. How to Lie with Statistics. Gollancz. Reid, C. From Zero to Infinity. Routledge.

TEXTBOOKS

Blatt, J. M. Introduction to Fortran IV Programming. Prentice-Hall. Fagg, S. V. Differential Equations. English Universities P. Shields, P. C. Elementary Linear Algebra. Worth. Spivak, M. Calculus. Benjamin.

REFERENCE BOOKS

As for 10.001 plus:

Abraham, R. Linear and Multilinear Algebra. Benjamin.

Brauer, F. & Nohel, J. Ordinary Differential Equations. Benjamin.

Burkhill, J. C. A First Course in Mathematical Analysis. C.U.P.

Crowell, R. H. & Williamson, R. E. Calculus of Vector Functions. Prentice-Hall.

Hochstadt, H. Differential Equations. Holt, Rinehart & Winston.

Lang, S. Linear Algebra. Addison-Wesley.

Murdoch, D. C. Linear Algebra for Undergraduates. Wiley.

Spivak, M. Calculus on Manifolds. Benjamin.

10.021 Mathematics IT

Calculus, analysis, analytic geometry, algebra, probability theory, elementary computing.

TEXTBOOKS

Blatt, J. M. Introduction to Fortran IV Programming. Prentice-Hall.

Greening, M. G. First Year General Mathematics. N.S.W.U.P.

Youse, B. K. & Stalnaker, A. W. Calculus for the Social and Natural Sciences. International Textbook Co.

REFERENCE BOOKS

Allendoerfer, C. B. & Oakley, C. O. Fundamentals of College Algebra. McGraw-Hill.

Bates, G. E. Probability. Addison-Wesley.

Burford, R. L. Introduction to Finite Probability. Merrill.

Christian, R. C. Logic and Sets. Blaisdell.

Fine, N. J. Introduction to Modern Mathematics. Rand McNally & Co.

- Hoyt, J. P. A Brief Introduction to Probability Theory. International Text Book Co.
- Johnson, W. G. & Zaccaro, L. N. Modern Introductory Mathematics. McGraw-Hill.

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10.031 Mathematics (one level II unit)*

Differential equations, use of Laplace transforms, solutions by series; partial differential equations and their solution for selected physical problems, use of Fourier series; multiple integrals, matrices and their application to theory of linear equations, eigenvalues; introduction to numerical methods.

TEXTBOOK

Keane, A. & Senior, S. A. Mathematical Methods. Science.

REFERENCE BOOKS

Grove, W. E. Brief Numerical Methods. Prentice-Hall.

Hildebrand, F. B. Advanced Calculus for Applications. Prentice-Hall.

Pipes, L. A. & Harvill, L. R. Applied Mathematics for Engineers and Physicists. 3rd ed. McGraw-Hill.

Spiegel, M. R. Advanced Mathematics for Engineers and Scientists. McGraw-Hill,

Wylie, C. R. Advanced Engineering Mathematics. 3rd ed. McGraw-Hill.

10.032 Mathematics (one level III unit)*

Vector Calculus; special functions; convolution theorem and applications; complex variable theory; Fourier integrals; Laplace transforms with application to ordinary and partial differential equations.

TEXTBOOK

Kreyszig, E. Advanced Engineering Mathematics. Wiley.

REFERENCE BOOKS

Grove, W. E. Brief Numerical Methods. Prentice-Hall.

Hildebrand, F. B. Advanced Calculus for Applications. Prentice-Hall. Jeffreys, G. V. & Jenson, V. G. Mathematical Methods in Chemical

Engineering. Academic.

Keane, A. & Senior, S. A. Mathematical Methods. Science.

10.111A Pure Mathematics II-Linear Algebra

Vector Spaces: inner products, linear operators, spectral theory, quadratic forms. Linear Programming: convex sets and polyhedra, feasible solutions, optimality, duality.

TEXTBOOKS

SESSION 1

Tropper, A. M. Linear Algebra. Nelson. Paperback.

SESSION 2

Gass, H. Linear Programming. I.S.E. McGraw-Hill.

REFERENCE BOOKS

Hoffman, K. & Kunze, R. Linear Algebra. Prentice-Hall. Lang, S. Linear Algebra. Addison-Wesley.

^{*} These units are also available to Faculty of Science students as a sequence of two units constituting a terminating service course in mathematics. As such they are mutually exclusive to any other level II or level III units in Pure and/or Applied Mathematics.

10.111B Pure Mathematics II—Analysis

Complex variables: analytic functions, elementary functions, Taylor and Laurent series, integrals, Cauchy's theorem, residues, evaluation of certain real integrals, maximum modulus principles. Linear differential equations of the second order: equations with constant coefficients, power series solutions, Laplace transforms, Bessel functions.

TEXTBOOKS

SESSION 1

Kolman, B. & Trench, W. F. Elementary Multivariable Calculus. Academic.

SESSION 2

Churchill, R. V. Complex Variables and Applications. I.S.E. McGraw-Hill.

REFERENCE BOOKS

Hilton, P. J. Partial Derivatives. Routledge. Thomas, G. B. Calculus and Analytic Geometry. 4th ed. Addison-Wesley

10.111C Pure Mathematics II—Algebra and Geometry

Abstract Algebra: Euclidean algorithm, unique factorization theorem, mathematical systems, groups, determination of small groups, homomorphisms and normal subgroups. Geometry: elementary concepts of Euclidean, projective and affine geometries.

TEXTBOOKS SESSION 1 Dean, R. A. Elements of Abstract Algebra. Wiley.

SESSION 2 Gans, D. Transformations and Geometrics. Appleton-Century-Crofts.

REFERENCE BOOK

Coxeter, H. S. M. Introduction to Geometry. Wiley.

10.121A Higher Pure Mathematics II—Algebra

Linear Algebra: vector spaces, commutative rings, polynomials, modules, linear transformations, eigenvectors, invariant subspaces, canonical forms, linear functions, bilinear and multi-linear algebra. Group Theory: subgroups, quotient groups, isomorphisms, Lagrange's theorem, Sylow's theorem.

TEXTBOOKS

Clark, A. Elements of Abstract Algebra. Wadsworth, 1971. Hoffman, K. & Kunze, R. Linear Algebra. Prentice-Hall.

REFERENCE BOOKS
Hartley, B. & Hawkes, T. O. Rings, Modules and Linear Algebra. Chapman & Hall.
Herstein, I. M. Topics in Algebra. Blaisdell.
Green, J. A. Sets and Groups. Macmillan.
Hall, M. The Theory of Groups. Macmillan.
Lang, S. Linear Algebra. W.S.S. Addison-Wesley.
Ledermann, W. The Theory of Finite Groups. Oliver & Boyd.

10.121B Higher Pure Mathematics II—Real and Complex Analysis

Construction of reals; uniform convergence; implicit and inverse function theorems; analytic functions; Laurent and Taylor series; calculus of residues; conformal mapping.

TEXTBOOKS

SESSION 1

Williamson, R. E. Crowell, R. H. & Trotter, H. F. Calculus of Vector Functions. Prentice-Hall.

SESSION 2

Jamieson, G. J. D. A First Course on Complex Functions. Chapman & Hall.

REFERENCE BOOKS

Derrick, W. Introductory Complex Analysis. Academic. Goldberg, R. R. Methods of Real Analysis. Blaisdell. Knopp, K. Elements of the Theory of Functions. Dover. Lang, S. Calculus of Several Variables. Addison-Wesley. Spivak, M. Calculus. Benjamin.

10.121C Higher Pure Mathematics II—Number Theory and Geometry

Galois fields, quadratic reciprocity, quadratic forms, continued fractions, number theoretic functions; axioms for a geometry, affine geometry, Desargues' theorem, projective geometry.

TEXTBOOK

Niven, I. & Zuckerman, H. S. Introduction to the Theory of Numbers. Wiley.

REFERENCE BOOK Hardy, G. H. & Wright, E. M. The Theory of Numbers. O.U.P.

10.112A Pure Mathematics III—Number Theory and Algebra

Euclidean algorithm, congruences, sums of squares, diophantine equations, rings, polynomials, fields.

TEXTBOOKS SESSION 1 Griffin, H. Elementary Theory of Numbers. McGraw-Hill. Mack, J. M. Number Theory Notes. Dept. Pure Maths, Univ. of Sydney. SESSION 2

Dean, R. A. Elements of Abstract Algebra. Wiley.

REFERENCE BOOKS

Birkhoff, G. & MacLane, S. A Survey of Modern Algebra. Macmillan. Hardy, G. H. & Wright, E. M. Introduction to the Theory of Numbers. O.U.P.

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10.112B Pure Mathematics III—Real Analysis

Convergence of sequences; Cauchy sequences; lim sup and lim inf; convergence tests for series; sequences and series of functions; uniform convergence; Taylor and Fourier series; evaluation of integrals and solutions of differential equations via series; metric spaces; contraction mapping principle and its applications.

TEXTBOOK Youse, B. K. Introduction to Real Analysis. Allyn & Bacon.

REFERENCE BOOKS Goffman, C. Introduction to Real Analysis. Harper. Spivak, M. Calculus. Benjamin.

10.112C Pure Mathematics III—Differential Geometry

Curves and surfaces in space. Differential forms. Curvature.

TEXTBOOK O'Neill, B. Elementary Differential Geometry. Academic.

REFERENCE BOOK Willmore, T. Differential Geometry. O.U.P.

10.112D Pure Mathematics III—Topology and Set Theory

Cardinal and ordinal numbers. Elementary topology of surfaces.

TEXTBOOKS

SESSION 1

Gray, J. D. Lecture Notes on Set Theory and Transfinite Arithmetic. Author, 1973.

SESSION 2

Frechet, M. & Fan, K. Initiation to Combinatorial Topology. Prindle, Weber & Schmidt.

Kamke, E. Theory of Sets. Dover.

REFERENCE BOOKS

Blackett, D. W. Elementary Topology: A Combinatorial and Algebraic Approach. Academic Press.

Wilder, R. L. Introduction to the Foundations of Mathematics. Wiley.

10.112E Pure Mathematics III—Complex Analysis and Differential Equations

Complex analysis and ordinary differential equations.

TEXTBOOKS SESSION 1 Churchill, R. V. Complex Variables and Applications. I.S.E. McGraw-Hill.

SESSION 2

Plaat, O. Ordinary Differential Equations. Holden-Day.

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10.122A Higher Pure Mathematics III—Algebra

Field theory and theory of rings and modules.

TEXTBOOK

Lang, S. Algebra. Addison-Wesley.

10.122B Higher Pure Mathematics III—Integration and Functional Analysis

Lebesgue Integration; Fourier series; normed vector spaces; Hilbert spaces; measure theory.

TEXTBOOK

Epstein, B. Linear Functional Analysis. Saunders.

10.122C Higher Pure Mathematics III—Topology and Differential Geometry

The axiom of choice, metric and topological spaces, compactness. Compact surfaces, triangulations, geodesics, Gauss-Bonet theorem.

TEXTBOOK

Dugundji, J. Topology. Allyn & Bacon.

10.122E Higher Pure Mathematics III—Complex Analysis and Differential Equations

Analytic continuation; entire and meromorphic functions; elliptic functions; normal families and further advanced topics in complex analysis. Existence and uniqueness theorems for ordinary differential equations; linear systems; qualitative theory of autonomous systems; equations on manifolds.

TEXTBOOKS

 Cartan, H. Elementary Theory of Analytic Functions of One or Several Complex Variables. Addison-Wesley.
 Roxin, E. O. Ordinary Differential Equations. Wadsworth.

10.211A Applied Mathematics II—Mathematical Methods

Review of functions of two and three variables, divergence, gradient, curl; line, surface, and volume integrals; Green's and Stokes' theorems. Special functions, including gamma and Bessel functions. Differential equations and boundary value problems, including vibrating string and vibrating circular membrane; Fourier series.

TEXTBOOKS

Blatt, J. M. Introduction to Fortran IV Programming. Prentice-Hall. Sneddon, I. N. Fourier Series. Routledge. Spiegel, M. R. Advanced Mathematics for Scientists and Engineers. Schaum. Spiegel, M. R. Theory and Problems of Vector Analysis. Schaum.

REFERENCE BOOKS

Betz, H., Burcham, P. B. & Ewing, G. M. Differential Equations with Applications. I.S.R. Harper.

Blatt, J. M. Basic Fortran IV Programming. Computer Systems (Aust.). Dettman, J. W. Mathematical Methods in Physics and Engineering.

McGraw-Hill. Smith, G. D. Vector Analysis Including the Dynamics of a Rigid Body.

Smith, G. D. Vector Analysis Including the Dynamics of a Rigid Body. O.U.P.

10.211B Applied Mathematics II—Analytical Dynamics

Kinematics of particles and rigid bodies. Dynamics of particles, including simple harmonic motion and motion in a central force field. Dynamics of systems of particles, conservation principles, collisions, rocket motion. Dynamics of rigid bodies, including compound pendulum and Euler's equations. Lagrange's and Hamilton's equations.

TEXTBOOK

Smith, R. C. & Smith, P. Mechanics. Wiley.

REFERENCE BOOK

Lawden, D. F. A Course in Applied Mathematics. Vol. I. English U.P.

10.211C Applied Mathematics II—Hydrodynamics

Conservation laws and Bernoulli's equation for one-dimensional flow. Equations of continuity and Euler's equation. Kelvin's theorem. Incompressible, irrotational flow in two and three dimensions, including applications of complex variables, method of images, harmonic functions, and axially symmetric flow. Introduction to compressible and viscous fluids.

TEXTBOOK

Brenkert, K. Jr. Elementary Theoretical Fluid Mechanics. Wiley.

10.221A Higher Applied Mathematics II-Mathematical Methods

As for 10.211A but in greater depth.

TEXTBOOKS

Queen, N. M. Vector Analysis. McGraw-Hill, 1967.

Williamson, R. E., Crowell, R. H. & Trotter, H. F. Calculus of Vector Functions. Prentice-Hall.

REFERENCE BOOKS

Blatt, J. M. Basic Fortran IV Programming. Computer Systems (Aust.). Dettman, J. W. Mathematical Methods in Physics and Engineering. McGraw-Hill.

10.221B Higher Applied Mathematics II—Analytical Dynamics

As for 10.211B but in greater depth.

TEXTBOOK

McCuskey, S. W. Introduction to Advanced Dynamics. Addison-Wesley.

REFERENCE BOOKS

Goldstein, H. Classical Mechanics. Addison-Wesley.

Lawden, D. F. A Course in Applied Mathematics. Vol I. English U.P.

B122 THE UNIVERSITY OF NEW SOUTH WALES

10.221C Higher Applied Mathematics II—Hydrodynamics

As for 10.211C but in greater depth.

TEXTBOOK

Curle, N. & Davies, H. J. Modern Fluid Dynamics. Vol. 1, Van Nostrand.

REFERENCE BOOKS

Chorlton, F. Textbook of Fluid Dynamics. Van Nostrand. Milne-Thomson, L. M. Theoretical Hydrodynamics. Macmillan.

10.212A Applied Mathematics III—Numerical Analysis

Polynomial approximation, interpolation and extrapolation, numerical quadrature, solution of ordinary differential equations, sets of linear equations, matrix eigenvalues and eigenvectors, boundary value problems, partial differential equations. A knowledge of FORTRAN programming is essential.

TEXTBOOK

Conte, S. D. Elementary Numerical Analysis. McGraw-Hill.

REFERENCE BOOKS

Fike, C. T. Computer Evaluation of Mathematical Functions. Prentice-Hall. Fox, L. & Mayers, D. F. Computing Methods for Scientists & Engineers. Ō.U.P.

Ralston, A. A First Course in Numerical Analysis. McGraw-Hill.

10.212B Applied Mathematics III—Continuum Mechanics

Cartesian tensors, stress and strain in continuous media. Equations of equilibrium and motion. Equations of elasticity. Bending and torsion of beams. Plane elasticity (if time available). Viscous flow of liquids (if time available).

TEXTBOOK

Mase, G. E. Continuum Mechanics. Schaum.

REFERENCE BOOK

Long. R. L. Mechanics of Solids and Fluids. Prentice-Hall.

10.212D Applied Mathematics III—Mathematical Methods

Sturm-Liouville equation, eigenvalues, expansion in orthonormal func-tions. Fourier, Fourier-Bessel and Legendre series as special cases. Fourier and Laplace transforms, with application to ordinary and partial differential equations, Diffusion equation and transmission-line equation. Wave equation.

TEXTBOOKS

Rabenstein, A. L. Introduction to Ordinary Differential Equations. Academic.

Stephenson, G. An Introduction to Partial Differential Equations for Science Students, 2nd ed. Longmans, Paperback,

REFERENCE BOOKS

Birkhoff, G. & Rota, G. Ordinary Differential Equations. Ginn & Co.

Carslaw, H. S. & Jaeger, J. C. Operational Methods in Applied Mathematics. Dover.

Raven, F. H. Mathematics of Engineering Systems. McGraw-Hill. Schelkunoff, S. A. Applied Mathematics for Engineers and Scientists. Van Nostrand.

FACULTIES OF BIOLOGICAL SCIENCES AND SCIENCE B123

10.212L Applied Mathematics III—Optimization Techniques

Origins of optimization problems. The Simplex Algorithm. Duality. Transportation problems. Numerical methods. Nonlinear programming. The Kuhn-Tucker theorem. Computational algorithms. Introduction to dynamic programming. Network problems.

TEXTBOOK

Box, M. J. et al. Nonlinear Optimization Techniques. Oliver & Boyd, 1969.

10.222B Higher Applied Mathematics III—Continuum Mechanics

As for 10.212B but in greater depth.

REFERENCE BOOKS

Fung, Y. C. A First Course in Continuum Mechanics. Prentice-Hall. Landau, L. D. & Lifshitz, E. M. Theory of Elasticity. Pergamon.

10.222C Higher Applied Mathematics III—Maxwell's Equations and Special Relativity

Electrostatic and quasi-static magnetic fields: mathematical formulation of basic laws, field equations, methods of solution, general theorems, polarization, energy and mechanical forces. Electromagnetic fields: Maxwell's equations, Poynting theorem, Maxwell stress tensor, electromagnetic momentum and radiation pressure, electromagnetic potentials, radiation, vector wave equation, solutions, cavity resonators, waveguides. Relativity: relativistic kinematics, dynamics and electrodynamics, radiation from moving charges, radiation damping.

TEXTBOOKS

Jackson, J. D. Classical Electrodynamics. Wiley. Lawden, D. F. Tensor Calculus and Relativity. Methuen.

REFERENCE BOOKS

Argence, E. & Kahan, T. Theory of Waveguides & Resonators. Blackie. Becker, R. Electromagnetic Fields and Interactions. Vol I. Blackie.

Einstein, A. et al. The Principle of Relativity. Dover.

Moller, C. The Theory of Relativity. O.U.P.

Panofsky, W. K. H. & Phillips, M. Classical Electricity and Magnetism. Addison-Wesley.

Pauli, W. Theory of Relativity. Pergamon.

10.222D Higher Applied Mathematics III—Mathematical Methods

Functions of a complex variable, contour integration. Fourier, Laplace and Mellin transforms, solutions of ordinary and partial differential equations. Asymptotic expansions.

REFERENCE BOOKS

Carrier, G. F., Crook, M. & Pearson, C. E. Functions of a Complex Variable: Theory and Technique. McGraw-Hill.

Courant, R. & Hilbert, D. Methods of Mathematical Physics. Vol. 1. Interscience.

Lighthill, M. J. Fourier Analysis and Generalised Functions. C.U.P. Paperback.

Watson, G. N. & Whittaker, E. T. A Course in Modern Analysis. C.U.P.

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10.222F Higher Applied Mathematics III-Quantum Mechanics

Review of physical basis for quantum mechanics, simple harmonic oscillator, hydrogen atom. General formalism, angular momentum, perturbation theory and other approximation methods. Scattering problems.

TEXTBOOK

Schiff, L. I. Quantum Mechanics. 3rd ed. I.S.E. McGraw-Hill.

REFERENCE BOOKS

Landau, L. D. & Lifshitz, E. M. Quantum Mechanics. Pergamon. Messiah, A. Quantum Mechanics. Vols. 1 & 2. North-Holland.

10.223 Applied Mathematics IV

Specialized study in selected topics for students who intend to graduate with honours. Includes preparation of an undergraduate thesis.

TEXTBOOK

No set text.

10.412 Dynamical and Physical Oceanography

(a) The physical properties of the oceans and their measurement, including: salinity, temperature, density, dynamic heights. Currents, waves and tides. (b) Theoretical models of current and waves.

Up to seven days field/laboratory work per year.

REFERENCE BOOK

Defant, A. Physical Oceanography. Vols I & II. Pergamon Press, 1960.

STATISTICS

10.311 Theory of Statistics II

An introduction to an axiomatic treatment of probability. Variates (univariates, multivariates, expectations, moment generating and characteristic functions). Standard distributions. Sampling distributions. Point estimation (moments, maximum likelihood, minimum χ^2 , etc.). Confidence interval estimation, exact and approximate. Elementary Neyman-Pearson theory of tests of significance, standard significance tests. Regression (including curvilinear) on a single fixed variable.

INTRODUCTORY READING

Moroney, M. J. Facts from Figures. Pelican. Tanur, J. M. ed. Statistics: A guide to the Unknown. Holden-Day.

TEXTBOOKS

Graybill, F. A. & Mood, A. M. Introduction to the Theory of Statistics. McGraw-Hill.

Hogg, R. V. & Craig, A. T. Introduction to Mathematical Statistics. 3rd ed. Collier-Macmillan.

Statistical Tables.

REFERENCE BOOKS

Anderson, R. L. & Bancroft, T. A. Statistical Theory in Research. McGraw-Hill.

Hartley, H. O. & Pearson, E. S. Biometrika Tables for Statisticians. C.U.P. Larson, H. J. Introduction to Probability and Statistical Inference. Wiley. Parzen, E. Modern Probability Theory and its Applications. Wiley.

10.321 Higher Theory of Statistics II

10.311 at greater depth and covering a slightly wider field.

TEXTBOOKS

As for 10.311, plus:

Kendall, M. G. & Stuart, A. The Advanced Theory of Statistics. Vols. I & II. 2nd ed. Griffin.

REFERENCE BOOK

Feller, W. An Introduction to Probability Theory and its Applications. Vol. 1. 3rd ed. Wiley.

10.312A Theory of Statistics III—Stochastic Processes and Applications

Conditional expectations, generating functions, branching processes, finite Markov chains, introduction to finite-state space Markov processes in continuous time, applications of stochastic processes in genetics.

TEXTBOOK

Bailey, N. J. T. Elements of Stochastic Processes with Applications to the Natural Sciences. Wiley.

REFERENCE BOOKS

Feller, W. An Introduction to Probability Theory and its Applications. Vol I. Wiley.

Karlin, S. A First Course in Stochastic Processes. Academic.

Ross, S. M. Introduction to Probability Models. Academic.

10.312B Theory of Statistics III—Experimental Design (Applications) and Sampling

Principles of good experimental design, analyses of fully randomized and randomized block designs, factorial treatment structure, components of variance, multiple comparisons; finite populations, simple random sampling, stratified random sampling, optimum allocation, estimation of sample size.

REFERENCE BOOKS

Brownlee, K. A. Statistical Theory & Methodology in Science & Engineering. 2nd ed. Wiley.

Cochran, W. G. Sampling Techniques. Wiley.

Cochran, W. G. & Cox, G. M. Experimental Designs. I. S. E. Wiley.

Cox, D. R. Planning of Experiments. Wiley.

10.312C Theory of Statistics III—Experimental Design (Theory) and Project

Multivariate normal distribution, quadratic forms, multiple regression, theory of the general linear hypothesis and its application to experimental designs.

TEXTBOOK

Graybill, F. A. An Introduction to Linear Statistical Models. McGraw-Hill.

REFERENCE BOOK

Anderson, T. W. An Introduction to Multivariate Statistical Analysis. Wiley.

10.312D Theory of Statistics III—Probability Theory and Contingency Tables

Characteristic functions, convergence of probability distributions, the central limit theorem, expansions related to the normal distributions, extreme value distributions; general theory of the 2×2 contingency table, χ^2 test and exact test, $m \times n$ contingency table, subdivision of χ^2 .

TEXTBOOK

Heathcote, C. R. Probability: Elements of the Mathematical Theory. Allen & Unwin.

REFERENCE BOOKS

Kendall, M. G. & Stuart, A. The Advanced Theory of Statistics. Vol. II. 2nd ed. Griffin.

Lamperti, J. Probability. Benjamin.

Lancaster, H. O. The Chi-squared Distribution. Wiley.

10.322A Higher Theory of Statistics III

As for 10.312A but in greater depth.

TEXTBOOK

Cox, D. R. & Miller, H. D. The Theory of Stochastic Processes. Methuen.

REFERENCE BOOKS

As for 10.312A, plus:

Bailey, N. J. T. Elements of Stochastic Processes with Applications to the Natural Sciences. Wiley.

10.322B Higher Theory of Statistics III

As for 10.312B but in greater depth.

TEXTBOOKS

Cochran, W. G. & Cox, G. M. Experimental Designs. I. S. E. Wiley. Statistical Tables.

REFERENCE BOOKS

Brownlee, K. A. Statistical Theory and Methodology in Science & Engineering. 2nd ed. Wiley.

Cochran, W. G. Sampling Techniques. Wiley. Cox, D. R. Planning of Experiments. Wiley.

10.322C Higher Theory of Statistics III

As for 10.312C but in greater depth.

TEXTBOOK As for 10.312C.

REFERENCE BOOKS

As for 10.312C plus:

Rao, C. R. Linear Statistical Inference and its Applications. Wiley.

10.322D Higher Theory of Statistics III

As for 10.312D but in greater depth.

TEXTBOOK

Heathcote, C. R. Probability: Elements of the Mathematical Theory. Allen & Unwin.

REFERENCE BOOKS

As for 10.312D, plus:

Chung, K. L. A Course in Probability Theory. Harcourt, Brace & World.

10.323 Theory of Statistics IV

Specialized study, from the topics set out, for students attempting honours in the Science or Arts courses with a major in Statistics. Mathematical basis. Experimental design; response surfaces. Stochastic processes. Theories of inference. Sequential analysis. Non-parametric methods. Multivariate analysis. Mathematical programming. Information theory. Discrete distributions. Project.

TEXTBOOKS

Anderson, T. W. Multivariate Statistical Analysis. Wiley.

Box, G. E. P. & Jenkins, G. Time Series Analysis, Forecasting and Control. Holden-Dav.

Cox, D. R. & Smith, W. Queues. Methuen.

Cramer, H. & Leadbetter, M. R. Stationary and Related Stochastic Processes. Wilev.

Feller, W. An Introduction to Probability Theory and its Applications. Vol. 2. Wiley.

Fisher, R. A. Statistical Methods and Scientific Inference. Oliver & Boyd.

Graybill, F. A. An Introduction to Linear Statistical Models. McGraw-Hill.

Hartley, H. O. & Pearson, E. S. Biometrika Tables for Statisticians. Vol. I. C.U.P

Jenkins, G. M. & Watts, D. G. Spectral Analysis and its Applications. Holden-Day.

John, P. W. M. Statistical Design and Analysis of Experiments. Macmillan.

Kempthorne, O. The Design and Analysis of Experiment. Wiley.

Searle, S. R. Linear Models. Wiley.

Wald, A. Sequential Analysis. Wiley.

REFERENCE BOOKS

Bradley, J. V. Distribution-free Statistical Tests. Prentice-Hall.

Cochran, W. G. & Cox, G. M. Experimental Designs. Wiley.

Cochran, W. G. Sampling Techniques. Wiley. Ferguson, T. S. Mathematical Statistics—A Decision Theoretic Approach. Academic.

Fraser, D. A. S. The Structure of Inference. Wiley.

- Ghosh, B. K. Sequential Tests of Statistical Hypotheses. Addison-Wesley. Godambe, V. P. & Sprott, D. A. Foundations of Statistical Inference. Holt, Rinehart & Winston.
- Kendall, M. G. & Stuart, A. The Advanced Theory of Statistics. Vol. II. 2nd ed. Griffin.

Lehmann, E. L. Testing Statistical Hypotheses. Wiley.

Moran, P. A. P. An Introduction to Probability Theory. O.U.P. Noether, G. E. Elements of Non-parametric Statistics. Wiley.

Raiffa, H. & Schlaifer, R. Applied Statistical Decision Theory. M.I.T. Rao, C. R. Advanced Statistical Methods in Biometric Research. Wiley.

Seal, H. Multivariate Statistical Analysis for Biologists. Methuen.

Sarhan, A. E. & Greenberg, B. G. Contributions to Order Statistics. Wiley. Scheffe, H. The Analysis of Variance. Wiley. Wald, A. Statistical Decision Functions. Wiley.

Wetherill, G. B. Sequential Methods in Statistics. Methuen.

Yaglom, A. M. An Introduction to the Theory of Stationary Random Functions. Prentice-Hall.

10.331 Statistics SS

An introduction to the theory of probability, with finite, discrete and continuous sample spaces. The standard elementary univariate distributions: binomial, Poisson and normal; an introduction to multivariate distributions. Standard sampling distributions, including those of χ^2 , t and F. Estimation by moments and maximum likelihood (including sampling variance formulae, and regression); confidence interval estimation. The standard tests of significance based on the above distributions, with a discussion of power where appropriate. An introduction to experimental design; fixed, random and mixed models, involving multiple comparisons and estimation of variance components.

TEXTBOOKS

Freund, J. E. Mathematical Statistics. 2nd ed. Prentice-Hall. Statistical Tables.

REFERENCE BOOKS

- Bennett, C. A. & Franklin, N. L. Statistical Analysis in Chemistry and the Chemical Industry. Wiley.
- Davies, O. L. Statistical Methods in Research and Production. Oliver & Boyd.
- Steel, R. G. D. & Torrie, J. H. Principles and Procedures of Statistics. McGraw-Hill.

SCHOOL OF MICROBIOLOGY

Microbiology is concerned with the nature and behaviour of the microscopic and submicroscopic forms of living organisms, particularly the fungi, bacteria and viruses. These agents have innumerable effects on human welfare, some of which are beneficial and others, clearly harmful. Thus, some microorganisms may aid in the decomposition of organic waste materials, increase the fertility of soils or be directly involved in the production of foodstuffs, beverages, pharmaceuticals (including antibiotics) and other industrially important compounds; others are important agents of human, animal and plant diseases, of food spoilage or destruction of a variety of structural materials. Knowledge of the behaviour of microorganisms can be applied directly both to increasing their benefits to man or minimizing their harmful effects. In the latter case, particularly human or animal disease, a knowledge of the "twin-science", immunology, is important for appreciating or developing methods for specific prevention or treatment of certain diseases. Additionally, fundamental studies in microbiology are being used to advance our knowledge of the nature of living organisms and the basic processes of life, particularly in the areas of molecular biology, genetics and metabolism.

Students may undertake Microbiology as a major or minor Science course, as part of the medical degree or in Food Technology, and in other courses in the Faculty of Applied Science, in Biological Technology, Public Health Engineering and Food and Drug Analysis or for the degree of BSc(Ed).

Microbiology and/or Immunology can be taken to the honours level and for the Master of Science and Doctor of Philosophy degrees. Medical students can interrupt their course to undertake the BSc(Med) degree in Microbiology while higher medical degrees may also be taken in this subject. Such advanced studies will include research work as well as additional reading programmes and in some cases, formal instruction. Students who have not majored in Microbiology but have otherwise suitable basic training may enrol for a higher degree.

It is essential that students who wish to enrol in any course offered by the School should ensure that they have satisfied all the prerequisites; only in special circumstances may these be waived or varied with the permission of the Head of School. The School's education advisory officer should be consulted for advice on the best course structure which might satisfy the student's particular interests.

For graduation with microbiology as a major subject students must have taken, as a minimum requirement, the two double units 44.102 General Microbiology and 44.112 Applied Microbiology. For those who specifically wish to make a career in this discipline it is advised they should also take the single units 44.122 Immunology and 44.132 Virology which along with 44.112 are offered in Session 2 of the year. As alternatives to 44.122 and 44.132 students could take the single units 43.102D Mycology (School of Botany), 42.102 Fermentation Technology (School of Biological Technology) or the double unit 41.102B Metabolic Pathways and Control Mechanisms (School of Biochemistry) assuming 41.102 has been taken; a double major in Microbiology and Biochemistry would be possible under the last alternative.

Students not wishing to major in Microbiology may choose one or more units from 44.102 Basic General Microbiology or 44.122 Immunology. Ordinarily 44.112 Applied Microbiology and 44.132 Virology cannot be taken without having done 44.102.

A student wishing to take an honours course will be expected to have achieved a high standard in courses taken for the pass degree and to have majored in Microbiology. Additionally he must receive the permission of the Head of School.

For details of level, unit value, when offered, hours per week and prerequisites, see page B22.

44.101 Introductory Microbiology

The general nature, occurrence and importance of microorganisms. A systematic review of the major groups of microorganisms: the eucaryotic protista (micro-algae, protozoa and fungi); procaryotic protista (blue-green algae, "higher" bacteria, typical unicellular bacteria and small bacteria-like forms); plant, animal and bacterial viruses. The relationship between microorganisms and their environment; ecological considerations. Interactions between microorganisms and higher organisms.

TEXTBOOK

Brock, T. D. Biology of Microorganisms. Prentice-Hall, 1970.

or

Pelczar, M. J. & Reid, R. D. Microbiology. 3rd ed. McGraw-Hill, 1972. or

Stanier, R. Y., Doudoroff, M. & Adelberg, E. A. General Microbiology. 3rd ed. Macmillan, 1971. (Also published as The Microbial World. 3rd ed. Prentice-Hall, 1970.)

or

Hawker, L. E. & Linton, A. H. eds. Micro-organisms: Function, Form and Environment. Arnold, 1971.

Note: Hawker and Linton is recommended for students wishing to major in Microbiology in the following year.

44.102 General Microbiology

Double unit, level III.

Systems for the isolation, identification and taxonomic description of microorganisms; fine structure, cyto-chemistry, genetics of bacteria and viruses; metabolic requirements of microorganisms; microorganisms and their environment; growth, inhibition and death; energy-yielding and biosynthesizing systems; genotypic and phenotypic control systems.

TEXTBOOKS

As for 44.101 if not taking other Microbiology units. Otherwise:

- Hawker, L. E. & Linton, A. H. eds. Microorganisms: Function, Form and Environment. Arnold, 1971.
- Davis, B. D., Dulbeco, R., Eisen, H. N., Ginsberg, H. S. & Wood, W. B. Microbiology. Complete ed. Harper & Row, 1968.

44.112 Applied Microbiology

Double unit, level III.

Endeavours to relate the basic facts about microorganisms to a variety of practical conditions. The occurrence, importance, activity and control of microorganisms in soil, air, water and in their relationship with higher organisms (other than man); their industrial applications including manufacture, preservation and spoilage of food and dairy products. The nature of bacterial and fungal diseases of man, their cultural and serological diagnosis, epidemiology, treatment and prevention will be discussed in some detail.

TEXTBOOKS

As for 44.102.

44.122 Immunology

Single unit, level III.

Basic immunology and immunological techniques. The interdisciplinary nature of the subject makes this unit suitable for students taking any major sequence in biological science and also for higher degree students who require a background training in immunology. The course includes phylogeny and ontogeny of the immune response; antigen and antibody structure; antigen-antibody reaction; immunochemistry; immunogenetics, clinical immunology; transplantation.

TEXTBOOKS

Roitt, I. Essential Immunology. Blackwell Scientific Publications, 1972.

44.132 Virology

Single unit, level III.

The structure, replication and behaviour of animal, plant and bacterial viruses; applications of virological techniques; virus diseases of animals and plants, their epidemiology and control.

TEXTBOOKS

Fenner, F. & White, D. O. Medical Microbiology. Academic, 1970. and As for 44.102.

44.103 Microbiology II (Honours)

Advanced formal study in approved subjects, together with a research project. The results of the latter are embodied in a thesis.

44.111 Microbiology

A short introductory course to microbiology which is designed to familiarize students, without previous biological training, with microorganisms and with the methods used in their isolation and identification. The content of the course is similar to that of 44.101.

TEXTBOOK

Brock, T. D. Biology of Microorganisms. Prentice-Hall, 1970.

Pelczar, M. J. & Reid, R. D. Microbiology. 3rd ed. McGraw-Hill, 1972.

SCHOOL OF PHILOSOPHY

The study of philosophy is partly the study of perennial problems of common interest to everyone; for example, the foundations of morality, the grounds of religious belief, the source and reliability of knowledge, and the relation between body and mind. Philosophy also leans out to and illuminates other fields of study. Consequently courses in philosophy are designed to make it possible for students to pursue a philosophical interest related to their other interests.

The first year course in philosophy is a wide-ranging course which is intended to give a broad introduction to the subject and assumes no previous acquaintance with it. It is broken into two sessions, with an examination at the end of each session, but Arts students take it as a whole and other students are advised to do so. There is no distinction between Pass and Honours.

For second year, Pass courses are presented and examined in sessionlength units. This arrangement makes it possible to offer a wide range of units from which students may select freely, subject only to certain stipulations regarding prerequisites. In a normal course, students take two course units in each session.

SELECTION OF UNITS

The pattern of courses after the first year is intended to give students a wide range. Constraints are imposed by, firstly, the prerequisites of the various subjects; secondly, the distribution of courses as between Sessions 1 and 2; and to some extent, timetabling. The following details will assist students with their initial choice.

The course-units available in Session 1 having no prerequisite apart from Introductory Philosophy A and B are:

Predicate Logic; Descartes; British Empiricism; Greek Philosophy; Scientific Method; Philosophy of Biology; Existentialism.

Of these, Predicate Logic is prerequisite to a range of advanced logic courses, and some of the others are also prerequisites, alone or as alternatives, to certain other subjects.

HONOURS COURSES

There is no division of students into Pass and Honours during the first year. From the second year special additional course-units are provided for Honours students.

52.111 Philosophy I

The course-units Introductory Philosophy A and Introductory Philosophy B as detailed below. There will be examinations at the end of each session but the course will be treated as an integrated whole-year one.

52.112 Philosophy II

Four course-units, normally two in each session.

52.122 Philosophy II (Honours)

Three course-units in Session 1 and two in Session 2, plus Honours Seminar A.

COURSE UNITS

Introductory Philosophy A (Session 1)

A first course for students new to the subject. The course divides into two strands: 1. Plato: A study of some dialogues of Plato, paying special attention to Socratic Definition and to Plato's Theory of Forms. 2. Informal Logic: An approach to logic by way of language, treating such topics as the uses of utterances, the truth and significance conditions of statements, the non-formal analysis of arguments, and the logical relations of propositions.

Students who would like to read some philosophy before starting the course might look at *Problems of Philosophy* by Bertrand Russell, or Philosophy Made Simple by R. H. Popkin and A. Stroll.

TEXTBOOKS

1. Plato

Passmore, J. Philosophical Reasoning. Duckworth. Plato. The Last Days of Socrates. Tredennick, H. trans. Penguin Classics. Vlastos, G. ed. The Philosophy of Socrates. Macmillan.

REFERENCE BOOKS

1. Plato

Allen, R. E. Plato's Euthyphro and the Earlier Theory of Forms. Routledge.

Bluck, R. S. Plato's Phaedo. Routledge.

Burnet, J. Greek Philosophy. Macmillan.

Cornford, F. M. The Republic of Plato. O.U.P.

Cross, R. C. & Woozley, A. D. Plato's Republic. Macmillan.

Guthrie, W. K. C. Socrates. C.U.P.

Hackforth, R. Plato's Phaedo. Bobbs-Merrill.

Plato. Parmenides and Other Dialogues. Warrington, J. trans. Everyman. Plato. Protagoras and Meno. Guthrie, W. K. C. trans. Penguin Classics. Plato. The Symposium. Hamilton, W. trans. Penguin Classics.

Robinson, R. Plato's Earlier Dialectic. O.U.P.

Taylor, A. E. Plato. Methuen.

Vlastos, G, ed. Plato: I. Metaphysics and Epistemology. Macmillan.

2. Informal Logic

Hamblin, C. L. Elementary Formal Logic-A Programmed Course. Hicks Smith and University Paperbacks.

Hospers, J. Introduction to Philosophical Analysis. 2nd ed. Prentice-Hall or Routledge, 1967.

Taylor, D. M. Explanation and Meaning. C.U.P.

Introductory Philosophy B (Session 2)

A continuation of Introductory Philosophy A. The course divides into two strands: 1. Hume: A study of some sections of Hume's Enquiry. Topics to be discussed may include: the miraculous and the existence of God, the mind-body problem and personal identity, the freedom of the will. 2. Formal Logic: An introduction to a system of Natural Deduction sufficient for the symbolization of such ordinary language arguments and the construction of such proofs as lie within the field of propositional logic and simple predicate logic.

TEXTBOOKS

1. Hume

Hume, D. On Human Nature and the Understanding. Flew, A. ed. Collier.

2. Formal Logic

Kalish, D. & Montague, R. Logic: Techniques of Formal Reasoning. Harcourt, Brace & World.

REFERENCE BOOKS

1. Hume

Ayer, A. J. The Concept of a Person. Macmillan. Berofsky, B. Free Will and Determinism. Harper & Row. Flew, A. Hume's Philosophy of Belief. Routledge. Flew, A. Body, Mind and Death. Macmillan. Hick, J. The Existence of God. Macmillan. Sesonske, A. Human Understanding. Wadsworth. Smart, N. Philosophers and Religious Truth. S.C.M. Taylor, D. M. Explanation and Meaning, C.U.P.

Predicate Logic (Session 1)

Prerequisite: Introductory Philosophy A and B.

A system of natural deduction is presented for the first order predicate calculus, including identity and definite descriptions. Emphasis is upon construction of formal derivations, methods of showing the invalidity of formal arguments, and the evaluation of informal arguments by symbolization.

TEXTBOOK

Kalish, D. & Montague, R. Logic: Techniques of Formal Reasoning. Harcourt, Brace & World.

REFERENCE BOOK

Church, A. Introduction to Mathematical Logic. Princeton.

Descartes (Session 1)

Prerequisite: Introductory Philosophy A and B.

A study of the main issues raised in the philosophy of Descartes and their importance for the development of modern philosophy. Emphasis is on the *cogito ergo sum* argument, the Cartesian method and the search for rational certainty, his theory of ideas, the body-mind problem, and his account of freedom.

TEXTBOOK

Anscombe, G. E. M. & Geach, P. T. eds. Descartes's Philosophical Writings. Nelson.

REFERENCE BOOKS

Beck, L. J. The Metaphysics of Descartes. Clarendon. Beck, L. J. The Method of Descartes. Clarendon. Buchdahl, G. Metaphysics and the Philosophy of Science. Blackwell. Copleston, F. A History of Philosophy. Vol. 4. Doubleday.

Doney, W. ed. Descartes: A Collection of Critical Essays. Doubleday. Frankfurt, H. G. Demons, Dreamers and Madmen: The Defence of Reason in Descartes's Meditations. Bobbs-Merrill.

Haldane, E. & Ross, G. R. T. eds. The Philosophical Works of Descartes. Dover.

Joachim, H. H. Descartes's Rules for the Direction of the Mind. Allen & Unwin.

Keeling, S. V. Descartes. O.U.P.

Kenny, A. Descartes: A Study of His Philosophy. Random House.

Popkin, R. H. Scepticism from Erasmus to Descartes. Van Gorcum.

Sesonske, A. & Fleming, N. Meta-Meditations: Studies in Descartes. Wadsworth.

Smith, N. K. New Studies in the Philosophy of Descartes. Macmillan.

Smith, N. K. Studies in the Cartesian Philosophy. Russell & Russell.

British Empiricism (Session 1)

Prerequisite: Introductory Philosophy A and B.

A survey of the empiricist tradition with special concentration on Locke and Berkeley.

TEXTBOOKS

Armstrong, D. M. Berkeley's Philosophical Writings. Collier. Paperback. Locke, J. An Essay Concerning Human Understanding. Fontana.

REFERENCE BOOKS

References will be given in lectures.

Greek Philosophy: Thales to Plato (Session 1)

Prerequisite: Introductory Philosophy A and B.

The leading ideas of the Greek Philosophers from Thales to Plato, with special reference to the Pre-Socratics.

PRINCIPAL REFERENCE BOOK

Guthrie, W. K. C. A History of Greek Philosophy. Vols. I & II. C.U.P.

REFERENCE BOOKS

Aristotle. The Works of Aristotle Translated into English. Vol. VIII. Metaphysics. O.U.P.

Burnet, J. Early Greek Philosophy. Black.

Burnet, J. Greek Philosophy. Macmillan.

Cornford, F. M. From Religion to Philosophy. Harper.

Cornford, F. M. Principium Sapientiae. Harper.

Cornford, F. M. Plato and Parmenides. Routledge.

Farrington, B. Greek Science. Penguin.

Furley, D. J. & Allen, R. E. eds. Studies in Presocratic Philosophy. Vols. I and II. Routledge.

Freeman, K. Ancilla to the Pre-Socratic Philosophers. Blackwell.

Gershenon, D. E. & Greenberg, D. A. Anaxagoras and the Birth of Physics. Blaisdell.

Grunbaum, A. Modern Science and Zeno's Paradoxes. Allen & Unwin, 1968.

Jaeger, W. The Theology of the Early Greek Philosophers. O.U.P. Kahn, C. H. Anaximander and the Origins of Greek Cosmology. Columbia. Kahn, C. H. The Verb 'Be' in Ancient Greek. Reidel.

Kirk, G. S. Heraclitus. The Cosmic Fragments. C.U.P.

Kirk, G. S. & Raven, G. E. The Pre-Socratic Philosophers. C.U.P.

Lee, H. D. P. Zeno of Elea. C.U.P.

O'Brien, D. Empedocles' Cosmic Cycle: A Reconstruction from the Fragments and Secondary Sources. C.U.P.

Philip, J. A. Pythagoras and Early Pythagoreanism. O.U.P.

Raven, J. E. Pythagoreans and Eleatics. C.U.P.

Robinson, R. Essays in Greek Philosophy. O.U.P.

Salmon, W. C. Zeno's Paradoxes. Bobbs-Merrill.

Sambursky, S. The Physical World of the Greeks. Routledge.

Seligman, P. The Apeiron of Anaximander. Univ. of London. Thomson, G. Studies in Ancient Greek Society. Vol. 2. The First Philosophers. Lawrence & Wishart.

Scientific Method (Session 1)

Prerequisite: Introductory Philosophy A and B.

A study of the nature of empirical knowledge as exemplified in the physical and social sciences and in history, with emphasis on the concept of explanation, the nature of induction and scientific laws, counterfactual statements, and the paradoxes of confirmation.

TEXTBOOKS

Hempel, C. G. Philosophy of Natural Science. Prentice-Hall. Rudner, R. S. Philosophy of Social Science. Prentice-Hall.

REFERENCE BOOKS

Barker, I. F. Induction and Hypothesis. Cornwall U.P.

Brown, R. Explanation in Social Science. Routledge.

Danto, A. & Morgenbesser, S. eds. Philosophy of Science-Readings. Meridian.

Dray, W. H. Laws and Explanation in History. O.U.P.

Dray, W. H. ed. Philosophical Analysis and History. Harper & Row.

Feigl, H. & Sellars, W. Readings in Philosophical Analysis. Appleton-Century-Crofts.

Hempel, C. G. Aspects of Scientific Explanation. Free Press.

Hintikka, J. & Suppes, P. Aspects of Inductive Logic. North Holland.

Hume, D. On Human Nature and the Understanding. Flew, A. ed. Collier.

Mill, J. S. A System of Logic. Longmans Green.

Nagel, E. The Structure of Science. Routledge.

Pap, A. An Introduction to the Philosophy of Science. Free Press.

Popper, K. R. The Logic of Scientific Discovery. Hutchinson.

Popper, K. R. The Poverty of Historicism. Routledge.

Popper, K. R. Conjectures and Refutations. Routledge.

Scheffler, I. The Anatomy of Inquiry. Knopf.

Strawson, P. F. Introduction to Logical Theory, Methuen,

Philosophy of Biology (Session 1)

Prerequisite: Introductory Philosophy A and B.

An introduction to some of the problems associated with the philosophy of biology. Main consideration is the autonomy of biology; i.e., whether biology is in principle reducible to the physical sciences and, ultimately, to physics, or whether the biologist necessarily employs types of description and explanation that have no application in the explanation and description of merely physical phenomena. No prior knowledge of biology is assumed but candidates will be expected to familiarize themselves with the attitudes of various biologists to these issues.

TEXTBOOK

Nagel, E. The Structure of Science. Routledge.

REFERENCE BOOKS

Beckner, M. The Biological Way of Thought. Columbia U.P.

Aga., W. E. A Contribution to the Theory of Living Organisms. Melbourne U.P. and C.U.P.

Bertalanffy, L. von. Problems of Life. Watts & Co. Bertalanffy, L. von. Modern Theories of Development. O.U.P. Braithwaite, R. B. Scientific Explanation. C.U.P. Haldane, J. S. Mechanism, Life and Personality. Murray.

Haldane, J. S. The Philosophical Basis of Biology. Hodder & Stoughton.

Lillie, R. S. General Biology and Philosophy of Organism. Chicago U.P.

Schubert-Soldern, R. Mechanism and Vitalism: Philosophical Aspects of Biology, Notre Dame U.P.

Russell, E. S. The Directiveness of Organic Activities. C.U.P.

Sommerhoff, G., Analytical Biology. O.U.P.

Tinbergen, N. The Study of Instinct. O.U.P.

Woodger, J. H. Biological Principles. Routledge.

Woodger, J. H. Biology and Language. C.U.P.

Oppression and Liberation (Session 2)

Prerequisite: Introductory Philosophy A and B.

A discussion of oppression and liberation, both in general and with special reference to the oppression of women and its ideology.

REFERENCE BOOKS

Althusser, L. For Marx. Penguin, 1972.

Bateson, G. Steps to an Ecology of Mind. Ballantine Books, N.Y., 1972. Cooper, D. ed. The Dialectics of Liberation. Penguin, 1969. de Beauvoir, S. The Second Sex. Penguin, 1972. Firestone, S. The Dialectic of Sex. Paladin, 1972.

Freire, P. Cultural Action for Freedom. Penguin, 1969.

Freire, P. Pedagogy of the Oppressed. Penguin, 1970.

Freud, S. Fragment of an Analysis of a Case of Hysteria. Standard edition of complete psychological works, vol. 7. Hogarth, 1953-66. Freud, S. On the Sexual Theories of Children. Standard edition, vol. 9.

Hogarth, 1953-66.

Freud, S. Some Psychical Consequences of the Anatomical Distinction between Sexes. Standard edition, vol. 19. Hogarth, 1953-66.

Freud, S. The Ego and the Id. Standard edition, vol. 19. Hogarth, 1953-66.

Freud, S. *Repression.* Standard edition, vol. 14. Hogarth, 1953-66.
Freud, S. *The Unconscious.* Standard edition, vol. 14. Hogarth, 1953-66.
Freud, S. *Three Essays on the Theory of Sexuality.* Standard edition, vol. 7. Hogarth, 1953-66.

Friedan, B. The Feminine Mystique. Penguin, 1972.

Hegel, G. W. F. The Phenomenology of Mind. Allen & Unwin, 1967.

Jenness, L. ed. Feminism and Socialism. Pathfinder Press, 1972. Jordan, Z. A. Karl Marx: Selections. Nelson, 1972.

Kollontai, A. Communism and the Family. Pluto Press, London, 1971. Laing, R. D. The Politics of the Family. Tavistock, 1971. Laing, R. D. & Cooper, D. Reason and Violence. Tavistock, 1964. Paperback.

Lefebvre, H. The Sociology of Marx. Penguin, 1972.

Marcuse, H. Five Lectures. Beacon Press, 1970. Marx, K. Capital. Vol. 1. Moscow, 1958. Marx, K. & Engels, F. The German Ideology. Progress Publishers, Moscow, 1964.

Marx, K. & Engels, F. Selected Works. Lawrence & Wishart, 1970.

Millett, K. Sexual Politics. Doubleday, 1970,

Mitchell, J. Woman's Estate. Penguin, 1971.

Rieff, P. Freud. Gollancz, London, 1959.

Rowbotham, S. Women, Resistance and Revolution. Allen Lane, 1972. Sartre, J.-P. The Problem of Method. Methuen.

Sartre, J.-P. Saint Genet. Actor and Martyr. Braziller, U.S.A., 1971. Schneir, M. ed. Feminism: The Essential Historical Writings. Vintage

Books, 1972. Szasz, T. The Myth of Mental Illness. Paladin, 1972.

Tanner, L. B. ed. Voices from Women's Liberation. Signet, 1971.

Wandor, M. ed. The Body Politic. Stage 1. London, 1972.

Existentialism (Session 1)

Prerequisite: Introductory Philosophy A and B.

Sartre's account of man-in-the-world. Sartre's ontology, his use of a phenomenological method and his ethics.

TEXTBOOKS

Manser, A. Sartre, A Philosophic Study. Athlone Press. Sartre, J.-P. Being and Nothingness. Methuen.

REFERENCE BOOKS

Cranston, M. Freedom. Longman.

Cumming, R. D. ed. The Philosophy of Jean-Paul Sartre. Methuen. Molina, F. Existentialism as Philosophy. Prentice-Hall.

Murdock, I. Sartre. Bowes & Bowes. Sartre, J.-P. The Transcendence of the Ego. Williams, F. & Kirkpatrick, R. trans. Noonday Press.

Sartre, J.-P. Imagination: A Psychological Critique. Williams, F. trans. Michigan U.P.

Sartre, J.-P., Nausea. Baldick, R. trans. Penguin. Sartre, J.-P. Sketch for a Theory of the Emotions. Mairet, P. trans. Methuen.

Sartre, J.-P. Intimacy. Alexander, L. trans. Panther Books.

Sartre, J.-P. The Psychology of the Imagination. Frechtman, B. trans. Rider. Sartre, J.-P. Two Plays. (The Flies, In Camera). Gilbert, S. trans. Hamish Hamilton.

Sartre, J.-P. Three Plays. Hamish Hamilton. Sartre, J.-P. Literary and Philosophical Essays. Michielson, A. trans. Rider. Sartre, J.-P. Portrait of an Anti-Semite. de Mauny, E. trans. Secker & Warburg.

Warnock, M. The Philosophy of Sartre. Hutchinson Uni. Library.

Foundations of Mathematics (Session 2)

Prequisite: Predicate Logic.

An introduction to a selection of problems concerning the foundations of Mathematics including the following topics: Non-Euclidean Geometry and consistency proofs, Axiomatics, Antinomies of naive set theory, Logicism, Intuitionism, Formalism, Gödel's Incompleteness result.

TEXTBOOK

Wilder, R. S. An Introduction to the Foundations of Mathematics. Wiley.

REFERENCE BOOKS

Benacerraf, P. & Putman, H. Philosophy of Mathematics. Prentice-Hall. Blanche, R. Axiomatics. Routledge.

Fraenkel, A. A. & Bar-Hillel, Y. Foundations of Set Theory. North-Holland.

Kleene, S. C. An Introduction to Meta-mathematics. Princeton. Korner, S. The Philosophy of Mathematics. Hutchinson.

Mendelson, E. Mathematical Logic. Van Nostrand.

Nagel, E. & Newman, J. R. Gödel's Proof. N.Y.U.P. or Routledge.

Argument (Session 2)

Prerequisite: Introductory Philosophy A and B.

A theoretical study of practical argumentation in the courtroom, politics and everyday life as compared with argument in logic, mathematics and theoretical science. Confirmation and probability, authority, testimony, of practical arguments; logical rationalism and scepticism.

REFERENCE BOOKS

Ackermann, R. Nondeductive Inference. Routledge. Hamblin, C. L. Fallacies. Methuen. Hart, H. L. A. The Concept of Law. O.U.P. Kneale, W. C. Probability and Induction. O.U.P. Passmore, J. A. Philosophical Reasoning. Duckworth. Toulmin, S. The Uses of Argument. C.U.P.

Logical Atomism (Session 2)

Prerequisite: Introductory Philosophy A and B.

A survey of the logical atomism of Russell and Wittgenstein and of the logical positivist movement.

TEXTBOOKS

Pears, D. ed. Russell's Logical Atomism. Fontana.

Wittgenstein, L. Tractatus Logico-Philosophicus. Pears, D. F. & McGuinness, B. F. trans, Routledge.

REFERENCE BOOKS

Reference books will be given in lectures.

Philosophy of Psychology (Session 2)

Not offered in 1974.

Prerequisite: Scientific Method.

A critical examination of some aspects of fundamental theory of psychology, with special emphasis on classical and contemporary behaviourism and behaviourist orientated psychology, and on the general conceptions of 'behaviour' and 'purpose'.

While Psychology I is not a prerequisite for this course, a preparatory survey of the introductory chapters of J. O. Whittaker's *Psychology* will be of value to students.

TEXTBOOK

Fodor, J. A. Psychological Explanation. Random House, 1968.

Aesthetics (Session 2)

Prerequisite: Introductory Philosophy A and B.

An examination of the central concepts, types of judgment and theories occurring in the fields of aesthetics, art criticism and literary criticism.

TEXTBOOK

Coleman, F. J. ed. Contemporary Studies in Aesthetics. McGraw-Hill,

REFERENCE BOOKS

- Boyce Gibson, A. Muse and Thinker. Methuen.
- Elton, W. ed. Aesthetics and Language. O.U.P.
- Gombrich, E. H. Art and Illusion. Phaidon.
- Goodman, N. Languages of Art. Bobbs-Merrill.
- Langer, S. K. Problems of Art. Scribners, N.Y.
- Margolis, J. ed. Philosophy Looks at the Arts. Scribners. Margolis, J. The Language of Art and Art Criticism. Wayne State U.P.
- Vivas, E. & Murray, K. eds. The Problems of Aesthetics. Holt. Rinehart & Winston.
- Wolheim, R. Art and Its Objects. Harper & Row.
- Wittgenstein, L. Lectures and Conversations on Aesthetics, Psychology, and Religion. Barrett, C. ed. Blackwell.
- Ziff. P. Philosophic Turnings: Essays in Conceptual Appreciation. O.U.P.

Plato (Session 2)

Prerequisite: Greek Philosophy: Thales to Plato.

A course centred around some of the later dialogues of Plato (Parmenides, Theaetetus, Sophist).

TEXTBOOK

Plato, Parmenides and Other Dialogues, Everyman,

REFERENCE BOOKS

Allen, R. E. Studies in Plato's Metaphysics. Routledge.

Bamborough, R. ed. New Essays on Plato and Aristotle. Routledge & Kegan Paul.

Burnet, J. Greek Philosophy. Macmillan.

- Cornford, F. M. Plato and Parmenides. Routledge. Cornford, F. M. Plato's Theory of Knowledge. Routledge.
- Crombie, I. M. An Examination of Plato's Doctrines. Vol. II. Plato on Knowledge and Reality. Routledge.

Robinson, R. Plato's Earlier Dialectic. O.U.P.

Robinson, R. Essays in Greek Philosophy. O.U.P.

Ross, W. D. Plato's Theory of Ideas. O.U.P.

Vlastos, G. ed. Plato. Vol. I. Metaphysics and Epistemology. Macmillan.

Spinoza and Leibniz (Session 2)

Prerequisite: Descartes.

A study of the main issues raised in the philosophy of the two great seventeenth century rationalists, with emphasis upon the development of their metaphysical systems in response to unresolved problems in the philosophy of Descartes and to contemporary scientific thinking. Their ethical views.

TEXTBOOKS

Leibniz, G. W. Selections. Weiner, P. P. ed. Scribner.

Spinoza, B. Ethics and On the Improvement of the Understanding. Both available in Works of Spinoza. Elwes, R. H. M. trans. Dover.

REFERENCE BOOKS

Alexander, H. G. ed. The Leibniz-Clarke Correspondence, Manchester U.P. Buchdahl, G. Metaphysics and the Philosophy of Science. Blackwell,

Copleston, A. A History of Philosophy. Vol. 4. Doubleday. Curley, E. M. Spinoza's Metaphysics: An Essay in Interpretation. Harvard Ú.P.

Hallett, H. F. Spinoza: The Elements of His Philosophy. Athlone.

Joseph, H. W. B. Lectures on the Philosophy of Leibniz. Clarendon.

Leibniz, G. W. Monadology and Other Writings. Latta, R. H. ed. O.U.P. Leibniz, G. W. Discourse on Metaphysics. Lucas, P. G. & Grint, L. eds.

Manchester U.P. Parkinson, G. H. R. Logic and Reality in Leibniz's Metaphysics. Clarendon. Parkinson, G. H. R. Spinoza's Theory of Knowledge. Clarendon. Rescher, N. The Philosophy of Leibniz. Prentice-Hall. Russell, B. The Philosophy of Leibniz. Allen & Unwin.

Saw, R. L. Leibniz. Pelican.

Saw, R. L. The Vindication of Metaphysics. Macmillan.

Spinoza, B. Earlier Philosophical Writings. Hayes, F. A. trans. Library of Liberal Arts.

Set Theory (Session 1)

Prerequisite: Predicate Logic.

An axiomatic development of Zermelo-Fraenkel set theory, including a construction of the natural numbers, equinumerosity, ordinal and cardinal numbers, the axiom of choice and some of its consequences.

TEXTBOOK

Suppes, P. Axiomatic Set Theory. Van Nostrand.

REFERENCE BOOKS

Bernays, P. & Fraenkel, A. A. Axiomatic Set Theory. North Holland.

Fraenkel, A. A. Abstract Set Theory. North Holland. Fraenkel, A. A. & Bar-Hillel, Y. Foundations of Set Theory. North Holland. Halmos, P. Naive Set Theory. Van Nostrand.

Kalish, D. & Montague, R. Logic: Techniques of Formal Reasoning. Harcourt, Brace & World.

Quine, V. W. Set Theory and Its Logic. Harvard U.P.

Russell, B. Introductions to Mathematical Philosophy. Allen & Unwin.

Sierpinski, W. Cardinal and Ordinal Numbers. Polish Scientific Publishers.

Honours Seminar A (Session 2)

For Honours students in their second year. The course is based on articles from recent issues of philosophy journals. Students will be expected to read and prepare papers on an individual basis.

REFERENCE BOOKS

To be advised in class.

SCHOOL OF PHYSICS

The School of Physics provides both pass and honours courses. The pass course with major studies is available by taking Physics or Higher Physics units and may be completed in three years. This course may include the core units which aim to present a broad and balanced treatment of all branches of physics without undue emphasis on topics which may be temporarily prominent, and also a choice of elective units which aim to present more specific and detailed study in certain specialized areas. The course including Higher Physics units is normally a prelude to entry into the Honours year. These studies which are completed within the framework of the Science Course (see earlier) provide unit groupings which are appropriate for students seeking qualification as professional physicists, whether they intend to engage in research, industrial practice or the teaching of science.

A student intending to take a pass degree with a major in Physics must complete Physics 1.001, Physics units 1.112A, B and C, and four level III Physics units of which three must be from 1.113A, B, C or D. Note that 10.001 Mathematics is a prerequisite of all Physics level II units and that 10.211A Applied Mathematics is a co-requisite of all Physics level II units. Students are also advised to take units 10.111A and 10.111B of Pure Mathematics in second year. Additional Mathematics units are prerequisite to Higher Physics level III units (see regulations). Students are also advised to complete supporting units in accordance with the Science Course regulations and will normally include 2.001 Chemistry I. It should be understood that units of corresponding higher subjects can often be substituted for those mentioned above.

HONOURS

A student intending to take Honours in Physics will normally complete the sequence of Higher Physics units 1.011; 1.122 A, B and C; 1.123 A, B, C and D. However, students with a very good record in Physics 1.001 or in 1.112 A, B and C may be considered for admission to Higher Physics units on application to the Head of School. Applied Mathematics 10.211A (or the Higher Applied Mathematics equivalent) is a co-requisite of Higher Physics level II units and Pure Mathematics 10.111A and B (or the Higher Pure Mathematics equivalents) are prerequisites to Higher Physics level III units. Students are also strongly advised to take Applied Mathematics units 10.212A and D (or equivalents) in their third year of study.

The following show typical programmes which, together with the prescribed General Studies subjects, complete requirements for a degree.

A. Pass Course Majoring in Physics (suitable for Science Teachers)

Level No. Units

FIRST YEAR

Physics	I	2
Mathematics	Ι	2
Chemistry	Ι	2
General Biology	I	2

SECOND YEAR			
		No. Units	
Physics	II	3	
Pure Mathematics	II	2	
Applied Mathematics	ų	$\frac{1}{2}$	
Geoscience	I	2	
THIRD YEAR			
Physics	ш	4	
Chemistry	II	3	
Other Units	II/III	1	
OR			
B. Pass Course Majoring in Physics			
D. 1855 Course Majoring in Thysics	Loval	No. Units	
FIRST YEAR	Level	NO. Onits	
	_	_	
*Physics	Ī	2 2 2 2	
Mathematics	I I	2	
Chemistry Other Units	Ĩ	2	
Other Onits	1	2	
SECOND YEAR			
*Physics	п	3	
Applied Mathematics Unit A	ÎÎ	ĭ	
Pure Mathematics Unit B	II	1	
Other Units	I/II	4	
THIRD YEAR			
†Physics	ш	4	
Pure Mathematics Unit A (if not previously taken)	11	1	
Other Units	II/III	2 or 3	
C. Leading to Honours in Physics			
C. Leading to Honours in Physics	Level	No. Units	
FIRST YEAR	Lever	NO: Onits	
	-	•	
Higher Physics	I	2 2 2	
Mathematics	Í	2	
Other Units	Î	$\overline{2}$	
SECOND YEAR			
Higher Physics	II	3 2	
Pure Mathematics Units A, B	II		
Applied Mathematics Other Units		1 2	
	11	<i>L</i>	
THIRD YEAR			
†Higher Physics A, B, C and D	III	4	
Applied Mathematics	III	1	
Physics Electives	III	1	
Or other Units	ш	3	
For footnotes, see overleaf.			

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- * Admission to Physics level II units requires completion of 1.001 Physics or 1.011 Higher Physics. Admission to Higher Physics II units normally requires completion of 1.011, Higher Physics I, but students who complete 1.001 Physics I at a high standard and pass 10.001 Mathematics I or 10.011 Higher Mathematics I may be admitted with the approval of the Head of School.
- [†] Students must note that certain Applied Mathematics level III units cannot be counted with certain Physics level III units.

Physics Prizes

The following prizes are offered annually:

The School Prize, for the best overall performance in Physics level II units or Higher Physics level II units, value \$40.

The Physics Staff Prize, for the best overall performance in Physics level III Units or Higher Physics level III Units. Value \$60.

The Head of School's Prize, for the best performance in laboratory work at level III in the School of Physics, value \$20.

The Physics IV Prize for the best performance in Physics IV, value \$40.

1.001 Physics I

Aims and nature of physics and the study of motion of particles under the influence of mechanical, electrical, magnetic and gravitational forces. Concepts of force, inertial mass, energy, momentum, charge, potential, fields. Application of the conservation principles to solution of problems involving charge, energy and momentum. Electrical circuit theory, application of Kirchoff's Laws to AC and DC circuits. Uniform circular motion, Kepler's Laws and rotational mechanics.

The application of wave and particle theories in physics. A review of the atomic theory of matter and the structure and properties of atomic nuclei. A molecular approach to energy transfer, kinetic theory, gas laws and calorimetry. The wave theories of physics, transfer of energy by waves, properties of waves. Application of wave theories to optical and acoustical phenomena such as interference, diffraction and polarization. Interaction of radiation with matter, photoelectric effect, Compton effect, spectroscopy. Resolution of the wave — particle paradox by means of wave mechanics and the uncertainty principle.

TEXTBOOK

Bueche, P. Introduction to Physics for Scientists and Engineers. McGraw-Hill.

REFERENCE BOOKS

- Ference, M., Lemon, H. & Stephenson, R. J. Analytical Experimental Physics. Chicago U.P.
- Halliday, D. & Resnick, R. Physics for Students of Science and Engineering. Vols. I & II. Wiley.
- Wiedner, R. T. V. & Sells, R. L. Elementary Classical Physics. Vols. I and II. Allyn & Bacon.

1.011 Higher Physics I

Kinematics—Non-uniformly accelerated systems. Centripetal acceleration. Laws of motion. Momentum. Impulse. Potential and kinetic energy. Power. Conditions of equilibrium. Elasticity. Young's bulk and shear moduli. Poisson's ratio. Strain energy. Hydrodynamics. Bernouilli's equation. Motion in resistive medium. Moments of inertia. Rotational dynamics. Simple harmonic motion. Pendulums. Motion about free axis. Progressive and stationary waves. Energy current. Superposition of waves. Doppler effect. Resonance. Huygen's principle. Reflection, refraction, interference and diffraction of waves. Electromagnetic spectrum. Polarization.

Electrostatics—Gauss' theorem. Electric intensity. Capacitance. Electromagnetism. Biot-Savart and Ampere's circuital laws. Force on moving charge and on conductor. Torque on coil. D.C. instruments. Electromagnetic induction. Faraday's and Lenz's laws. Self and mutual inductance. D.C. circuits. Kirchoff's rules and Thevenin's theorem. Growth and decay of current. A.C. circuits. Resonance. Diode. Triode. Amplifiers and oscillators. Electronic measuring instruments.

TEXTBOOKS

Halliday, D. & Resnick, R. *Physics for Students of Science and Engineering*. Vols. I & II, or combined volume. Wiley.

Russell, G. J. & Mann, K. Alternating Current Circuit Theory. N.S.W.U.P. Spiegel, M. R. Theory and Problems of Theoretical Mechanics. Schaum.

REFERENCE BOOKS

Brophy, J. J. Basic Electronics for Scientists. McGraw-Hill. Paperback.

- Feynman, R. P., Leighton, R. B. & Sands, M. The Feynman Lectures on Physics. Vols. I & II. Addison-Wesley.
- Tomboulian, D. H. Electric and Magnetic Fields. Harcourt, Brace & World, 1965.

Physics Level II Units

1.112A Electromagnetism

Electrostatics in vacuum and in dielectrics. Magnetostatics in vacuum and in magnetic materials. Maxwell's equations and simple applications.

TEXTBOOK

Reitz, J. R. & Milford, F. J. Foundation of Electromagnetic Theory. 2nd ed. Addison-Wesley.

1.112B Modern Physics

Special relativity. Quantum theory. Schrödinger wave equation and simple applications. Atomic and nuclear physics. Nuclear reactions.

TEXTBOOK

Beiser, A. Perspectives of Modern Physics. Rev. ed. McGraw-Hill, 1969.

1.112C Waves in Continuous Media and Thermodynamics

Waves in continuous media: Oscillations and forced vibrations, Fourier analysis, travelling waves and wave packets. *Thermodynamics*: First and second laws of thermodynamics. Thermodynamic functions and simple applications. Statistical foundations of thermodynamics.

TEXTBOOKS

French, A. P. Vibrations and Waves. Nelson, 1971. Mandl, F. Statistical Physics. Wiley, 1971.

1.122A Electromagnetism

Further electrostatics. Poisson's and Laplace's equations. Ferro-magnetism. Maxwell's equations and application to waves in isotropic dielectrics. Poynting vector.

TEXTBOOK

Corson, D. & Lorrain, P. Introduction to Electromagnetic Fields and Waves. Freeman, 1962.

1.122B Quantum Physics

Syllabus as for 1.112B but treated at a higher level and including some solid state physics.

TEXTBOOK

Eisberg, R. M. Fundamentals of Modern Physics. Wiley, 1961.

1.122C Thermodynamics and Mechanics

Thermodynamics: as for 1.112C Thermodynamics but at higher level and with some additional topics. *Mechanics:* oscillations and forced vibrations. Lagrange's equation, variational principles, Hamilton's equations. *Note:* 1.122A, B and C are units of Higher Physics II and the prerequisite is normally 1.011 Physics.

TEXTBOOKS

Symon, K. R. Mechanics. 2nd ed. Addison-Wesley, 1965. Reif, F. Fundamentals of Statistical and Thermal Physics. McGraw-Hill, 1965.

1.212 Physics IIT

Any two of the following half-units.

1.212A Geometrical Optics

Reflection, Refraction. Thin and thick lenses and lens systems. Instruments and their aberrations. Photometry.

TEXTBOOK

Fincham, W. Optics. Hatton.

1.212B Electronics

Vacuum tubes and applications. Conduction in solids; solid state diodes, transistors, amplifiers, feed back.

TEXTBOOK

Smith, R. J. Circuits, Devices and Systems. 2nd ed. Wiley, 1972.

1.212C Introduction to Solids

Introductory quantum mechanics and atomic physics; crystal structure; point and line defects; introductory band theory; conductors, semiconductor and insulators; energy level diagrams.

TEXTBOOK

Rudden, M. N. & Wilson, J. A Simplified Approach to Solid State Physics. Butterworths, 1971.

1.212D Biophysics

Thermodynamic equilibrium, the first and second law of thermodynamics, chemical potential. Transport phenomena. Diffusion of electrolytes through membranes, bioelectric potentials. Excitability in nerve. Fixed charge systems. Active transport. Experimental techniques of Biophysics.

TEXTBOOK

To be announced in class.

Physics Level III Units

1.113A Wave Mechanics and Spectroscopy

Concepts, harmonic oscillator, uncertainty principle, the free particle, barriers, the hydrogen atom, many electron atoms, removal of degeneracy, spectroscopy, molecules, periodic potentials, band structure, perturbations.

TEXTBOOK

Beiser, A. Perspectives of Modern Physics. Rev. ed. McGraw-Hill, 1969.

1.113B Electromagnetic Fields and Physical Optics

Wave equation; propagation in dielectrics and ionized media; reflection and transmission; guided waves; coherence of radiation; interaction of radiation with matter; stimulated emission; laser oscillators; properties of laserlight; interferometry; diffraction; convolution theorem X-ray and neutron diffraction.

TEXTBOOK

Lipson, H. & S. S. Optical Physics. C.U.P., 1969.

1.113C Statistical Mechanics and Solid State

Thermodynamic potentials, ensembles and partition functions, lattice vibrations, the grand canonical ensemble, Pauli exclusion principle, Bose-Einstein and Fermi-Dirac distributions.

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Structure of crystals, imperfections, specific heat. Band theory of solids, semiconductors.

TEXTBOOKS

Blakemore, J. S. Solid State Physics. Saunders, 1969. Jackson, E. A. Equilibrium Statistical Mechanics. Prentice-Hall, 1968.

1.113D Astrophysics and Nuclear Physics

The observational environment, optical astronomy, radio astronomy, X-ray astronomy, stellar evolution, radio sources, the sun.

Detecting instruments and accelerators for nuclear particles, radioactive processes, nuclear reactions, angular distributions, mesons, baryons, excited nuclear states.

TEXTBOOK

Tayler, R. J. The Stars: Their Structure and Evolution. Wykeham Sci. Series.

Higher Physics Level III Units

1.123A Quantum Mechanics

Concepts, measurements, expectation values, wave mechanics, matrix mechanics, free particle and barrier problems, hydrogen atom spin, exclusion principle, stationary and time dependent perturbation methods, scattering. Born approximation and partial waves.

TEXTBOOK

Schiff, L. I. Quantum Mechanics. 2nd ed. McGraw-Hill.

1.123B Electromagnetic Theory and Statistical Mechanics

Metallic boundary conditions, eigenfunctions and eigenvalues, cavities, wave guides, scattering by a conductor, wave equation for potentials, radiation fields, Hertz potential, dipole and multi-pole radiation, radiated energy and angular momentum.

Statistical mechanics: Kinetic theory, the Boltzmann equation, Maxwell-Boltzmann distribution, Boltzmann's H-theorem; classical statistical mechanics: postulates, equipartition, ensembles, difficulties; quantum statistical mechanics; postulates, ensembles, Fermi and Bose statistics.

TEXTBOOK

Corson, D. & Lorrain, P. Introduction to Electromagnetic Fields and Waves. Freeman, 1962.

REFERENCE BOOK

Knox, J. H. Molecular Thermodynamics. Wiley.

1.123C Solid State and Nuclear Physics

Crystallography, binding energy, phonons, lattice conduction, free electron gas, band theory.

Nuclear models, binding energy, nuclear forces, elementary particles, nuclear reactions, radioactive decay.

TEXTBOOKS

Burcham, W. E. Nuclear Physics, an Introduction. Longmans, 1963. Kittel, C. Introduction to Solid State Physics. 3rd ed. Wiley, 1967.

1.123D Atomic Physics and Spectroscopy

Collision parameters, transport coefficients, potential functions, atomic collisions, scattering of heavy particles, scattering of electrons, avalanche formation, recombination, radiation processes, stimulated emission, detectors.

Spectrum of hydrogen, fine structure, electron spin, vector treatment of spectroscopy, emission and absorption of radiation, diatomic molecules.

TEXTBOOKS

McDaniel, E. W. Collision Processes in Ionised Gases. Wiley, 1964. White, H. W. Introduction to Atomic Spectra. McGraw-Hill, 1934.

Physics Level III Supplementary Units

1.133A Electronics

A.C. circuit analysis, band theory of semiconductors, diode, field effect transistor, rectifier circuits, power supplies, single and multistage amplifiers, positive feedback, oscillators.

TEXTBOOKS

Delaney, C. F. G. Electronics for the Physicist. Penguin, 1969. Russell, G. J. & Mann, K. Alternating Current Circuit Theory. N.S.W.U.P. Transistor Manual. General Electric Co. 1972 or 1971.

1.143A Biophysics

Ear and sound, eye and light, impulses by nerves, the brain, hearing, vision muscles, heart-beat, structure of proteins, nucleic acid, radiation effects, enzymes, diffusion and permeability.

TEXTBOOK

Ackerman, E. Biophysical Science. Prentice-Hall, 1962.

1.143B Solid State Devices and Electronics

(Syllabus follows on from 1.133A which is a prerequisite.)

Generalized amplifiers, negative feedback, special amplifiers, regulated power supplies, modulation, pulse circuits, silicon-controlled rectifier circuits, instruments.

TEXTBOOKS

Gibbons, J. F. Semiconductor Electronics. McGraw-Hill, 1966. S.C.R. Handbook. General Electric Co. Van der Ziel, A. Introduction to Electronic Circuits. Allyn & Bacon, 1969.

1.143C Magnetism

Diamagnetism, paramagnetism, ferromagnetism, antiferromagnetism, domains, technical magnetism, geomagnetism, magnetic resonances as a tool in solid state research.

TEXTBOOK No set text.

1.143D Conceptual Framework of Physics

Physics and Metaphysics: The place of speculation in theory formation. Space and Time: Systems of coordinates, the nature and arrow of time, parity, micro causality. Fundamental Physical Phenomena: The fundamental phenomena on which physical theories have been based; electrical, gravitational, inertial nuclear and entropy/probability. Field Theory: In particular e.m. and gravitational field theory. Mathematical formalization of physical phenomena, action at a distance, field propagation, field energy, connection to relativity. Relativity: The fundamental postulates, simultaneity, limiting speeds, connection with field theory, mass and energy. Relationship between Micro- and Macro-Cosmos: Divisibility of matter (molecules, atoms, nuclei, nucleon), matter and anti matter, statistical nature of the behaviour of large aggregates or systems, the concept of entropy, the second law of thermodynamics. The place of determinism in physics. Matter and Energy: Conservation laws, inertial mass, equivalence principle, field energy, spatial delimitation of material particles. Theory of Quantum Processes: Granularity effects, uncertainty principle, effects of measurements, virtual processes. Determinism vs. indeterminism in physics, application to nuclear phenomena.

TEXTBOOK

No set text.

1.143E Electrical and Optical Properties of Solids

Equilibrium properties of semiconductors and insulators, conductivity, excess carriers, flow equations, contact barriers; luminescence, relaxation phenomena.

TEXTBOOKS

Adler, R. B., Smith, A. C. & Longini, L. D. Introduction to Semiconductor Physics. Wiley, 1964.

Blakemore, J. S. Solid State Physics. Saunders, 1969.

Gray, P. E., De Witt, D., Boothroyd, A. R. & Gibbons, J. F. Physical Electronics and Circuit Models of Transistors. Wiley.

1.143F Marine Acoustic and Seismic Methods (Oceanography Unit)

Prerequisite: 10.211A or 10.221A or 10.031.

Cave Theory: General wave equation for fluids, viscoelastic media and solids. Travelling and standing wave solutions. Wave Guides: Fluid and solid waveguides, ray and mode theories. Sound Transmission in the Ocean: Applications of reflection and refraction theory, scattering and diffraction effects. Experiments relating to the above: Including fluid waveguide, solid waveguide, measurement of absorption and reflection coefficients.

Ray Theory Interpretation and Applications: Seismic refraction methods, seismic reflection methods, computational methods. Instrumentation and Processing: Seismic and acoustic sources, recording systems, signal processing. Geological and Physical Interpretation. Practical work relating to the above: Instrumentation, recording and interpretation of field data.

REFERENCE BOOKS

Ewing, W. M., Jardetzky, W. S. & Press, F. Elastic Waves in Layered Media. McGraw-Hill, 1957.

Officer, C. B. Introduction to the Theory of Sound Transmission. McGraw-Hill, 1958.

Redwood, M. Mechanical Waveguides. Pergamon, 1960.

Physics Higher Level III Supplementary Units 1.153A Hydrodynamics and Magnetohydrodynamics

Not offered in 1974.

1.153B Relativity and Electromagnetism

Scalars and vectors in non-Cartesian frames. Principle of relativity and signal propagation. Space-time. Four-vectors. Mass-energy. Four-momentum. Electromagnetic field equations. Gauges. Wave equation. Solutions. Introductory tensors. Field tensor. Stress tensor. Four-momentum of free field. Moving charges. Electromagnetic mass.

TEXTBOOK No set text.

1.114 Physics IV (Honours)

Four compulsory courses:

Quantum Mechanics TEXTBOOK Schiff, L. F. Quantum Mechanics. 3rd ed. McGraw-Hill, 1968.

Solid State Physics TEXTBOOKS Goldsmid, H. J. Problems in Solid State Physics. Pion, 1968. Kittel, C. Introduction to Solid State Physics. 4th ed. Wiley, 1971.

Low Energy Nuclear Physics TEXTBOOK Blatt, J. M. & Weisskopf, V. F. Theoretical Nuclear Physics. Wiley, 1952.

Statistical Mechanics TEXTBOOK Huang, K. Statistical Mechanics. Wiley, 1963.

Four electives chosen from: Non-equilibrium Statistical Mechanics TEXTBOOK No set text.

Methods of Solid State Physics TEXTBOOK No set text.

Solid State Applications TEXTBOOK Beeforth, T. H. & Goldsmid, H. J. Physics Solid State Devices. Pion-London, 1970.

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Dielectric and Defect Properties of Solids TEXTBOOK No set text.

Biophysics
TEXTBOOKS
Katchalsky, A. & Curran, P. F. Non Equilibrium Thermodynamics in Biophysics. Harvard U.P., 1965.
Prigogine, I. Introduction to Thermodynamics of Irreversible Processes. Interscience, 1967.

Physical Acoustics TEXTBOOK No set text.

Quantum Electrodynamics TEXTBOOK No set text.

Nuclear Fields TEXTBOOK No set text.

Diffraction Theory TEXTBOOK No set text.

Structural Properties of Materials TEXTBOOK No set text.

SCHOOL OF PHYSIOLOGY AND PHARMACOLOGY

FOR STUDENTS IN THE SCIENCE COURSE

Courses in physiology for students in the Science Course were commenced in 1963. During the second and third years of their BSc course, students may take Principles of Physiology and Physiology II. Students reaching an adequate standard in these subjects may proceed to a BSc degree with honours by taking Physiology III during the fourth year of the course.

For details of level, unit value, when offered, hours per week, prerequisites and co-requisites, see page B27.

73.011A Principles of Physiology

An introductory course in physiology. It considers in some detail the basic problems of homeostasis encountered in man and animals. Function is considered at cellular and systemic levels, and examples are drawn from mammalian and invertebrate species.

TEXTBOOK

Vander, A. J., Sherman, J. H. & Luciano, D. S. Human Physiology. McGraw-Hill, 1970. (\$12.50).

73.012 Physiology II

An advanced course in the principles of physiology, centred on four major areas: circulation, respiration, membrane biophysics and neuro-physiology.

TEXTBOOKS

Lists given from time to time.

The course consists of lectures, tutorials, seminars and laboratory work. On some occasions students may be required to attend at other times for the maintenance and treatment of experimental animals. The study of psychology as a formal discipline in undergraduate courses is traditional in Australian Universities. Psychology as a subject is concerned with the systematic study of human behaviour and associated mental processes. The School of Psychology offers psychology as a major subject in the full-time Arts and Science Courses and also offers a fulltime undergraduate professional degree course in psychology in the Faculty of Biological Sciences.

In the Science course, Psychology may be studied as a major sequence in accordance with Science course regulations. A student who wishes to proceed to Honours in Psychology should refer to the requirements set out in Clause 2(b) of the same regulations.

The four-year course in Psychology, which leads to the degree of Bachelor of Science in the Faculty of Biological Sciences, is a full-time course designed to meet the requirements of students who intend to become professional psychologists, as either practitioners or research workers. It provides extensive study of psychological theory and practice, supported by an appropriate selection of other subjects. Full details of this course are given on pages B17-19.

Prizes in Psychology

The Australian Psychological Society Prize in Psychology is awarded annually to a fourth-year student. The Society also awards annually twoyears Student Subscriberships to the Australian Journal of Psychology to two outstanding students at the end of second year.

The Staff Prize in Psychology is awarded annually to an outstanding second-year student.

12.001 Psychology I

An introduction to the content and methods of psychology as a behavioural science, with special emphasis on (a) the biological and social bases of behaviour, (b) learning, and (c) individual differences.

The course includes training in methods of psychological enquiry, and the use of elementary statistical procedures.

Part A: Theory

TEXTBOOKS

There is no set text for this course, but students are advised to obtain, as a general reference, an introductory text such as one of the following:— C.R.M. *Psychology Today*. CRM Books, Del Mar, 1970, or 2nd ed., 1972. Dember, W. N. & Jenkins, J. J. *General Psychology*. Prentice-Hall, 1970. Morgan, C. T. & King, R. A. *Introduction to Psychology*. 4th ed. McGraw-Hill, 1971.

Recommended as an additional text for intending honours students: Hebb, D. O. *Textbook of Psychology*. 3rd ed. Saunders, 1972.

REFERENCE BOOKS

Allport, G. W. Pattern and Growth in Personality. Holt, 1961.

Allport, G. W. Personality. Holt. 1937.

Anastasi, A. Individual Differences. Wiley, 1965.

Beech, H. R. Changing Man's Behaviour. Penguin, 1969. Coopersmith, S. Frontiers of Psychological Research. Readings from Scientific American. Freeman, 1964.

Cronbach, L. J. Essentials of Psychological Testing. 2nd ed., 1960, or 3rd ed., 1970. Harper & Row.

Deese, J. Psycholinguistics. Allyn & Bacon, 1970.

Lazarus, R. S. Personality. 2nd ed. Prentice-Hall, 1971.

McGaugh, J. L., Weinberger, N. M. & Whalen, R. E. Psychobiology: The Biological Bases of Behavior. Readings from Scientific American. Freeman, 1966.

McKinney, F. Understanding Personality: Cases in Counselling. Houghton, 1965.

Reynolds, G. Primer of Operant Conditioning. Scott, Foresman, 1968.

Vernon, P. E. Personality Assessment: A Critical Survey. Methuen, 1964.

Vernon, P. E. Personality Tests and Assessments. Methuen, 1953. Walker, E. L. Conditioning and Instrumental Learning. Brooks/Cole, 1967.

Part B: Practical

TEXTBOOK

Lumsden, J. Elementary Statistical Method. Univ. of W.A. Press, 1969.

12.004 Psychology IV

Psychology IV in the BSc (Psychology) course. A program of selected study from the School's Advanced Electives. In consultation with the Head of School or his representative, plus a research thesis or project. Combina-tions of electives are available for students intending to specialize in areas of professional practice (e.g., clinical, industrial, educational psychology) or in research.

TEXTBOOKS

To be determined in consultation with Head of School.

12.042 Psychology IIA

(BSc Psychology students only.)

Observational methods and laboratory instrumentation in applied psychology.

REFERENCE BOOKS

Bradford, L. P., Gibb, J. R. & Benne, K. D. T-Group Theory and Laboratory Method. Wiley, 1964.

Kleinmuntz, B. Personality Measurement. Dorsey, 1967.

Richardson, S. A., Dohrenwend, E. S. & Klein, D. Interviewing: Its Forms and Functions. Basic Books, 1965.

Shouksmith, G. Assessment Through Interviewing. Pergamon, 1968. Sidowski, J. B. Experimental Methods & Instrumentation in Psychology. McGraw-Hill, 1966.

Webb, E. J., Campbell, D. T., Schwartz, R. D. & Seechrest, L. Unobtrusive Measures: Non-reactive Research in the Social Sciences. Rand McNally, 1966.

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12.152 Research Methods II

General introduction to the design and analysis of experiments; hypothesis testing, estimation, power analysis; general treatment of simple univariate procedures; correlation and regression.

TEXTBOOKS

Anderson, B. F. The Psychology Experiment: An Introduction to Scientific Method, 2nd ed. Brooks/Cole, Belmont, California, 1971.

Welkowitz, J., Ewen, R. B. & Cohen, J. Introductory Statistics for the Behavioral Sciences. Academic, 1971.

REFERENCE BOOKS

Bachrach, A. J. Psychological Research: An Introduction. 3rd ed. Random House, 1972.

Hays, W. L. Statistics. Holt Int. Ed. Holt, Rinehart & Winston, 1969. Lumsden, J. Elementary Statistical Method. Univ. of W.A. Press, 1969.

McGuigan, F. J. Experimental Psychology: A Methodological Approach. 2nd ed. Prentice-Hall, 1968.

Underwood, B. J. Psychological Research. Appleton-Century-Crofts, 1957.

12.153 Research Methods IIIA

Introduction to analysis of variance-one way and complete factorial designs. Elementary Fortran programming, emphasizing editing of data for use in package programmes.

TEXTBOOKS

Hays, W. L. Statistics. International ed. Holt, Rinehart & Winston, 1969. and either

Blatt, J. M. Introduction to Fortran IV Programming: Using the Watfor Computer. Goodyear Pub. Coy, 1968 (or later).

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Lee, R. M. A Short Course in Fortran IV Programming. McGraw-Hill, 1967.

12.163 Research Methods IIIB

Experimental Design; complex analysis of variance; planned and post hoc comparisons: multivariate procedures as data reduction techniques.

TEXTBOOKS

Hays, W. L. Statistics. International Ed. Holt, Rinehart & Winston, 1969. Lieberman, B. Contemporary Problems in Statistics. O.U.P., 1971.

REFERENCE BOOKS

Nunnally, J. Psychometric Theory. McGraw-Hill, 1967.

Winer, B. J. Statistical Principles in Experimental Design, McGraw-Hill, 1962.

12.252 Learning II

History of the study of learning. Basic phenomena of conditioning and verbal and perceptual motor learning. Problems and applications.

REFERENCE BOOKS

As for 12.253 Learning IIIA.

12.253 Learning IIIA

Current experimental and theoretical problems in learning; classical and operant conditioning; reinforcement issues.

REFERENCE BOOKS

Kimble, G. A. Hilgard and Marauis' Conditioning and Learning. Appleton. 1961.

Marx, M. H. ed. Learning: Processes. Macmillan, 1969.

Mednick, S. A. Learning. Foundations of Modern Psychology series. Prentice-Hall, 1964.

Pavlov, I. P. Conditioned Reflexes. Dover, 1960.

Reynolds, G. S. A Primer of Classical Conditioning. Scott, Foresman, 1968. Smith, W. S. & Moore, J. W. Conditioning and Instrumental Learning.

McGraw-Hill, 1966.

Smith, W. S. & Rohrman, N. L. Human Learning. McGraw-Hill, 1970.

12.263 Learning IIIB

Human learning processes. Biochemistry of learning, and aversive control of behaviour.

REFERENCE BOOKS

Black, A. H. & Prokasy, W. F. eds. Classical Conditioning II. Appleton. 1972.

Bolles, R. C. Theory of Motivation. Harper, 1967. Honig, W. K. ed. Operant Conditioning. Appleton, 1966.

Kimble, G. A. Hilgard and Marquis' Conditioning and Learning. Appleton, 1961.

Pavlov, I. P. Conditioned Reflexes. Dover, 1960.

Prokasy, W. F. ed. Classical Conditioning. Appleton, 1965.

Walters, R. H., Cheyne, J. A. & Banks, R. K. Punishment. Penguin, 1972.

12.302 Personality II

History of the study of personality. Developmental influences. Tests and assessments and their applications in the prediction of behaviour.

REFERENCE BOOKS

As for 12.303 Personality IIIA.

12.303 Personality IIIA

The development and structure of personality, with reference to bio-logical and social determinants. Problems of theory and measurement.

REFERENCE BOOKS

Maior Reference Books

Lazarus, R. S. & Opton, E. M. Personality. Penguin, 1967.

Mischel, W. Introduction to Personality. Holt, Rinehart & Winston, 1971. Sarason, I. G. Personality: An Objective Approach. Wiley, 1966. Sarnoff, I. Personality Dynamics and Development. Wiley, 1962.

This course cannot be covered adequately by any one book. Conse-quently reference books only are listed. The first four of these are considered to be the more significant. It is suggested that students could form themselves into syndicates and thereby acquire these four reference books.

Other Reference Books

Baughman, E. E. Personality: The Psychological Study of the Individual. Prentice-Hall, 1972.

Bischof, L. J. Adult Psychology. Harper & Row. 1969.

- Blum, G. S. Psychodynamics: The Science of Unconscious Mental Forces. Wadsworth, 1966.
 Donelson, E. Personality: A Scientific Approach. Meredith Corporation, N.Y., 1973.
- Lazarus, R. S. Patterns of Adjustment & Human Effectiveness. McGraw-Hill, 1969.

Lundin, R. W. Personality: A Behavioral Analysis. Collier-Macmillan, 1969.

McCurdy, H. G. The Personal World. Harcourt, 1961. McNeil, E. B. The Concept of Human Development. Wadsworth. Martin, W. & Stendler, C. R. Readings in Child Development. Harcourt, Mathin, W. & Steinler, C. R. Realings in Control Decomposition of the process of the proces of the pro

Vernon, P. E. Personality Assessment. Methuen, 1964.

12.313 Personality IIIB

The psychology of interpersonal relationships and transactions. Techniques of interpersonal influence.

TEXTBOOKS

Blank, L., Gottsengen, G. B. & Gottsengen, M. G. eds. Confrontation-Encounters in Self and Interpersonal Awareness. Macmillan, 1971.

Rogers, C. R. & Stevens, B. Person to Person: The Problem of Being Human. A New Trend in Psychology. Real People Press, Lafayette, California, 1967.

12.322 Motivation II

The spectrum of human motivation and emotion: hunger, sex, fear, stress, achievement, altruism, personal causation.

TEXTBOOK

Cofer, C. N. Motivation and Emotion. Scott, Foresman, 1972.

12.323 Motivation IIIA

A study of the conditions governing the arousal and direction of behavioural sequences, with particular reference to the social determinants of the goals of behaviour.

TEXTBOOKS

Cofer, C. N. Motivation and Emotion. Scott, Foresman, 1972. Russell, W. A. ed. Milestones in Motivation: Contribution to the Psychology of Drive and Purpose. Appleton-Century-Crofts, 1970.

REFERENCE BOOKS

Atkinson, J. W. ed. Motives in Fantasy, Action and Society. Van Nos-trand, 1958.

- Atkinson, J. W. & Feather, N. eds. Theory of Achievement Motivation. Wiley, 1966. Berkowitz, L. Aggression: A Social Psychological Analysis. McGraw-Hill,
- 1962.
- Bolles, R. C. Theory of Motivation. Harper, 1966. Cofer, C. N. & Appley, M. H. Motivation Theory and Research. Wiley, 1964.

- De Charms, R. Personal Causation. Academic, 1969. Haber, R. N. ed. Current Research in Motivation. Holt, 1966. Heckhausen, H. The Anatomy of Achievement Motivation. Academic, 1967. Irwin, F. W. Intentional Behaviour and Motivation: A Cognitive Theory.
- Lippincott, 1971.
- Lazarus, R. S. Psychological Stress and the Coping Process. McGraw-Hill, 1966.
- Macaulay, J. & Berkowitz, L. eds. Altruism and Helping Behaviour. Academic, 1970.
- Mednick, M. T. & Mednick, S. A. eds. Research in Personality. Holt, 1963. McClelland, D. C. ed. Studies in Motivation. Appleton, 1955. Murray, E. J. Motivation and Emotion. Foundations of Modern Psycho-
- logy Series. Prentice-Hall, 1964.
- Peters, R. S. The Concept of Motivation. Routledge, 1958. Ryan, T. A. Intentional Behaviour: An Approach to Human Motivation. Ronald Press, 1970.
- Stacey, C. L. & De Martino, M. F. eds. Understanding Human Motivation. Allen, 1958 & 1963.
- Wright, D. Psychology of Moral Behaviour. Penguin, 1971.
- Yates, A. J. Frustration and Conflict. Methuen, 1963.

Young, P. T. Motivation and Emotion. Wiley, 1961.

12.372 Psychological Assessment II

Introduction to the theory of psychological measurement. Properties of scales; elementary scaling methods. Test theory.

TEXTBOOKS

Gathercole, C. E. Assessment in Clinical Psychology. Penguin, 1968. Hays, W. L. Quantification in Psychology. Brooks/Cole, 1967.

12.373 Psychological Assessment (Testing) IIIA

Principles and techniques of psychological assessment. Types of tests and their application in selection and allocation procedures.

TEXTBOOK

Anastasi, A. Psychological Testing. Macmillan, 1968.

REFERENCE BOOKS

Cronbach, L. J. Essentials of Psychological Testing. 3rd ed. Harper, 1970. Helmstadter, G. C. Principles of Psychological Measurement. Appleton, 1964.

Miller, D. M. Interpreting Test Scores. Wiley, 1972.

Nunnally, J. C. Introduction to Psychological Measurement. McGraw-Hill, 1970.

Tyler, L. E. Tests and Measurements. Prentice-Hall, 1971.

Vernon, P. E. The Structure of Human Abilities. Methuen, 1961.

12.383 Psychological Assessment (Psychometric Theory) IIIB

Uni-dimensional and multi-dimensional scaling models. Test theory, factor analysis, and other multivariate methods.

TEXTBOOKS

Hammer, A. G. Elementary Matrix Algebra for Psychologists. Pergamon. Nunnally, J. Psychometric Theory. McGraw-Hill, 1967.

REFERENCE BOOKS

Cattell, R. B. ed. Handbook of Multivariate Experimental Psychology. Rand McNally, 1966.

Coombs, C. H. et al. Mathematical Psychology. An Elementary Introduction. Prentice-Hall, 1970.

Torgerson, W. S. Theory and Methods of Scaling. Wiley, 1958. Van der Geer, J.P. Introduction to Multivariate Analysis for the Social Sciences. Freeman, 1971.

Whitla, D. K. ed. Handbook of Measurement and Assessment in Behavioral Sciences. Addison-Wesley, 1968.

12.402 Physiological Psychology

(BSc Psychology students only.)

Neural and endocrinal bases of behaviour, psychophysiology, comparative behaviour (including ethology), behaviour genetics, with brief treat-ments of the behavioural effects of drugs and physiological malfunction.

TEXTBOOK

Milner, P. M. Physiological Psychology. Holt, 1970.

REFERENCE BOOK

Thompson, R. Foundations of Physiological Psychology. Harper, 1967.

12.412 Physiological Psychology II

Elementary neurophysiology and neuroanatomy. Neural and endocrinal bases of sensory and motor processes, motivation, emotion and learning.

TEXT AND REFERENCE BOOKS

As for 12.402 Physiological Psychology.

12.413 Physiological Psychology IIIA

Advanced treatment of physiology of motivation, emotion and learning with practical work. Psychopharmacology; psychophysiology; genetics and behaviour.

TEXT AND REFERENCE BOOKS

As for 12.402 Physiological Psychology.

12.452 Human Information Processing II

An introduction to psychophysical methods by which man's ability to make discriminations is measured, and the application of these methods to problems of perception, learning and memory.

TEXTBOOKS

McNicol, D. A Primer of Signal Detection Theory. Allen & Unwin, 1971. Norman, D. Memory and Attention. Wiley, 1969.

REFERENCE BOOKS

Annett, J. Feedback and Human Behaviour. Penguin, 1969. Mackworth, J. F. Vigilance and Habituation. Penguin, 1969. Mackworth, J. F. Vigilance and Attention. Penguin, 1970. Moray, N. Listening and Attention. Penguin, 1969. Neisser, U. Cognitive Psychology. Appleton-Century-Crofts, 1967.

12.453 Human Information Processing IIIA

A study of the stages involved in the reception of stimulus information from the environment, its analysis, storage, and translation into responses. Particular emphasis will be given to the perception and storage of verbal information. Topics include attention, vigilance, discrimination, memory and retrieval.

TEXT AND REFERENCE BOOKS

As for 12.452 Human Information Processing II.

12.472 Perception II

Modern approaches to the study of perception and their historical antecedants. Consideration will be given to the differing emphases of theories in terms of the phenomena considered and the variables utilized in the attempted explanations.

TEXTBOOK

Hochberg, J. E. Perception. Prentice-Hall, 1964.

REFERENCE BOOKS

Boring, E. G. A History of Experimental Psychology. 2nd ed. Appleton, 1950.

Dember, W. N. The Psychology of Perception. Holt, 1960.

Dodwell, P. C. ed. Perceptual Learning and Adaptation. Penguin, 1970. Gibson, E. J. Principles of Perceptual Learning and Development. Appleton-Century-Crofts, 1969. Gibson, J. J. The Senses Considered as Perceptual Systems. Allen &

Unwin, 1966.

Gregory, R. L. Eye and Brain. Weidenfeld & Nicolson, 1966.

Gregory, R. L. The Intelligent Eye. Weidenfeld & Nicolson, 1970.

Haber, R. N. ed. Contemporary Theory and Research in Visual Percep-

tion. Holt, Rinehart & Winston, 1968.

O'Neil, W. M. The Beginnings of Modern Psychology, Penguin, 1968.

12.473 Perception IIIA

The characteristics and processes of visual perception. Topics include the basic requirement for visual perception and the relative contributions of the observer and the stimulus in a range of visual situations.

TEXT AND REFERENCE BOOKS

As for 12.472 Perception II.

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12.483 Perception IIIB

Man in a spatial environment. A study of the organization and stability of the visual world with particular reference to the constancies, object movement, eye movement and locomotion.

TEXTBOOK

Haber, R. N. ed. Contemporary Theory and Research in Visual Perception. Holt, Rinehart & Winston, 1968.

REFERENCE BOOKS

Cornsweet, T. N. Visual Perception. Academic, 1970. Day, R. H. Human Perception. Wiley Australasia, 1969. Dodwell, P. C. ed. Perceptual Learning and Adaptation. Penguin, 1970. Graham, C. H. ed. Vision and Visual Perception. Wiley, 1965.

12.502 Social Psychology II

A study of behaviour in a variety of interpersonal situations. Culture and social structure as determinants of the range and content of behaviour.

TEXTBOOKS

Freedman, J. L., Carlsmith, J. M. & Sears, D. O. Social Psychology. Prentice-Hall, 1970.

Proshansky, H. M., Ittelson, W. H. & Rivlin, L. G. eds. Environmental Psychology: Man and his Physical Setting. Holt, Rinehart & Winston, 1970.

12.503 Social Psychology IIIA

The bases of interpersonal attraction; class and race as determinants of behaviour; the experimental study of social processes in small groups; cross-cultural studies of social influence.

TEXTBOOKS

As for 12.502 Social Psychology II.

12.552 Developmental Psychology II

A study of the acquisition, maintenance and modification of behaviour in the developing human organism.

TEXTBOOKS

Kennedy, W. A. Child Psychology. Prentice-Hall, 1971.

Telford, C. W. & Sawrey, J. M. The Exceptional Individual. Prentice-Hall. 1967.

REFERENCE BOOKS

Bischof, L. J. Adult Psychology. Harper & Row, 1969.

Clarke, A. & Clarke, A. D. B. Mental Deficiency: The Changing Outlook. Methuen, 1958. Critchley, M. The Diplexic Child. Heinemann, 1970.

C.R.M. Developmental Psychology Today. CRM Books, Del Mar, 1971. Danziger, K. Socialization. Penguin, 1971.

Ellis, N. ed. Handbook of Mental Deficiency. McGraw-Hill, 1963.

Garrison, K. C. & Force, D. G. Jnr. The Psychology of Exceptional Children, Ronald Press, 1965.

Hurlock, E. B. Developmental Psychology. 3rd ed. or later. McGraw-Hill, 1968

Lidz, T. The Person. Basic Books, 1968.

Luria, A. R. & Yudovich, F. La. Speech and the Development of Mental Processes in the Child. Penguin, 1971.

Mittler, P. Aspects of Autism. British Psychological Society, 1968.

Mussen, P. H., Conger, J. J. & Kagan, J. Child Development and Person-ality. 3rd ed. Harper International, 1969.

Nash, J. Developmental Psychology: A Psychobiological Approach. Prentice-Hall, 1970.

Rayner, E. Human Development: An Introduction to the Psychodynamics of Growth, Maturity and Ageing. Allen & Unwin, 1971. Sebald, H. Adolescence. Appleton-Century-Crofts, 1968.

Sluckin, W. Early Learning in Man and Animal. Allen & Unwin, 1970. Vernon, M. D. Reading and its Difficulties. C.U.P., 1971.

Wesley, F. Child-rearing Psychology. Behavioral Publications, N.Y., 1971. Woodward, W. M. The Development of Behaviour. Penguin, 1971.

Wright, B. A. Physical Disability: A Psychological Approach. Harper, 1960.

12.553 Developmental Psychology IIIA

The study of individual differences and basic psychological processes within a developmental framework. Psychological processes at various stages from infancy to senescence. The study of exceptional individuals within a developmental framework.

TEXT AND REFERENCE BOOKS

As for 12.552 Developmental Psychology II.

12.602 Abnormal Psychology II

Concepts of normality, neurosis and psychosis. The range of behaviour disorders. Medical, dynamic and behavioural models of psyochopathology.

TEXTBOOKS

Cashdan, S. Abnormal Psychology. Prentice-Hall, 1972.

REFERENCE BOOKS

Freedman, A. M. & Kaplan, H. I. A Comprehensive Textbook of Psychiatry. Williams & Wilkins, 1967.

Kisker, G. W. The Disorganized Personality. McGraw-Hill, 1972. Maher, B. A. Principles of Pyschopathology. McGraw-Hill, 1966.

12.603 Abnormal Psychology IIIA

Conflict, anxiety and avoidance behaviour. Anti-social behaviour, psychosomatic disorders, brain pathology, mental deficiency, schizophrenia, depression, methods of diagnosis and treatment.

TEXTBOOKS

C.R.M. Abnormal Psychology: Current Perspectives. CRM Books, Del Mar, 1972.

Gorlow, L. & Katkovsky, W. Readings in the Psychology of Adjustment. McGraw-Hill, 1968.

Wolpe, J. The Practice of Behaviour Therapy. Permagon, 1969.

REFERENCE BOOKS

Beck, A. T. Depression. Staples, 1967.

Eysenck, H. J. Experiments in Behaviour Therapy. Pergamon, 1964. Eysenck, H. J. & Rachman, S. The Causes and Cures of Neurosis. Routledge, 1965.

Fenichel, O. The Psychoanalytic Theory of Neurosis. Norton, 1945. Frank, C. Behaviour Therapies. McGraw-Hill, 1969.

- Freedman, A. M. & Kaplan, H. I. A Comprehensive Textbook of Psychiatry. Williams & Wilkins, 1967. Jackson, D. D. ed. The Aetiology of Schizophrenia. Basic, 1960. Kisker, G. W. The Disorganized Personality. McGraw-Hill, 1972. Laing, R. D. The Divided Self: An Existential Study in Sanity and

- Madness. Quadrangle Press, 1959.
- Laing, R. D. The Politics of Experience. Pantheon. 1967.
- Lidz, T. The Person. Basic Books, 1968.
- Maher, B. A. Principles of Psychopathology. Mc-Graw Hill, 1966.
- Pronko, N. K. Textbook of Abnormal Psychology. Williams & Wilkins, 1963.

Rachman, S. Critical Essays on Psychoanalysis. Pergamon, 1963. Shirley, H. F. Pediatric Psychiatry. Harvard U.P., 1963.

12.623 Guidance and Counselling III

The application of Psychological Tests and other techniques of appraisal to educational and vocational selection and guidance. Advice, and other procedures used to assist client decision making.

TEXTBOOK

Tyler, L. The Work of the Counsellor, Appleton, 1961.

REFERENCE BOOKS

- Carkhuff, R. R. & Berenson, B. G. Beyond Counselling and Therapy. Holt, Rinehart & Winston, 1967.
- Carkhuff, R. R. Helping and Human Relations. Vol. I. Selection and
- Training. Holt, Rinehart & Winston, 1969. Carkhuff, R. R. Helping and Human Relations. Vol. II. Practice and Research. Holt, Rinehart & Winston, 1969.
- Carkhuff, R. R. The Development of Human Resources. Holt, Rinehart & Winston, 1971.

12.653 Industrial Psychology III

The role of the psychologist in industry. Social, psychological and physical conditions affecting work behaviour. Selected aspects of human factors engineering and of human and industrial relations.

TEXTBOOKS

Goldthorpe, J. H. et al. The Affluent Worker. C.U.P., 1968.

McCormick E. J. Human Factors Engineering. 3rd ed. McGraw-Hill, 1970.

REFERENCE BOOKS

Boritz, L. Servants of Power. Science Editions, 1960. Herzberg, F. Work and the Nature of Man. Staples Press, 1966. White, W. H. The Organization Man. Penguin, 1956.

12.703 Psychological Techniques III

Laboratory techniques, including the use of recording stimulating, and control equipment. Observation, and other forms of appraisal.

REFERENCE BOOKS

- Bradford, L. P., Gibb, J. R. & Benne, K. D. eds. T-Group Theory and Laboratory Method. Wiley, 1964.
- Cooper, C. L. & Mangham, I. L. eds. T-groups: A Survey of Research. Wiley, 1971.

Egan, G. ed. Encounter Groups: Basic Readings. Wadsworth, 1971.

Gorden, R. L. Interviewing: Strategy, Techniques and Tactics. Dorsev. Illinois, 1969.

Kleinmuntz, B. Personality Measurement, Dorsey, 1967.

Shouksmith, G. Assessment Through Interviewing. Pergamon, 1968. Webb, E. J., Campbell, D. T., Schwartz, R. D. & Seechrest, L. Unobstrusive Measures: Non-reactive Research in the Social Sciences. Rand McNally, 1966.

12.713 Behaviour Control and Modification III

Aversive and appetitive reinforcement in the control and modification of undesirable behaviour. The conditions of attitude change and behavioural influence. Ethical Issues.

TEXTBOOKS

- Bandura, A. The Principles of Behavior Modification. Holt, Rinehart & Winston, 1969.
- Brenman, M. & Gill, M. Hypnotherapy. Wiley, 1964.
- Jahoda, M. & Warren, N. Attitudes, Modern Psychology Series. Penguin, 1966.

Weitzenhoffer, A. M. & Hilgard, E. R. Stanford Hypnotic Susceptibility Scale, Forms A and B. Consulting Psychologists, 1959.

Wolpe, J. The Practice of Behavior Therapy. Pergamon, 1969.

REFERENCE BOOKS

- Ayllon, T. & Azrin, N. The Token Economy. Appleton-Century-Crofts, 1968.
- Bergin, A. E. & Garfield, S. L. eds. Handbook of Psychotherapy and Behaviour Change. Wiley, 1971.
- Gordon, J. E. ed. Handbook of Clinical and Experimental Hypnosis. Macmillan, 1967.

Lifton, R. J. Thought Reform & The Psychology of Totalism. Pelican, 1961.

Rosnow, R. L. & Robinson, E. J. eds. Experiments in Persuasion. Academic, 1967.

Truax, C. B. & Carkhuff, R. R. Toward Effective Counselling and Psychotherapy. Aldine, 1967.

12.741 Psychology (Optometry)

Visual Perception-The nature and characteristics of visual perception. Topics to be discussed include: psychophysics, the organization of visual perception, the influence of context, and the effects of learning and motivation on perception. Throughout the course emphasis will be placed on an examination of relevant experimental data. Abnormal Psychology— The concept of normality-abnormality, and an examination of the principle psychodynamic processes. Causes and symptoms of various mental

disorders are introduced with some emphasis on symptoms and their importance in optometrical practice.

Part A-Visual Perception

TEXT AND REFERENCE BOOKS As for 12.472 Perception II.

Part B—Abnormal Psychology

TEXTBOOK

Coleman, J. C. Abnormal Psychology and Modern Life. 4th ed. Scott, Foresman, 1971.

SCHOOL OF ZOOLOGY

The School provides undergraduate courses in Zoology and Entomology taught as part of a Unit pattern. The School offers nine units of Zoology and four units of Entomology and contributes to a unit of Biometry and Genetics offered jointly with the School of Botany. All courses leading to a Science degree in Zoology are dependent on an adequate background in Biochemistry and in Genetics and Biometry. The units offered place an emphasis on experimental Animal Physiology, experimental and applied Entomology and an ecological approach to Marine Science.

Graduates at the bachelor level may find employment in scientific and technical departments of various State and Commonwealth organizations, in certain industries and in teaching. Students intending to pursue research careers in any of the various branches of Zoology are urged to complete the requirements for Honours, Master's or Doctor's degrees.

Courses in Zoology are taught mainly in the lecture theatres and laboratories, but field work, including field camps and excursions, is an essential part of all courses. To this end, the School maintains an undergraduate teaching Field Station at Smith's Lake, where compulsory courses are taught during vacations. The animal physiology teaching is strongly oriented towards Australian invertebrate and vertebrate animals, and the School has interests in field stations at which marsupials are studied.

For details of level, unit value, when offered, hours per week, prerequisites and co-requisites, see pages B22-23.

HONOURS IN ZOOLOGY OR ENTOMOLOGY

Students must receive permission of the Head of School before proceeding to Honours. Generally speaking they should have completed all subjects or units required for a Bachelor's degree and have achieved a consistently high standard in relevant units offered by the School of Zoology.

43.101A/45.101A Genetics and Biometry

For details of this course, which is offered jointly by the Schools of Zoology and Botany, see under School of Botany.

45.101B Invertebrate Zoology

A comparative study of the major invertebrate phyla with emphasis on morphology, systematics and phylogeny. Practical work to illustrate the lecture course. Obligatory field camp.

TEXTBOOK

Meglitsch, P. A. Invertebrate Zoology. 2nd ed. O.U.P., 1972.

REFERENCE BOOKS

Al-Hussaini, A. H. & Demian, E. J. Practical Animal Morphology. MacMillan, 1967. Barnes, R. D. Invertebrate Zoology. Saunders, 1968.

Fraenkel, G. S. & Gunn, D. L. Orientation of Animals. Dover, 1961.

Hyman, L. The Invertebrata. Vols. 1-6. McGraw-Hill. Imms, A. D. Textbook of Entomology. Methuen, 1962.

MacGinitie, G. E. & MacGinitie, N. Natural History of Marine Animals. McGraw-Hill, 1968.

Nicol, J. A. C. The Biology of Marine Animals. Pitman, 1967. Prosser, C. L. & Brown, F. A. Comparative Animal Physiology. Saunders, 1961.

Wilbur, K. & Yonge, C. M. Physiology of a Mollusc. Vols. 1 & 2. Academic, 1966

45.101C Vertebrate Zoology

A comparative study of the Chordata. Morphology, systematics, evolution, natural history, with reference to selected aspects of physiology and reproduction. Practical work to supplement the lecture course. Field excursions as arranged.

TEXTBOOKS

Romer, A. S. The Vertebrate Story. 4th ed. Chicago U.P., 1959. (Sixth Impression 1971.)

Weichert, C. K. Anatomy of the Chordates. 3rd ed. McGraw-Hill, 1969. Young, J. Z. The Life of the Vertebrates. O.U.P., 1958.

REFERENCE BOOKS

Bellairs, A. The Life of Reptiles. Vols. 1 & 2. Weidenfeld & Nicolson Natural History, 1969.

Berrill, N. J. The Tunicata. Ray Society Monograph, 1950.

Kurten, B. The Age of the Dinosaurs. Weidenfeld & Nicolson, 1968.

Kurten, B. The Age of Mammals. Weidenfeld & Nicolson, 1971.

Marshall, A. J. Biology and Comparative Physiology of the Birds. Vols. 1 & 2. Academic, 1960-1961.

Saunders, J. T. & Manton, S. M. A Manual of Vertebrate Morphology. 4th ed. O.U.P., 1969.

Walker, E. P. Mammals of the World. 2nd ed. Johns Hopkins U.P., 1968. Young, J. Z. The Life of Mammals. Clarendon, 1957.

45.101D Field Ecology

A lecture series on the basic principles of ecology followed by an examination and evaluation of the field methods used to measure the environment and the distribution and abundance of organisms.

TEXTBOOK

Southwood, T. R. E. Ecological Methods. Methuen, 1966.

This unit is offered as a lecture series (two per week) and a two-week camp in November/December at the University's Smith's Lake Field Station.

45.102A Marine Ecology

A study of the metabolic, regulatory and reproductive activities of marine organisms with particular reference to the physical, chemical and biological environment in which they occur. Both field and laboratory practical work are included.

TEXTBOOK

Tait, R. V. Elements of Marine Ecology. An Introductory Course, 2nd ed. Butterworths, 1972.

REFERENCE BOOKS

- McConnaughey, Bayard H. Introduction to Marine Biology. The C. V. Mosby Co., 1970.
- Moore, H. B. Marine Ecology. Wiley, 1958. Newell, R. C. Biology of Intertidal Animals. Elek, 1970.

Pickard, G. L. Descriptive Physical Oceanography. Pergamon, 1963. Winberg, G. G. Methods for the Estimation of Production of Aquatic Animals. Academic, 1971.

45.102B Animal Behaviour

An introduction to Ethology, the biological study of behaviour. Physiological, ecological, developmental and evolutionary aspects of behaviour are examined as important elements of the study of causal factors underlying behaviour. Both field and laboratory work are included.

TEXTBOOKS

Smith, F. V. Purpose in Animal Behaviour. Hutchinson, 1971.

Manning, A. An Introduction to Animal Behaviour. 2nd ed. Arnold, 1972.

REFERENCE BOOKS

Aronson, L., Tobach, E., Lehrman, D. S. & Rosenblatt, J. S. Development and Evolution of Behaviour. Freeman, 1970.
Crook, J. H. Social Behaviour in Birds and Mammals. Academic, 1970.
Hinde, R. A. Animal Behaviour. McGraw-Hill, 1970.

Hinde, R. A. ed. Bird Vocalization. C.U.P., 1970.

Klopfer, P. H. Habitats and Territories. Basic Books, 1969.

Marler, P. & Hamilton, W. J. Mechanisms of Animal Behaviour. Wiley, 1966.

Sluckin, W. J. Early Learning in Animals and Man. Allen & Unwin, 1969.

45.102C Comparative and Environmental Physiology

A study of the physiology of the various classes of vertebrate animals with particular emphasis on the adaptation of the animal to its environment. Includes: osmotic and ionic regulation, respiration and circulation, temperature regulation, nerve and muscle function, digestion and metabolism.

TEXTBOOKS

Gordon, M. S. Animal Function: Principles and Adaptations. 2nd ed. Macmillan, 1972.

Tyndale-Biscoe, H. Life of Marsupials. Arnold, 1973.

REFERENCE BOOK

Hoar, W. S. General and Comparative Physiology. Prentice-Hall, 1966.

45.102D Comparative Reproductive Physiology

A survey of reproductive mechanisms, reproductive histology, reproductive endocrinology, and embryology, with particular reference to the comparative aspects in vertebrate species. A detailed treatment of marsupial and monotreme reproduction.

TEXTBOOKS Gilchrist, F. G. A Survey of Embryology. McGraw-Hill, 1968. Nalbandov, A. V. Reproductive Physiology. Freeman, 1964.

REFERENCE BOOK Tyndale-Biscoe, H. Life of Marsupials. Arnold, 1973.

45.102E Invertebrate Physiology

An examination of certain aspects of general and reproductive physiology of invertebrates, including studies on body water and salts, excretion, vascular systems, respiration, digestion and absorption, the effects of temperature on invertebrate physiology, gametogenesis, fertilization, egg cleavage, reproductive cycles and endocrinology. The embryonic and evolutionary aspects of modes of larval development are also considered.

TEXTBOOKS

Barrington, E. J. W. Invertebrate Structure and Function. Nelson, 1967. Yapp, W. B. An Introduction to Animal Physiology. 3rd ed. O.U.P., 1970.

REFERENCE BOOKS

Florey, E. An Introduction to General and Comparative Animal Physiology.

Saunders, 1966. Hoar, W. S. General and Comparative Physiology. Prentice-Hall, 1966. Newell, R. C. Biology of Intertidal Animals. Elek, 1970.

45.102F Invertebrate Behaviour

Phylogenetic examination of behaviour in relation to the increasing complexity of invertebrates, with emphasis on orientation and movement; feeding, defensive, reproductive, social and rhythmic behaviour. Involves both exogenous and endogenous contributions to invertebrate behaviour.

TEXTBOOKS

Carthy, J. D. The Behaviour of the Arthropods. Oliver & Boyd, 1965. Marler, P. & Hamilton, W. J. Mechanisms of Animal Behaviour. Wiley, 1965.

REFERENCE BOOKS

Fraenkel, G. S. & Gunn, D. C. Orientation of Animals. Dover, 1961. Harker, J. E. The Physiology of Diurnal Rhythms. C.U.P., 1964. Haskell, P. T. Insect Sounds. Witherby, 1961. Highnam, K. C. & Hill, L. The Comparative Endocrinology of the Invertebrates. Arnold, 1969.

Howse, P. E. Termites: A Study in Social Behaviour. Hutchinson, 1970. Wells, M. J. Brain and Behaviour in Cephalopods. Heinemann, 1962.

45.201A Insect Structure and Classification

A comparative study of the internal anatomy and external morphology of insects. Classification and bionomics of major groups and families. A collection of insects is to be made. Practical work to include dissections, a study of mouthparts, wing venations, segmentation, etc. Field excursions as arranged.

TEXTBOOK C.S.I.R.O. The Insects of Australia. M.U.P., 1969. **REFERENCE BOOKS** Guthrie, E. M. & Tindall, A. R. The Cockroach. Arnold. Imms, A. D. Textbook of Entomology. Methuen. Ross, H. H. Textbook of Entomology, Wiley. Snodgrass, R. E. Principles of Insect Morphology. McGraw-Hill.

45.201B Insect Physiology

A study of the functions of the various organ systems and of the whole insect. Various aspects of reproduction, growth and metabolism. Experimental work to illustrate the lecture course.

TEXTBOOK

Chapman, P. F. The Insects, Structure and Function, E.U.P., 1969.

REFERENCE BOOKS

Beck, S. D. Insect Photoperiodism. Academic. Dethier, V. G. The Physiology of Insect Senses. Methuen. Highnam, K. C. & Hill, L. The Comparative Endocrinology of the Invertebrates. Arnold.

Roeder, K. D. Nerve Cells and Insect Behaviour. Harvard U.P.

Wigglesworth, V. B. Principles of Insect Physiology. Methuen.

45.201C Applied Entomology

Fundamentals of insect control. Pest species and types of damage caused. Control by insecticides, physical and biological means. Insect toxicology. Insecticide resistance. Practical work to illustrate the above and also various aspects of bioassay in Entomology. Field excursions as arranged.

TEXTBOOK

Woods, A. Pest Control: A Survey, McGraw-Hill, 1974.

45.201D Project

Detailed studies of selected aspects of insect physiology; ecology and toxicology. Treatment of topics will be in depth rather than breadth. Practical work will illustrate the lectures and will place emphasis on design and planning of experiments.

REFERENCE BOOKS

Beament, J. W. L., Treherne, J. E. & Wigglesworth, V. B. eds. Advances in Insect Physiology. Academic.
Bunning, E. The Physiological Clock. Springer Verlag.
Dethier, V. G. To Know a Fly. Holden-Day.

Johanssen, O. A. & Butt, F. H. Embryology of Insects & Myriopods. McGraw-Hill.

Lees. A. D. Diapause in Arthropods. C.U.P.

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FACULTIES OF BIOLOGICAL SCIENCES AND SCIENCE B173

B174 THE UNIVERSITY OF NEW SOUTH WALES

FACULTIES OF BIOLOGICAL SCIENCES AND SCIENCE B175

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(b) Why was the information	n of so little	use to you?	
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